Chapter 3

Molecular Components of Cells



Molecular Components of Cells

- Chemical composition
 - C, H, N, O and small amount of other elements
- Molecular building blocks
 - Lipids
 - Carbohydrates
 - Proteins
 - Nucleic acid
 - DNA (deoxyribonucleic acid)
 - RNA (ribonucleic acid)

Atoms, Ions, and Molecules

Atoms

- Smallest piece of an recognizable element
- Nucleus (proton + neutron) + electron
- Same number of protons and electrons : no charge
- Biologically important atoms: C, H, O, N, S, P, Na, K, Ca, Cl

lons

- Electrically charged atoms
- Biological importance: electrical impulse, ion balance
 - Ca²⁺, Na⁺, K⁺, Cl⁻

Molecules

Generated from chemical bonding of atoms

Subunits of Biological Molecules

Class of Molecules	Examples	Repeating Unit
Lipid	Fats, oils	Glycerol, fatty acid
Carbohydrate	Sugars, starch, cellulose	Simple sugars
Nucleic acid	DNA, RNA	Nucleotide
Proteins	Enzymes	Amino acids

- Hydrophobic fats, oils, and cholesterol etc.
- High energy C-H, C-C bonds → good energy storage
- Fats : glycerol + fatty acids

Triglyceride

Fatty acid

- Saturated: tight packing → solid at room temperature
- Unsaturated: more than one cis-double bond → liquid

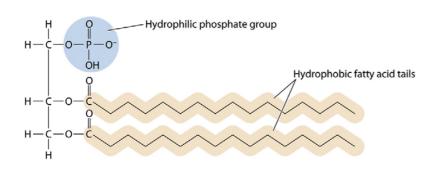
A. A saturated fatty acid

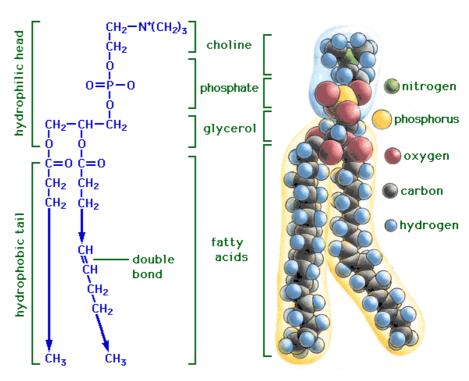
B. A monounsaturated fatty acid

C. A polyunsaturated fatty acid

Phospholipid

- Major component of cellular membrane
- Glycerol backbone
- two fatty acids (hydrophobic)+ phosphate (hydrophilic)
- Additional small polar molecules bound to phosphate group





Sterols

- Cyclic hydrocarbon compounds
- Cholesterol
 - Component of animal cell membranes
 - decrease membrane fluidity
 - Starting material for steroid hormones and bile synthesis

C. Cortisol
$$H-C-OH$$
 $C=O$
 CH_3
 OH

CH₃

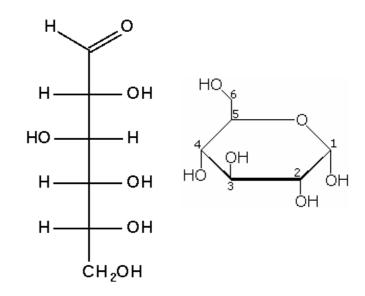
CH₃

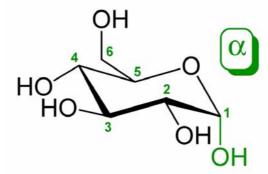
CH₃

 CH_3

Carbohydrates

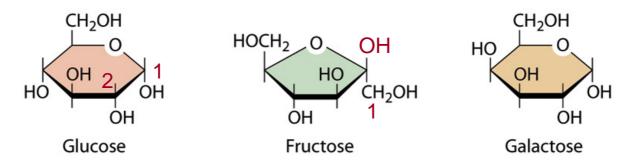
- C:H:O = 1:2:1
- Simple sugars (monosaccharide)
 - Glucose, fructose, galactose
- Disaccharide
 - Simple sugars linked by glycosidic bond
 - Sucrose (glucose-α-1,2-fructose)
 - Lactose (galactose-β-1,4-glucose)
 - Maltose (glucose-α-1,4-glucose)



