



Chromatography Column Mass Balance with Negligible Dispersion



Examples: α and elution volume

■ Example 1

: large sample volume

- Sample emerging after $1/(1-\alpha)$ column volume
- Sample volume = 5 x column volume (V_c) and washing with 5 x column volume
 - α of a protein to remain in the column
 - $1/(1-\alpha) \times V_c > 10 V_c$
 - $\alpha > 0.9$

■ Example 2

: Protein of interest has α of near 0

- Purification of protein in the flow-through fraction

Example : Linear equilibrium

■ Example3 : Chromatographic separation of two solutes

■ Conditions

- Linear equilibrium constant $K_{eq,1} = 7.5, K_{eq,2} = 7.8$
- Flow rate : 1.5 liter/min
- Column diameter: 63 cm
- Void fraction : 0.33

■ (Q) What column length (L) is required to separate the two solute by 5 min ?

■ (S)

- $u_1 =$

- $v =$

- Elution time for solute 1 =

$$u_i = \frac{v}{\varepsilon + (1 - \varepsilon)q'_i(c_i)}$$

Example : Shock Wave Velocity

■ Effective solute velocity at the front

- The concentration change is not continuous but is a step change

- $u_{i,sh} = v / [\varepsilon + (1 - \varepsilon)(\Delta q_1 / \Delta c_1)]$

■ Example4 : Shock wave velocity for a nonlinear isotherm

■ Conditions

- Solute 1 with Langmuir isotherm characteristics, $K_{eq} = 7.5$ ml/mg
- $S_{tot} = 120 \mu\text{g/ml}$
- Same column conditions as Example 3

- (Q) Calculate shock wave velocity for an injection of 1 mg/ml

■ (S)

- $[CS] = \{K_{eq} S_t [C]\} / \{1 + K_{eq}[C]\} =$
- $\Delta q_1 / \Delta c_1 =$
- $u_{i,sh} =$

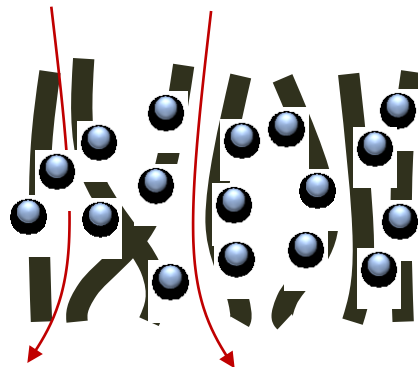


Membrane Adsorbents: Radial Flow Columns



Membrane adsorbent

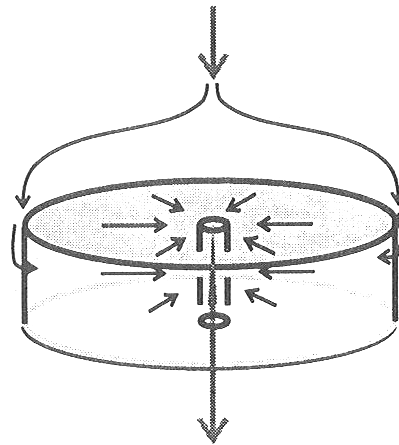
- **A stack of many membranes**
- **Advantage of membrane**
 - Large surface area per unit volume
 - No diffusion of solute away from the buffer flow
 - Fast flow rate
 - Resistance against pressure is determined mainly by pore size
 - The increase in required pressure is proportional to the stack thickness
 - Can apply fast flow rate (100-500 cm/h)
 - No significant time delay for binding
- **Capacity**
 - $0.1 \sim 0.3 \text{ mg/cm}^2$, 10 mg/cm^3 (50 membrane $\sim 1 \text{ cm}$)
 - 5 to 10 times less than conventional column beads (ion exchange)



Radial Flow Column

■ Increase flow rate

- Flow from outside of the column cylinder to the inside
- Collection tube receives the buffer flow
- High flow rate
 - Because of large cross-sectional area of initial contact





Batch Adsorption



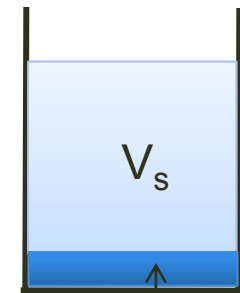
General Principles

■ Methods

- Mix adsorbent with proteins
- Recovery and washing by flocculation, filtration, or centrifugation
- Can be used to remove unnecessary proteins

