

Lecture Note of Innovative Ship and Offshore Plant Design

**Innovative Ship and Offshore Plant Design**  
**Part II. Offshore Plant Design**  
**Ch. 2 Sizing and Configuration of Topside Systems**

Spring 2019

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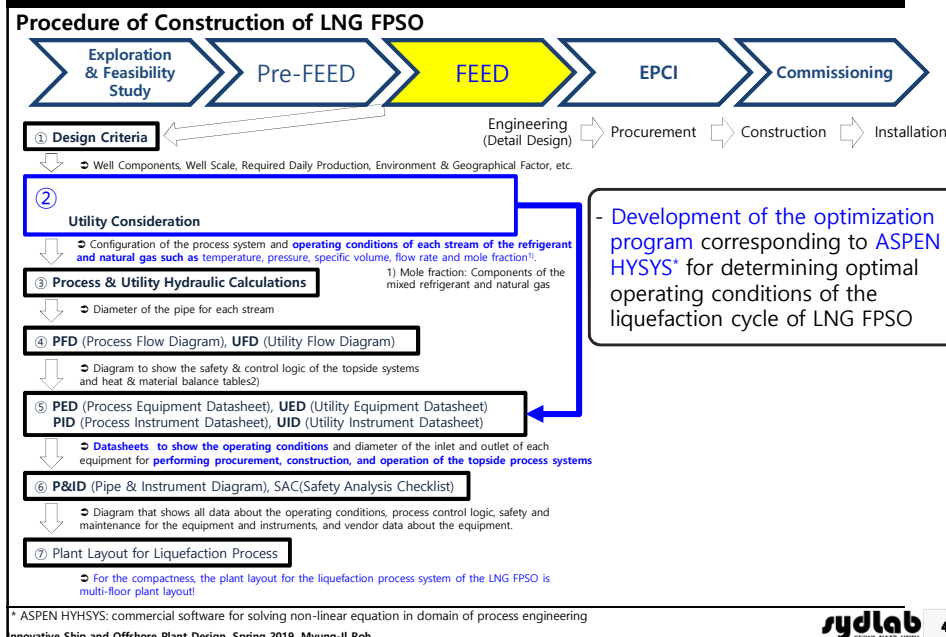
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## Ch. 2 Sizing and Configuration of Topping Systems

1. Determination of Optimal Operating Condition
2. Optimal Synthesis of Liquefaction Cycle
3. Determination of Optimal Operating Condition of the Liquefaction Cycle for LNG FPSO

### Procedures of Process FEED of Liquefaction System of LNG FPSO and Importance of Optimal Operating Condition



## Major Considerations for the Selection of the Liquefaction Cycle in Offshore Application (1/2)



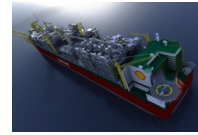
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<LNG FPSO>

- All major oil companies required that liquefaction cycles shall have reliability based on the results from **previous onshore projects**.
- **Dual Mixed Refrigerant(DMR)** cycle was verified from the SAKHALIN onshore liquefaction cycle in 2005.
  
- **Safety studies: HAZard and Operability(HAZOP), HAZard Identification(HAZID), Failure Modes and Effects Analysis(FMEA), Fault Tree Analysis(FTA), Event Tree Analysis(ETA), CFD Exhausts Dispersion Study - Helideck Study Report, Dropped Object Study, Explosion Risk Analysis, Failure, etc.**

\* HAZOP: Structured and systematic examination of a planned or existing process or operation in order to identify and evaluate problems that may represent risks to personnel or equipment, or prevent efficient operation

\* HAZID: The process of identifying hazards, which forms the essential first step of a risk assessment

\* FMEA: This analysis technique is used to systematically analyze postulated component failures and identify the resultant effects on system operations.

\* FTA: This analysis technique is used to understand how systems can fail, to identify the best ways to reduce risk or to determine event rates of a safety accident or a particular system level failure.

\* ETA: This analysis technique is used to analyze the effects of functioning or failed systems given that an event has occurred.

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## Major Considerations for the Selection of the Liquefaction Cycle in Offshore Application (2/2)



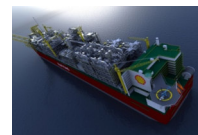
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<LNG FPSO>

- If the LNG FPSO is **inclined more than 1.5 degrees**, the capacity of **LNG production can be reduced by 10%**.
- Therefore, the liquefaction cycle in the LNG FPSO has to be designed by **considering mechanical damping devices, internal turret system, and dynamic positioning system**.
  
- Available area for the liquefaction cycle in offshore application is smaller than that of onshore plant.
- **By determining the optimal operating conditions and doing the optimal synthesis of the liquefaction cycle**, the required power for the compressors can be reduced which will result in **the reduction of the compressor size and the flow rate of the refrigerant**. Thus, **the overall sizes of the liquefaction cycle including the pipe diameter, equipment and instrument can be reduced**.
- Therefore, the compactness can be achieved by optimization studies such as determination of the optimal operating condition or optimal synthesis of the liquefaction cycle.

➡ **Optimum design can be used!**

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# 1. Determination of Optimal Operating Condition

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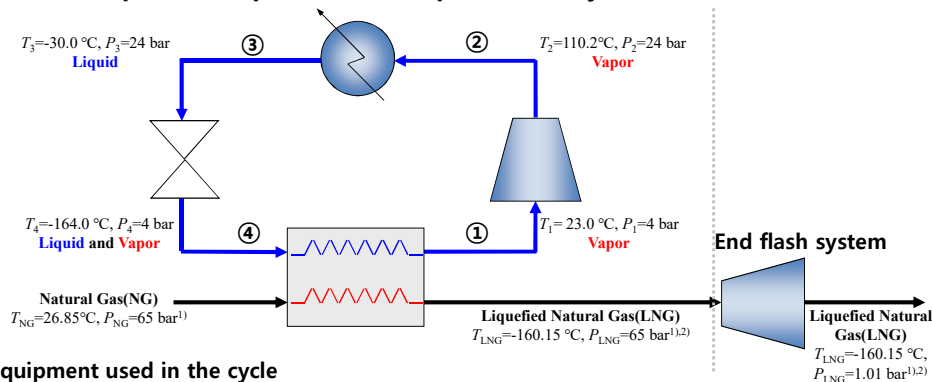
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## Introduction to Liquefaction Cycle

### • Goal of the LNG Liquefaction Cycle

To liquefy NG to LNG for decreasing the volume of the NG

### • An example of Simplified LNG Liquefaction Cycle



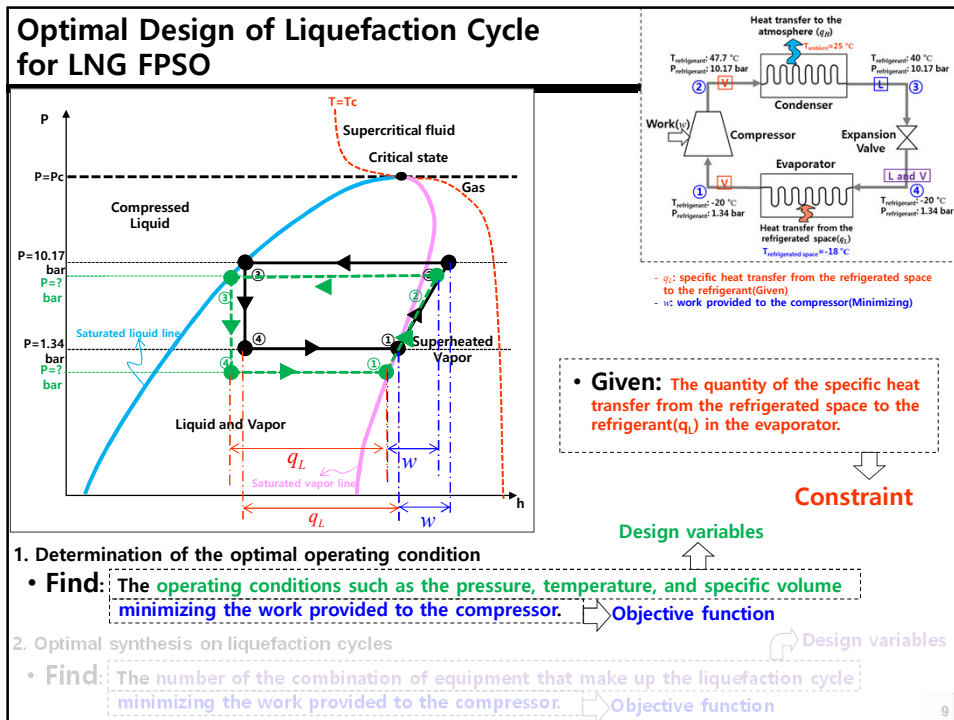
### Equipment used in the cycle

- 1) Compressor: brings the vapor refrigerant to a high pressure, which raises its temperature as well.
- 2) Sea Water Cooler(a kind of condenser): transfer heat from the hot vapor refrigerant to the sea water.
- 3) Valve: decreases the pressure of the liquid refrigerant, which decreases its temperature as well.
- 4) Heat Exchanger(a kind of evaporator): absorbs heat from the natural gas to cool down the NG, while the refrigerant is vaporized.

1) The temperature and pressure of the natural gas and liquefied natural gas are the values of the general case.

2) In the end flash system, the pressure of LNG expanded to the atmospheric pressure (1.01 bar) to be stored in the LNG tank.

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### Mathematical Model of the Liquefaction Cycle

#### - Calculation of Specific Enthalpy ( $h$ ) $h = u + P \cdot v$

• **Physical Constraint based on Thermodynamics #1**

**Energy conservation**

Calculation of

Many tables of thermodynamics properties does not give values for internal energy. To allow calculation of enthalpy from the pressure, specific volume, and temperature, the following equation is derived by using the definition ( $h = u + Pv$ ), the equation of state, and the experiment.

$$h = h^{IG} + h^R$$

$h^{IG}$ : Ideal gas value of the specific enthalpy  
 $h^R$ : Residual specific enthalpy (correction of the ideal gas state values to the real gas values)

\* Enthalpy: Internal energy plus the product of pressure and volume. Meaning of thermodynamic potential. Having the same dimension with the amount of heat or energy  
 \* Reference: Smith, J.M., Introduction to Chemical Engineering Thermodynamics, 7<sup>th</sup> edition, McGraw-Hill, 2005, pp.199-253

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## Mathematical Model of the Liquefaction Cycle - Equations of State

### Physical Constraint based on Thermodynamics #2

#### Equations of state

$$P_1 v_1 = RT_1$$

[Equation of state for an ideal gas (Boyle-Charles' Law)]

To improve the equation of state for the liquids and vapors, the equation of state for an ideal gas is modified by using the experiment and experience.

Example) Soave, Redlich, Kwong (SRK) equation

$$v = \frac{RT}{P} + b - \frac{a(T)}{P} \frac{v-b}{(v-\epsilon b)(v-\sigma b)}$$

$$v_1 = \frac{RT_1}{P} + b - \frac{a(T_1)}{P_1} \frac{v_1-b}{(v_1-\epsilon b)(v_1-\sigma b)}$$

$$v_{1,v} = \frac{RT_{1,v}}{P} + b - \frac{a(T_{1,v})}{P_1} \frac{v_{1,v}-b}{(v_{1,v}-\epsilon b)(v_{1,v}-\sigma b)}$$

$$v_{1,l} = \frac{RT_{1,l}}{P} + b - \frac{a(T_{1,l})}{P_1} \frac{v_{1,l}-b}{(v_{1,l}-\epsilon b)(v_{1,l}-\sigma b)}$$

$s^*$ : specific entropy

$$a(T) = \psi \frac{a(T_c) R^2 T_c^2}{P_c}$$

$$\psi = 0.42748 \text{ for SRK equation}$$

$R$ : gas constant (8.314 J/mol·K)

$P_c$ : critical pressure of the refrigerant

$T_c$ : critical temperature of the refrigerant

$\Omega = 0.08664$  for SRK equation

$$b = \Omega \frac{RT_c}{P_c} \quad \epsilon = 0 \text{ for SRK equation}$$

$$\sigma = 1 \text{ for SRK equation}$$

Example) Ammonia:

$$\omega = 0.253, P_c = 112.80 \text{ (bar)}, T_c = 405.7 \text{ (K)}$$

\* Equations of state: Thermodynamic equation describing the state of matter under a given set of physical conditions

\* Classical ideal gas law (Boyle-Charles' law) : Combination of Boyle's law (gas volume varies inversely with the pressure) and Charles' law (linear relationship between volume and temperature)

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## Mathematical Model of the Liquefaction Cycle - Calculation of Specific Entropy (s)

$$ds = \frac{dq}{T}$$

### Physical Constraint based on Thermodynamics #3

#### Criteria for quality of the energy

Calculation of

To allow calculation of entropy from the pressure, specific volume, and temperature, the following equation is derived by using the definition ( $ds = dq/T$ ), the equation of state, and the experiment.

$$s = s^{IG} + s^R$$

$s^{IG}$ : Ideal gas value of the entropy

$s^R$ : Residual entropy (correction of the ideal gas state values to the real gas values)

\* Entropy: Unavailable energy. Total energy = Available energy (enthalpy) + Unavailable energy (Entropy).

In a natural thermodynamic process, there is an increase in the sum of the entropies of the participating systems (The second law of thermodynamics).