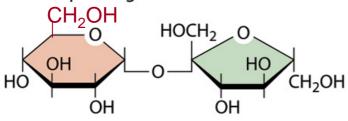


Carbohydrates

A. Simple sugars

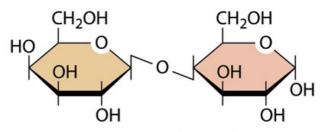


B. Complex sugars



Sucrose (table sugar)
Glucose + fructose

Glc- α -1,2-Fru



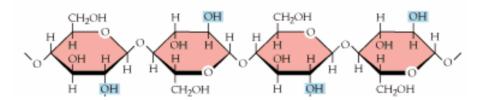
Lactose (milk sugar) Galactose + glucose

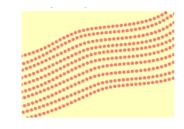
Gal-β-1,4-Glc

Carbohydrates

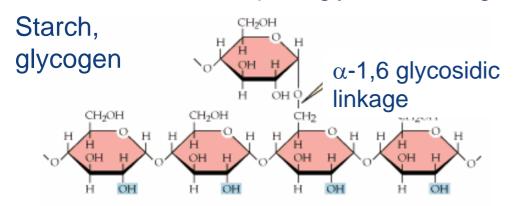
- Polysaccharide
 - Starch, Glycogen, cellulose from glucose
 - Agar, carrageenan (thickener)

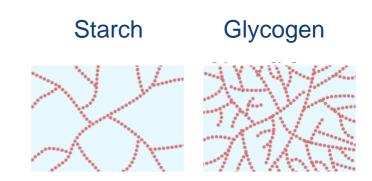
Cellulose





β-1,4 glycosidic linkage





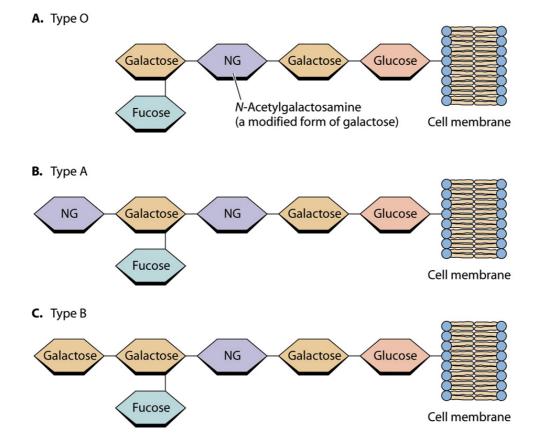
 α -1,4 glycosidic linkage

Roles of Carbohydrates I

- Carbohydrates in energy metabolism
 - Plant
 - Glucose synthesis by photosynthesis
 6CO₂ + 6H₂O + energy → C₆H₁₂O₆ + 6O₂
 - Starch for energy storage
 - cellulose for structural compound
 - Animals
 - Intake glucose from food
 - Glycogen for energy storage

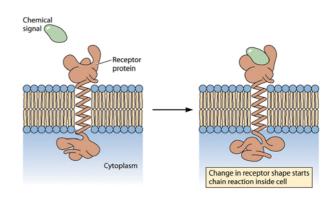
Roles of Carbohydrates II

- Carbohydrates in molecular recognition
 - Cellular recognition, cell signaling, cell adhesion
 - e.g. blood typing : sugar chains in the membrane of RBC



Proteins

- Roles of proteins
 - Most of the cellular functions
 - Enzymes : chemical reactions
 - Receptors : signal transduction
 - Antibody: recognition of foreign molecules and trigger immune responses
 - Transporters: e.g. hemoglobin for oxygen
 - Structural proteins: keratin (hair and nails), actin and myosin (muscle)
 - Diversity of organism
 - Diversity of organization of proteins within an organism



Proteins

Amino acids

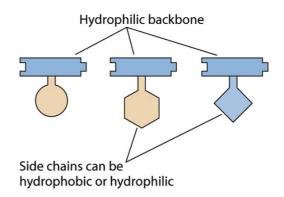
- Building blocks of proteins
- Hydrophilic backbone + 20 side chains

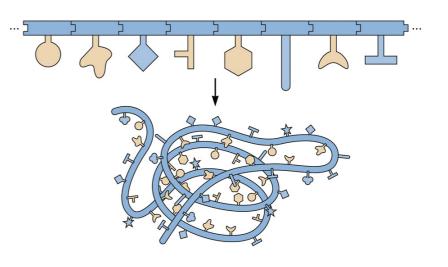
Polypeptide

 Amino acid chains linked by peptide bond

Tree-dimensional structure

- Determines protein function
- Determined by amino acid sequence





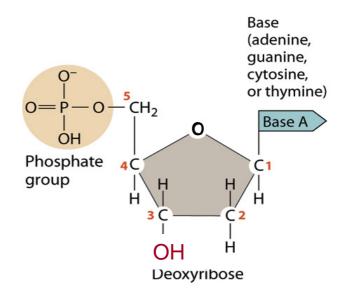
Nucleic acids

Nucleotides

- Building blocks of nucleic acids
- (deoxy)ribose + phosphate group + 4 bases
- Bases: adenine (A), guanine (G), cytosine (C), thymine (T)

