

Development and Property evaluation of high entropy bulk metallic glass in RE-Al-Co system

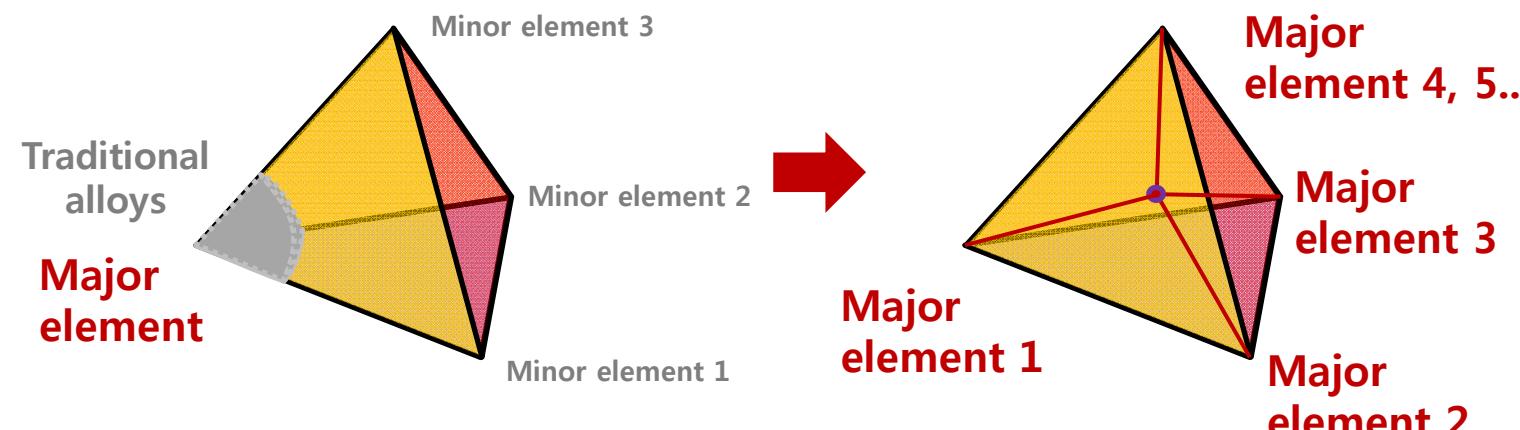
Current Status of Structural Materials
– Advanced Research of Structural Materials –

2014-031814
Jinyeon Kim

Contents

- Basic concepts of high entropy alloys and bulk metallic glasses
 - History of high entropy bulk metallic glasses (HE BMGs)
 - Development of novel HE BMGs in $\text{RE}^{(1)}_{18}\text{RE}^{(2)}_{18}\text{RE}^{(3)}_{20}\text{Al}_{24}\text{Co}_{20}$ alloys
- 1) Fragility and mechanical properties → Characteristic of BMG
- 2) Crystallization behaviors → Characteristic of HEA
(Stable liquid to solid **during cooling** & metastable liquid to solid **during heating**)
- “ HE BMGs ” → Bridging between HEAs and BMGs**

Basic concepts of high entropy alloy (HEA)



Conventional alloy system

Ex) 304 steel - Fe₇₄Cr₁₈Ni₈

(1) Thermodynamic : high entropy effect

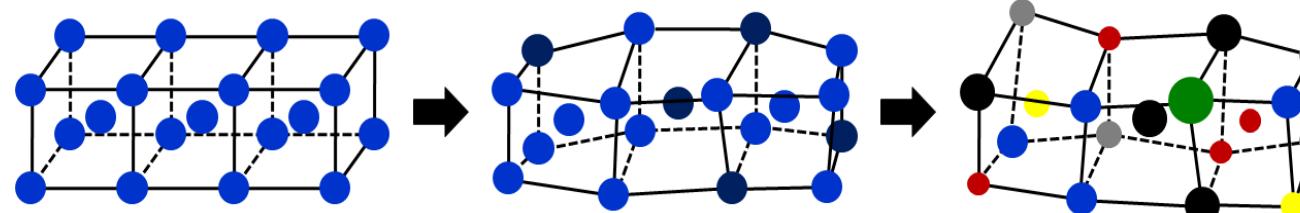
(3) Structure : severe lattice distortion effect

High entropy alloy system

Ex) Al₂₀Co₂₀Cr₂₀Fe₂₀Ni₂₀

(2) Kinetics : sluggish diffusion effect

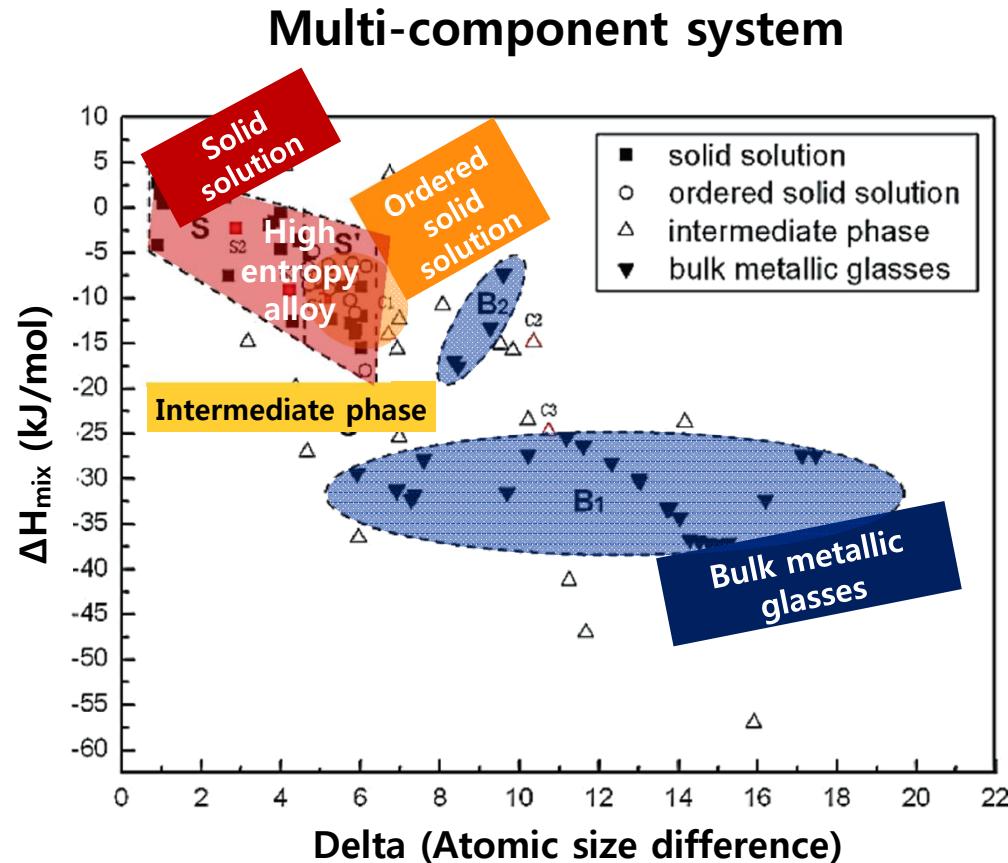
(4) Property : cocktail effect



Yong Zhang et al., Adv. Eng. Mat. P534-538, 2008

→ Unique dislocation behavior & High temperature stability

Empirical rules: high entropy alloy vs bulk metallic glass



High entropy alloy (HEA)

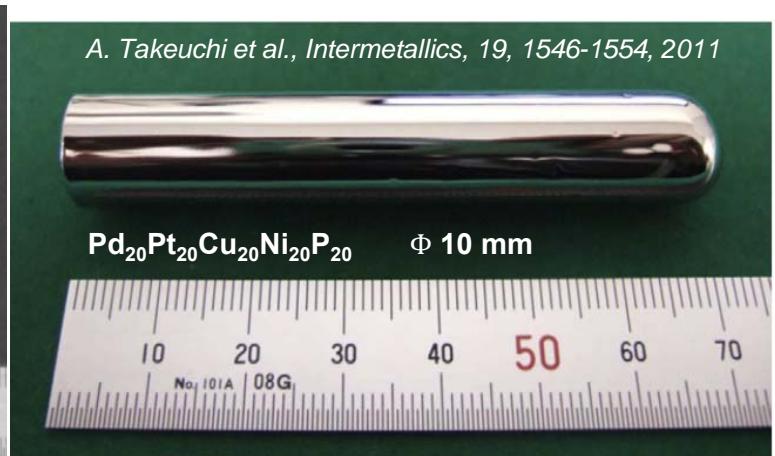
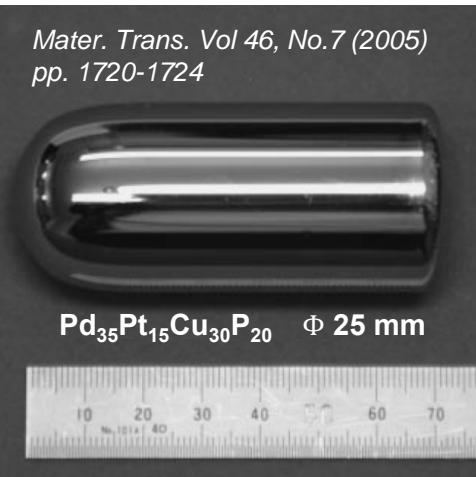
- ▶ Multi-component systems consisting of more than five elements
- ▶ Small difference of atomic size ratio under 12%
- ▶ Almost zero value of heats of mixing among the three main constituent elements

Bulk metallic glass (BMG)

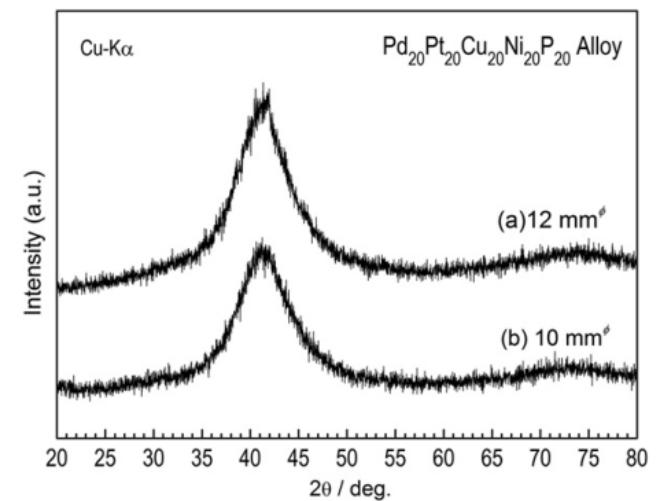
- ▶ multi-component systems consisting of more than three elements
- ▶ Significant difference in atomic size ratios above about 12% among the three constituent elements
- ▶ Negative heats of mixing among the three main constituent elements

→ By linking the BMGs and HEAs, high entropy bulk metallic glass could be developed.
→ We can understand more about multi-component alloy systems.

First report for high entropy bulk metallic glass



- Most BMGs have major constituent element. but, in this case, BMG has similar atomic percent of constituents.
- $\mathbf{Pd_{20}Pt_{20}Cu_{20}Ni_{20}P_{20}}$ HE-BMG was firstly reported by combining $\mathbf{Pd_{40}Cu_{30}Ni_{10}P_{20}}$, $\mathbf{Pt_{57.5}Cu_{14.7}Ni_{5.3}P_{22.5}}$ ($D_{\max}=20$ mm) and $\mathbf{Pd_{35}Pt_{15}Cu_{30}P_{20}}$ BMGs with high GFA.