\* Reference: Smith, J.M., Introduction to Chemical Engineering Thermodynamics, 7th edition, McGraw-Hill, 2005, pp.199-253

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## Mathematical Model of the Liquefaction Cycle - Physical Assumptions


- **Physical Constraint based on Thermodynamics #4**

Physical assumptions for the liquefaction process

**“Isobaric process”**  
- There is no pressure drop.

**“Adiabatic process”**  
- There is no sufficient time to transfer much heat.

**“Isentropic process”**  
- “Entropy” does not change.  
- “Adiabatic process” and “Reversible”

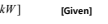

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**[Summary] Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator (1/2)**

- $T$ : Temperature,  $h$ : specific enthalpy,  $s$ : specific entropy
- $P$ : Pressure
- $v$ : Specific volume
- $w$ : Power provided to the compressor per mass
- $q_H$ : Specific heat transfer from the refrigerant to the atmosphere
- $q_L$ : Specific heat transfer from the refrigerated space to the refrigerant(Given)
- $f$ : Vapor fraction

<p><b>1. Compressor</b></p> <p>1) Design Variables: <math>P_1, v_1, T_1, P_2, v_2, T_2, T_3, w</math></p> <p>2) Constraint:</p> $h_1(P_1, v_1, T_1) + w = h_2(P_2, v_2, T_2) \quad \text{[The first law of the thermodynamics]}$ $\eta = \frac{h_2(P_2, v_2, T_2) - h_1(P_1, v_1, T_1)}{h_2(P_2, v_2, T_2) - h_1(P_1, v_1, T_1)} \quad \text{[Efficiency of the compressor]}$ $s_1(P_1, v_1, T_1) = s_2(P_2, v_2, T_2) \quad \text{[The second law of the thermodynamics]}$ $v_1 = \frac{RT_1}{P} + b - \frac{a(T_1)}{P_1} \frac{v_1 - b}{(v_1 - \sigma b)(v_1 - \sigma b)} \quad \text{[Equation of state]}$ $v_2 = \frac{RT_2}{P} + b - \frac{a(T_2)}{P_2} \frac{v_2 - b}{(v_2 - \sigma b)(v_2 - \sigma b)} \quad \text{[Equation of state]}$	<p><b>2. Condenser</b></p> <p>1) Design Variables: <math>P_2, v_2, T_2, P_3, v_3, T_3, q_H</math></p> <p>2) Constraint:</p> $h_2(P_2, v_2, T_2) = q_H + h_3(P_3, v_3, T_3) \quad \text{[The first law of the thermodynamics]}$ $P_2 = P_3 \quad \text{[Isobaric process]}$ $v_2 = \frac{RT_2}{P} + b - \frac{a(T_2)}{P_2} \frac{v_2 - b}{(v_2 - \sigma b)(v_2 - \sigma b)} \quad \text{[Equation of state]}$ $v_3 = \frac{RT_3}{P} + b - \frac{a(T_3)}{P_3} \frac{v_3 - b}{(v_3 - \sigma b)(v_3 - \sigma b)} \quad \text{[Equation of state]}$ $P_3 = 10^{\frac{A}{T_3 + C - 273.15}} \quad \text{[Saturated pressure and temperature]}$ $T_3 > T_{amb} + \Delta T_{min} \quad \text{[Outlet temperature of the condenser]}$
<p><b>3. Expansion Valve</b></p> <p>1) Design Variables: <math>P_3, v_3, T_3, P_4, v_4, T_4, v_{4j}, v_{4s}, v_{-f}</math></p> <p>2) Constraint:</p> $h_3(P_3, v_3, T_3) = (1 - v_{-f}) \cdot h_{4j}(P_4, v_{4j}, T_4) + v_{-f} \cdot h_{4s}(P_4, v_{4s}, T_4) \quad \text{[The first law of the thermodynamics]}$ $P_4 = 10^{\frac{A}{T_4 + C - 273.15}} \quad \text{[Saturated pressure and temperature]}$ $v_3 = \frac{RT_3}{P} + b - \frac{a(T_3)}{P_3} \frac{v_3 - b}{(v_3 - \sigma b)(v_3 - \sigma b)} \quad \text{[Equation of state]}$ $v_4 = (1 - v_{-f}) \cdot v_{4j} + v_{-f} \cdot v_{4s}$ $v_{4j} = \frac{RT_4}{P} + b - \frac{a(T_4)}{P_4} \frac{v_{4j} - b}{(v_{4j} - \sigma b)(v_{4j} - \sigma b)} \quad \text{[Equation of state]}$ $v_{4s} = \frac{RT_4}{P} + b - \frac{a(T_4)}{P_4} \frac{v_{4s} - b}{(v_{4s} - \sigma b)(v_{4s} - \sigma b)} \quad \text{[Equation of state]}$	<p><b>4. Evaporator</b></p> <p>1) Design Variables: <math>P_4, v_4, T_4, P_1, v_1, T_1, v_{1j}, v_{1s}, v_{-f}, M, q_L</math></p> <p>2) Constraint:</p> $M \cdot (1 - v_{-f}) \cdot h_{1j}(P_1, v_{1j}, T_1) + M \cdot v_{-f} \cdot h_{1s}(P_1, v_{1s}, T_1) + M \cdot q_L = M \cdot h_1(P_1, v_1, T_1) \quad \text{[The first law of the thermodynamics]}$ $P_4 = P_1 \quad \text{[Isobaric process]}$ $v_4 = \frac{RT_4}{P} + b - \frac{a(T_4)}{P_4} \frac{v_4 - b}{(v_4 - \sigma b)(v_4 - \sigma b)} \quad \text{[Equation of state]}$ $v_1 = (1 - v_{-f}) \cdot v_{1j} + v_{-f} \cdot v_{1s}$ $v_{1j} = \frac{RT_1}{P} + b - \frac{a(T_1)}{P_1} \frac{v_{1j} - b}{(v_{1j} - \sigma b)(v_{1j} - \sigma b)} \quad \text{[Equation of state]}$ $v_{1s} = \frac{RT_1}{P} + b - \frac{a(T_1)}{P_1} \frac{v_{1s} - b}{(v_{1s} - \sigma b)(v_{1s} - \sigma b)} \quad \text{[Equation of state]}$ $P_1 = 10^{\frac{A}{T_1 + C - 273.15}} \quad \text{[Saturated pressure and temperature]} \quad M \cdot q_L = 20 [kW] \quad \text{[Given]}$


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### [Summary] Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator (2/2)

$T$ : Temperature,  $h$ : specific enthalpy,  $s$ : specific entropy  
 $P$ : Pressure  
 $v$ : Specific volume  
 $w$ : Power provided to the compressor per mass  
 $q_H$ : Specific heat transfer from the refrigerant to the atmosphere  
 $q_L$ : Specific heat transfer from the refrigerated space to the refrigerator(Given)  
 $U$ : Heat transfer coefficient of the evaporator  
 $A$ : Area of the evaporator

**1. Design variables (Operating conditions) [21]:**  $P_H, T_H, v_H, T_M, v_S, v_{4,H}, v_{4,M}, v_{4,S}, v_{4,0}, v_{4,1}, v_{4,2}, v_{4,3}, v_{4,4}, w, M, q_L, q_H$  ( $i=1,2,3,4$ )

**2. Equality constraints [19]**

**1) Compressor [6]**  
 $\dot{m} \cdot h_1(P_1, v_1, T_1) + \dot{m} \cdot w = \dot{m} h_2(P_2, v_2, T_2)$  [The first law of the thermodynamics]  
 $\eta = \frac{h_2(P_2, v_2, T_2) - h_1(P_1, v_1, T_1)}{h_2(P_2, v_2, T_2) - h_1(P_1, v_1, T_1)}$  [The second law of the thermodynamics]  
 $s_1(P_1, v_1, T_1) = s_2(P_2, v_2, T_2)$   
 $v_1 = \frac{RT_1}{P_1} + b - \frac{a(T_1)}{P_1} \frac{v_1 - b}{(v_1 - \epsilon b)(v_1 - \sigma b)}$  [Equation of state]  
 $v_2 = \frac{RT_2}{P_2} + b - \frac{a(T_2)}{P_2} \frac{v_2 - b}{(v_2 - \epsilon b)(v_2 - \sigma b)}$   $v_3 = \frac{RT_3}{P_3} + b - \frac{a(T_3)}{P_3} \frac{v_3 - b}{(v_3 - \epsilon b)(v_3 - \sigma b)}$

**2) Condenser [4]**  
 $h_3(P_3, v_3, T_3) = q_H + h_4(P_4, v_4, T_4)$  [The first law of the thermodynamics]  
 $P_3 = P_4$  [Isobaric process]  
 $v_3 = \frac{RT_3}{P_3} + b - \frac{a(T_3)}{P_3} \frac{v_3 - b}{(v_3 - \epsilon b)(v_3 - \sigma b)}$  [Equation of state]  
 $\frac{P_3}{10^5} = 10^{\frac{A}{T_3 + C} - \frac{B}{T_3 + D}}$  [Saturated pressure and temperature]

**3) Expansion valve [5]**  
 $h_4(P_4, v_4, T_4) = (1 - v_{4,f}) \cdot h_{4,L}(P_4, v_{4,L}, T_{4,L}) + v_{4,f} \cdot h_{4,V}(P_4, v_{4,V}, T_{4,V})$  [The first law of the thermodynamics]  
 $\frac{P_4}{10^5} = 10^{\frac{A}{T_4 + C} - \frac{B}{T_4 + D}}$  [Saturated pressure and temperature]  
 $v_{4,L} = \frac{RT_{4,L}}{P_4} + b - \frac{a(T_{4,L})}{P_4} \frac{v_{4,L} - b}{(v_{4,L} - \epsilon b)(v_{4,L} - \sigma b)}$   $v_{4,V} = \frac{RT_{4,V}}{P_4} + b - \frac{a(T_{4,V})}{P_4} \frac{v_{4,V} - b}{(v_{4,V} - \epsilon b)(v_{4,V} - \sigma b)}$   
 $v_4 = (1 - v_{4,f}) \cdot v_{4,L} + v_{4,f} \cdot v_{4,V}$

**4) Evaporator [4]**  
 $M \cdot (1 - v_{4,f}) \cdot h_{4,L}(P_4, v_{4,L}, T_{4,L}) + M \cdot v_{4,f} \cdot h_{4,V}(P_4, v_{4,V}, T_{4,V}) + M \cdot q_L = M \cdot h_1(P_1, v_1, T_1)$  [The first law of the thermodynamics]  
 $P_4 = P_1$  [Isobaric process]  
 $\frac{P_4}{10^5} = 10^{\frac{A}{T_4 + C} - \frac{B}{T_4 + D}}$   $M \cdot q_L = 20[kW]$  [Saturated pressure and temperature]  
 $T_3 > T_{amb} + \Delta T_{min}$  [Heat transfer in the evaporator]

**3. Inequality constraint [1]**  
 $T_3 > T_{amb} + \Delta T_{min}$  [Outlet temperature of the condenser]

Since the number of equality constraints is less than the number of design variables, these equations form an **indeterminate problem**.  
 We need a certain criteria to determine the proper solution. By introducing the criteria(objective function), this problem can be formulated as an **optimization problem**.

**4. Objective function (f)**  
 Minimize the power provided to the compressor.  
 $f = M \cdot w$

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### Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator - Optimization Method

**Optimization Problem**

1. Design variables (Operating conditions) [21]  
 :  $P_H, T_H, v_H, T_M, v_S, v_{4,H}, v_{4,M}, v_{4,S}, v_{4,0}, v_{4,1}, v_{4,2}, v_{4,3}, v_{4,4}, w, M, q_L, q_H$  ( $i=1,2,3,4$ )

2. Equality constraints [19]  
 1) Compressor (6)  
 2) Condenser (4)  
 3) Expansion valve (5)  
 4) Evaporator (4)

3. Inequality constraints [1]

4. **Objective function: Minimize the compressors power**  
 Minimize  $f = M \cdot w$

$T$ : Temperature,  $h$ : specific enthalpy,  $s$ : specific entropy  
 $P$ : Pressure  
 $v$ : Specific volume  
 $w$ : Power provided to the compressor per mass  
 $q_H$ : Specific heat transfer from the refrigerant to the atmosphere  
 $q_L$ : Specific heat transfer from the refrigerated space to the refrigerator(Given)

**Modified optimization problem**

- **Free variables** [2 = 21 - 19]  
 :  $T_1, P_2$
- **Inequality constraints** [1]

④ Calculation of objective function  
 Minimize  $f = M \cdot w$

⑤ Assume the free variable to calculate smaller value of the objective function using sequential quadratic programming(SQP) method

① Assume the free variable  $T_1, P_2$       ③ Determined 19 variables

**System of nonlinear equations**

- **Design variables** [19]
- **Equality constraints** [19]

② → Determine the 19 variables using Newton-Raphson method

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### Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator - Optimization Result

**Problem<sup>1)</sup>**  
**Given:**  $M \cdot q_L = 20 [kW]$ ,  $U = 1,000 [Wm^2 / K]$ ,  $A = 4.0 [m^2]$ ,  
 $T_C = -12^\circ C$ ,  $\eta = 95\%$ ,  $T_{amb} = 25^\circ C$ ,  $\Delta T_{min} = 5^\circ C$

where  $\dot{m} \cdot q_L$ : Rate of heat transfer from the refrigerated space to the refrigerator  
 $T_C$ : temperature of the refrigerated space  
 $\eta$ : efficiency of the compressor where  
 $T_{amb}$ : ambient temperature  
 $\Delta T_{min}$ : minimum value of the difference between the ambient temperature and outlet temperature in the condenser

**Find: Operating condition**  
**Minimize**  $f = M \cdot W$  : Power provided to the compressor [kW]

**Optimization result**

Items	Result obtained in this study
$P_1$ [bar]	2.113
$T_1$ [K]	256.152
$v_1$ [m <sup>3</sup> /mol]	0.0098128
$P_2$ [bar]	11.717
$T_2$ [K]	390.278
$v_2$ [m <sup>3</sup> /mol]	0.0026551
$P_3$ [bar]	11.717
$T_3$ [K]	303.273
$v_3$ [m <sup>3</sup> /mol]	0.0000365
$P_4$ [bar]	2.114
$T_4$ [K]	256.151
$v_4$ [m <sup>3</sup> /mol]	0.0017003
$v \cdot f$	0.1703
$v4v$ [m <sup>3</sup> /mol]	0.0098130
$v4l$ [m <sup>3</sup> /mol]	0.0000327
$T_1$ [K]	384.793
$v_3$ [m <sup>3</sup> /mol]	0.0026119
$M$ [kg/s]	0.0176
$w$ [J/g]	265.698
$q_L$ [J/g]	1,136.364
$q_H$ [J/g]	1,402.676
Objective function(W)	4.672

\* Reference: 1) Jensen, J.B., 2008, Optimal Operation of Refrigeration Cycles, Ph.D. thesis, Norwegian University of Science and Technology.  
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### Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator - Verification of the Mathematical Model and Optimization Result (1/3)

**How can we verify the mathematical model of this research for refrigerator?**

**<Input>**  
 Assume the 2 design variables

**Mathematical model:**  
 - Design variables [21]  
 - Equality constraints [19]

**<Output>**  
 The values of the other design variables

	Mathematical Model
$P_1$ [bar]	2.113
$T_1$ [K]	256.150
$v_1$ [m <sup>3</sup> /mol]	0.0098157
$P_2$ [bar]	11.698
$T_2$ [K]	390.001
$v_2$ [m <sup>3</sup> /mol]	0.0026570
$P_3$ [bar]	11.698
$T_3$ [K]	303.180
$v_3$ [m <sup>3</sup> /mol]	0.0000365
$P_4$ [bar]	2.113
$T_4$ [K]	256.150
$v_4$ [m <sup>3</sup> /mol]	0.0017269
$v \cdot f$	0.1732
$v4v$ [m <sup>3</sup> /mol]	0.0098157
$v4l$ [m <sup>3</sup> /mol]	0.0000327
$T_1$ [K]	384.296
$v_3$ [m <sup>3</sup> /mol]	0.0026153
$M$ [kg/s]	0.0176
$w$ [J/g]	268.657
$q_L$ [J/g]	1,136.364
$q_H$ [J/g]	1,402.189

- Mathematical Model of this research
  - Design variables (Operating conditions) [21]
  - Equality constraints [19]
 ➔ Indeterminate problem

To verify the mathematical model this research, we assume the values of the two design variables, solve and compare the result with that of the Aspen HYSYS.