Terminology

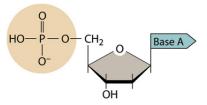
- Base
- Nucleoside : sugar + base
- Nucleotide : sugar + base + phosphate



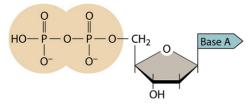
Nucleotide Chains

 Linkage of 5' carbon to 3' carbon through phosphodiester bond

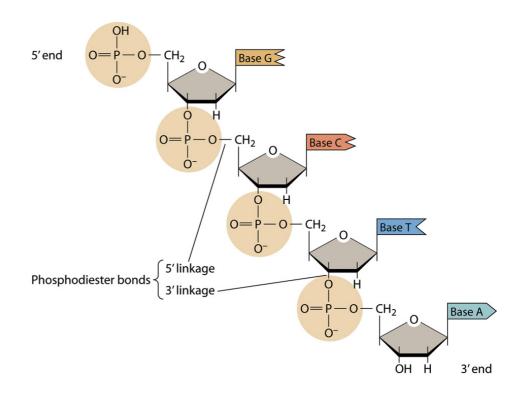
A. Adenosine monophosphate (AMP)



B. Adenosine diphosphate (ADP)

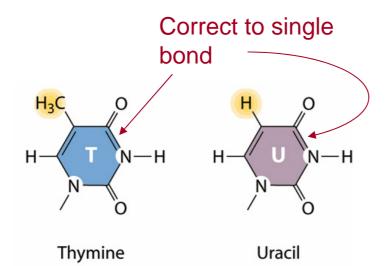


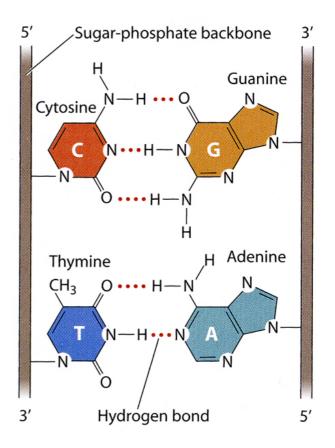
C. Adenosine triphosphate (ATP)



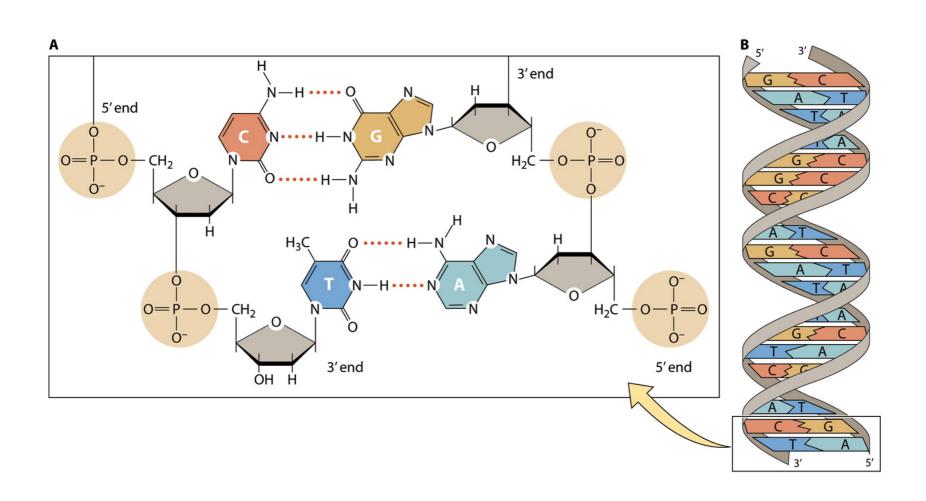
Nucleotide Chains

- Base pairing
 - C=G, T=A : hydrogen bonding
 - Complementary base pairs
 - Antipararallel strand in DNA molecule





Nucleotide Chains



DNA Replication

Synthesis of a complementary strand using the other strand as a template

DNA polymerase

