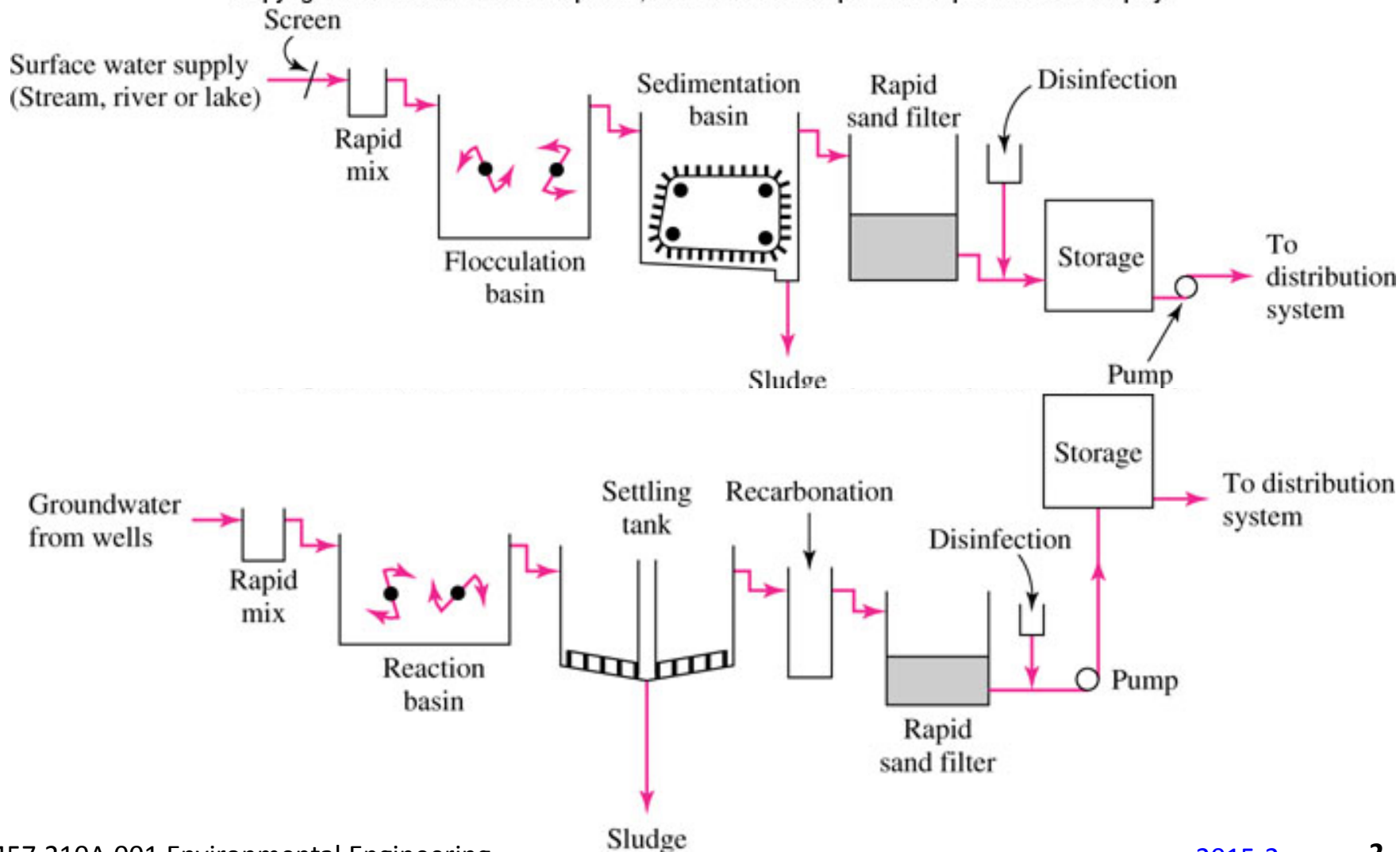


# Final review

# Water treatment processes

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# Water treatment - hardness

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**Q:** A water sample having a pH of 7.4 is analyzed to have the following ion concentrations. Determine the alkalinity, total hardness, carbonate hardness, and noncarbonate hardness of the sample in mg/L as CaCO<sub>3</sub>.