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### Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator - Verification of the Mathematical Model and Optimization Result (2/3)

How can we verify the mathematical model of this research for refrigerator?

**<Input>**

Assume the 2 design variables

↓

Mathematical model:  
- Design variables (21)  
- Equality constraints (19)

↓

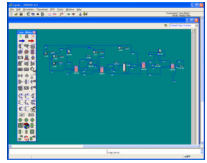
**<Output>**

The values of the other design variables

**<Input>**

Assume the 2 design variables

↓



Aspen HYSYS<sup>1)</sup>  
There are equality constraints, but it is difficult to find the equality constraints (Black box).

↓

**<Output>**

The values of the other design variables

	Mathematical Model
P <sub>1</sub> [bar]	2.113
T <sub>1</sub> [K]	256.150
v <sub>1</sub> [m <sup>3</sup> /mol]	0.0098157
P <sub>2</sub> [bar]	11.698
T <sub>2</sub> [K]	390.001
v <sub>2</sub> [m <sup>3</sup> /mol]	0.0026570
P <sub>3</sub> [bar]	11.698
T <sub>3</sub> [K]	303.180
v <sub>3</sub> [m <sup>3</sup> /mol]	0.0000365
P <sub>4</sub> [bar]	2.113
T <sub>4</sub> [K]	256.150
v <sub>4</sub> [m <sup>3</sup> /mol]	0.0017269
v · f	0.1732
v <sub>4v</sub> [m <sup>3</sup> /mol]	0.0098157
v <sub>4l</sub> [m <sup>3</sup> /mol]	0.0000327
T <sub>4</sub> [K]	384.296
v <sub>4</sub> [m <sup>3</sup> /mol]	0.0026153
M[kg/s]	0.0176
w[J/g]	268.657
q <sub>l</sub> [J/g]	1,136.364
q <sub>d</sub> [J/g]	1,402.189

	Aspen HYSYS	Difference[%]
P <sub>1</sub> [bar]	2.117	0.16%
T <sub>1</sub> [K]	256.150	0.00%
v <sub>1</sub> [m <sup>3</sup> /mol]	0.0098003	0.16%
P <sub>2</sub> [bar]	11.698	0.00%
T <sub>2</sub> [K]	389.863	0.04%
v <sub>2</sub> [m <sup>3</sup> /mol]	0.0026562	0.03%
P <sub>3</sub> [bar]	11.698	0.00%
T <sub>3</sub> [K]	303.150	0.01%
v <sub>3</sub> [m <sup>3</sup> /mol]	0.0000365	0.00%
P <sub>4</sub> [bar]	2.117	0.16%
T <sub>4</sub> [K]	256.151	0.00%
v <sub>4</sub> [m <sup>3</sup> /mol]	0.0017234	0.20%
v · f	0.1731	0.04%
v <sub>4v</sub> [m <sup>3</sup> /mol]	0.0098003	0.16%
v <sub>4l</sub> [m <sup>3</sup> /mol]	0.0000327	0.01%
T <sub>4</sub> [K]	384.165	0.03%
v <sub>4</sub> [m <sup>3</sup> /mol]	0.0026116	0.14%
M[kg/s]	0.0176	0.05%
w[J/g]	268.353	0.11%
q <sub>l</sub> [J/g]	1,136.364	0.00%
q <sub>d</sub> [J/g]	1,401.532	0.05%
Max difference		0.20%

1) It is a kind of simulation program to solve the simultaneous equations by considering equality constraint related with thermodynamics and made by AspenTech.

The mathematical model can be verified by comparing between the values of the other design variables obtained by mathematical model of this research and Aspen HYSYS.

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**sydlab** 19

### Mathematical Model for the Determination of Optimal Operating Condition for the Refrigerator - Verification of the Mathematical Model and Optimization Result (3/3)

**Indeterminate problem**

1. Design variables (Operating conditions) [21]  
: P<sub>p</sub>, T<sub>p</sub>, v<sub>p</sub>, T<sub>s</sub>, v<sub>s</sub>, v<sub>d,p</sub>, v<sub>s,r</sub>, v<sub>f</sub>, w, M, q<sub>L</sub>, q<sub>H</sub> (i=1,2,3,4)

2. Equality constraints [19]  
1) Compressor [6]  
2) Condenser [4]  
3) Expansion valve [5]  
4) Evaporator [4]

↓

• Assumption of the values of the free variables [2 = 21 - 19]  
: T<sub>1</sub> = 256.150[K]  
P<sub>2</sub> = 11.698[bar]

↓

**Problem of nonlinear equations**

- Design variables [19]
- Equality constraints [19]

➔ Use of Newton-Raphson method

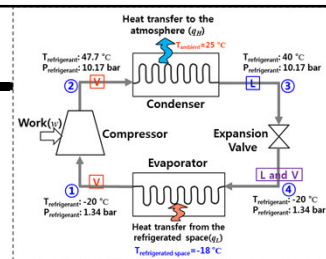
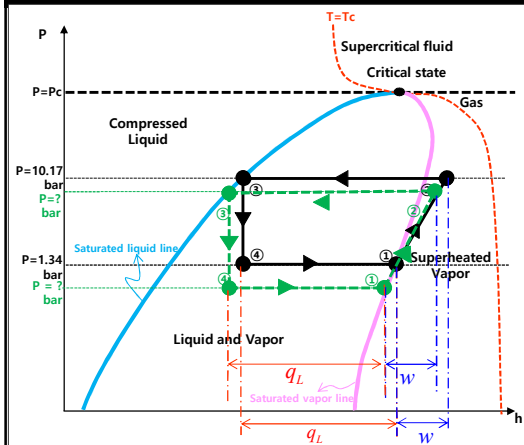
	Mathematical Model
P <sub>1</sub> [bar]	2.113
T <sub>1</sub> [K]	256.150
v <sub>1</sub> [m <sup>3</sup> /mol]	0.0098157
P <sub>2</sub> [bar]	11.698
T <sub>2</sub> [K]	390.001
v <sub>2</sub> [m <sup>3</sup> /mol]	0.0026570
P <sub>3</sub> [bar]	11.698
T <sub>3</sub> [K]	303.180
v <sub>3</sub> [m <sup>3</sup> /mol]	0.0000365
P <sub>4</sub> [bar]	2.113
T <sub>4</sub> [K]	256.150
v <sub>4</sub> [m <sup>3</sup> /mol]	0.0017269
v · f	0.1732
v <sub>4v</sub> [m <sup>3</sup> /mol]	0.0098157
v <sub>4l</sub> [m <sup>3</sup> /mol]	0.0000327
T <sub>4</sub> [K]	384.296
v <sub>4</sub> [m <sup>3</sup> /mol]	0.0026153
M[kg/s]	0.0176
w[J/g]	268.657
q <sub>l</sub> [J/g]	1,136.364
q <sub>d</sub> [J/g]	1,402.189

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**sydlab** 20

## 2. Optimal Synthesis of Liquefaction Cycle

### Optimal Design of Liquefaction Cycle for LNG FPSO



- $q_L$ : specific heat transfer from the refrigerated space to the refrigerant (Given)
- $w$ : work provided to the compressor (Minimizing)

• **Given:** The quantity of the specific heat transfer from the refrigerated space to the refrigerant ( $q_L$ ) in the evaporator.

**Constraint**

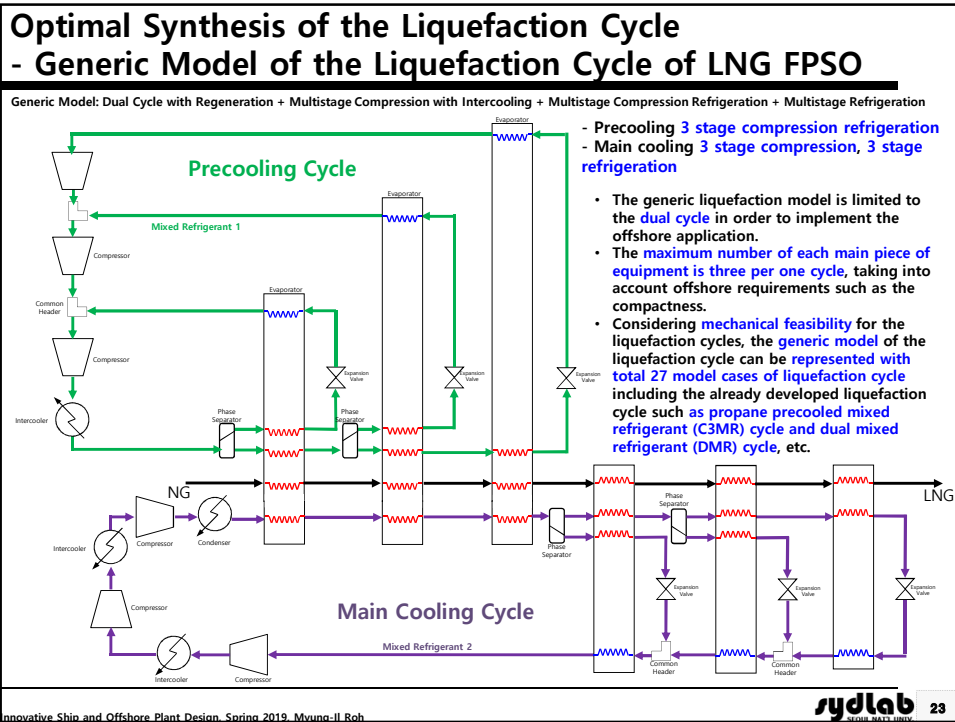
Design variables

1. Determination of the optimal operating condition

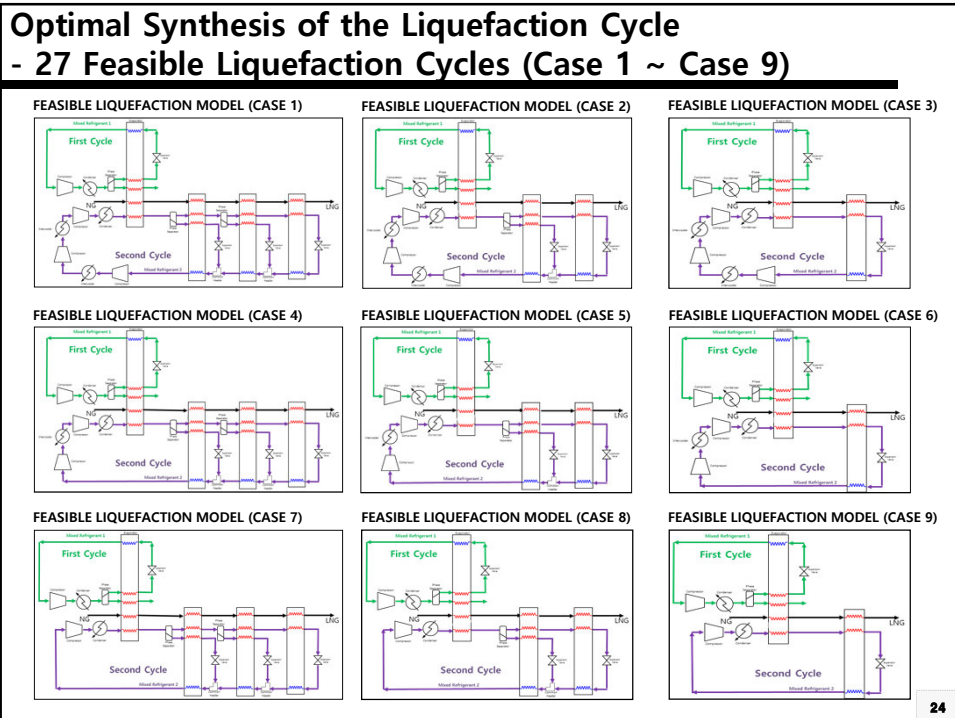
• **Find:** The operating conditions such as the pressure, temperature and specific volume minimizing the work provided to the compressor. **Objective function**