■ Requirement

- α should be very close to 1
- Concentration of free protein is the same throughout the volume
 - The amount of protein adsorbed : qV_a
 - q : conc. of adsorbed protein
 - The amount of protein not adsorbed : $p(V_a + V_s)$
 - p : conc. of free protein
 - Partition coefficient
$$\alpha = q/(p + q), \quad p = q(1 - \alpha)/\alpha$$
 - Fraction protein adsorbed
$$f = qV_a/[p(V_a + V_s) + qV_a] = V_a \alpha / [V_a + (1 - \alpha)V_s]$$



V_a : Volume of
adsorbent when
settled

Application of Batch Adsorption

■ Requirement for high α

- $\alpha \approx 1$ for 80-90% yield
 - Need to optimize α by changing binding buffer
- $K_p < 10^{-6}$ M for $\alpha > 0.98$

■ Suitable for highly specific adsorbents

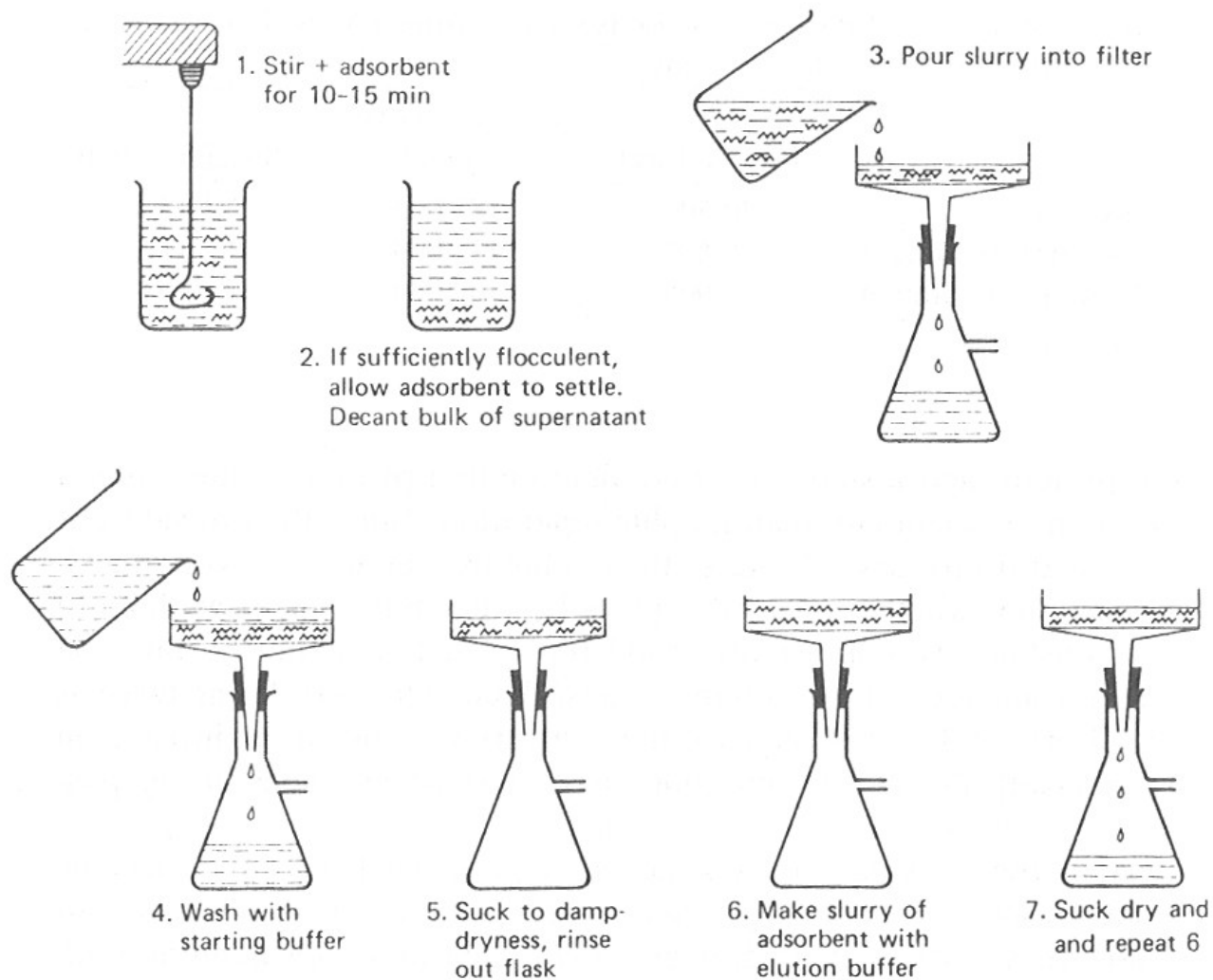
- Immunoabsorbents
- Dye ligand adsorbents