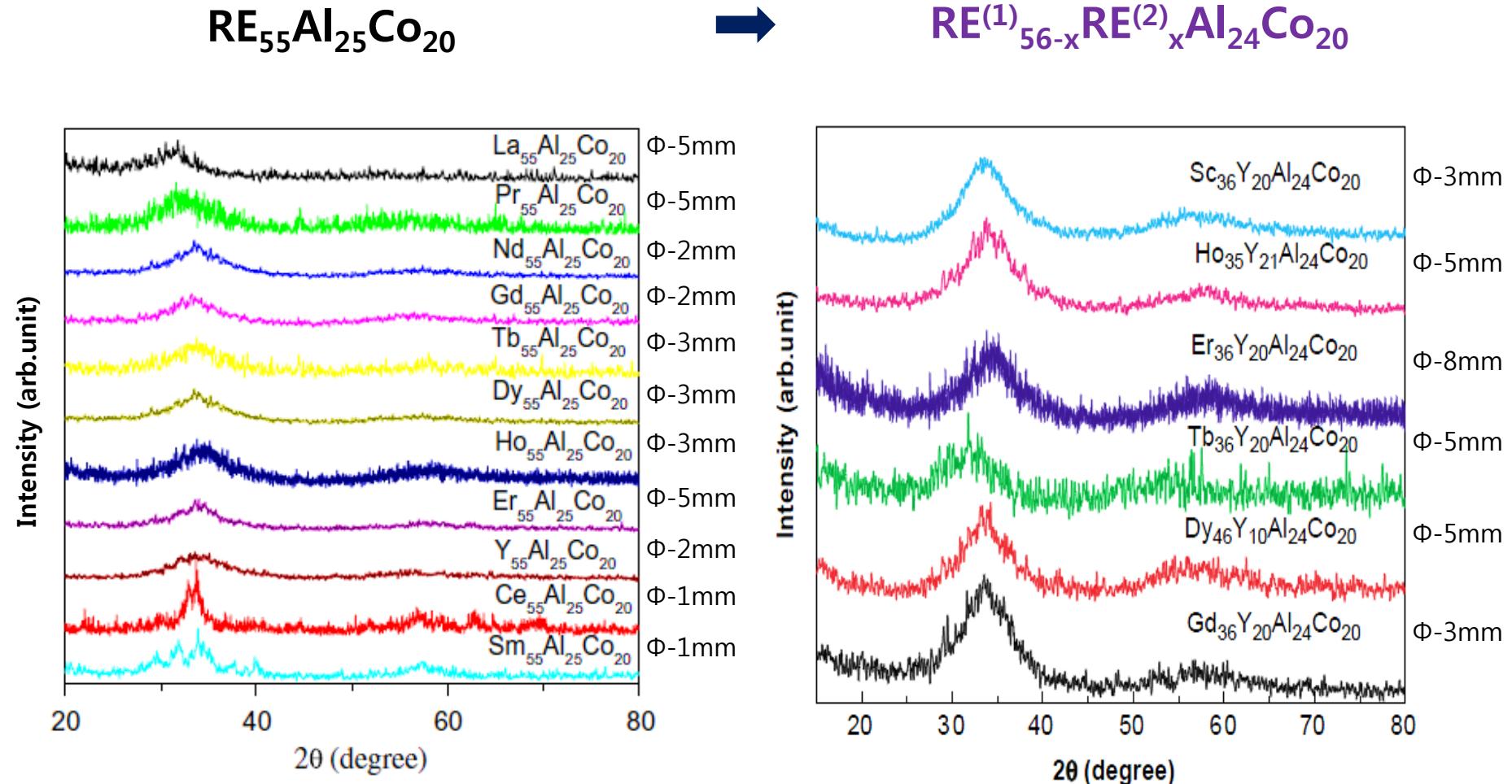


Reported high entropy bulk metallic glasses

Composition (at%)	D _c (mm)	T _g (K)	T _x (K)	ΔT (K)	σ _y (MPa)
Ti ₂₀ Zr ₂₀ Hf ₂₀ Cu ₂₀ Ni ₂₀	1.5	658	711	53	1920
Zn ₂₀ Ca ₂₀ Sr ₂₀ Yb ₂₀ (Li _{0.55} Mg _{0.45}) ₂₀	3	323	348	25	<500
Sr ₂₀ Ca ₂₀ Yb ₂₀ Mg ₂₀ Zn ₂₀	~5	353	389	36	<500
Sr ₂₀ Ca ₂₀ Yb ₂₀ Mg ₂₀ (Zn _{0.5} Cu _{0.5}) ₂₀	~5	351	391	40	<500
Er ₂₀ Tb ₂₀ Dy ₂₀ Ni ₂₀ Al ₂₀	2	-	-	-	-

→ Development of Rare earth high entropy bulk metallic glass

Rare earth element based BMGs: (a) relatively high glass former

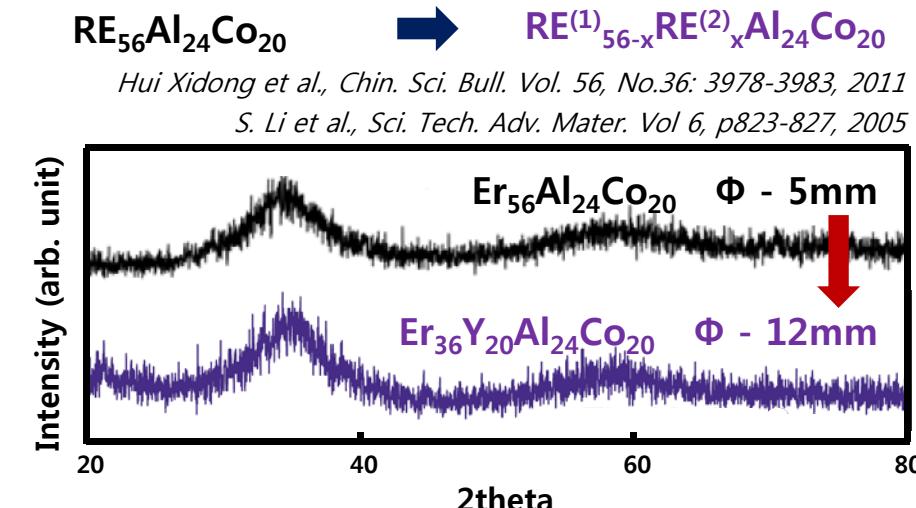


S. Li et al., Journal of Non-Crystalline Solids 354 (2008) 1080-1088

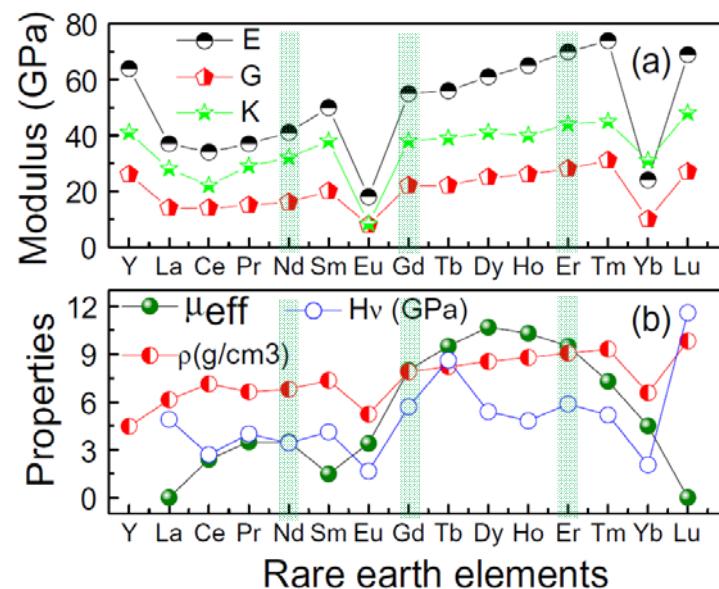
S. Li et al., Science and technology of Advanced Materials 6 (2005) 823-827

► RE based BMGs → high GFA in simple ternary and quarternary system

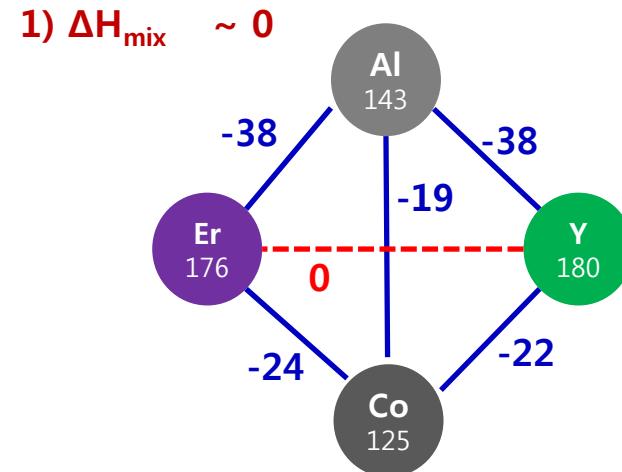
Rare earth element based BMGs: (b) Characteristics of HE alloys



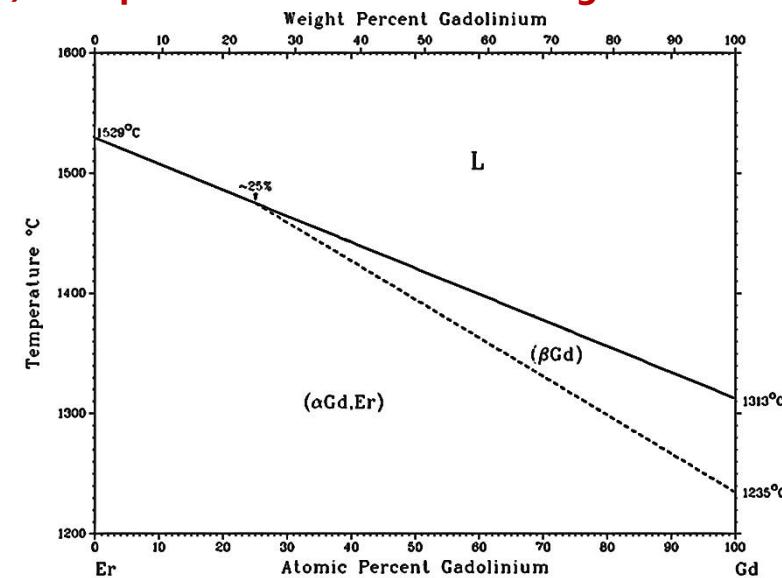
2) Similar properties among RE elements



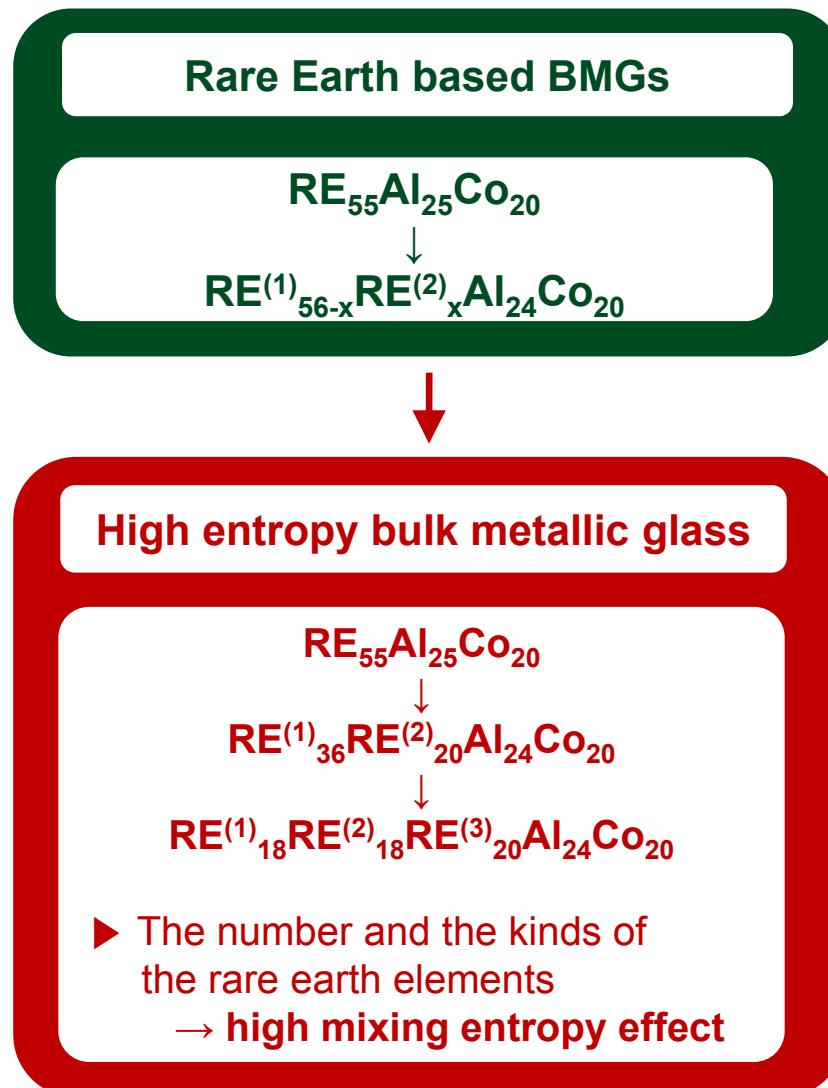
Q. Luo et al., J. Non-Cryst. Solids, 355, p759-775, 2009



