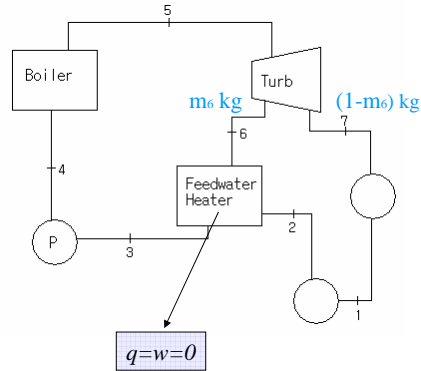
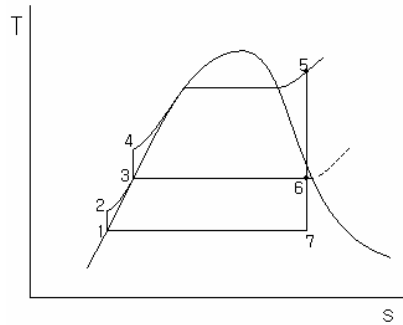


## From last time...

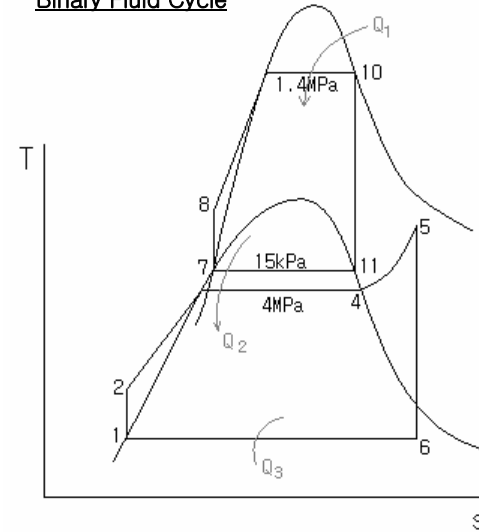
- Regeneration Cycle  
uses feed-water heaters  
Efficiency is identical to that of Carnot cycle!



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## Binary Fluid Cycle



- 고온부에 적합한 유체(mercury)로 고온 사이클 형성.
- 저온부에 적합한 유체로 다른 사이클 형성.

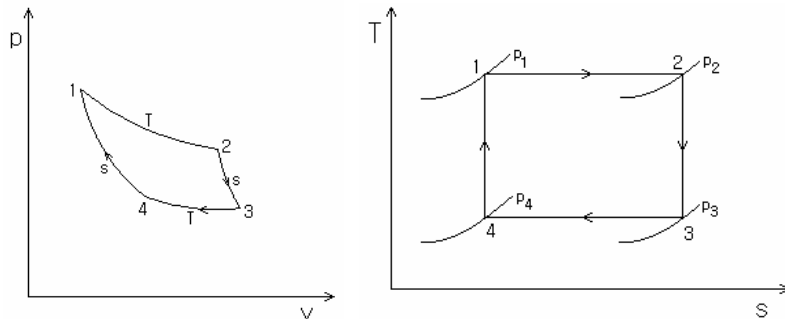
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## Chapter 9 Air Standard Cycle

Steam cycle: 상변화 (L-G)  
Air cycle: 단일상(G)

- Air-standard (공기표준) Carnot Cycle  
(ie Carnot cycle all at the superheated vapor zone)



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Recall

$$\eta_{th} = 1 - \frac{T_L}{T_H} = 1 - \frac{T_3}{T_2} = 1 - \frac{T_4}{T_1}$$

Using isentropic relation for ideal gas

$$\frac{T_3}{T_2} = \left( \frac{p_2}{p_3} \right)^{\frac{1-\gamma}{\gamma}} = \left( \frac{p_1}{p_4} \right)^{\frac{1-\gamma}{\gamma}}$$

$$\frac{T_3}{T_2} = \left( \frac{V_3}{V_2} \right)^{1-\gamma} = \left( \frac{V_4}{V_1} \right)^{1-\gamma}$$