■ Batch vs. column

- Batch
 - Fast
 - Can treat large volume
 - Loss of protein if $\alpha < 0.98$
- Column
 - 100 % adsorption even at $\alpha = 0.95$
 - Slower
 - Possible clogging with crude extracts

V_a/N_s	α	f
0.05	0.9	0.30
0.1	0.9	0.45
0.2	0.9	0.60
0.5	0.9	0.75
0.1	0.95	0.63
0.2	0.95	0.76
0.5	0.95	0.86
0.05	0.98	0.70
0.1	0.98	0.82
0.1	0.98	0.89
0.1	0.99	0.99
0.02	0.998	0.91
0.05	0.998	0.96

Practical Approach



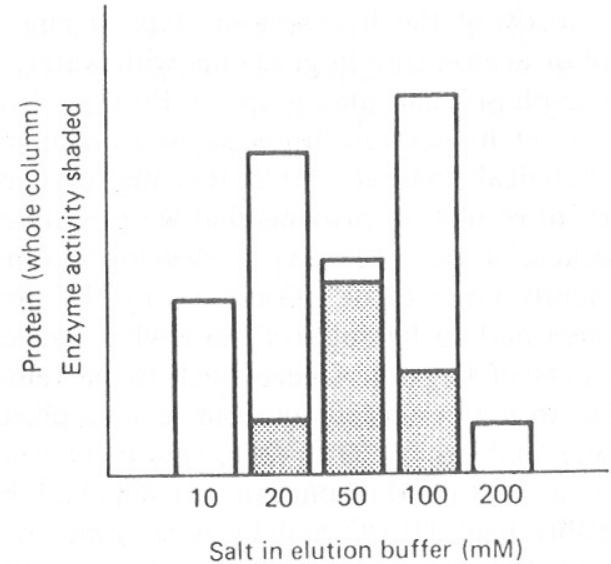
Practical Approach

■ Step-wise elution

- Wash with 20 mM salt
- Elution with 50-60 mM salt

■ Typical batch adsorbents

- Hydroxyapatite
- Ion exchangers
- Affinity adsorbents
- Dye ligand adsorbents
- Hydrophobic adsorbents
- Immunoadsorbents





High-Performance Liquid Chromatography



■ High pressure liquid chromatography

- Developed for separation of low-molecular weight compound
 - Fast flow rate to prevent diffusion
 - Smaller beads for faster flow rates
 - Higher pressure for smaller beads
- Reverse-phase chromatography
 - Using organic solvents methanol, acetonitrile in gradients with water or acidic solutions

■ High performance liquid chromatography

- High pressure is not as important as high quality of the adsorbent material

Bead Size, Flow Rate, Pressure, and Optimum Performance

■ Ways to increase plate number

- Long column → increase in operation time
- Small and spherical beads → decrease in flow rate
→ need to develop pressure-resistant column

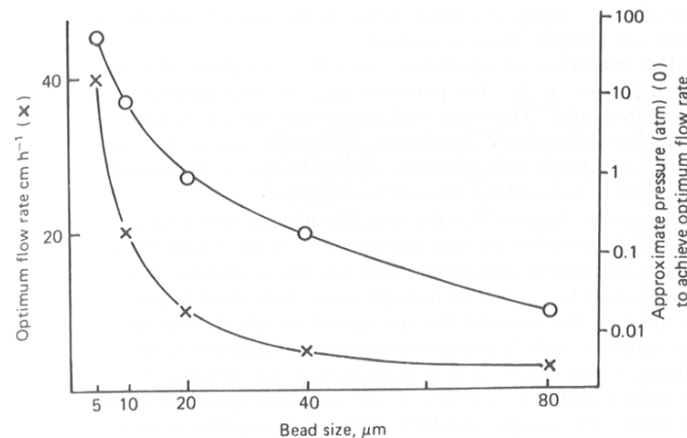
■ Relationship between parameters

- Rate of attainment of equilibrium
 - $\propto 1/D$ (diffusion coefficient of solute)
 - $\propto 1/d_p$ (bead diameter)
- Flow rate
 - $\propto d_p^2$ (assuming ideal packing)
- Resolution
 - $\propto \sqrt{L}$ (column length)
- Optimum performance
 - At the smallest value of plate height which depending on flow rate