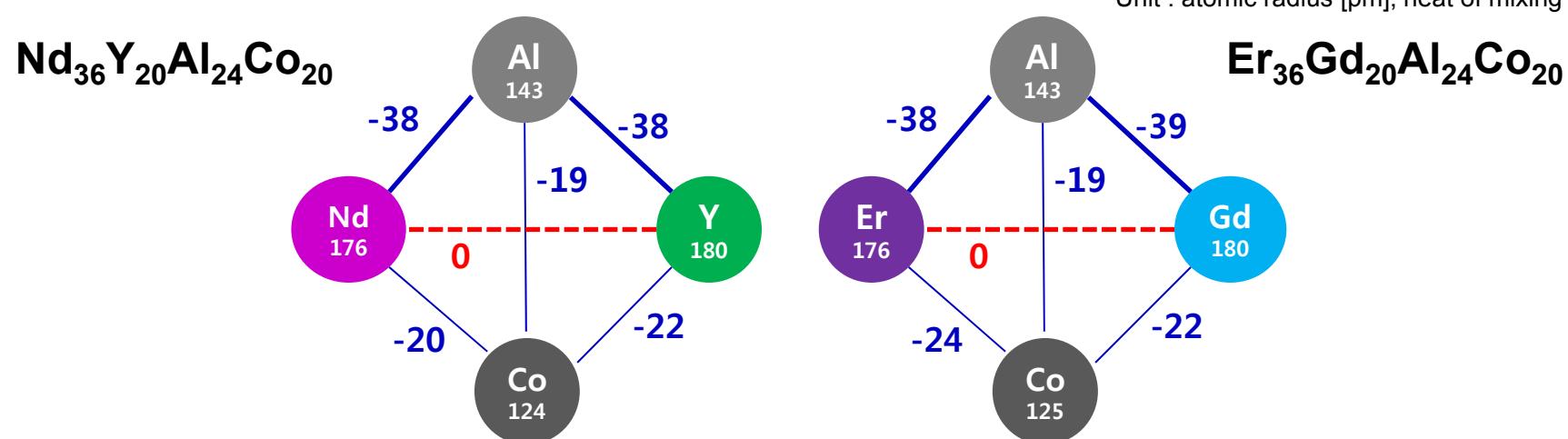
3) Complete solid solution among RE elements



Alloy design of high entropy bulk metallic glasses



Newly developed quarternary $\text{RE}^{(1)}_{36}\text{RE}^{(2)}_{20}\text{Al}_{24}\text{Co}_{20}$ BMGs



- Combination of a heavy rare earth elements, Al and transition metal lead to the large mismatch in atomic size and the large heat of mixing of the constituent elements → **high GFA**

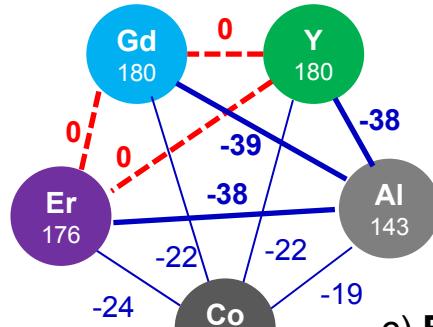
Composition	T _g (K)	T _x (K)	ΔT _x (K)	ΔH (J/g)	D _{max} (mm)	
Nd ₃₆ Y ₂₀ Al ₂₄ Co ₂₀	567	634	67	-92	≥2	Present work
Er ₃₆ Gd ₂₀ Al ₂₄ Co ₂₀	639	677	38	-72	≥2	Present work



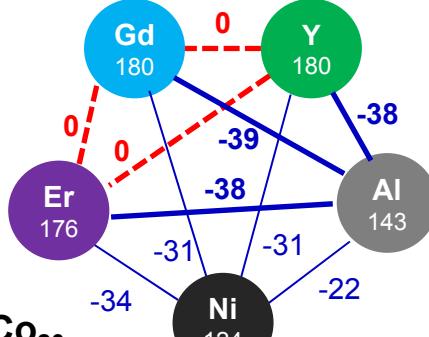
Extension of alloy design for RE¹RE²RE³-Al-TM HE BMG system

Alloy design : RE⁽¹⁾₁₈RE⁽²⁾₁₈RE⁽³⁾₂₀Al₂₄Co₂₀ HE BMGs

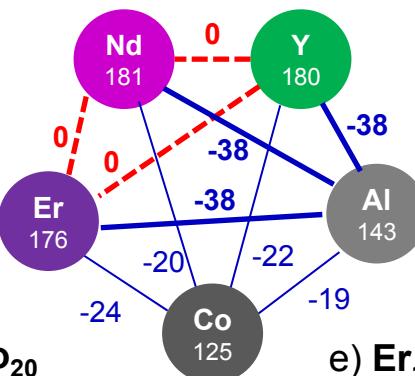
a) Er₁₈Gd₁₈Y₂₀Al₂₄Co₂₀



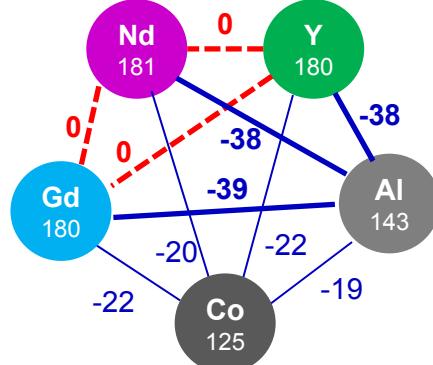
b) Er₁₈Gd₁₈Y₂₀Al₂₄Ni₂₀



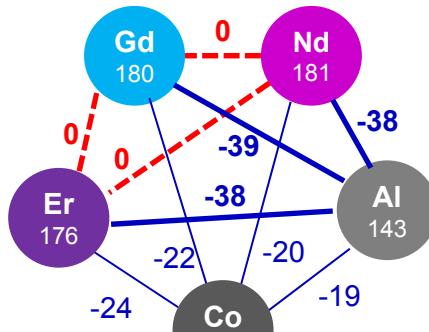
c) Er₁₈Nd₁₈Y₂₀Al₂₄Co₂₀



d) Gd₁₈Nd₁₈Y₂₀Al₂₄Co₂₀



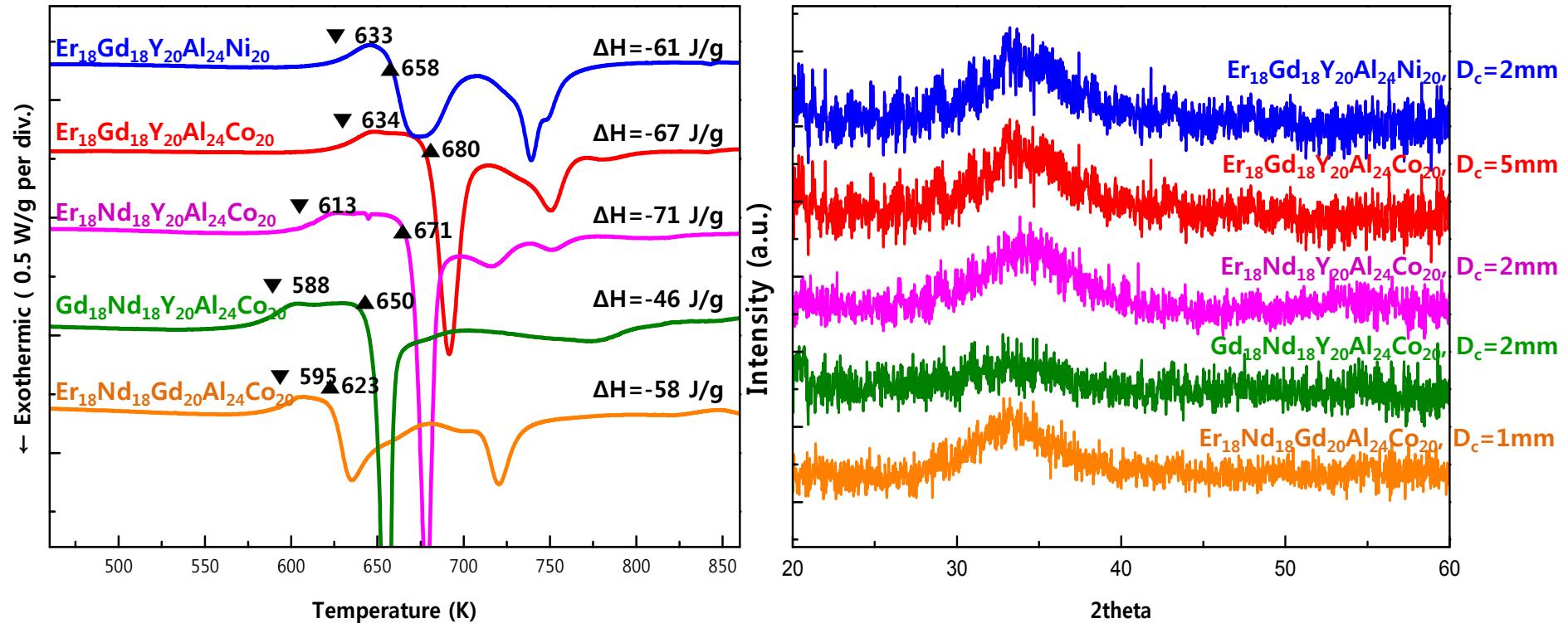
e) Er₁₈Nd₁₈Gd₂₀Al₂₄Co₂₀



Unit : atomic radius [pm], heat of mixing [kJ/mol]

Glass formation in $\text{RE}^{(1)}_{18}\text{RE}^{(2)}_{18}\text{RE}^{(3)}_{20}\text{Al}_{24}\text{Co}_{20}$ HE samples

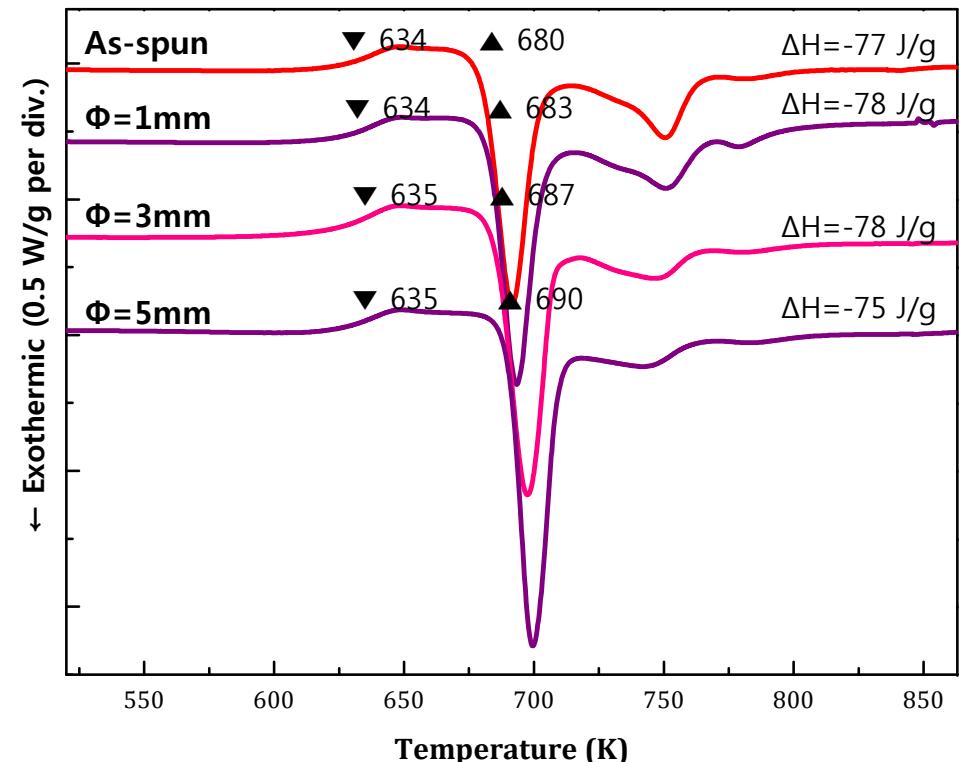
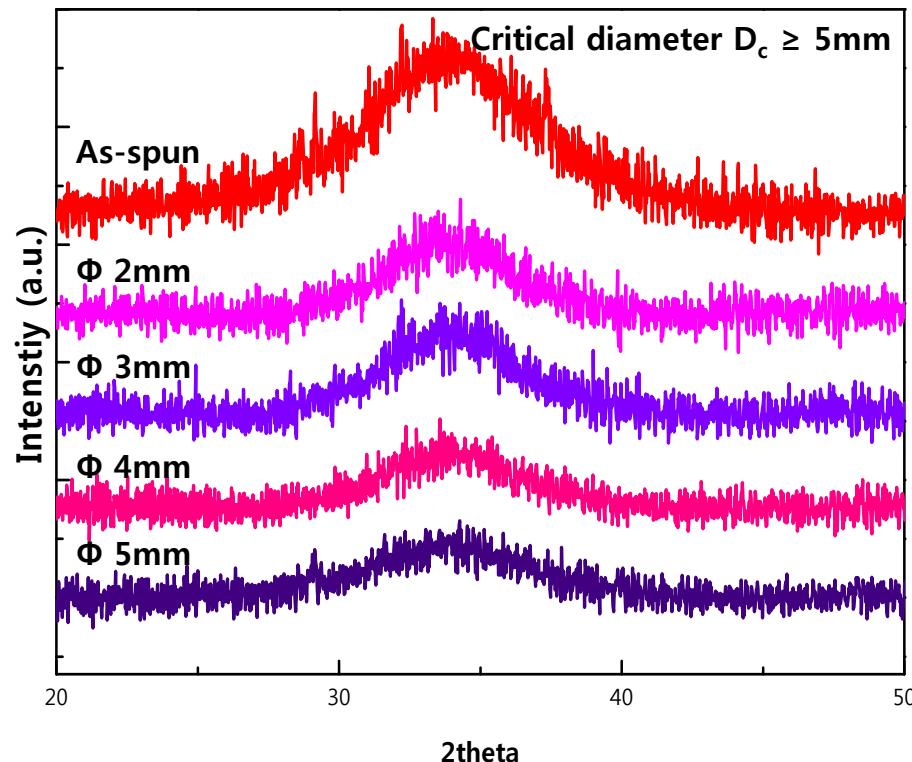
Er & Gd & Nd & Y : Select 3 elements + Al + Co & Ni : Select 1 element



- ▶ Above five HE compositions show clear supercooled liquid region before crystallization in DSC traces and broad diffuse peaks in XRD patterns of ribbons and bulk samples.
- Fully amorphous ribbons and bulk samples are fabricated in above five RE-based high entropy compositions.