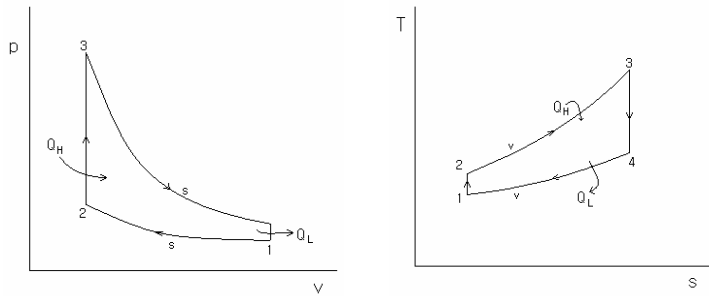
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•Air-standard Otto Cycle

Heat is transferred at constant -  $v$ .

- ideal spark-ignition internal combustion engine.



- 1-2: Isentropic compression of air
- 2-3: Piston motion stops – heat is added. (e.g. ignition of the fuel-air mixture)
- 3-4: Isentropic expansion
- 4-1: Rejection of heat from the air while piston motion is stopped.

$$\eta = 1 - \frac{Q_L}{Q_H} = 1 - \frac{mc_v(T_4 - T_1)}{mc_v(T_3 - T_2)} = 1 - \frac{T_1 \left( \frac{T_4}{T_1} - 1 \right)}{T_2 \left( \frac{T_3}{T_2} - 1 \right)}$$

Using isentropic process relations

$$\frac{T_2}{T_1} \left( \frac{v_1}{v_2} \right)^{\gamma-1} = \frac{T_3}{T_4} \left( \frac{v_4}{v_3} \right)^{\gamma-1}$$

$$v_1 = v_4 \quad v_2 = v_3$$

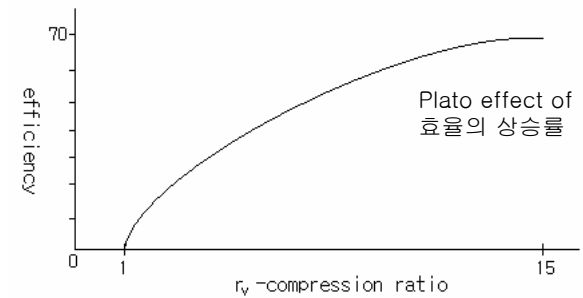
$$\therefore \frac{T_2}{T_1} = \frac{T_3}{T_4} \quad \text{or} \quad \frac{T_3}{T_2} = \frac{T_4}{T_1}$$

$$\eta_{th} = 1 - \frac{T_1}{T_2} = 1 - \frac{1}{\left( \frac{v_1}{v_2} \right)^{\gamma-1}}$$

$$= 1 - r_v^{1-\gamma}$$

$$= 1 - \frac{1}{r_v^{\gamma-1}}$$

where  $r_v =$  compression ratio  $= \frac{v_1}{v_2} = \frac{v_4}{v_3}$   
 $\eta \uparrow$  with  $\uparrow$  compression ratio:



One can only  $\uparrow r_v$  to a certain value to avoid explosion (엔진폭발).

- 연료의 연소특성 및 제작문제로 제한이 있다.
- spark knock- how to avoid?
- Anti-knock gasoline → 무연, add lead

• Diesel Cycle (as compared to Otto Cycle)

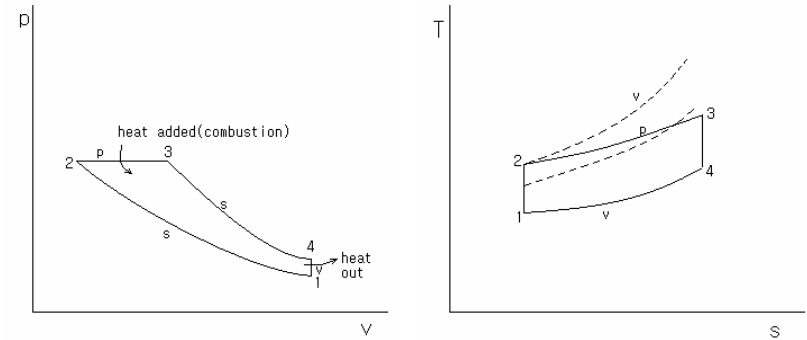
Heat is added at constant  $p$  / rejects at constant  $v$

