Optimal Design of Energy Systems (M2794.003400)

Chapter 5. MODELING THERMAL EQUIPMENT

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5.1 Using physical insight

- Major concerns of this chapter : actual thermal equipment

Heat exchanger :

It is important to select the type of heat exchanger and calculate how a certain heat exchanger will perform

Distillation separator :

Understanding of separation of binary mixtures expands the horizons of applications of the simulation and optimization

Turbomachinery :

Studying the turbomachinery shows how the use of dimensionless group can simplify the equation

5.2 Selecting vs. simulating a heat-exchanger

- **Selecting** the heat exchanger:
 - ① Choosing type of the heat exchanger (Shell & tube, Finned, compact, etc.)
 - ② Specifying the details (number of tubes, tube diameter, core size, etc.)
 - ③ Heat transfer duty is specified already
- **Simulating** the heat exchanger:
 - 1) Heat exchanger already exists, either in actual hardware or specific design
 - ② Simulation of a heat exchanger consists of predicting outlet conditions
 - ③ Performance charicteristics of the heat exchanger are available (such as the area and overall heat transfer coefficients)

5.3 Counterflow heat exchanger

- Most favorable ΔT is achieved with a counterflow arrangement

(Hot side fluid)

$$q = w_h c_{ph}(t_{h,i} - t_{h,o})$$

(Cold side fluid)

$$q = w_c c_{pc}(t_{c,o} - t_{c,i})$$

(Heat transfer rate)

$$q = UA\Delta T_{lm}$$



Fig. Typical counterflow heat exchanger

5.3.1 LMTD (Log Mean Temperature Difference) method for a counter arrangement

(Hot side fluid) $dq = -w_h c_{ph} dT_h = -W_h dT_h$

(Cold side fluid)

$$dq = -w_c c_{pc} dT_c = -W_c dT_c$$

(Heat transfer)

$$dq = UdA(T_h - T_c) = UdA\Delta T$$

$$d(\Delta T) = d(T_h - T_c) = dT_h - dT_c = -dq(\frac{1}{W_h} - \frac{1}{W_c})$$



5.3.1 LMTD method for a counter arrangement

- represent the heat transfer rate as

$$dq = \frac{-d(\Delta T)}{1/W_h - 1/W_c} = UdA\Delta T \rightarrow \frac{d(\Delta T)}{\Delta T} = -UdA(\frac{1}{W_h} - \frac{1}{W_c})$$

- integrating on both side,

$$\int \frac{d(\Delta T)}{\Delta T} = \ln\left(\frac{T_{h,o} - T_{c,i}}{T_{h,i} - T_{c,o}}\right) = -\frac{UA}{q} (T_{h,i} - T_{h,o} - T_{c,o} + T_{c,i})$$



5.3.1 LMTD method for a counter arrangement

- finally, heat transfer rate at the counter flow hx is represented as

$$q = UA\Delta T_{lm} = UA\frac{((T_{h,o} - T_{c,i}) - (T_{h,i} - T_{c,o}))}{\ln((T_{h,o} - T_{c,i})/(T_{h,i} - T_{c,o}))} = UA\frac{\Delta T_2 - \Delta T_1}{\ln(\Delta T_2/\Delta T_1)}$$

- When '**select**' the heat exchanger under a certain fluid condition, LMTD method is a good way to specify the required UA

5.3.2 ε-NTU method for a counter flow HX

- Effectiveness, ε :

$$\varepsilon = \frac{q}{q_{max}} \quad (0 < \varepsilon < 1)$$
$$q = \varepsilon q_{max} = \varepsilon W_{min}(T_{h,i} - T_{c,i})$$

- Number of Transfer unit, NTU :

$$NTU = \frac{UA}{W_{min}}$$

- Heat capacity ratio, W_r:

$$W_r = \frac{W_{min}}{W_{max}}$$

5.3.2 ε-NTU method for a counter flow HX

- It is possible to represent ϵ as a function of NTU and heat capacity ratio for all the types of heat exchanger

$$\varepsilon = f(NTU, W_r)$$

- To '**simulate**' the existing heat exchanger, ε-NTU method is a useful way to obtain heat transfer rate of the heat exchanger

5.3.2 ε-NTU method for a counter flow HX

- To get an ϵ -NTU relation for a counter flow HX ($W_{min} = W_h$), effectiveness is given as

$$\boldsymbol{\varepsilon} = \frac{q}{q_{max}} = \frac{W_h(T_{h,i} - T_{h,o})}{W_{min}(T_{h,i} - T_{c,i})} = \frac{T_{h,i} - T_{h,o}}{T_{h,i} - T_{c,i}}$$

- In a fact that heat transfer rate of each side is same, heat capacity ratio is represented as

$$q = W_h (T_{h,i} - T_{h,o}) = W_c (T_{c,o} - T_{c,i})$$
$$W_r = \frac{W_{min}}{W_{max}} = \frac{W_h}{W_c} = \frac{T_{c,o} - T_{c,i}}{T_{h,i} - T_{h,o}}$$