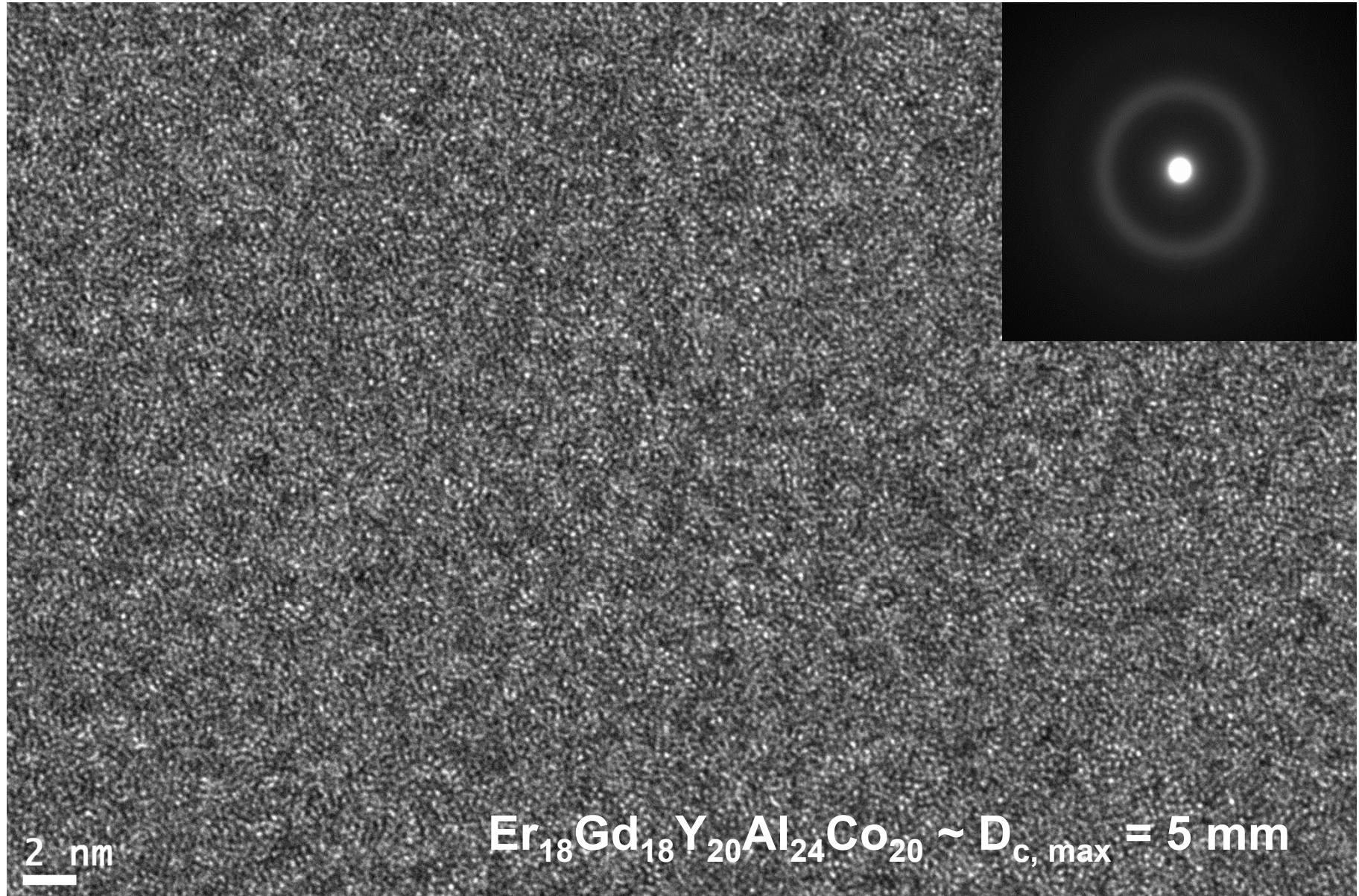
Development of novel RE⁽¹⁾₁₈RE⁽²⁾₁₈RE⁽³⁾₂₀Al₂₄Co₂₀ HE BMGs

$\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ high entropy bulk metallic glass: $D_{\max} \geq 5 \text{ mm}$



- ▶ Fully amorphous bulk samples up to 5 mm in diameter can be fabricated by suction casting method in $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ high entropy alloy compositions.

HR-TEM image for $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ 5mm bulk sample



Newly developed RE⁽¹⁾₁₈RE⁽²⁾₁₈RE⁽³⁾₂₀Al₂₄Co₂₀ HE BMGs

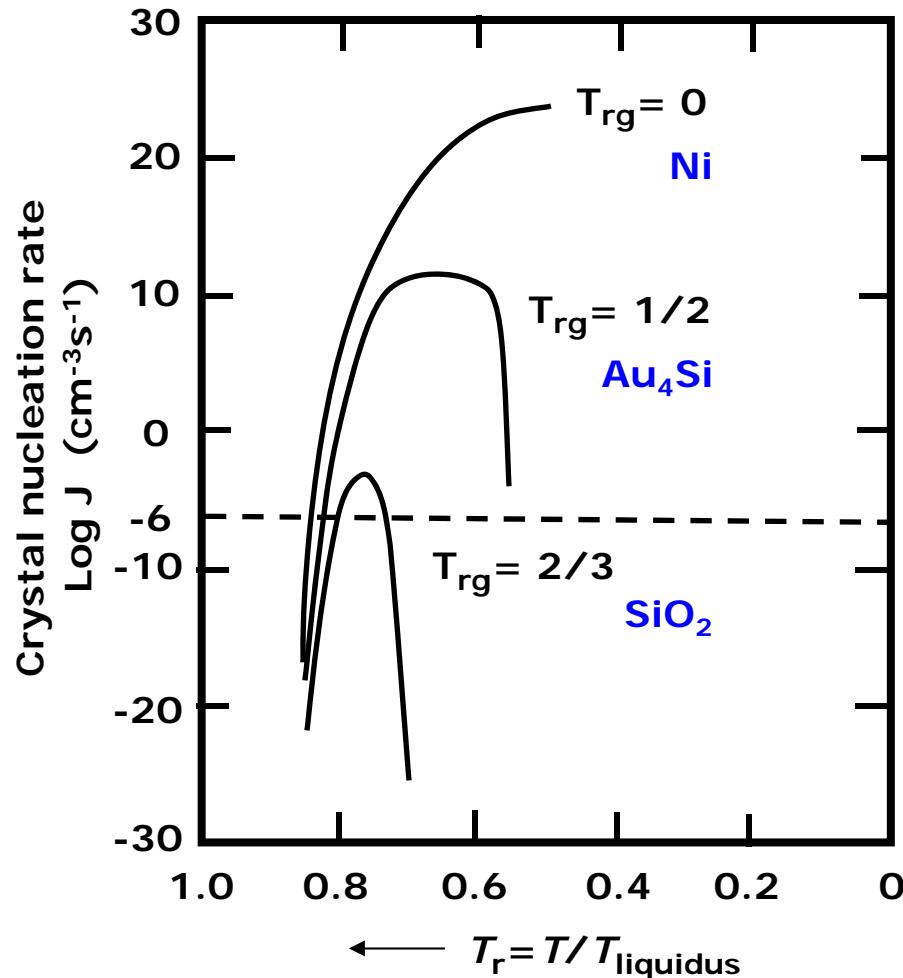
Composition	T _g (K)	T _x (K)	T _I (K)	T _{rg}	γ	D _{max} (mm)
Er ₁₈ Gd ₁₈ Y ₂₀ Al ₂₄ Ni ₂₀	633	656	1021	0.620	0.397	2
Er ₁₈ Gd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀	634	680	1031	0.615	0.408	5
Er ₁₈ Nd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀	613	671	988	0.621	0.419	2
Gd ₁₈ Nd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀	588	651	1004	0.586	0.409	1
Er ₁₈ Nd ₁₈ Gd ₂₀ Al ₂₄ Co ₂₀	595	624	962	0.619	0.401	1

- ▶ Five RE-based HE BMGs are newly developed in this study.
- ▶ GFA parameters (T_{rg} and γ) do not show a linear correlation with the GFA of developed RE-base HE BMGs.

Criterion for glass formation

T_{rg} parameter = T_g/T_m

: ability to avoid crystallization during cooling



$$T_{rgNi} < T_{rgAu4Si} < T_{rgSiO2}$$

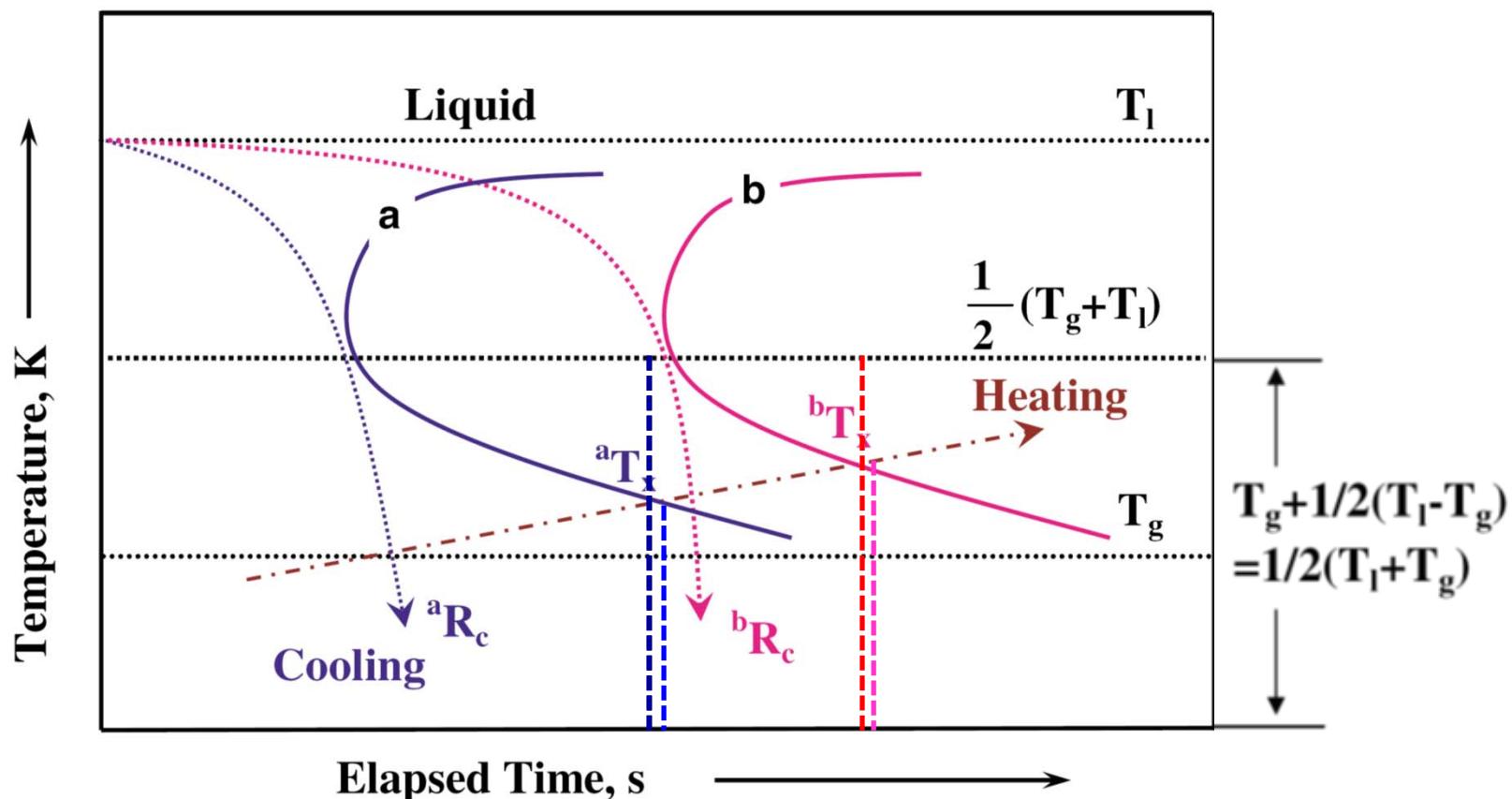
$$R_{Ni} > R_{Au4Si} > R_{SiO2}$$

Turnbull, 1959 ff.