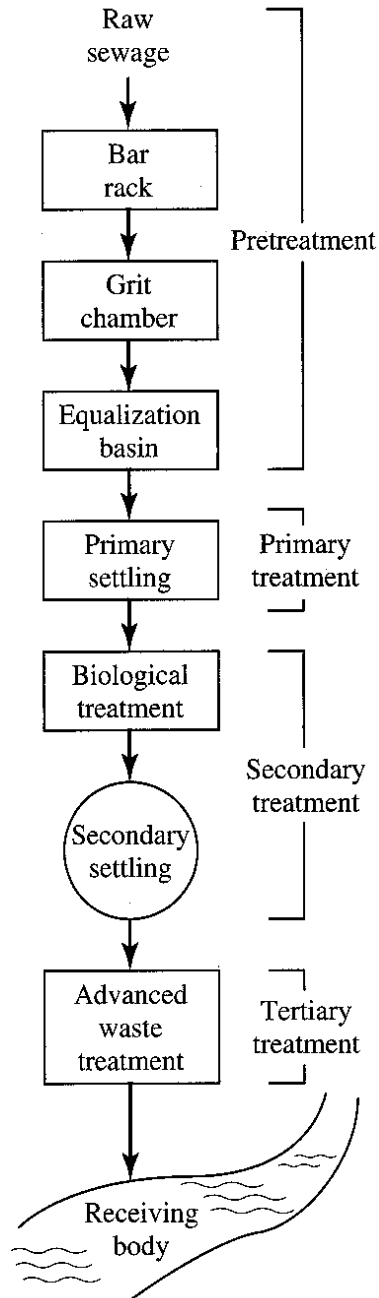
Ion	Concentration (mg/L)	Ion	Concentration (mg/L)
Ca <sup>2+</sup>	53.2	HCO <sub>3</sub> <sup>-</sup>	132
Mg <sup>2+</sup>	13.1	SO <sub>4</sub> <sup>2-</sup>	58.4
Na <sup>+</sup>	17.5	Cl <sup>-</sup>	21.2
K <sup>+</sup>	4.8		
Fe <sup>3+</sup>	0.32		

# Water treatment - disinfection

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**Q:** A chlorine disinfection is applied for a reactor having a dimension of 5 m x 2 m x 25 m (W x H x L) and receiving an influent flow rate of 1000 m<sup>3</sup>/hr. The first-order decay coefficient under the current chlorine dose is 0.31 min<sup>-1</sup>. Assuming that the chlorine concentration is constant in the reactor and the reactor works as an ideal PFR, would it be possible to achieve 2-log removal of pathogens? What if the reactor cannot be assumed as an ideal PFR?

# Wastewater treatment processes



- Pretreatment: removal of materials that may damage mechanical devices & flow equalization
- Primary treatment: sedimentation basin, remove SS & particulate BOD by gravity
- Secondary treatment: remove BOD by microorganisms, further SS removal
- Tertiary treatment: polishing of secondary effluent for improved quality

# Wastewater treatment - analysis

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**Q:** An aeration tank with a volume of 4000 m<sup>3</sup> is receiving primary effluent at a flow rate of 10000 m<sup>3</sup>/d having a BOD<sub>5</sub> of 500 mg/L. Calculate the effluent BOD<sub>5</sub> and the biomass concentration in the tank when there is no sludge recycle. Does the effluent BOD<sub>5</sub> meet the secondary effluent standard of BOD<sub>5</sub> = 30 mg/L? If not, determine the solids retention time required to achieve the standard. What is the biomass concentration in the tank according to the calculated solids retention time? Use following parameters.

$$K_s = 50 \text{ mg/L BOD}_5$$

$$Y = 0.5 \text{ mg VSS/mgBOD}_5$$

$$k_d = 0.10 \text{ d}^{-1}$$

$$\mu_m = 3 \text{ d}^{-1}$$

# Hazardous waste management

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- US regulation
  - RCRA (Resource Conservation and Recovery Act)
  - CERCLA (Comprehensive Environmental Response, Compensation, and Liability Act)
- Soil & groundwater treatment technologies
  - In situ vs. ex situ
  - Pump-and-treat, soil vapor extraction, air sparging, permeable reactive barrier, soil washing, thermal desorption, in situ bioremediation

# Air pollution

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- Micro-, meso-, and macro-scale
- Primary and secondary pollutants
- Air pollutants: CO, NO<sub>x</sub>, SO<sub>x</sub>, lead, photochemical oxidants, particulates, other hazardous pollutants (hydrocarbons, heavy metals, dioxins, asbestos, ...)
- Indoor air pollution, acid rain, ozone depletion, global warming
- Montreal protocol vs. Kyoto protocol



# Air pollution control

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- Absorption vs. adsorption
- Combustion
- Cyclones
- Filter
- Liquid scrubbing
- Electrostatic precipitation

# Solid waste management

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- Classification of wastes (Korea)
  - By sources
  - By hazard
- Recycle
- Composting
- Incineration
- Sanitary landfill
  - Landfill operation – area method
  - Leachate and landfill gas control

# Noise pollution

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**Q:** A sound is measured over 1-minute period and the average dB(A) values are taken at 10 sec intervals as shown below. Calculate the  $L_{eq}$  value of the sound over the period. Also, calculate the maximum value of N for  $L_N = 55$  dB(A) assuming linear change in dB(A) between the time intervals.

Time intervals	Average dB(A) at time intervals
0-10 sec	50
10-20 sec	55
20-30 sec	60
30-40 sec	55
40-50 sec	50
50-60 sec	50