Reduced Velocity

■ Determination of an ideal flow rate

- $V_{red} = v d_p / D$
 - v = linear flow rate
 - d_p = bead diameter
 - D = diffusion coefficient of solute
- Optimum V_{red} : 3~10
 - 50-200 kDa protein, d_p 100 μm (conventional low-pressure chromatography) $\rightarrow v < 2\text{cm/h}$: normal operation : 20 cm/h
 - d_p 10 μm (HPLC) $\rightarrow v < 20\text{cm/h}$, normal operation: 10 times faster
 - For d_p 20-40 μm
 - Optimum flow rate $\propto 1/d_p$
 - Pressure to obtain the same flow rate $\propto 1/d_p^2$



■ Bead size and Resolution

- Optimum resolution in the presence of 5×10^8 beads
 - 0.5 ml column for 10 μm beads
 - 200 ml for 80 μm beads

■ HPLC columns for protein works

- Adsorbents
 - Silica particles
 - Synthetic organic polymers with hydrophilic groups
- 1-20 ml total volume
- Theoretical plate height > 0.05 mm
- 0.5-2 ml/min \rightarrow protein elution after 5-60 min
- Cautions
 - Do not use particulate material nor unstable compounds in the sample to prevent precipitation or clogging
 - Filtration and degassing of the buffer before use
 - Use high quality reagents
 - Equilibrate the sample to the operation conditions and filtration through a 0.2 μm filter



Types of Adsorbent Used in Protein Chromatography



Adsorbents

■ Essential features of adsorbents

- The nature of the matrix
- The types of functional group that carries out the adsorption

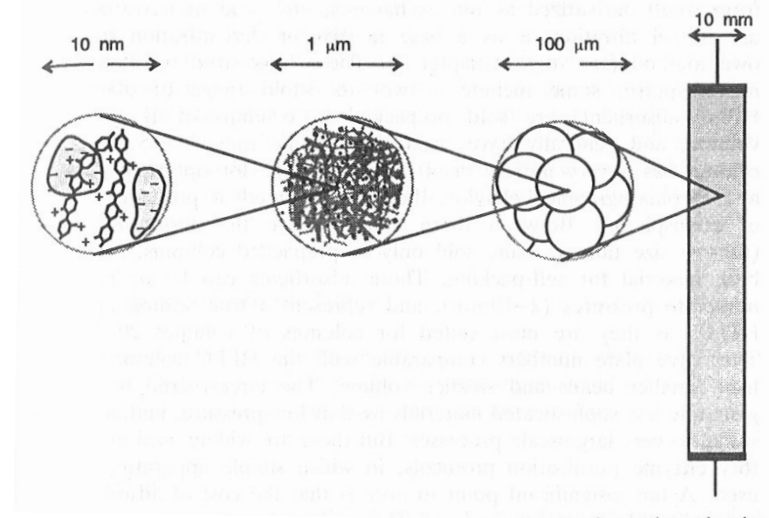
■ Nature of bead matrix

■ Matrix

- Insoluble polymer which is cast into bead form
- Requirements
 - Porosity : allow protein penetration
 - Rigidity

■ Matrix materials

- Carbohydrate polymer
 - Agarose, cellulose, dextran
- Synthetic organic polymers
- Inorganic matrices
 - Silica
- Composite structures



Adsorbents

■ Bead size

- For operation at low pressure (up to 2 atm)
 - 50-150 μm , not monodisperse
- For operation at moderate pressure (2-10 atm)
 - 30-40 μm
 - Most suited for columns of 20-500 ml volume
- HPLC adsorbents (10s atm)
 - 5-15 μm
 - Narrow bead diameter range for optimum packing
 - Packed in 1-20 ml volume

■ Attachment of ligand to the bead matrix

- By simple chemical reaction
 - Ion exchangers, hydrophobics
- Introduction of spacer arms for affinity ligands

Adsorbent Types 1: Ion Exchangers

■ Principles

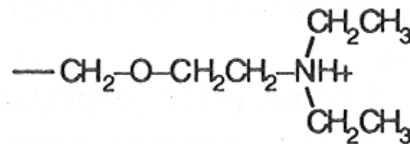
- Interaction of proteins by electrostatic attraction