Criterion for glass formation

$$\gamma \propto T_x \left[\frac{1}{2(T_g + T_l)} \right] \propto \frac{T_x}{T_g + T_l}$$

$$\gamma_a < \gamma_b \rightarrow R_a > R_b$$



Newly developed RE⁽¹⁾₁₈RE⁽²⁾₁₈RE⁽³⁾₂₀Al₂₄Co₂₀ HE BMGs

Composition	T_g (K)	T_x (K)	T_l (K)	≥ 0.6			≥ 0.35		
				T_{rg}	γ	D_{max} (mm)	T_{rg}	γ	D_{max} (mm)
Er ₁₈ Gd ₁₈ Y ₂₀ Al ₂₄ Ni ₂₀	633	656	1021	0.620	0.397	2	0.615	0.408	5
Er ₁₈ Gd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀	634	680	1031	0.621	0.419	2	0.586	0.409	1
Er ₁₈ Nd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀	613	671	988	0.619	0.401	1	0.619	0.401	1
Gd ₁₈ Nd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀	588	651	1004						
Er ₁₈ Nd ₁₈ Gd ₂₀ Al ₂₄ Co ₂₀	595	624	962						

- ▶ Five RE-based HE BMGs are newly developed in this study.
- ▶ GFA parameters (T_{rg} and γ) do not show a linear correlation with the GFA of developed RE-base HE BMGs.

Property evaluation of RE based HE-BMG

1. Mechanical property

- Nano-indentation
- Fragility

2. Phase transformation

- From **stable** liquid to solid ; Solidification
- From **meta-stable** liquid to solid ; Crystallization

I. Fragility and Mechanical properties : Characteristic of BMG

Fragility

- Fragility ~ extensively used to figure out liquid dynamics and glass properties corresponding to "frozen" liquid state

< Classification of glass >

Strong network glass : Arrhenius behavior



$$\eta = \eta_0 \exp\left[\frac{E_a}{RT}\right]$$

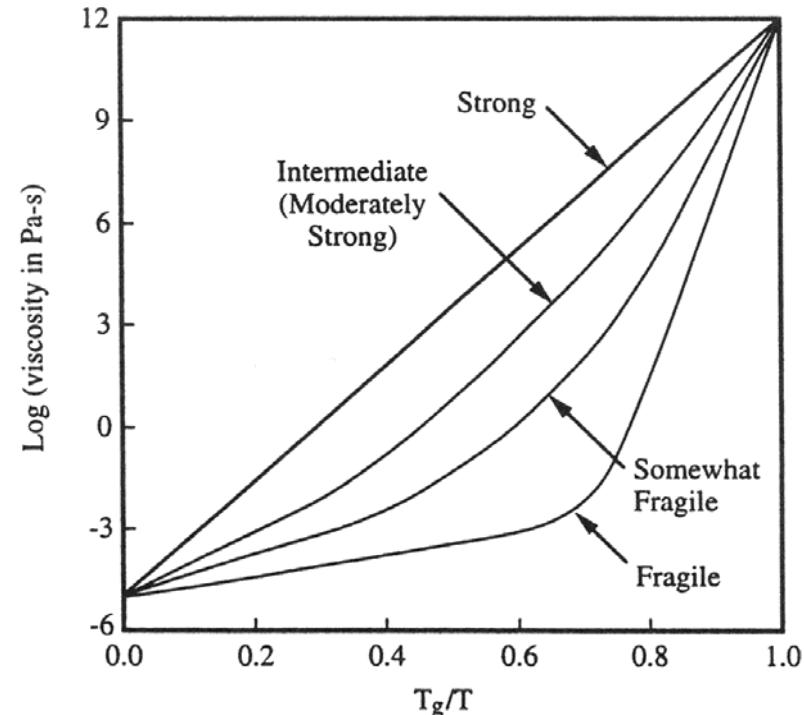
Fragile network glass : Vogel-Fulcher relation

$$\eta = \eta_0 \exp\left[\frac{B}{T - T_0}\right]$$

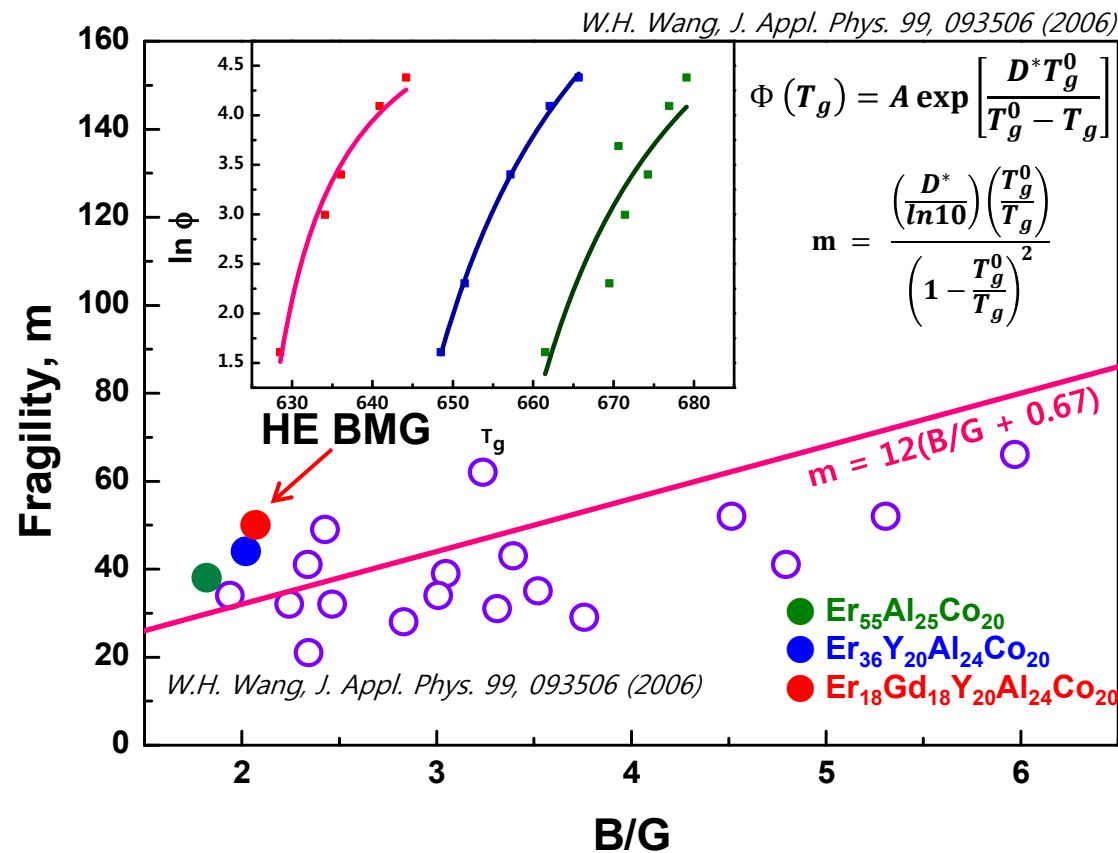
< Quantification of Fragility >

$$m = \left. \frac{d \log \eta(T)}{d(T_{g,n}/T)} \right|_{T=T_{g,n}} = \left. \frac{d \log \tau(T)}{d(T_g/T)} \right|_{T=T_g}$$

Slope of the logarithm of viscosity, η (or structural relaxation time, τ) at T_g

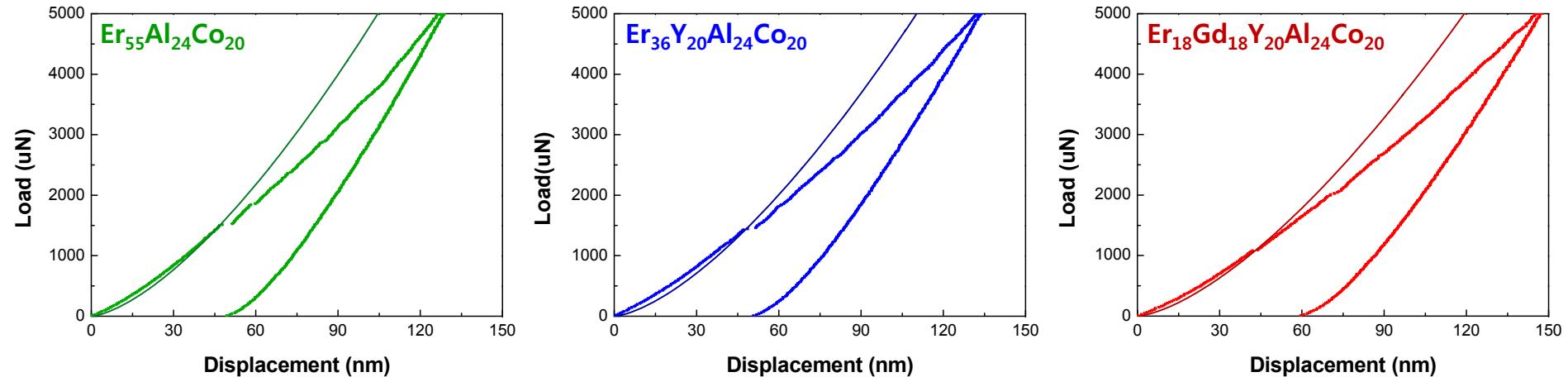


Correlation between fragility and elastic properties in BMGs



Mechanical properties for RE-based (HE) BMGs

Relatively low yield strength of HE BMG than RE BMGs: “Softening behavior”



Unloading/loading rate : 1mN/sec, Maximum Load : 5mN, Conical tip ($r=2\mu\text{m}$), as-spun samples

Strain rate : 10^{-4} /sec, 1mm rod samples

	Nano-indentation			Compression test	
	Reduced elastic modulus [GPa]	Nano-hardness [GPa]	Yield strength [GPa]	Elastic modulus [GPa]	Yield strength [GPa]
Er ₅₅ Al ₂₅ Co ₂₀ [Er55]	73.31±6.6	5.50±0.5	2.18±0.2	79.79	1.60
Er ₃₆ Y ₂₀ Al ₂₄ Co ₂₀ [Er36]	73.0±9.0	5.37±0.3	2.17±0.2	72.94	1.46
Er ₁₈ Gd ₁₈ Y ₂₀ Al ₂₄ Co ₂₀ [Er18]	65.18±4.2	4.81±0.2	2.0±0.2	65.71	1.24

‘ Fragility → Characteristic of BMG ’

II. Crystallization behavior: Characteristic of HEA

- 1) Stable liquid to crystalline phase during cooling: Solidification**
- 2) Metastable liquid to crystalline phase during heating: Crystallization**

Property evaluation of $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE BMG