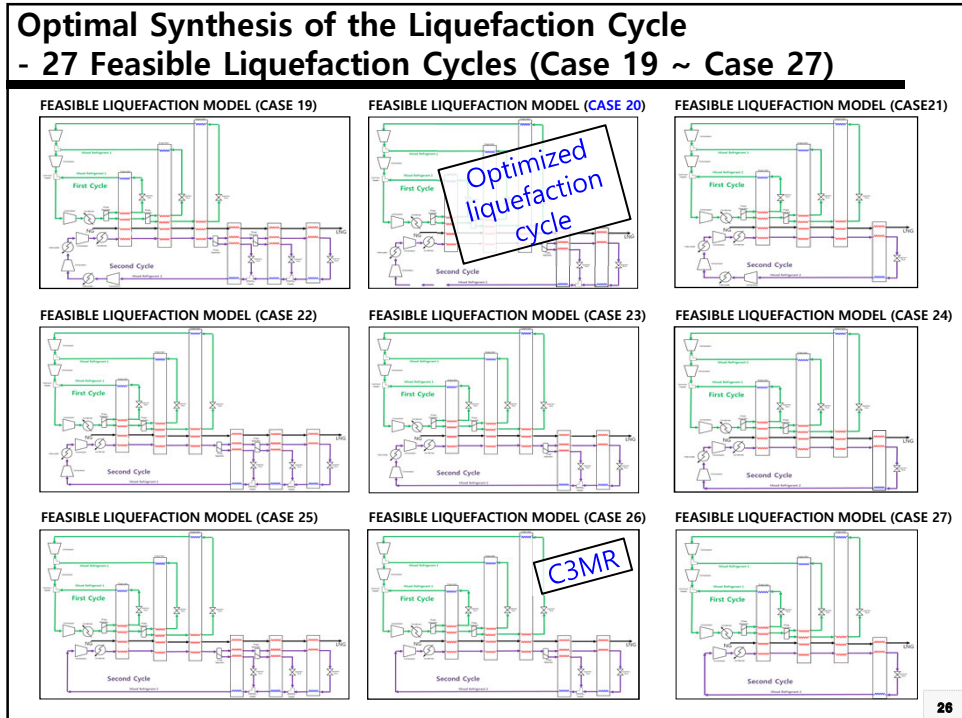
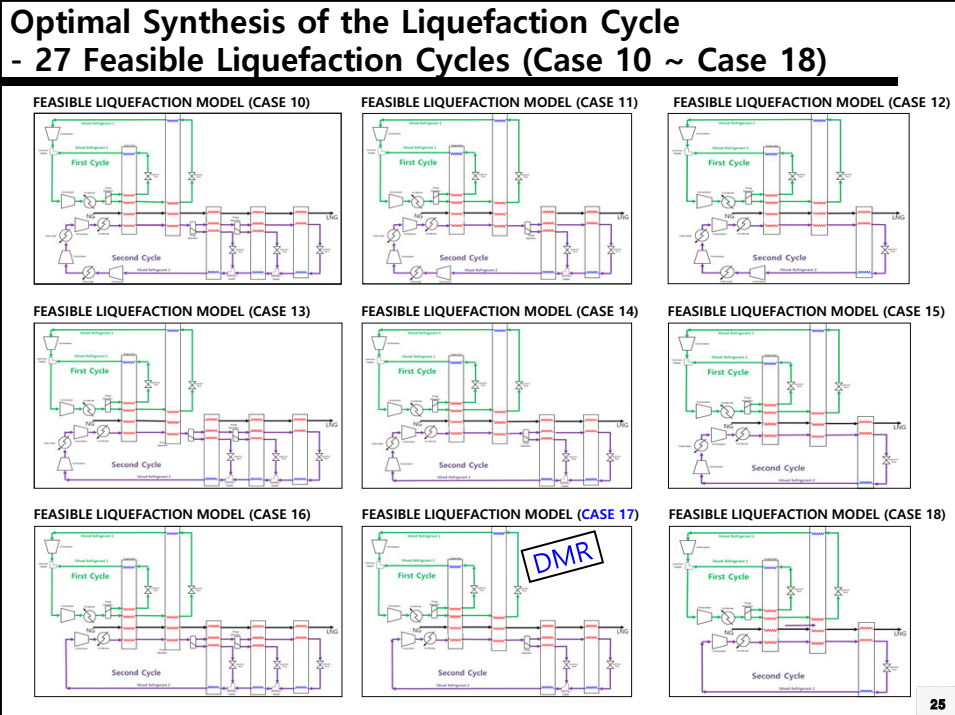
2. Optimal synthesis on liquefaction cycles

• **Find:** The number of the combination of equipment that make up the liquefaction cycle minimizing the work provided to the compressor. **Objective function**



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## Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle (1/6)

**1. Design variables (Operating conditions) [187]** T: Temperature / P: Pressure / v: Specific volume / z<sub>pre</sub>: mole fraction of the component j at the precooling part / w: work input to the compressor per mass / c: flow rate ratio between inlet and outlet 4 / ṁ<sub>pre</sub>: mass flow rate at the precooling refrigerant  

$$: P_j, T_j, V_j (j = 1_p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG}),$$

$$T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, V_{S,2p}, V_{S,19p}, V_{S,21p}, V_{S,2m}, V_{S,4m}, V_{S,6m},$$

$$W_1, W_2, W_3, W_4, W_5, W_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, V_{-f10}, V_{-f15}, z_{j,pre} (j = 1, 2, 3), z_{k,max} (k = 1, 2, 3, 4)$$
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant

### 2. Equality constraints [165]

#### 2.1 Equality constraints of precooling part [83]

##### 1) Compressor 1: [6]

$$h_{1p}(P_{1p}, T_{1p}, v_{1p}, z_{j,pre}) + w_1 = h_{2p}(P_{2p}, T_{2p}, v_{2p}, z_{j,pre})$$

$$\eta = \frac{h_{2p}(P_{2p}, T_{2p}, v_{2p}, z_{j,pre}) - h_{1p}(P_{1p}, T_{1p}, v_{1p}, z_{j,pre})}{h_{2p}(P_{2p}, T_{2p}, v_{2p}, z_{j,pre}) - h_{1p}(P_{1p}, T_{1p}, v_{1p}, z_{j,pre})}$$

$$s_{1p}(P_{1p}, T_{1p}, v_{1p}, z_{j,pre}) = s_{2p}(P_{2p}, T_{2p}, v_{2p}, z_{j,pre})$$

$$v_{1p} = v_{1p}(P_{1p}, T_{1p}, z_{j,pre})$$

$$v_{S,2p} = v_{S,2p}(P_{S,2p}, T_{S,2p}, z_{j,pre})$$

$$v_{2p} = v_{2p}(P_{2p}, T_{2p}, z_{j,pre})$$

##### 2) Condenser 1: [3]

The temperature of the outlet of the sea water cooler is usually given.  
**T=310K**

$$P_{2p} = P_{2p}$$

$$v_{3p} = v_{3p}(T_{3p}, P_{3p}, z_{j,pre})$$

##### 3) Tee 1: [6]

$$h_{1p}(P_{1p}, T_{1p}, v_{1p}, z_{j,pre}) = c_1 \cdot h_{2p}(P_{2p}, T_{2p}, v_{2p}, z_{j,pre}) + (1-c_1) \cdot h_{3p}(P_{3p}, T_{3p}, v_{3p}, z_{j,pre})$$

$$P_{3p} = P_{4p}, P_{3p} = P_{8p}$$

$$T_{4p} = T_{8p}$$

$$v_{4p} = v_{4p}(T_{4p}, P_{4p}, z_{j,pre}), v_{8p} = v_{8p}(T_{8p}, P_{8p}, z_{j,pre})$$

##### 4) Evaporator 1: [14]

$$c_1 \cdot \dot{m}_{pre} \cdot h_{1p}(P_{1p}, T_{1p}, v_{1p}, z_{j,pre}) + c_1 \cdot \dot{m}_{pre} \cdot h_{2p}(P_{2p}, T_{2p}, v_{2p}, z_{j,pre})$$

$$+ (1-c_1) \cdot \dot{m}_{pre} \cdot h_{3p}(P_{3p}, T_{3p}, v_{3p}, z_{j,pre})$$

$$+ \dot{m}_{main} \cdot h_{9m}(P_{9m}, T_{9m}, v_{9m}, z_{k,max}) + \dot{m}_{NG} \cdot h_{NG}(P_{NG}, T_{NG}, v_{NG}, z_{i,NG})$$

$$= c_1 \cdot \dot{m}_{pre} \cdot h_{4p}(P_{4p}, T_{4p}, v_{4p}, z_{j,pre}) + c_1 \cdot \dot{m}_{pre} \cdot h_{8p}(P_{8p}, T_{8p}, v_{8p}, z_{j,pre})$$

$$+ (1-c_1) \cdot \dot{m}_{pre} \cdot h_{9p}(P_{9p}, T_{9p}, v_{9p}, z_{j,pre}) + \dot{m}_{main} \cdot h_{9m}(P_{9m}, T_{9m}, v_{9m}, z_{k,max})$$

$$+ \dot{m}_{NG} \cdot h_{NG}(P_{NG}, T_{NG}, v_{NG}, z_{i,NG})$$

$$P_{4p} = P_{5p}, P_{4p} = P_{7p}, P_{4p} = P_{8p}, P_{4m} = P_{8m}, P_{4NG} = P_{8NG}$$

$$T_{5p} = T_{9p}, T_{5p} = T_{9m}, T_{5p} = T_{9NG}$$

$$v_{5p} = v_{5p}(T_{5p}, P_{5p}, z_{j,pre}), v_{7p} = v_{7p}(T_{7p}, P_{7p}, z_{j,pre})$$

$$v_{9p} = v_{9p}(T_{9p}, P_{9p}, z_{j,pre}), v_{9m} = v_{9m}(T_{9m}, P_{9m}, z_{k,max})$$

$$v_{9NG} = v_{9NG}(T_{9NG}, P_{9NG}, z_{i,NG})$$

##### 5) Expansion valve 1: [2]

$$h_{5p}(P_{5p}, T_{5p}, v_{5p}, z_{j,pre}) = h_{6p}(P_{6p}, T_{6p}, v_{6p}, z_{j,pre})$$

$$v_{6p} = v_{6p}(T_{6p}, P_{6p}, z_{j,pre})$$

##### 6) Tee 2: [6]

$$(1-c_1) \cdot h_{4p}(P_{4p}, T_{4p}, v_{4p}, z_{j,pre})$$

$$= c_1 \cdot (1-c_1) \cdot h_{10p}(P_{10p}, T_{10p}, v_{10p}, z_{j,pre})$$

$$+ (1-c_1) \cdot (1-c_1) \cdot h_{14p}(P_{14p}, T_{14p}, v_{14p}, z_{j,pre})$$

$$P_{10p} = P_{10p}, P_{10p} = P_{14p}$$

$$T_{10p} = T_{14p}$$

$$v_{10p} = v_{10p}(T_{10p}, P_{10p}, z_{j,pre}),$$

$$v_{14p} = v_{14p}(T_{14p}, P_{14p}, z_{j,pre})$$

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## Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle (2/6)

**1. Design variables (Operating conditions) [187]** T: Temperature / P: Pressure / v: Specific volume / z<sub>pre</sub>: mole fraction of the component j at the precooling part / w: work input to the compressor per mass / c: flow rate ratio between inlet and outlet 4 / ṁ<sub>pre</sub>: mass flow rate at the precooling refrigerant  

$$: P_j, T_j, V_j (j = 1_p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG}),$$

$$T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, V_{S,2p}, V_{S,19p}, V_{S,21p}, V_{S,2m}, V_{S,4m}, V_{S,6m},$$

$$W_1, W_2, W_3, W_4, W_5, W_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, V_{-f10}, V_{-f15}, z_{j,pre} (j = 1, 2, 3), z_{k,max} (k = 1, 2, 3, 4)$$
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant

### 2. Equality constraints [165]

#### 2.1 Equality constraints of precooling part [83]

##### 7) Evaporator 2: [14]

$$c_1 \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{10p}(P_{10p}, T_{10p}, v_{10p}, z_{j,pre}) + c_1 \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{12p}(P_{12p}, T_{12p}, v_{12p}, z_{j,pre})$$

$$+ (1-c_2) \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{14p}(P_{14p}, T_{14p}, v_{14p}, z_{j,pre}) +$$

$$+ \dot{m}_{main} \cdot h_{9m}(P_{9m}, T_{9m}, v_{9m}, z_{k,max}) + \dot{m}_{NG} \cdot h_{NG}(P_{NG}, T_{NG}, v_{NG}, z_{i,NG})$$

$$= c_1 \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{11p}(P_{11p}, T_{11p}, v_{11p}, z_{j,pre}) + c_1 \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{13p}(P_{13p}, T_{13p}, v_{13p}, z_{j,pre})$$

$$+ (1-c_2) \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{15p}(P_{15p}, T_{15p}, v_{15p}, z_{j,pre}) +$$

$$+ \dot{m}_{main} \cdot h_{9m}(P_{9m}, T_{9m}, v_{9m}, z_{k,max}) + \dot{m}_{NG} \cdot h_{NG}(P_{NG}, T_{NG}, v_{NG}, z_{i,NG})$$

$$P_{10p} = P_{11p}, P_{12p} = P_{13p}, P_{14p} = P_{15p}, P_{9m} = P_{9m}, P_{9NG} = P_{9NG}$$

$$T_{11p} = T_{15p}, T_{11p} = T_{9m}, T_{11p} = T_{9NG}$$

$$v_{11p} = v_{11p}(T_{11p}, P_{11p}, z_{j,pre}), v_{13p} = v_{13p}(T_{13p}, P_{13p}, z_{j,pre}),$$

$$v_{15p} = v_{15p}(T_{15p}, P_{15p}, z_{j,pre}), v_{9m} = v_{9m}(T_{9m}, P_{9m}, z_{k,max}),$$

$$v_{9NG} = v_{9NG}(T_{9NG}, P_{9NG}, z_{i,NG})$$

##### 8) Expansion valve 2: [2]

$$h_{11p}(P_{11p}, T_{11p}, v_{11p}, z_{j,pre}) = h_{12p}(P_{12p}, T_{12p}, v_{12p}, z_{j,pre})$$

$$v_{12p} = v_{12p}(T_{12p}, P_{12p}, z_{j,pre})$$

##### 9) Evaporator 3: [11]

$$(1-c_1) \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{15p}(P_{15p}, T_{15p}, v_{15p}, z_{j,pre})$$

$$+ (1-c_2) \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{17p}(P_{17p}, T_{17p}, v_{17p}, z_{j,pre})$$

$$+ \dot{m}_{main} \cdot h_{9m}(P_{9m}, T_{9m}, v_{9m}, z_{k,max}) + \dot{m}_{NG} \cdot h_{NG}(P_{NG}, T_{NG}, v_{NG}, z_{i,NG})$$

$$= (1-c_2) \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{16p}(P_{16p}, T_{16p}, v_{16p}, z_{j,pre})$$

$$+ (1-c_2) \cdot (1-c_1) \cdot \dot{m}_{pre} \cdot h_{18p}(P_{18p}, T_{18p}, v_{18p}, z_{j,pre})$$

$$+ \dot{m}_{main} \cdot h_{9m}(P_{9m}, T_{9m}, v_{9m}, z_{k,max}) + \dot{m}_{NG} \cdot h_{NG}(P_{NG}, T_{NG}, v_{NG}, z_{i,NG})$$

$$P_{15p} = P_{16p}, P_{17p} = P_{18p}, P_{9m} = P_{9m}, P_{9NG} = P_{9NG}$$

$$T_{16p} = T_{9m}, T_{16p} = T_{9NG}$$

$$v_{16p} = v_{16p}(T_{16p}, P_{16p}, z_{j,pre}), v_{18p} = v_{18p}(T_{18p}, P_{18p}, z_{j,pre}),$$

$$v_{9m} = v_{9m}(T_{9m}, P_{9m}, z_{k,max}), v_{9NG} = v_{9NG}(T_{9NG}, P_{9NG}, z_{i,NG})$$