■ Types

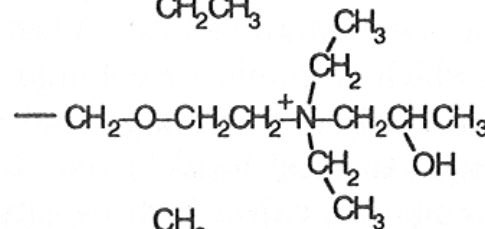
■ Anion exchangers

- Positively charged matrix
- Substituents attached to -OH on the matrix
 - Diethylaminoethyl(DEAE)
 - » NH^+ : complete deprotonation above pH 9.5
 - » Partly charged at pH7-8 : buffering effect for protein separations
 - Quaternary amino ethyl (QAE) : attached
 - » N^+

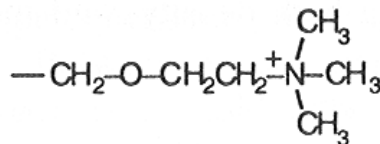
DEAE



Hydroxypropyl diethyl
aminoethyl, Q-



Trimethyl aminoethyl,
TMAE-, Q-



Adsorbent Types 1: Ion Exchangers

- Cation exchangers
 - Negatively charged matrix
 - Substituents
 - Weak carboxymethyl substituents (CM)
 - » Fully charged above pH 4.5
 - Strong sulfonate groups (S- and SP-)
 - » < pH 4
 - Phosphates : certain affinity actions
 - » Interaction with enzymes that bind multiply-phosphorylated substrates (e.g. nucleic acids), sugar phosphates

Carboxymethyl (CM) $\text{—CH}_2\text{—O—CH}_2\text{COO}^-$

Sulfopropyl (SP- or S-) $\text{—CH}_2\text{—O—CH}_2\text{CH}_2\text{CH}_2\text{SO}_3^-$

Hydrophobic and Reverse Phase Adsorbents

■ Principles

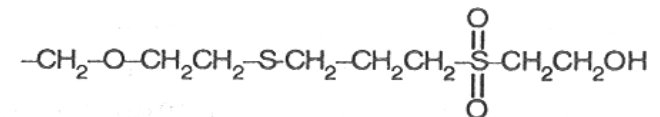
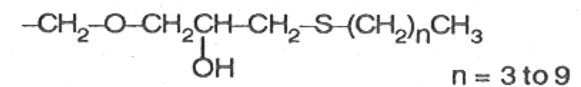
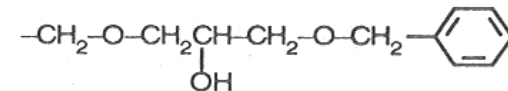
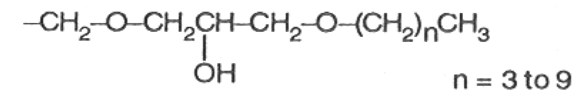
- Binding to proteins by hydrophobic interactions
- Strength of interaction
 - Higher interaction: high salt, high temperature
 - Weak interaction: presence of detergents or miscible organic solvents
- Poorer resolution than ion exchange chromatography

■ Substituents to -OH

- Short aliphatic chains (C₄ to C₁₀)
- Benzyl group

■ Reverse phase adsorbents

- Short aliphatic chains (C₈ to C₁₈)
- For HPLC, hydrophobic matrix, hydrophilic solvent
- Useful for small, stable proteins up to 30 kDa
 - Binding of denatured protein to the adsorbent
 - Bioactivity depends on the reversibility of the denaturation
- High resolving power



Other Adsorbents

■ Inorganic adsorbents

- Inorganic oxides and phosphates : batch mode
- Hydroxyapatite : column

■ Affinity adsorbents

- Mimicking protein-ligand interaction
- Biomimetic
 - Dyes, relatively nonspecific multifunctional adsorbents, metal chelators binding to target proteins

■ Immunoadsorbents

- Antibody adsorbents for protein purification
- Protein A/G for antibody purification