1. Mechanical property

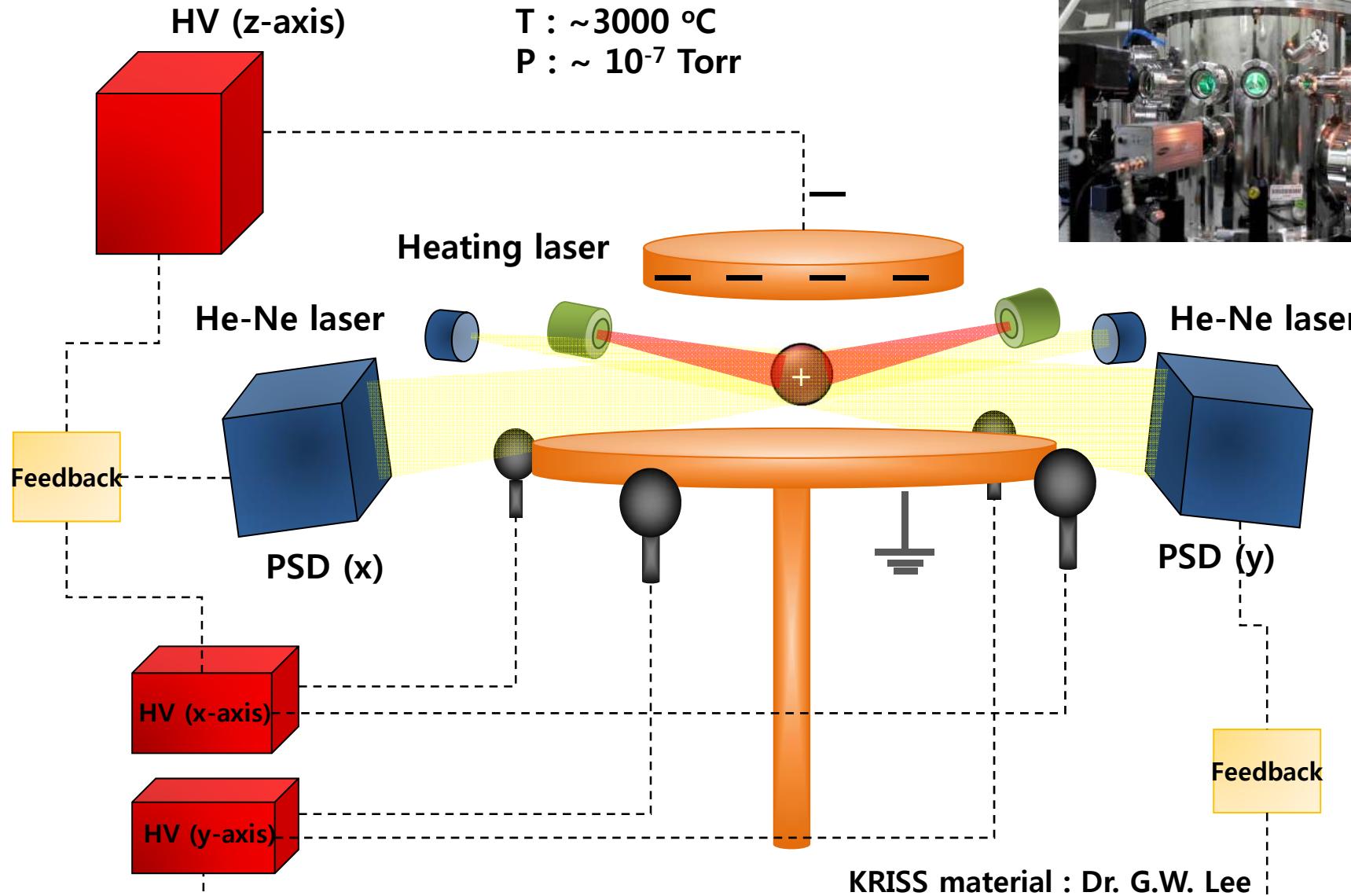
- Nanoindentation / compression test
- Fragility

2. Thermal property (phase transformation)

- Cooling : stable liq. → solid ; Electrostatic levitation
- Heating : supercooled liq. → solid ; DSC & TEM

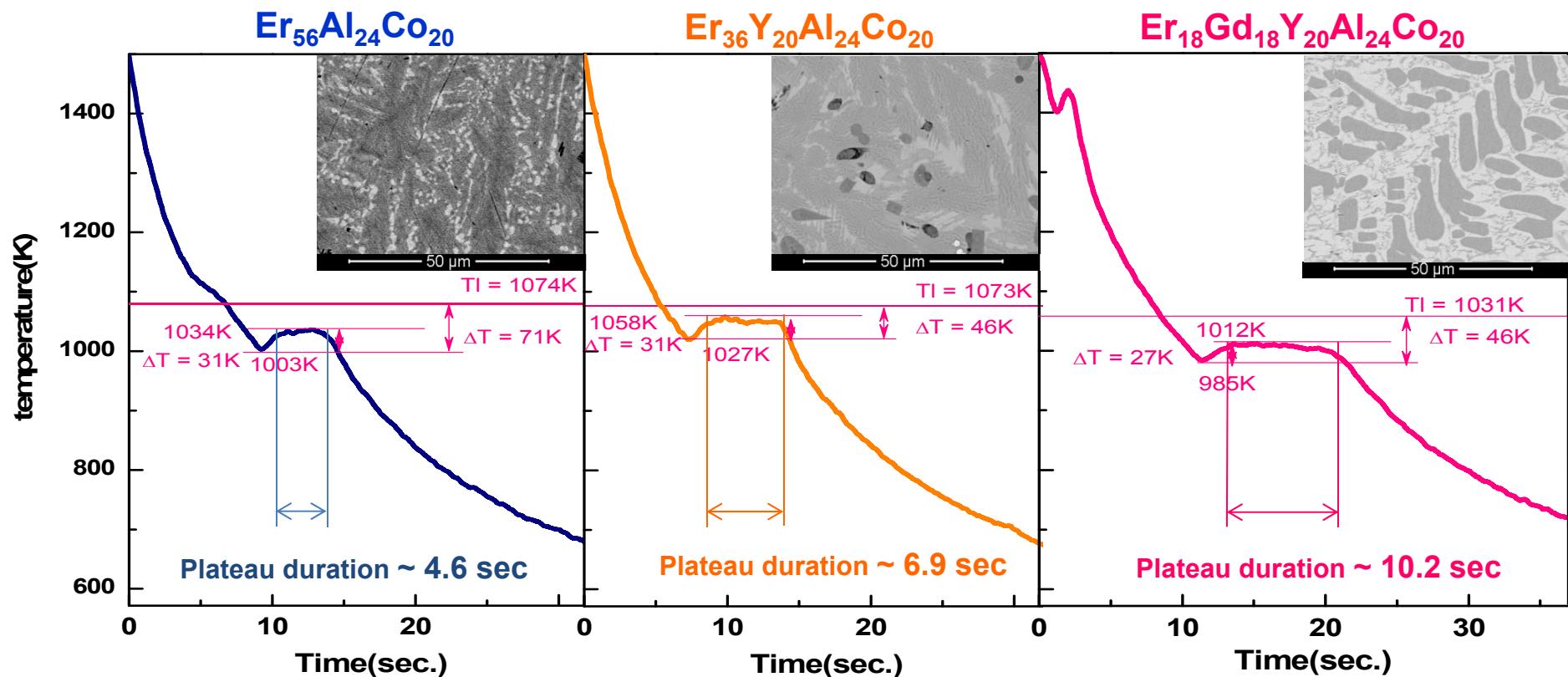
Electrostatic levitation in KRISS

Containerless equipment: close to homogeneous nucleation



Cooling curve for Er(-Gd-Y)-Al-Co BMG & HE BMG system

1) Phase transformation from stable liquid to solid : cooling



HE BMG: Relatively longer time to growth

' Sluggish diffusion behavior in stable liquid → characteristic of HEA '

Property evaluation of $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE BMG

1. Mechanical property

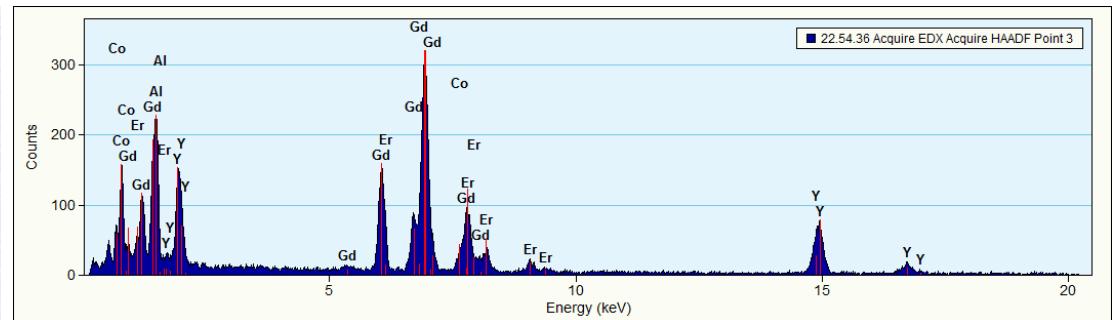
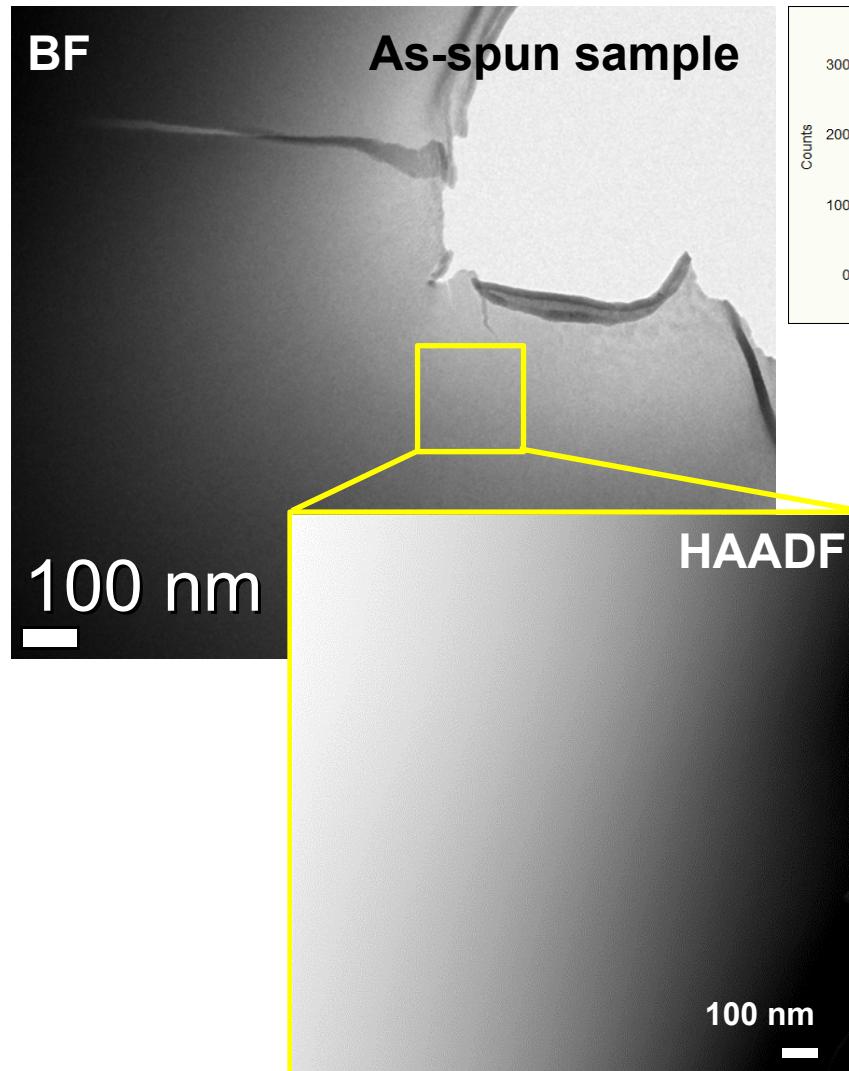
- Nanoindentation / compression test
- Fragility

2. Thermal property (phase transformation)

- Cooling : stable liq. → solid ; Electrostatic levitation
- Heating : supercooled liq. → solid ; DSC & TEM

TEM image of $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE BMG

2) Phase transformation from metastable liquid to solid : heating

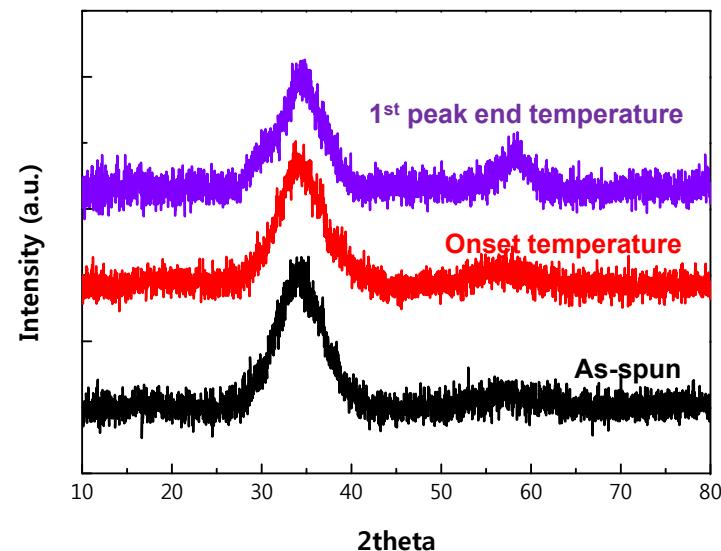
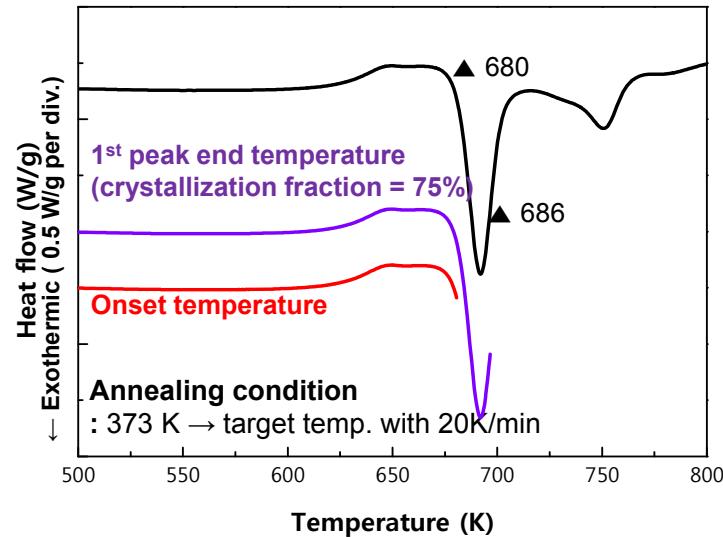