##### 10) Expansion valve 3: [2]

$$h_{16p}(P_{16p}, T_{16p}, v_{16p}, z_{j,pre}) = h_{17p}(P_{17p}, T_{17p}, v_{17p}, z_{j,pre})$$

$$v_{17p} = v_{17p}(T_{17p}, P_{17p}, z_{j,pre})$$

## Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle (3/6)

**1. Design variables (Operating conditions) [187]** T: Temperature / P: Pressure / v: Specific volume /  $z_{pre}$ : mole fraction of the component j at the precooling part / w: work input to the compressor per mass / c: flow rate ratio between inlet and outlet 4 /  $\dot{m}_{pre}$ : mass flow rate at the precooling refrigerant  
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant

$P_j, T_j, v_j (j = 1_p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG})$ ,  
 $T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, v_{S,2p}, v_{S,19p}, v_{S,21p}, v_{S,2m}, v_{S,4m}, v_{S,6m}$ ,  
 $W_1, W_2, W_3, W_4, W_5, W_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, v_{-f10}, v_{-f15}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [165]**

**2.1 Equality constraints of precooling part [83]**

**11) Compressor 3: [5]**

$$(1 - c_2) \cdot (1 - c_1) \cdot \dot{m}_{pre} \cdot h_{18p}(P_{18p}, T_{18p}, v_{18p}, z_{j,pre}) + W_3$$

$$= (1 - c_2) \cdot (1 - c_1) \cdot \dot{m}_{pre} \cdot h_{19p}(P_{19p}, T_{19p}, v_{19p}, z_{j,pre})$$

$$\eta = \frac{h_{S,19p}(P_{19p}, T_{S,19p}, v_{S,19p}, z_{j,pre}) - h_{18p}(P_{18p}, T_{18p}, v_{18p}, z_{j,pre})}{h_{19p}(P_{19p}, T_{19p}, v_{19p}, z_{j,pre}) - h_{18p}(P_{18p}, T_{18p}, v_{18p}, z_{j,pre})}$$

$$s_{19p}(P_{19p}, T_{19p}, v_{19p}, z_{j,pre}) = s_{18p}(P_{18p}, T_{18p}, v_{18p}, z_{j,pre})$$

$$v_{19p} = v_{18p}(P_{19p}, T_{19p}, z_{j,pre})$$

$$v_{S,19p} = v_{S,18p}(P_{19p}, T_{S,19p}, z_{j,pre})$$

**13) Compressor 2: [5]**

$$(1 - c_1) \cdot \dot{m}_{pre} \cdot h_{20p}(P_{20p}, T_{20p}, v_{20p}, z_{j,pre}) + W_2$$

$$= (1 - c_1) \cdot \dot{m}_{pre} \cdot h_{21p}(P_{21p}, T_{21p}, v_{21p}, z_{j,pre})$$

$$\eta = \frac{h_{S,21p}(P_{21p}, T_{S,21p}, v_{S,21p}, z_{j,pre}) - h_{20p}(P_{20p}, T_{20p}, v_{20p}, z_{j,pre})}{h_{21p}(P_{21p}, T_{21p}, v_{21p}, z_{j,pre}) - h_{20p}(P_{20p}, T_{20p}, v_{20p}, z_{j,pre})}$$

$$s_{20p}(P_{20p}, T_{20p}, v_{20p}, z_{j,pre}) = s_{21p}(P_{21p}, T_{21p}, v_{21p}, z_{j,pre})$$

$$v_{21p} = v_{20p}(P_{21p}, T_{21p}, z_{j,pre})$$

$$v_{S,21p} = v_{S,20p}(P_{21p}, T_{S,21p}, z_{j,pre})$$

**12) Common Header 2: [4]**

$$c_2 \cdot (1 - c_1) \cdot h_{13p}(P_{13p}, T_{13p}, v_{13p}, z_{j,pre}) + (1 - c_2) \cdot (1 - c_1) \cdot h_{19p}(P_{19p}, T_{19p}, v_{19p}, z_{j,pre})$$

$$= (1 - c_1) \cdot h_{20p}(P_{20p}, T_{20p}, v_{20p}, z_{j,pre})$$

$$P_{13p} = P_{19p} = P_{13p} = P_{20p}$$


$$v_{20p} = v_{20p}(P_{20p}, T_{20p}, z_{j,pre})$$

**14) Common Header 1: [3]**

$$(1 - c_1) \cdot h_{21p}(P_{21p}, T_{21p}, v_{21p}, z_{j,pre}) + c_1 \cdot h_{1p}(P_{7p}, T_{7p}, v_{7p}, z_{j,pre})$$

$$= h_{1p}(P_{7p}, T_{7p}, v_{7p}, z_{j,pre})$$

$$P_{7p} = P_{21p} = P_{7p} = P_{7p}$$



## Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle (4/6)

**1. Design variables (Operating conditions) [187]** T: Temperature / P: Pressure / v: Specific volume /  $z_{pre}$ : mole fraction of the component j at the precooling part / w: work input to the compressor per mass / c: flow rate ratio between inlet and outlet 4 /  $\dot{m}_{pre}$ : mass flow rate at the precooling refrigerant  
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant

$P_j, T_j, v_j (j = 1_p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG})$ ,  
 $T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, v_{S,2p}, v_{S,19p}, v_{S,21p}, v_{S,2m}, v_{S,4m}, v_{S,6m}$ ,  
 $W_1, W_2, W_3, W_4, W_5, W_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, v_{-f10}, v_{-f15}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [165]**

**2.2 Equality constraints of main cooling part [80]**

**1) Compressor 6: [6]**

$$h_{1m}(P_{1m}, T_{1m}, v_{1m}, z_{k,main}) + W_6 = h_{2m}(P_{2m}, T_{2m}, v_{2m}, z_{k,main})$$

$$\eta = \frac{h_{S,2m}(P_{2m}, T_{S,2m}, v_{S,2m}, z_{k,main}) - h_{1m}(P_{1m}, T_{1m}, v_{1m}, z_{k,main})}{h_{2m}(P_{2m}, T_{2m}, v_{2m}, z_{k,main}) - h_{1m}(P_{1m}, T_{1m}, v_{1m}, z_{k,main})}$$

$$s_{1m}(P_{1m}, T_{1m}, v_{1m}, z_{k,main}) = s_{2m}(P_{2m}, T_{2m}, v_{2m}, z_{k,main})$$

$$v_{1m} = v_{1m}(P_{1m}, T_{1m}, z_{k,main})$$

$$v_{S,2m} = v_{S,2m}(P_{2m}, T_{S,2m}, z_{k,main})$$

$$v_{2m} = v_{2m}(P_{2m}, T_{2m}, z_{k,main})$$

**3) Compressor 5: [5]**

$$h_{3m}(P_{3m}, T_{3m}, v_{3m}, z_{k,main}) + W_5 = h_{4m}(P_{4m}, T_{4m}, v_{4m}, z_{k,main})$$

$$\eta = \frac{h_{S,4m}(P_{4m}, T_{S,4m}, v_{S,4m}, z_{k,main}) - h_{3m}(P_{3m}, T_{3m}, v_{3m}, z_{k,main})}{h_{4m}(P_{4m}, T_{4m}, v_{4m}, z_{k,main}) - h_{3m}(P_{3m}, T_{3m}, v_{3m}, z_{k,main})}$$

$$s_{3m}(P_{3m}, T_{3m}, v_{3m}, z_{k,main}) = s_{4m}(P_{4m}, T_{4m}, v_{4m}, z_{k,main})$$

$$v_{S,4m} = v_{S,4m}(P_{4m}, T_{S,4m}, z_{k,main})$$

$$v_{4m} = v_{4m}(P_{4m}, T_{4m}, z_{k,main})$$

**5) Compressor 4: [5]**

$$h_{5m}(P_{5m}, T_{5m}, v_{5m}, z_{k,main}) + W_4 = h_{6m}(P_{6m}, T_{6m}, v_{6m}, z_{k,main})$$

$$\eta = \frac{h_{S,6m}(P_{6m}, T_{S,6m}, v_{S,6m}, z_{k,main}) - h_{5m}(P_{5m}, T_{5m}, v_{5m}, z_{k,main})}{h_{6m}(P_{6m}, T_{6m}, v_{6m}, z_{k,main}) - h_{5m}(P_{5m}, T_{5m}, v_{5m}, z_{k,main})}$$

$$s_{5m}(P_{5m}, T_{5m}, v_{5m}, z_{k,main}) = s_{6m}(P_{6m}, T_{6m}, v_{6m}, z_{k,main})$$

$$v_{S,6m} = v_{S,6m}(P_{6m}, T_{S,6m}, z_{k,main})$$

$$v_{6m} = v_{6m}(P_{6m}, T_{6m}, z_{k,main})$$

**2) Intercooler 2: [3]**

The temperature of the outlet of the sea water cooler is usually given.  
T=305K

$$P_{2m} = P_{2m}$$

$$v_{3m} = v_{3m}(T_{3m}, P_{3m}, z_{k,main})$$

**4) Intercooler 1: [3]**

The temperature of the outlet of the sea water cooler is usually given.  
T=305K

$$P_{2m} = P_{2m}$$


$$v_{5m} = v_{5m}(T_{5m}, P_{5m}, z_{k,main})$$

**6) Condenser 2: [3]**

The temperature of the outlet of the sea water cooler is usually given.  
T=305K

$$P_{6m} = P_{6m}$$

$$v_{7m} = v_{7m}(T_{7m}, P_{7m}, z_{k,main})$$



## Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle (5/6)

**1. Design variables (Operating conditions) [187]**  $T$ : Temperature /  $P$ : Pressure /  $v$ : Specific volume /  $z_{pre}$ : mole fraction of the component  $j$  at the precooling part /  $w$ : work input to the compressor per mass /  $c$ : flow rate ratio between inlet and outlet 4 /  $\dot{m}_{pre}$ : mass flow rate at the precooling refrigerant  
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant

$P_j, T_j, v_j (j = 1, p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG}),$   
 $T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, v_{S,2p}, v_{S,19p}, v_{S,21p}, v_{S,2m}, v_{S,4m}, v_{S,6m},$   
 $W_1, W_2, W_3, W_4, W_5, W_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, v_{-f_{10}}, v_{-f_{15}}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [165]**  
**2.2 Equality constraints of main cooling part [80]**

**7) Phase Separator 1: [7]**

$$\begin{aligned} & h_{10m} (P_{10m}, T_{10m}, v_{10m}, z_{k,main}) \\ &= v_{-f_{10}} \cdot h_{14m} (P_{14m}, T_{14m}, v_{14m}, v_{-f_{10}}, z_{k,main}) \\ &+ (1 - v_{-f_{10}}) \cdot h_{11m} (P_{11m}, T_{11m}, v_{11m}, (1 - v_{-f_{10}}), z_{k,main}) \\ & P_{10m} = P_{11m}, P_{10m} = P_{14m} \\ & T_{10m} = T_{11m}, T_{10m} = T_{14m} \\ & v_{11m} = v_{11m} (P_{11m}, T_{11m}, (1 - v_{-f_{10}}), z_{k,main}), v_{14m} = v_{14m} (P_{14m}, T_{14m}, v_{-f_{10}}, z_{k,main}) \end{aligned}$$

**8) Evaporator 4: [10]**

$$\begin{aligned} & (1 - v_{-f_{10}}) \cdot \dot{m}_{main} \cdot h_{11m} (P_{11m}, T_{11m}, v_{11m}, (1 - v_{-f_{10}}), z_{k,main}) \\ &+ v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{14m} (P_{14m}, T_{14m}, v_{14m}, v_{-f_{10}}, z_{k,main}) + \dot{m}_{main} \cdot h_{25m} (P_{25m}, T_{25m}, v_{25m}, z_{k,main}) \\ &+ \dot{m}_{NG} \cdot h_{3NG} (P_{3NG}, T_{3NG}, v_{3NG}, z_{1,NG}) \\ &= (1 - v_{-f_{10}}) \cdot \dot{m}_{main} \cdot h_{12m} (P_{12m}, T_{12m}, v_{12m}, (1 - v_{-f_{10}}), z_{k,main}) \\ &+ v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{15m} (P_{15m}, T_{15m}, v_{15m}, v_{-f_{10}}, z_{k,main}) + h_{1m} (P_{1m}, T_{1m}, v_{1m}, z_{k,main}) \\ &+ \dot{m}_{NG} \cdot h_{4NG} (P_{4NG}, T_{4NG}, v_{4NG}, z_{1,NG}) \\ & P_{11m} = P_{12m}, P_{14m} = P_{15m}, P_{25m} = P_{1m}, P_{3NG} = P_{4NG} \\ & T_{12m} = T_{15m}, T_{12m} = T_{4NG} \\ & v_{12m} = v_{12m} (P_{12m}, T_{12m}, (1 - v_{-f_{10}}), z_{k,main}), v_{15m} = v_{15m} (P_{15m}, T_{15m}, v_{-f_{10}}, z_{k,main}), \\ & v_{4NG} = v_{4NG} (P_{4NG}, T_{4NG}, z_{1,NG}) \end{aligned}$$