- This is also called the compression ignition engine.
- Injection and burning of the fuel in the actual engine.

$$\eta = 1 - \frac{Q_L}{Q_H} = 1 - \frac{c_v(T_4 - T_1)}{c_p(T_3 - T_2)} = 1 - \frac{T_1 \left( \frac{T_4}{T_1} - 1 \right)}{RT_2 \left( \frac{T_3}{T_2} - 1 \right)}$$

Note:

$$\begin{aligned} \delta Q &= dE + pdV \\ &= c_v dT + pdV \rightarrow \text{use this if } V - \text{const.} \\ \delta Q &= dH - Vdp \\ &= c_p dT - Vdp \rightarrow \text{use if } p - \text{const.} \end{aligned}$$



Note isentropic compression (1→2) is greater than the expansion (3→4).

• Comparison of two cycles

$$\eta_{otto} > \eta_{Diesel}$$

With some rearranging (Prove It!)

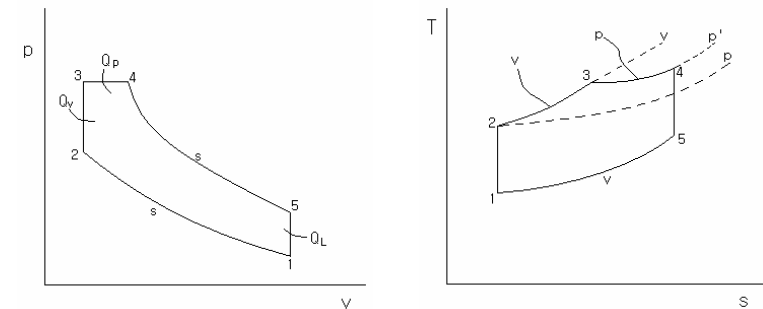
$$\eta = 1 - \underbrace{\frac{1}{r_v^{\gamma-1}}}_{Otto} \left\{ \frac{r_c^{\gamma-1}}{\gamma(r_c - 1)} \right\}$$

Note:

불꽃점화 기관 (Otto cycle) 에서는  
압축점화 기관 (Diesel) 에서는

공기+연료를 압축.  
공기만을 압축한 후 연료를 분사.

• Mixed or Dual Cycle



$$\begin{aligned} Q_H &= Q_v + Q_p \\ &= mc_v(T_3 - T_2) + mc_p(T_4 - T_3) \end{aligned}$$

$$Q_L = mc_v(T_5 - T_1)$$

$$\eta = 1 - \frac{c_v(T_5 - T_1)}{c_v(T_3 - T_2) + c_p(T_4 - T_3)}$$

$$r_v(\text{압축비}) = \frac{v_1}{v_2} \quad r_c(\text{차단비}) = \frac{v_4}{v_3} \quad r_p(\text{압력비}) = \frac{p_3}{p_2}$$

$$1-2: \quad T_2 = \left(\frac{v_1}{v_2}\right)^{k-1} T_1 = r_v^{k-1} T_1$$

$$2-3: \quad T_3 = \left(\frac{p_3}{p_2}\right) T_2 = (r_p)(r_v^{k-1}) T_1$$

$$3-4: \quad T_4 = \left(\frac{v_4}{v_3}\right) T_3 = r_c (r_p v_2^{k-1}) T_1$$

$$\begin{aligned} 4-5: \quad T_5 &= \left(\frac{v_4}{v_5}\right)^{k-1} T_4 = \left(\frac{v_4}{v_3} \frac{v_3}{v_5}\right)^{k-1} T_4 \\ &= \left(\frac{r_c}{r_v}\right)^{k-1} r_c r_p r_v^{k-1} T_1 \\ &= r_c^k r_p T_1 \end{aligned}$$

$$\eta_{th} = 1 - \frac{1}{r_v^{k-1}} \frac{r_p r_c^k - 1}{(r_p - 1) + k r_p (r_c - 1)}$$

열효율

Otto Cycle > Dual Cycle > Diesel Cycle