Element	Atomic %
Al (K)	21.58
Co (K)	18.95
Y (K)	17.2
Gd (L)	18.36
Er (L)	23.88
RE	59.44

No composition contrast
→ Fully amorphous state

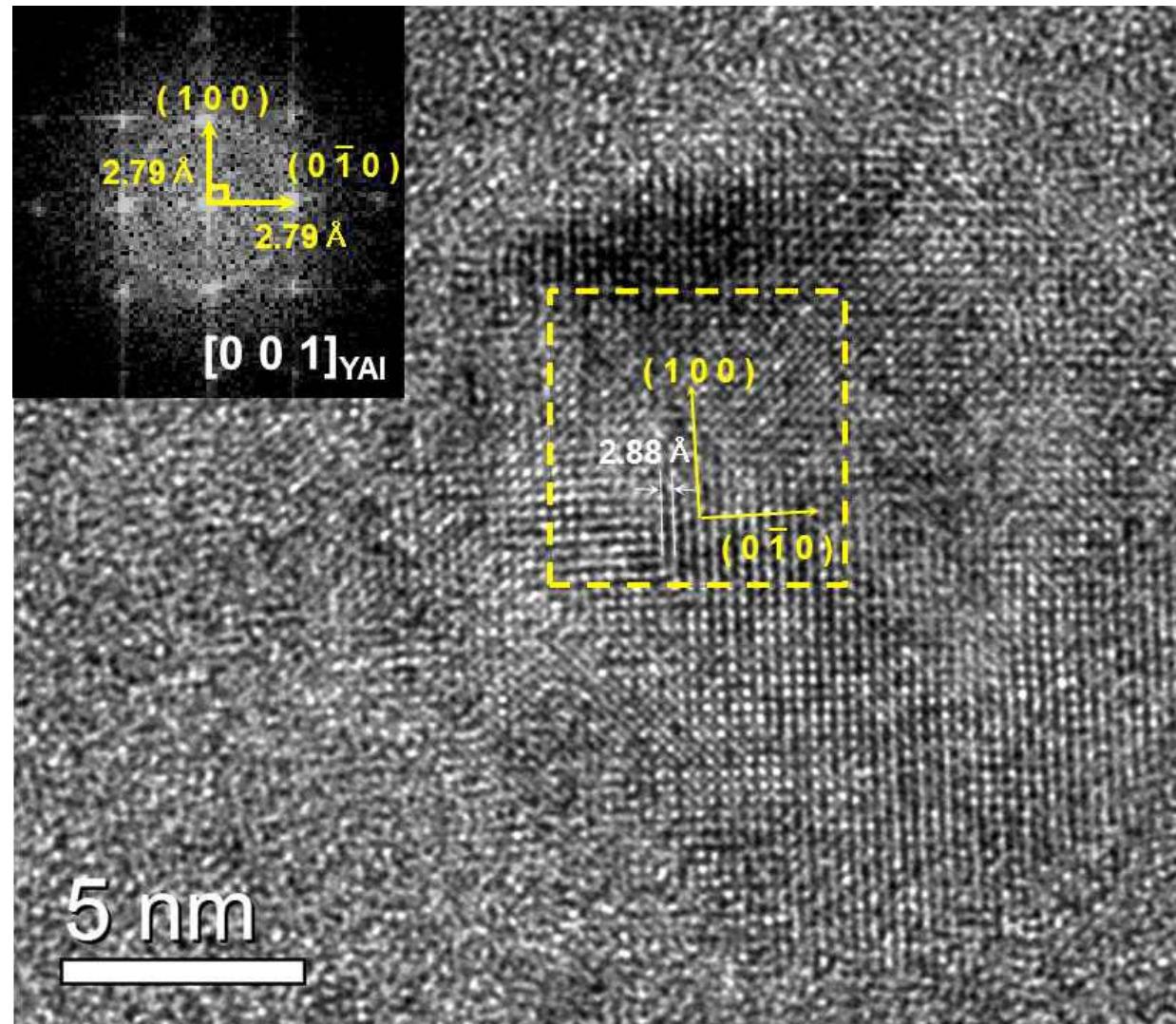
Crystallization behavior of $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE-BMG

2) Phase transformation from metastable liquid to solid : heating



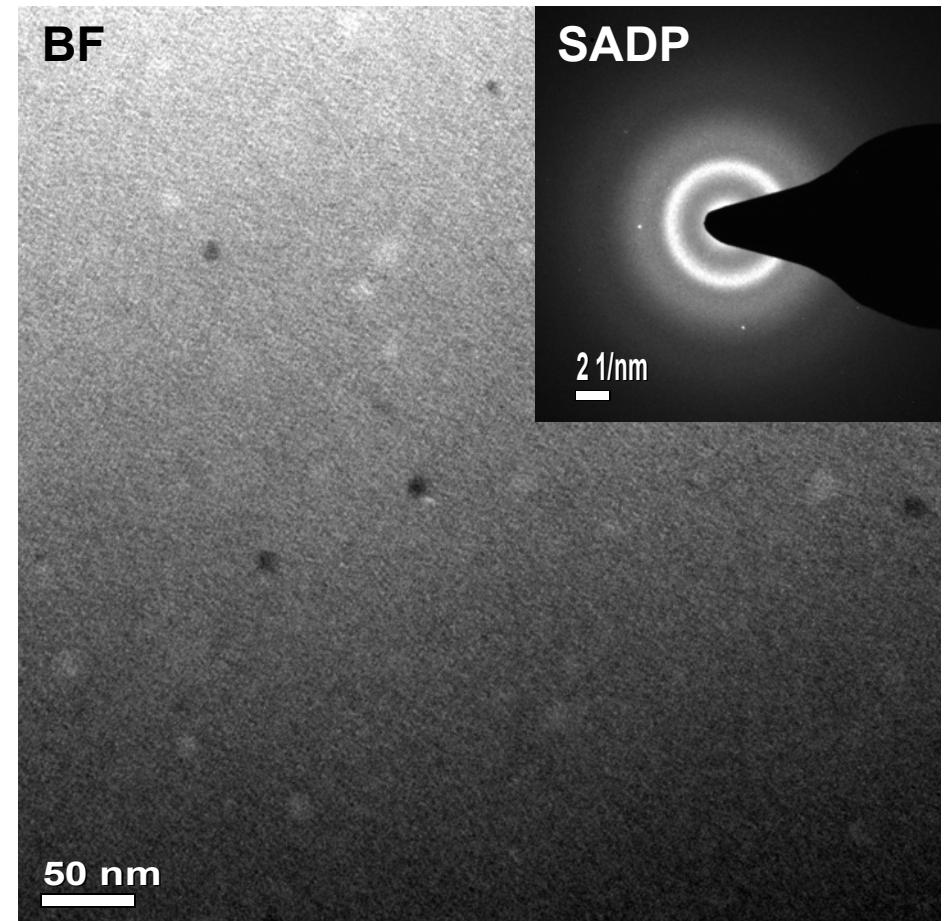
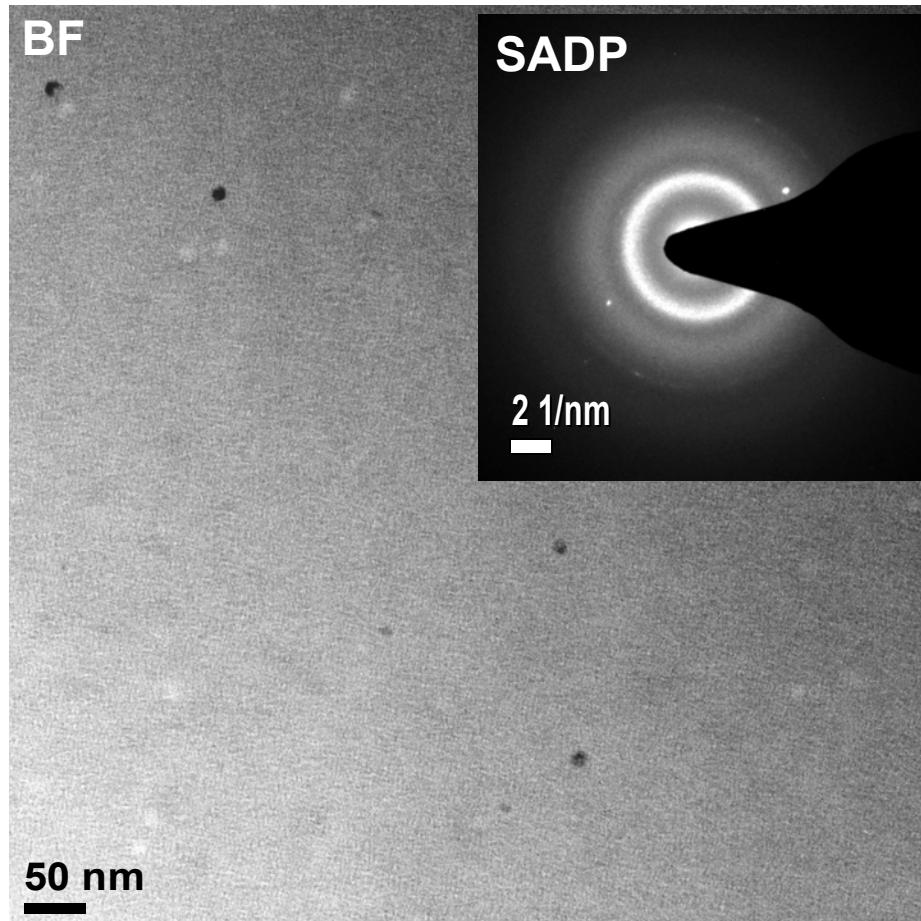
AlY precipitation of HE BMG annealed at 1st peak mid temp. (686 K)

2) Phase transformation from metastable liquid to solid : heating



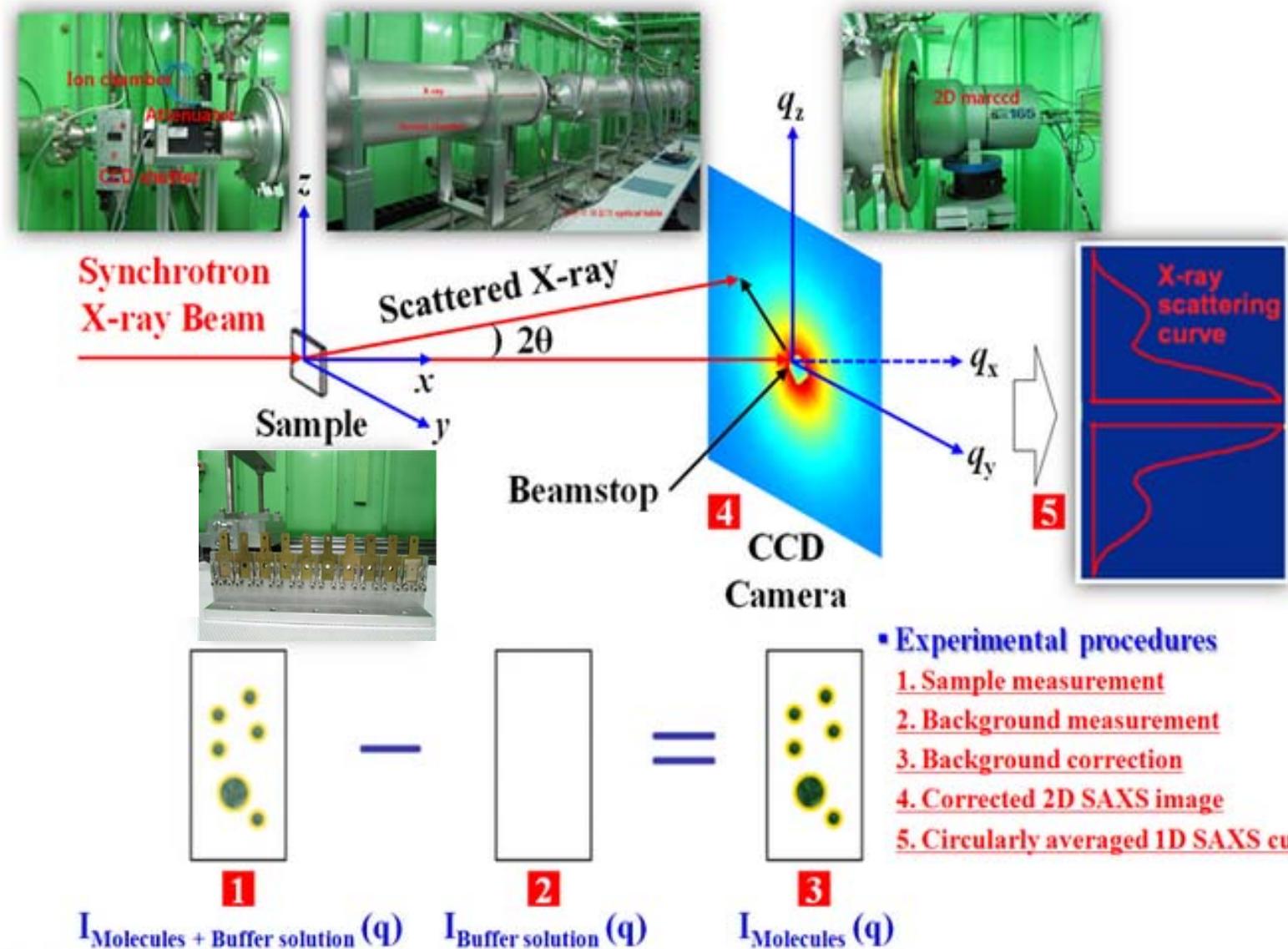
$\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE BMG annealed to onset temp. of T_x (680K)