**9) Phase Separator 2: [7]**

$$\begin{aligned} & v_{-f_{10}} \cdot h_{15m} (P_{15m}, T_{15m}, v_{15m}, v_{-f_{10}}, z_{k,main}) \\ &= v_{-f_{15}} \cdot v_{-f_{10}} \cdot h_{19m} (P_{19m}, T_{19m}, v_{19m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ (1 - v_{-f_{15}}) \cdot v_{-f_{10}} \cdot h_{16m} (P_{16m}, T_{16m}, v_{16m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) \\ & P_{15m} = P_{16m}, P_{15m} = P_{19m} \\ & T_{15m} = T_{16m}, T_{16m} = T_{19m} \\ & v_{16m} = v_{16m} (P_{16m}, T_{16m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}), v_{19m} = v_{19m} (P_{19m}, T_{19m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \end{aligned}$$

**10) Evaporator 5: [11]**

$$\begin{aligned} & (1 - v_{-f_{15}}) \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{16m} (P_{16m}, T_{16m}, v_{16m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{15}} \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{19m} (P_{19m}, T_{19m}, v_{19m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{24m} (P_{24m}, T_{24m}, v_{24m}, v_{-f_{10}}, z_{k,main}) + \dot{m}_{NG} \cdot h_{4NG} (P_{4NG}, T_{4NG}, v_{4NG}, z_{1,NG}) \\ &= (1 - v_{-f_{15}}) \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{17m} (P_{17m}, T_{17m}, v_{17m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{15}} \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{20m} (P_{20m}, T_{20m}, v_{20m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{25m} (P_{25m}, T_{25m}, v_{25m}, v_{-f_{10}}, z_{k,main}) + \dot{m}_{NG} \cdot h_{5NG} (P_{5NG}, T_{5NG}, v_{5NG}, z_{1,NG}) \\ & P_{16m} = P_{17m}, P_{19m} = P_{20m}, P_{24m} = P_{25m}, P_{4NG} = P_{5NG} \\ & T_{17m} = T_{20m}, T_{17m} = T_{5NG} \\ & v_{17m} = v_{17m} (P_{17m}, T_{17m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}), v_{20m} = v_{20m} (P_{20m}, T_{20m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}), \\ & v_{25m} = v_{25m} (P_{25m}, T_{25m}, v_{-f_{10}}, z_{k,main}), v_{5NG} = v_{5NG} (P_{5NG}, T_{5NG}, z_{1,NG}) \end{aligned}$$

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## Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle (6/6)

**1. Design variables (Operating conditions) [187]**  $T$ : Temperature /  $P$ : Pressure /  $v$ : Specific volume /  $z_{pre}$ : mole fraction of the component  $j$  at the precooling part /  $w$ : work input to the compressor per mass /  $c$ : flow rate ratio between inlet and outlet 4 /  $\dot{m}_{pre}$ : mass flow rate at the precooling refrigerant  
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant

$P_j, T_j, v_j (j = 1, p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG}),$   
 $T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, v_{S,2p}, v_{S,19p}, v_{S,21p}, v_{S,2m}, v_{S,4m}, v_{S,6m},$   
 $W_1, W_2, W_3, W_4, W_5, W_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, v_{-f_{10}}, v_{-f_{15}}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [165]**  
**2.2 Equality constraints of main cooling part [80]**

**11) Evaporator 6: [6]**

$$\begin{aligned} & v_{-f_{15}} \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{20m} (P_{20m}, T_{20m}, v_{20m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{15}} \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{22m} (P_{22m}, T_{22m}, v_{22m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ \dot{m}_{NG} \cdot h_{3NG} (P_{3NG}, T_{3NG}, v_{3NG}, z_{1,NG}) \\ &= v_{-f_{15}} \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{21m} (P_{21m}, T_{21m}, v_{21m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{15}} \cdot v_{-f_{10}} \cdot \dot{m}_{main} \cdot h_{23m} (P_{23m}, T_{23m}, v_{23m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &+ \dot{m}_{NG} \cdot h_{4NG} (P_{4NG}, T_{4NG}, v_{4NG}, z_{1,NG}) \\ & P_{20m} = P_{21m}, P_{22m} = P_{23m} \\ & T_{21m} = T_{4NG} \\ & v_{21m} = v_{21m} (P_{21m}, T_{21m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}), \\ & v_{23m} = v_{23m} (P_{23m}, T_{23m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \end{aligned}$$

**12) Expansion valve 4: [2]**

$$\begin{aligned} & h_{12m} (P_{12m}, T_{12m}, v_{12m}, (1 - v_{-f_{10}}), z_{k,main}) \\ &= h_{13m} (P_{13m}, T_{13m}, v_{13m}, (1 - v_{-f_{10}}), z_{k,main}) \\ & v_{13m} = v_{13m} (P_{13m}, T_{13m}, (1 - v_{-f_{10}}), z_{k,main}) \end{aligned}$$

**13) Expansion valve 5: [2]**

$$\begin{aligned} & h_{17m} (P_{17m}, T_{17m}, v_{17m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) = h_{18m} (P_{18m}, T_{18m}, v_{18m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) \\ & v_{18m} = v_{18m} (P_{18m}, T_{18m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) \end{aligned}$$

**14) Expansion valve 6: [2]**

$$\begin{aligned} & h_{21m} (P_{21m}, T_{21m}, v_{21m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) = h_{22m} (P_{22m}, T_{22m}, v_{22m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ & v_{22m} = v_{22m} (P_{22m}, T_{22m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \end{aligned}$$

**15) Common Header 3: [4]**

$$\begin{aligned} & (1 - v_{-f_{10}}) \cdot h_{13m} (P_{13m}, T_{13m}, v_{13m}, (1 - v_{-f_{10}}), z_{k,main}) + v_{-f_{10}} \cdot h_{25m} (P_{25m}, T_{25m}, v_{25m}, v_{-f_{10}}, z_{k,main}) \\ &= h_{26m} (P_{26m}, T_{26m}, v_{26m}, z_{k,main}) \\ & P_{13m} = P_{25m}, P_{13m} = P_{26m}, v_{26m} = v_{26m} (P_{26m}, T_{26m}, z_{k,main}) \end{aligned}$$

**16) Common Header 4: [4]**

$$\begin{aligned} & (1 - v_{-f_{15}}) \cdot v_{-f_{10}} \cdot h_{18m} (P_{18m}, T_{18m}, v_{18m}, (1 - v_{-f_{15}}), v_{-f_{10}}, z_{k,main}) \\ &+ v_{-f_{15}} \cdot v_{-f_{10}} \cdot h_{23m} (P_{23m}, T_{23m}, v_{23m}, v_{-f_{15}}, v_{-f_{10}}, z_{k,main}) \\ &= v_{-f_{10}} \cdot h_{24m} (P_{24m}, T_{24m}, v_{24m}, v_{-f_{10}}, z_{k,main}) \\ & P_{18m} = P_{23m}, P_{18m} = P_{24m}, v_{24m} = v_{24m} (P_{24m}, T_{24m}, v_{-f_{10}}, z_{k,main}), \end{aligned}$$

$$\sum_{j=1}^3 z_{j,pre} = 1, \sum_{k=1}^4 z_{k,main} = 1$$

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**[Summary] Mathematical Model for the Determination of Optimal Operating Condition of the Generic Liquefaction Cycle**

**1. Design variables (Operating conditions) [187]**  $T$ : Temperature /  $P$ : Pressure /  $v$ : Specific volume /  $z_{pre}$ : mole fraction of the component  $j$  at the precooling part /  $w$ : work input to the compressor per mass /  $c$ : flow rate ratio between inlet and outlet 4 /  $\dot{m}_{pre}$ : mass flow rate at the precooling refrigerant  
\*Subscript 'NG': natural gas, Subscript 'main': main cooling refrigerant


$: P_i, T_i, v_i (i = 1_p, \dots, 21_p, 1_m, \dots, 26_m, 1_{NG}, \dots, 5_{NG}),$   
 $T_{S,2p}, T_{S,19p}, T_{S,21p}, T_{S,2m}, T_{S,4m}, T_{S,6m}, v_{S,2p}, v_{S,19p}, v_{S,21p}, v_{S,2m}, v_{S,4m}, v_{S,6m},$   
 $w_1, w_2, w_3, w_4, w_5, w_6, c_1, c_2, \dot{m}_{pre}, \dot{m}_{main}, v_{-f10}, v_{-f15}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [165]**  
**2.1 Equality constraints of precooling part [83]**  
**2.2 Equality constraints of main cooling part [80]**

➔


**3. Objective Function: Minimize the compressors power**

$$\text{Minimize } \dot{m}_{pre} \cdot w_1 + \dot{m}_{pre} \cdot w_2 + \dot{m}_{pre} \cdot w_3 + \dot{m}_{main} \cdot w_4 + \dot{m}_{main} \cdot w_5 + \dot{m}_{main} \cdot w_6$$

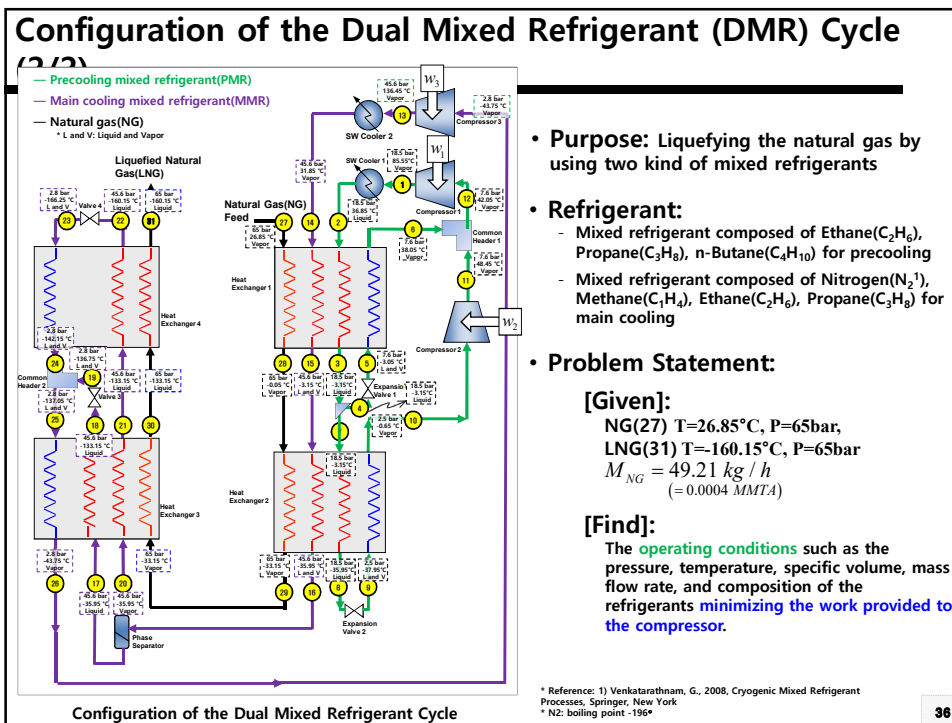
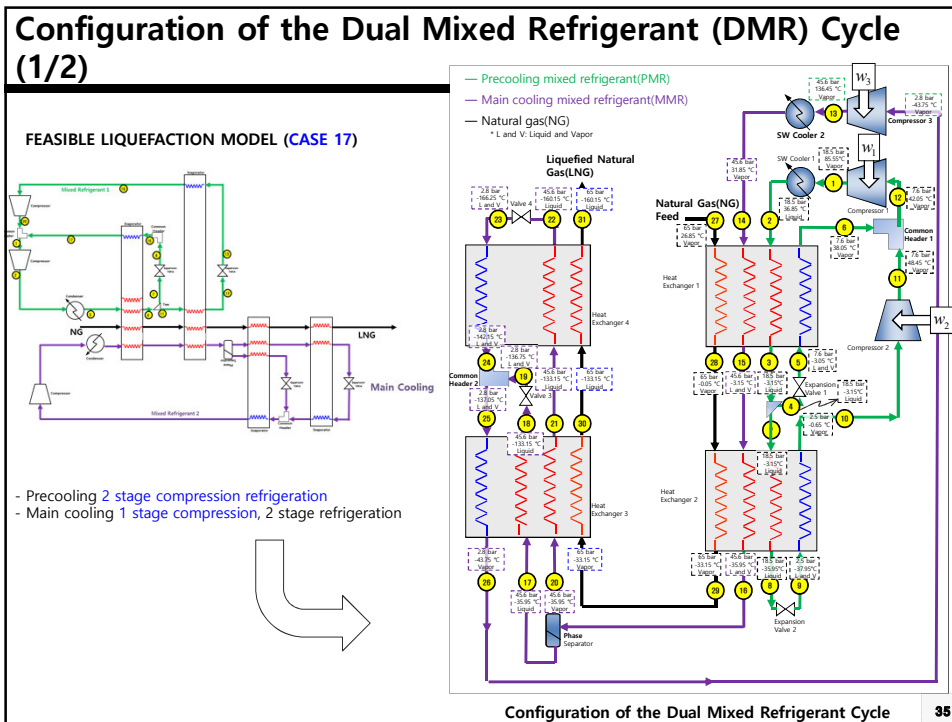
 ➔