Operating Conditions for Column Chromatography



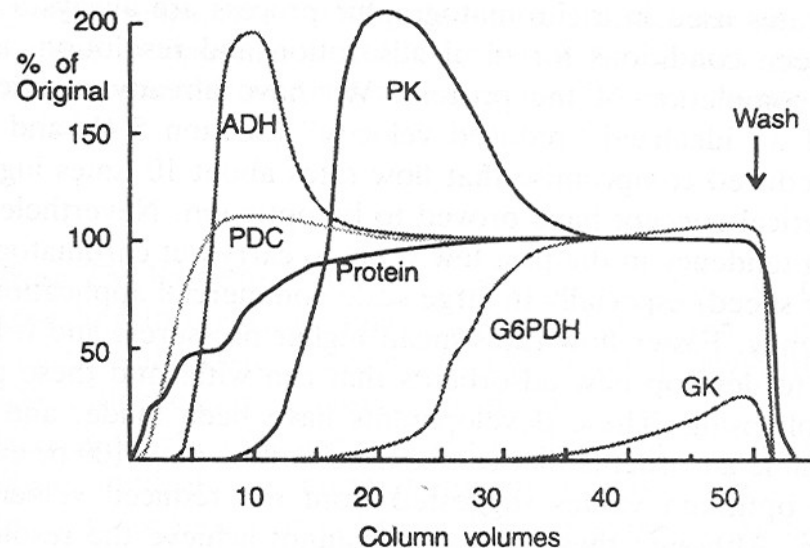
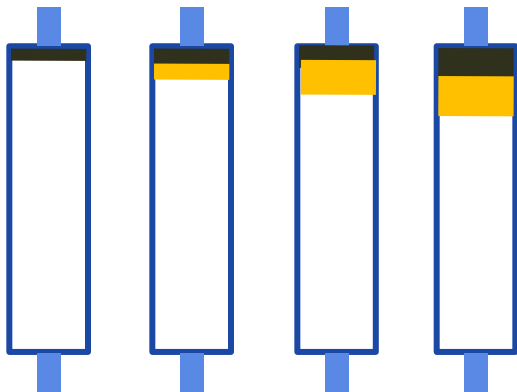
Sample Application

- **No particulate matter in the sample**
 - Filtration
 - Prevent using conditions (pH, temperature, salt concentration) which can induce protein aggregation
- **Adjust protein concentration of the sample**
 - Ultrafiltration or dilution
 - <10 mg/ml in general
 - At higher concentration
 - Proteins may affect the adsorption behavior
 - Generate osmotic pressure
 - For protein with low α value
 - High protein concentration and low volume
- **Determining the size of column**
 - Consider capacity of adsorbents
 - Ion exchangers : 1-10 mg/ml of adsorbent
 - Smaller loading → high loss, unnecessary cost, reduction of speed
 - If resolution is important : limited loading volume
 - If resolution is good or not very important : high loading is possible (e.g. affinity chromatography)

Overload and Displacement Chromatography

■ Principle

- Displacement of weakly binding proteins by strongly binding proteins
- Use insufficient adsorbent to bind all proteins
- Suitable for purification of proteins with high binding affinity



Purification of glucokinase (GK) using dye adsorbent

Flow Rates

■ Optimal flow rate

- 10 times faster flow rate than that suggested from reduced velocity (V_{red})
- For large beads (80-150 μm) : 20-30 cm / h at 20 °C, half of the rate in the cold room
- For 30-40 μm beads : 50-100 cm / h at 20 °C
- HPLC column with 10 μm beads : 2-4 cm / min

■ Fast flow rate

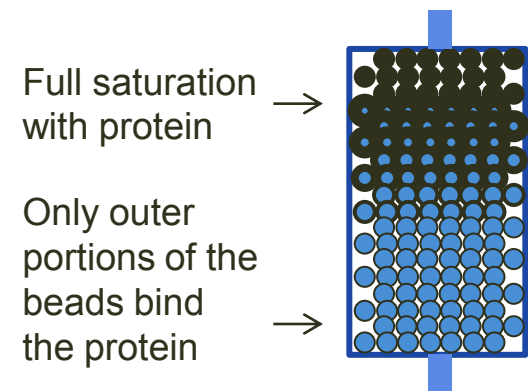
- 100~1,000 times faster flow rate than that suggested from V_{red}
- Need adsorbents resistant to high pressure
- Loss of resolution
- Gain of operational and economical advantage

■ Dynamic capacity

- Reduction of capacity at high flow rate
- Depending on the time needed for proteins to penetrate the beads and find the binding sites

Flow rate, cm/h	Mg IgG bound/ml of adsorbent
0	23
30	18
100	12.3
200	7.1
300	4.7
500	2.3
1000	0.83

- ProteinG Sepharose™ Fast Flow
- V_{red} : 1-3 cm/h



Effect of fast flow rate during column loading