2) Phase transformation from metastable liquid to solid : heating



‘ Sluggish diffusion behavior in SCL \rightarrow Slow growth rate ’

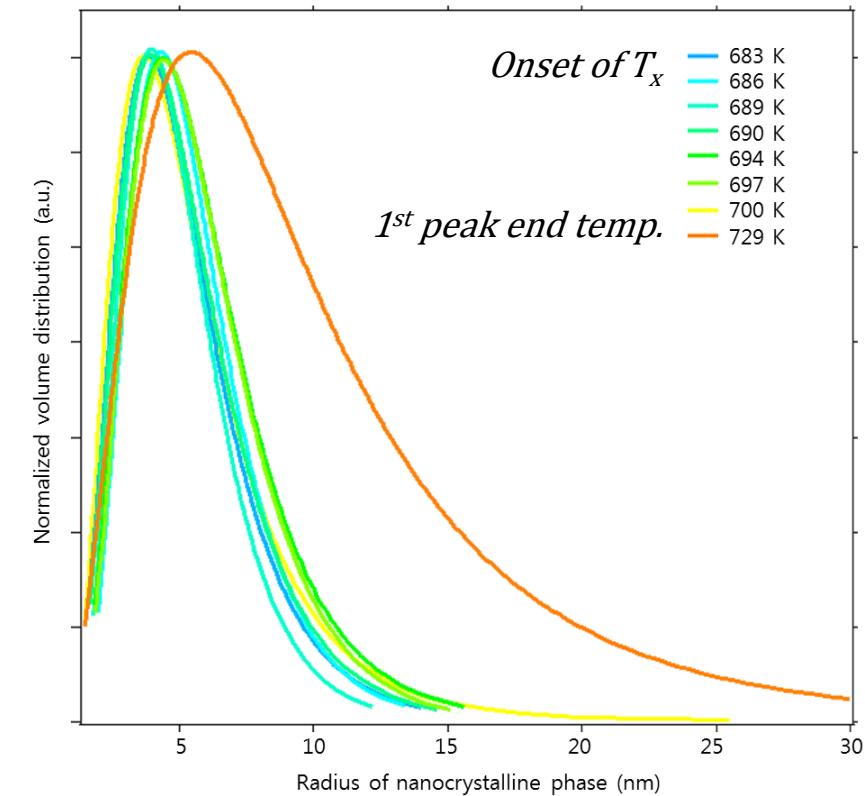
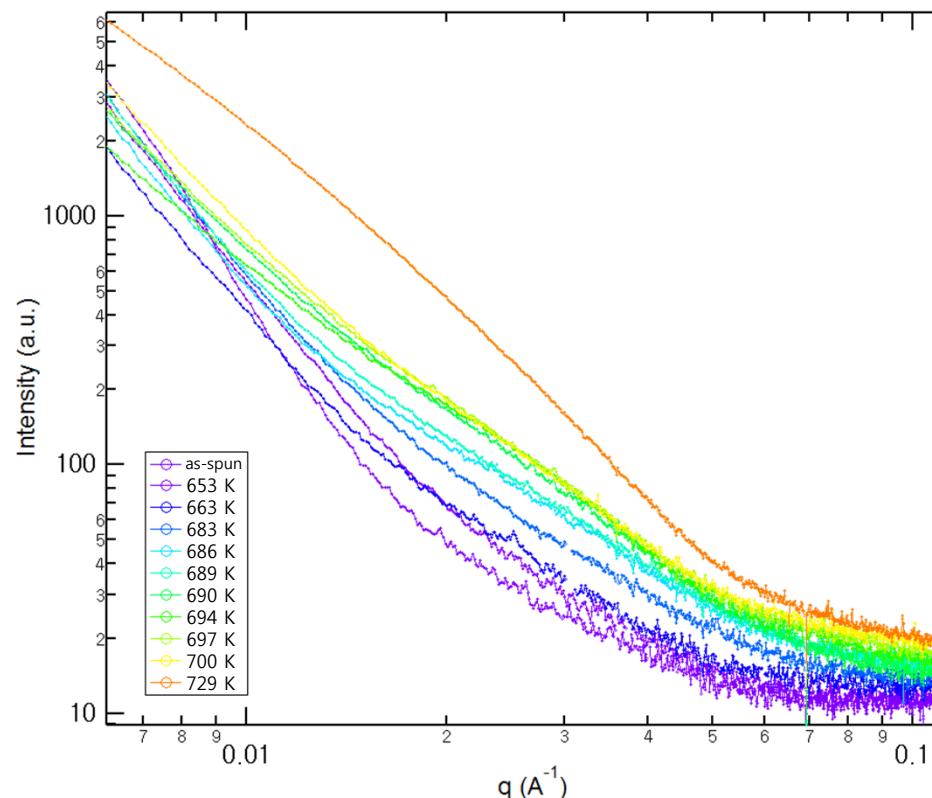
SAXS experiment in PALS



SAXS patterns and Size distribution of crystalline phases

- $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ ribbon samples annealed at different T

Nanoscale precipitation starts at ~ 680 K Average size of precipitate: $10\text{ nm} \rightarrow 15\text{ nm}$



' Sluggish diffusion behavior in SCL \rightarrow Slow growth rate '

Summary

1. Novel high entropy BMGs can be developed in $\text{RE}^{(1)}_{18}\text{RE}^{(2)}_{18}\text{RE}^{(3)}_{20}\text{Al}_{24}\text{Co}_{20}$ alloys.

In particular, $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE BMG can be fabricated over 5 mm in dia.

Composition	T _g (K)	T _x (K)	T _l (K)	T _{rg}	γ	D _{max} (mm)
$\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Ni}_{20}$	633	656	1021	0.620	0.397	2
$\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$	634	680	1031	0.615	0.408	5
$\text{Er}_{18}\text{Nd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$	613	671	988	0.621	0.419	2
$\text{Gd}_{18}\text{Nd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$	588	651	1004	0.586	0.409	1
$\text{Er}_{18}\text{Nd}_{18}\text{Gd}_{20}\text{Al}_{24}\text{Co}_{20}$	595	624	962	0.619	0.401	1

2. Characteristic of bulk metallic glass : “more fragile” and “Softening behavior”

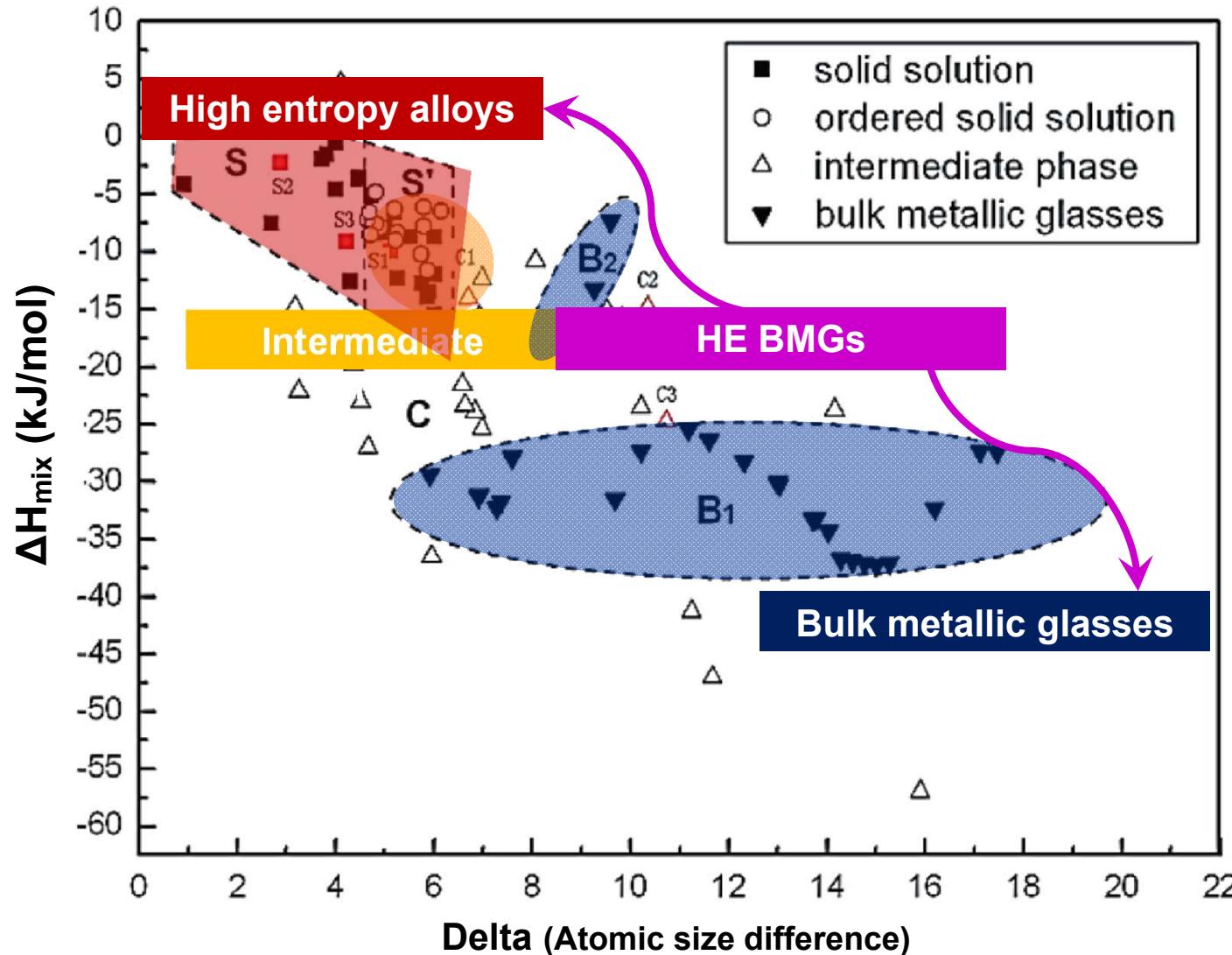
- $\text{Er}_{18}\text{Gd}_{18}\text{Y}_{20}\text{Al}_{24}\text{Co}_{20}$ HE BMG is more fragile glass former and, as a consequence, has relatively low elastic modulus, hardness and yield strength.

3. Characteristic of high entropy alloy : “ Sluggish diffusion behavior ”

- Phase transformation from stable liquid to solid during cooling
: longer plateau → relatively long time to growth during solidification
- Phase transformation from metastable liquid to solid during heating
: average size of precipitates during 1st crystallization ~ 10 nm – relatively slow growth rate

Summary

' HE BMGs ' → Bridging between HEAs and BMGs



Thanks for your kind attention

④ 4C Current Specifications

Beamline	4C SAXS II
Storage ring energy	3.0 GeV
Source	In-vacuum Undulator 20 (IVU 20)
Monochromator	DCM Si (111) Crystal
Standard energy (\AA / keV)	1.24 / 10.00
Energy resolution ($\Delta E/E$)	$\sim 2 \times 10^{-4}$
Energy range (keV)	8 ~ 20
Flux at CCD (ph/s/0.1%BW)	2×10^{13}
Beam size at CCD (μm)	23 (V) \times 300(H)
Mirror	Vertical focusing toroidal, rhodium coated
Sample-to-detector distance (m)	0.2 ~ 5.0
Observable minimum angle($^\circ$)	0.00128 $^\circ$ (400 nm @ 10 keV)
Observable resolution (nm)	0.3
Detector	Rayonix 2D SX165
Experimental methods	WAXS & SAXS (Bulk, Solution, Powder, Sol-gel, etc)