**4. Free variables [22 = 187 - 165]**

$: P_{1p}, P_{2p}, P_{12p}, P_{17p}, T_{5p}, T_{11p}, T_{16p}, c_1, c_2, z_{1,pre}, z_{2,pre}, \dot{m}_{pre},$   
 $P_{1m}, P_{2m}, P_{4m}, P_{6m}, T_{12m}, T_{17m}, z_{1,main}, z_{2,main}, z_{3,main}, \dot{m}_{main}$

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### 3. Determination of Optimal Operating Condition of the Liquefaction Cycle for LNG FPSO



## Mathematical Model for the Determination of Optimal Operating Condition of the DMR Cycle (1/2)

Composition of the refrigerant  
 Precooling: Ethane, Propane, n-Butane  
 Main cooling: Nitrogen, Methane, Ethane, Propane  
 Natural Gas: Methane(0.75%), Ethane(5.5%), Nitrogen(4.0%),  
 Propane(2.1%), n-Butane(0.5%), i-Butane(0.3%), i-Pentane(0.1%)

**1. Design variables (Operating conditions) [107]**  
 $P_1, T_1, v_i (i = 1, \dots, 26, 28, 29, 30), T_{S,11}, T_{S,11}, T_{S,13}, v_{S,1}, v_{S,11}, v_{S,13}, w_1, w_2, w_3, c, M_{pre}, M_{main}, v_{-f}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [91]**  $\sum_{j=1}^3 z_{j,pre} = 1, \sum_{k=1}^4 z_{k,main} = 1$

### 2.1 Equality constraints of precooling part [49]

**1) Compressor 1: [6]**

$$h_{12}(P_{12}, T_{12}, v_{12}, z_{j,pre}) + w_1 = h_1(P_1, T_1, v_1, z_{j,pre})$$

$$\eta = \frac{h_3(P_1, T_1, v_1, z_{j,pre}) - h_{12}(P_{12}, T_{12}, v_{12}, z_{j,pre})}{h_1(P_1, T_1, v_1, z_{j,pre}) - h_{12}(P_{12}, T_{12}, v_{12}, z_{j,pre})}$$

$$s_{12}(P_{12}, T_{12}, v_{12}, z_{j,pre}) = s_1(P_1, T_1, v_1, z_{j,pre})$$

$$v_{12} = v_{12}(T_{12}, P_{12}, z_{j,pre})$$

$$v_{S,1} = v_{S,1}(T_{S,1}, P_1, z_{j,pre})$$

$$v_1 = v_1(T_1, P_1, z_{j,pre})$$

**2) Sea water cooler 1: [3]**

The temperature of the outlet of the sea water cooler is usually given.  
T=310K

$$P_1 = P_2$$

$$v_2 = v_2(T_2, P_2, z_{j,pre})$$

**3) Heat exchanger 1: [11]**

$$M_{pre} \cdot h_2(P_2, T_2, v_2, z_{j,pre}) + c \cdot M_{pre} \cdot h_3(P_2, T_2, v_2, z_{j,pre}) + M_{main} \cdot h_4(P_{14}, T_{14}, v_{14}, z_{k,main}) + M_{NG} \cdot h_{28}(P_{28}, T_{28}, v_{28}, z_{NG})$$

$$= M_{pre} \cdot h_1(P_1, T_1, v_1, z_{j,pre}) + c \cdot M_{pre} \cdot h_2(P_1, T_1, v_1, z_{j,pre}) + M_{main} \cdot h_5(P_{15}, T_{15}, v_{15}, z_{k,main}) + M_{NG} \cdot h_{28}(P_{28}, T_{28}, v_{28}, z_{NG})$$

$$P_2 = P_1, P_3 = P_2, P_{14} = P_{15}, P_{28} = P_{28}, T_2 = T_{15}, T_3 = T_{28}$$

$$v_3 = v_3(T_3, P_3, z_{j,pre}), v_6 = v_6(T_6, P_6, z_{j,pre}), v_{15} = v_{15}(T_{15}, P_{15}, z_{k,main}), v_{28} = v_{28}(T_{28}, P_{28}, z_{NG})$$

**4) Tee: [6]**

$$h_3(P_3, T_3, v_3, z_{j,pre}) = c \cdot h_4(P_3, T_3, v_3, z_{j,pre}) + (1-c) \cdot h_2(P_2, T_2, v_2, z_{j,pre})$$

$$P_3 = P_2, P_3 = P_3$$

$$T_4 = T_3$$

$$v_4 = v_4(T_4, P_4, z_{j,pre}), v_5 = v_5(T_5, P_5, z_{j,pre})$$

**5) Expansion Valve 1: [2]**

$$h_4(P_4, T_4, v_4) = h_5(P_5, T_5, v_5)$$

$$v_5 = v_5(T_5, P_5)$$

**6) Heat exchanger 2: [11]**

$$(1-c) \cdot M_{pre} \cdot h_4(P_4, T_4, v_4, z_{j,pre}) + (1-c) \cdot M_{pre} \cdot h_3(P_3, T_3, v_3, z_{j,pre}) + M_{main} \cdot h_5(P_{15}, T_{15}, v_{15}, z_{k,main}) + M_{NG} \cdot h_{28}(P_{28}, T_{28}, v_{28}, z_{NG})$$

$$= (1-c) \cdot M_{pre} \cdot h_2(P_2, T_2, v_2, z_{j,pre}) + (1-c) \cdot M_{pre} \cdot h_1(P_1, T_1, v_1, z_{j,pre}) + M_{main} \cdot h_6(P_{16}, T_{16}, v_{16}, z_{k,main}) + M_{NG} \cdot h_{29}(P_{29}, T_{29}, v_{29}, z_{NG})$$

$$P_7 = P_8, P_9 = P_{10}, P_{15} = P_{16}, P_{28} = P_{29}, T_8 = T_{16}, T_8 = T_{29}$$

$$v_9 = v_9(T_9, P_9, z_{j,pre}), v_{10} = v_{10}(T_{10}, P_{10}, z_{j,pre}), v_{16} = v_{16}(T_{16}, P_{16}, z_{k,main}), v_{29} = v_{29}(T_{29}, P_{29}, z_{NG})$$

**7) Expansion Valve 2: [2]**

$$h_6(P_6, T_6, v_6, z_{j,pre}) = h_7(P_7, T_7, v_7, z_{j,pre})$$

$$v_7 = v_7(T_7, P_7, z_{j,pre})$$

**8) Compressor 2: [5]**

$$(1-c) \cdot h_8(P_{10}, T_{10}, v_{10}, z_{j,pre}) + w_2 = (1-c) \cdot h_7(P_{11}, T_{11}, v_{11}, z_{j,pre})$$

$$\eta = \frac{h_8(P_{11}, T_{11}, v_{11}, z_{j,pre}) - h_{10}(P_{10}, T_{10}, v_{10}, z_{j,pre})}{h_7(P_{11}, T_{11}, v_{11}, z_{j,pre}) - h_{10}(P_{10}, T_{10}, v_{10}, z_{j,pre})}$$

$$s_{10}(P_{10}, T_{10}, v_{10}, z_{j,pre}) = s_{11}(P_{11}, T_{11}, v_{11}, z_{j,pre})$$

$$v_{S,11} = v_{S,11}(P_{11}, T_{S,11}, z_{j,pre})$$

$$v_{11} = v_{11}(T_{11}, P_{11}, z_{j,pre})$$

**9) Common header 1: [3]**

$$c \cdot h_1(P_1, T_1, v_1, z_{j,pre}) + (1-c) \cdot h_2(P_2, T_2, v_2, z_{j,pre}) = h_3(P_3, T_3, v_3, z_{j,pre})$$

$$= h_4(P_4, T_4, v_4, z_{j,pre}) \quad P_1 = P_{11}, P_1 = P_2$$

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## Mathematical Model for the Determination of Optimal Operating Condition of the DMR Cycle (2/2)

Composition of the refrigerant  
 Precooling: Ethane, Propane, n-Butane  
 Main cooling: Nitrogen, Methane, Ethane, Propane  
 Natural Gas: Methane(0.75%), Ethane(5.5%), Nitrogen(4.0%),  
 Propane(2.1%), n-Butane(0.5%), i-Butane(0.3%), i-Pentane(0.1%)

**1. Design variables (Operating conditions) [107]**  
 $P_1, T_1, v_i (i = 1, \dots, 26, 28, 29, 30), T_{S,11}, T_{S,11}, T_{S,13}, v_{S,1}, v_{S,11}, v_{S,13}, w_1, w_2, w_3, c, M_{pre}, M_{main}, v_{-f}, z_{j,pre} (j = 1, 2, 3), z_{k,main} (k = 1, 2, 3, 4)$

**2. Equality constraints [91]**  $\sum_{j=1}^3 z_{j,pre} = 1, \sum_{k=1}^4 z_{k,main} = 1$

### 2.2 Equality constraints of main cooling part [40]

**10) Compressor 3: [6]**

$$h_{26}(P_{26}, T_{26}, v_{26}, z_{k,main}) + w_3 = h_3(P_{13}, T_{13}, v_{13}, z_{k,main})$$

$$\eta = \frac{h_{S,13}(P_{13}, T_{S,13}, v_{S,13}, z_{k,main}) - h_{26}(P_{26}, T_{26}, v_{26}, z_{k,main})}{h_3(P_{13}, T_{13}, v_{13}, z_{k,main}) - h_{26}(P_{26}, T_{26}, v_{26}, z_{k,main})}$$

$$s_{26}(P_{26}, T_{26}, v_{26}, z_{k,main}) = s_{13}(P_{13}, T_{S,13}, v_{S,13}, z_{k,main})$$

$$v_{26} = v_{26}(T_{26}, P_{26}, z_{k,main})$$

$$v_{S,13} = v_{S,13}(T_{S,13}, P_{13}, z_{k,main})$$

$$v_{13} = v_{13}(T_{13}, P_{13}, z_{k,main})$$

**11) Sea water cooler 2: [3]**

The temperature of the outlet of the sea water cooler is usually given.  
T=305K

$$P_{13} = P_{14} \quad v_{14} = v_{14}(T_{14}, P_{14}, z_{k,main})$$

**12) Phase separator: [7]**

$$h_{16}(P_{16}, T_{16}, v_{16}, z_{k,main}) = v_{-f} \cdot h_{20}(P_{20}, T_{20}, v_{20}, z_{k,main}) + (1-v_{-f}) \cdot h_7(P_{17}, T_{17}, v_{17}, z_{k,main})$$

$$P_{16} = P_{17}, P_{16} = P_{20} \quad T_{17} = T_{20} \quad T_{16} = T_{17}$$

$$v_{17} = v_{17}(T_{17}, P_{17}, z_{k,main}), v_{20} = v_{20}(T_{20}, P_{20}, z_{k,main})$$

**13) Heat exchanger 3: [10]**

$$v_{-f} \cdot M_{main} \cdot h_{20}(P_{20}, T_{20}, v_{20}, z_{k,main}) + (1-v_{-f}) \cdot M_{main} \cdot h_7(P_{17}, T_{17}, v_{17}, z_{k,main})$$

$$+ M_{main} \cdot h_{25}(P_{25}, T_{25}, v_{25}, z_{k,main}) + M_{NG} \cdot h_{29}(P_{29}, T_{29}, v_{29}, z_{NG})$$

$$= v_{-f} \cdot M_{main} \cdot h_{21}(P_{21}, T_{21}, v_{21}, z_{k,main}) + (1-v_{-f}) \cdot M_{main} \cdot h_{18}(P_{18}, T_{18}, v_{18}, z_{k,main})$$

$$+ M_{main} \cdot h_{26}(P_{26}, T_{26}, v_{26}, z_{k,main}) + M_{NG} \cdot h_{30}(P_{30}, T_{30}, v_{30}, z_{NG})$$

$$P_{20} = P_{21}, P_{17} = P_{18}, P_{25} = P_{26}, P_{29} = P_{30} \quad T_{21} = T_{18}, T_{21} = T_{18}$$

$$v_{21} = v_{21}(T_{21}, P_{21}, z_{k,main}), v_{18} = v_{18}(T_{18}, P_{18}, z_{k,main}), v_{30} = v_{30}(T_{30}, P_{30}, z_{NG})$$

**14) Expansion Valve 3: [2]**

$$h_{18}(P_{18}, T_{18}, v_{18}, z_{k,main}) = h_{19}(P_{19}, T_{19}, v_{19}, z_{k,main})$$

$$v_{19} = v_{19}(T_{19}, P_{19})$$

**15) Heat exchanger 4: [6]**

$$v_{-f} \cdot M_{main} \cdot h_{21}(P_{21}, T_{21}, v_{21}, z_{k,main}) + v_{-f} \cdot M_{main} \cdot h_{23}(P_{23}, T_{23}, v_{23}, z_{k,main}) + M_{NG} \cdot h_{30}(P_{30}, T_{30}, v_{30}, z_{NG})$$

$$= v_{-f} \cdot M_{main} \cdot h_{22}(P_{22}, T_{22}, v_{22}, z_{k,main}) + v_{-f} \cdot M_{main} \cdot h_{24}(P_{24}, T_{24}, v_{24}, z_{k,main}) + M_{NG} \cdot h_{30}(P_{30}, T_{30}, v_{30}, z_{NG})$$

$$P_{21} = P_{22}, P_{23} = P_{24}, T_{22} = T_{31} \quad v_{22} = v_{22}(T_{22}, P_{22}, z_{k,main}), v_{24} = v_{24}(T_{24}, P_{24}, z_{k,main})$$

**16) Expansion Valve 4: [2]**

$$h_{22}(P_{22}, T_{22}, v_{22}, z_{k,main}) = h_{23}(P_{23}, T_{23}, v_{23}, z_{k,main})$$

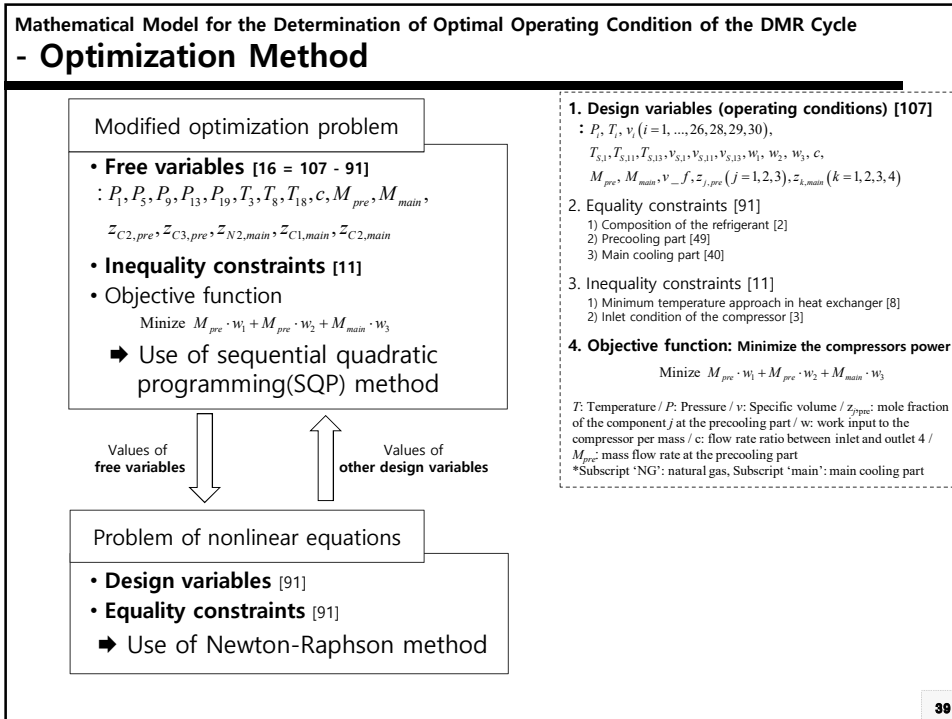
$$v_{23} = v_{23}(T_{23}, P_{23}, z_{k,main})$$

**17) Common header 2: [4]**

$$(1-v_{-f}) \cdot h_{19}(P_{19}, T_{19}, v_{19}, z_{k,main}) + v_{-f} \cdot h_{24}(P_{24}, T_{24}, v_{24}, z_{k,main}) = h_{25}(P_{25}, T_{25}, v_{25}, z_{k,main})$$

$$P_{19} = P_{24}, P_{19} = P_{25} \quad v_{25} = v_{25}(T_{25}, P_{25}, z_{k,main})$$

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### Mathematical Model for the Determination of Optimal Operating Condition of the DMR Cycle - Comparison between Optimization Result and Existing Study

P[bar], T[K], v[m<sup>3</sup>/mol], w[J/mol], m[mol/s], W[kW]

Result obtained by <a href="#">this study</a> :										Result obtained by <a href="#">Venkatarathnam<sup>1)</sup></a> :										[%]: Difference				
											[%]		[%]		[%]		[%]		[%]		[%]			
P1	19.17	P11	7.59	P21	48.57	Ts1		349.62		P1	19.20	0.2%	P11	7.60	0.2%	P21	48.60	0.1%	Ts1	349.59	0.0%			
T1	355.42	T11	312.51	T21	143.54	Ts11		305.49		T1	360.25	1.3%	T11	313.59	0.3%	T21	144.70	0.8%	Ts11	305.75	0.1%			
v1	0.001274	v11	0.003078	v21	0.000045	Ts13		388.66		v1	0.001291	1.3%	v11	0.003089	0.3%	v21	0.000045	0.8%	Ts13	389.24	0.1%			
P2	19.20	P12	7.59	P22	48.57	vs1		0.0014673		P2	19.20	0.0%	P12	7.60	0.2%	P22	48.60	0.1%	vs1	0.001467	0.0%			
T2	310.00	T12	308.94	T22	113.00	vs11		0.0032314		T2	310.00	0.0%	T12	313.79	1.5%	T22	113.00	0.0%	vs11	0.003234	0.1%			
v2	0.000122	v12	0.003044	v22	0.000040	vs13		0.0006245		v2	0.000123	0.0%	v12	0.003092	1.5%	v22	0.000040	0.0%	vs13	0.000625	0.1%			
P3	19.20	P13	48.57	P23	3.18	w1[J/mol]		2738.87		P3	19.20	0.0%	P13	48.60	0.1%	P23	3.00	6.0%	w1[J/mol]	2767.71	1.0%			
T3	273.33	T13	413.82	T23	106.95	w2		1099.84		T3	273.10	0.1%	T13	418.13	1.0%	T23	106.89	0.1%	w2	1103.87	0.4%			
v3	0.000087	v13	0.000663	v23	0.000304	w3		8233.20		v3	0.000087	0.1%	v13	0.000669	1.0%	v23	0.000304	0.1%	w3	8441.32	2.5%			
B4	19.20	P14	48.57	P24	3.18	c		0.6000		P4	19.20	0.0%	P14	48.60	0.1%	P24	3.00	6.0%	c	0.5980	0.3%			
T4	273.33	T14	305.00	T24	140.47	mpre[mol/s]		0.8982		T4	273.10	0.1%	T14	305.00	0.0%	T24	141.79	0.9%	mpre[mol/s]	0.913	1.6%			
v4	0.000087	v14	0.000389	v24	0.002884	mmain		0.9776		v4	0.000087	0.1%	v14	0.000389	0.0%	v24	0.002911	0.9%	mmain	1	2.2%			
P5	7.59	P15	48.57	P25	3.18	zpre Ethane		0.2481		P5	7.60	0.2%	P15	48.60	0.1%	P25	3.00	6.0%	z1 Ethane	0.2481	0.0%			
T5	269.94	T15	273.33	T25	140.36	zpre Propane		0.6410		T5	269.73	0.1%	T15	273.10	0.1%	T25	140.26	0.1%	z2 Propane	0.6416	0.1%			
v5	0.000158	v15	0.000248	v25	0.001089	zpre n-Butane		0.1110		v5	0.000159	0.1%	v15	0.000248	0.1%	v25	0.001090	0.1%	z3 n-Butane	0.1103	0.6%			
P6	7.59	P16	48.57	P26	3.18	zmain Nitrogen		0.0710		P6	7.60	0.2%	P16	48.60	0.1%	P26	3.00	5.9%	z1 Nitrogen	0.070	1.4%			
T6	306.51	T16	240.00	T26	236.43	zmain Methane		0.4153		T6	313.93	2.4%	T16	240.00	0.0%	T26	236.95	0.2%	z2 Methane	0.4180	0.6%			
v6	0.003020	v16	0.000141	v26	0.006349	zmain Ethane		0.2887		v6	0.003093	2.4%	v16	0.000141	0.0%	v26	0.006363	0.2%	z3 Ethane	0.2990	3.5%			
P7	19.20	P17	48.57	P28	65.00	<b>zmain Propane</b>		<b>0.225054</b>		P7	19.20	0.0%	P17	48.60	0.1%	P28	65.00	0.0%	<b>z4 Propane</b>	<b>0.2130</b>	<b>5.7%</b>			
T7	273.23	T17	240.00	T28	273.33	<b>Objective function(W)</b>		<b>11.497</b>		T7	273.00	0.1%	T17	240.00	0.0%	T28	273.10	0.1%	<b>Objective function(W)</b>	<b>11.976</b>	<b>4.0%</b>			
v7	0.000087	v17	0.000071	v28	0.000286					v7	0.000087	0.1%	v17	0.000071	0.0%	v28	0.000286	0.1%						
P8	19.20	P18	48.57	P29	65.00					P8	19.20	0.0%	P18	48.60	0.1%	P29	65.00	0.0%						
T8	240.00	T18	143.54	T29	240.00					T8	240.00	0.0%	T18	144.70	0.8%	T29	240.00	0.0%						
v8	0.000079	v18	0.000052	v29	0.000210					v8	0.000079	0.0%	v18	0.000053	0.8%	v29	0.000210	0.0%						
P9	2.80	P19	<b>3.18</b>	P30	65.00					P9	2.80	0.1%	P19	<b>3.00</b>	<b>6.0%</b>	P30	65.00	0.0%						
T9	236.90	T19	139.21	T30	143.54					T9	236.50	0.2%	T19	139.14	0.1%	T30	144.70	0.8%						
v9	0.000257	v19	0.000368	v30	0.000043					v9	0.000258	0.2%	v19	0.000368	0.1%	v30	0.000044	0.8%						
P10	2.80	P20	48.57							P10	2.80	0.1%	P20	48.60	0.1%									
T10	267.67	T20	240.00							T10	268.67	0.4%	T20	240.00	0.0%									
v10	0.007509	v20	0.000312							v10	0.007537	0.4%	v20	0.000312	0.0%									

➔ The result of the optimal operating condition of the DMR cycle obtained by this study **saves 4.0% of the total required power consumption** compared with the past relevant research.

\* Reference: 1) Venkatarathnam, G., 2008. Cooegenic Mixed Refrigerant Processes. Springer, New York.

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### Mathematical Model for the Determination of Optimal Operating Condition of the DMR Cycle - Validation of the Optimization Result

**1. Design variables (Operating Conditions) [107]**  $P$ [bar],  $T$ [K],  $v$ [m<sup>3</sup>/mol],  $w$ [l/mol],  $m$ [mol/s],  $W$ [kW]

**2. Equality constraints of Precooling part [91]**

► To verify the mathematical model of this research for the DMR Cycle, 16 design variables are assumed and the values of the other design variables are compared between the result obtained by the mathematical model and HYSYS. [%]: Difference

Mathematical model of this study for the DMR Cycle							HYSYS								
								[%]		[%]		[%]		[%]	
P1	19.64	P11	8.19	P21	48.92	Ts1	346.97	P1	19.64	P11	8.19	P21	48.92	Ts1	347.36
T1	352.31	T11	313.47	T21	140.36	Ts11	306.24	T1	352.71	T11	313.95	T21	140.36	Ts11	306.71
v1	0.001208	v11	0.002846	v21	0.000043	Ts13	395.94	v1	0.001209	v11	0.002847	v21	0.000043	Ts13	394.75
P2	19.64	P12	8.19	P22	48.92	vs1	0.001467	P2	19.64	P12	8.19	P22	48.92	vs1	0.001467
T2	310.00	T12	307.89	T22	113.00	vs11	0.003234	T2	310.00	T12	308.27	T22	113.00	vs11	0.003234
v2	0.000092	v12	0.002774	v22	0.000039	vs13	0.000625	v2	0.000092	v12	0.002774	v22	0.000039	vs13	0.000625
P3	19.64	P13	48.92	P23	2.79	w1[l/mol]	2505.86	P3	19.64	P13	48.92	P23	2.79	w1[l/mol]	2514.02
T3	275.01	T13	422.24	T23	105.80	w2	1187.98	T3	275.01	T13	420.94	T23	105.74	w2	1191.01
v3	0.000081	v13	0.000668	v23	0.000360	w3	8746.96	v3	0.000081	v13	0.000668	v23	0.000360	w3	8705.69
P4	19.64	P14	48.92	P24	2.79	c	0.584643	P4	19.64	P14	48.92	P24	2.79	c	0.584643
T4	275.01	T14	305.00	T24	137.74	Mpre[mol/s]	0.932866	T4	275.01	T14	305.00	T24	137.87	Mpre[mol/s]	0.932866
v4	0.000081	v14	0.000374	v24	0.003016	Mmain	0.957021	v4	0.000081	v14	0.000374	v24	0.003016	Mmain	0.957021
P5	8.19	P15	48.92	P25	2.79	zpre_Ethane	0.253899	P5	8.19	P15	48.92	P25	2.79	zpre_Ethane	0.253899
T5	272.01	T15	275.01	T25	137.41	zpre_Propane	0.63883	T5	272.01	T15	275.01	T25	137.26	zpre_Propane	0.63883
v5	0.000142	v15	0.000242	v25	0.001016	zpre_n-Butane	0.107275	v5	0.000142	v15	0.000242	v25	0.001016	zpre_n-Butane	0.107275
P6	8.19	P16	48.92	P26	2.79	zmain_Nitrogen	0.069317	P6	8.19	P16	48.92	P26	2.79	zmain_Nitrogen	0.069317
T6	303.97	T16	239.64	T26	237.63	zmain_Methane	0.405874	T6	304.21	T16	239.64	T26	236.62	zmain_Methane	0.405874
v6	0.002722	v16	0.000131	v26	0.006835	zmain_Ethane	0.2964	v6	0.002722	v16	0.000131	v26	0.006835	zmain_Ethane	0.2964
P7	19.64	P17	48.92	P28	65.00	zmain_Propane	0.228409	P7	19.64	P17	48.92	P28	65.00	zmain_Propane	0.228409
T7	275.01	T17	239.64	T28	275.01			T7	275.01	T17	239.64	T28	275.01		
v7	0.000081	v17	0.000068	v28	0.000290			v7	0.000081	v17	0.000068	v28	0.000290		
P8	19.64	P18	48.92	P29	65.00			P8	19.64	P18	48.92	P29	65.00		
T8	239.64	T18	140.36	T29	239.64			T8	239.64	T18	140.36	T29	239.64		
v8	0.000074	v18	0.000050	v29	0.000209			v8	0.000074	v18	0.000050	v29	0.000209		
P9	2.86	P19	2.79	P30	65.00			P9	2.86	P19	2.79	P30	65.00		
T9	236.58	T19	136.08	T30	140.36			T9	236.58	T19	135.97	T30	140.36		
v9	0.000232	v19	0.000344	v30	0.000042			v9	0.000232	v19	0.000344	v30	0.000042		
P10	2.86	P20	48.92					P10	2.86	P20	48.92				
T10	265.92	T20	239.64					T10	266.38	T20	239.64				
v10	0.007301	v20	0.000311					v10	0.007303	v20	0.000311				