5.3.2 ε-NTU method for a counter flow HX

- Meanwhile, rearranging the relation for heat transfer rate and LMTD yields

$$q = UA \frac{\Delta T_2 - \Delta T_1}{\ln(\Delta T_2 / \Delta T_1)} \implies \frac{\Delta T_2}{\Delta T_1} = \exp\left[\frac{UA}{q}(\Delta T_2 - \Delta T_1)\right]$$
$$\frac{T_{h,o} - T_{c,i}}{T_{h,i} - T_{c,o}} = \exp\left[\frac{UA}{q}\left[\left(T_{h,o} - T_{c,i}\right) - \left(T_{h,i} - T_{c,o}\right)\right]\right]$$
$$= \exp\left[-UA\left(\frac{\left(T_{h,i} - T_{h,o}\right)}{q} - \frac{\left(T_{c,o} - T_{c,i}\right)}{q}\right)\right] = \exp\left[-UA\left(\frac{1}{W_{min}} - \frac{1}{W_{max}}\right)\right]$$

- Right hand side of the equation is represented as

$$\frac{T_{h,o} - T_{c,i}}{T_{h,i} - T_{c,o}} = \exp\left[\frac{UA}{W_{min}}\left(1 - \frac{W_{min}}{W_{max}}\right)\right] = \exp\left[-\mathsf{NTU}(1 - W_r)\right]$$

5.3.2 ε-NTU method for a counter flow HX

- To eliminate the outlet temperature of the left hand side, following sequence is needed.

$$\frac{T_{h,o} - T_{c,i}}{T_{h,i} - T_{c,o}} = \frac{\left(T_{h,o} - T_{h,i}\right) + \left(T_{h,i} - T_{c,i}\right)}{\left(T_{h,i} - T_{c,i}\right) + \left(T_{c,i} - T_{c,o}\right)} = \frac{1 - \frac{\left(T_{h,i} - T_{h,o}\right)}{\left(T_{h,i} - T_{c,i}\right)}}{1 - \frac{\left(T_{c,o} - T_{c,i}\right)}{\left(T_{h,i} - T_{c,i}\right)}} = \frac{1 - \frac{\left(T_{h,i} - T_{h,o}\right)}{\left(T_{h,i} - T_{c,i}\right)}}{1 - \frac{\left(T_{h,i} - T_{h,o}\right)}{\left(T_{h,i} - T_{c,i}\right)}}$$

- In a fact that
$$W_r = \frac{T_{c,o} - T_{c,i}}{T_{h,i} - T_{h,o}}$$
 and $\varepsilon = \frac{T_{h,i} - T_{h,o}}{T_{h,i} - T_{c,i}}$

$$\frac{T_{h,o} - T_{c,i}}{T_{h,i} - T_{c,o}} = \frac{1 - \varepsilon}{1 - \varepsilon W_r}$$

5.3.2 ε-NTU method for a counter flow HX

- Finally, by reconnecting the left hand side and right hand side, the relation for heat transfer rate obtained from the LMTD method is represented as

$$\frac{\Delta T_2}{\Delta T_1} = \exp\left[\frac{UA}{q}(\Delta T_2 - \Delta T_1)\right] \qquad \Longrightarrow \qquad \frac{1-\varepsilon}{1-\varepsilon W_r} = \exp\left[-\mathrm{NTU}(1+W_r)\right]$$

- Thus, it is obvious that LMTD relation and ϵ -NTU relation are two different form of one heat transfer system. Rearranging the relation for ϵ yields

$$\varepsilon = \frac{1 - exp[-NTU(1 + W_r)]}{1 - exp[-NTU(1 - W_r)]}$$

5.3.2 ε-NTU method for a counter flow HX

- It is possible to get same relation when $W_{min} = W_h$



Fig. Temperature profiles in a counterflow heat exchanger

5.5 <u>Evaporator</u> and <u>Condensers</u> Liquid \rightarrow Vapor Vapor \rightarrow Liquid

One of the fluid changes phase, and no superheating or subcooling

→ Its temperature or pressure remains constant



Fig. Temperature distribution in fluids in a condenser

5.5 Evaporator and Condensers

- When secondary fluid(hot side) is at a two phase state, temperature is at a constant state.

$$q = UA \frac{(t_{h,o} - t_c) - (t_{h,i} - t_c)}{\ln[(t_{h,o} - t_c)/(t_{h,i} - t_c)]} \rightarrow \frac{(t_{h,o} - t_c)}{(t_{h,i} - t_c)} = \exp[\frac{UA}{q}(t_{h,i} - t_{h,o})]$$

- Thus, $t_{h,o}$ is represented as

$$t_{h,o} = t_{h,i} - (t_{h,i} - t_{c,i})(1 - e^{-NTU})$$

5.5 Evaporator and Condensers

- ϵ -NTU relation for the case is represented as

$$\frac{t_{h,i} - t_{h,o}}{t_{h,i} - t_{c,i}} = \varepsilon = 1 - e^{-NTU}$$

- Or as an alternative form

$$NTU = -\ln(1-\varepsilon)$$





Fig. Effectiveness of counter HX



Fig. Effectiveness of parallel HX

5.6 ε-NTU method for several cases

Flow Arrangement	Relation	
Parallel ow	$\mathrm{NTU} = -\frac{\ln\left[1 - \varepsilon(1 + C_t)\right]}{1 + C_t}$	(11.28b)
Counterow	$NTU = \frac{1}{C_r - 1} \ln\left(\frac{\varepsilon - 1}{\varepsilon C_r - 1}\right) \qquad (C_r < 1)$	
	$NTU = \frac{\varepsilon}{1 - \varepsilon} \qquad (C_r = 1)$	(11.29b)
Shell-and-tube		
One shell pass (2, 4, tube passes)	$(NTU)_1 = -(1 + C_r^2)^{-1/2} \ln \left(\frac{E-1}{E+1}\right)$	(11.30b)
	$E = \frac{2\ell_{\mathcal{B}_1} - (1 + C_{\ell})}{(1 + C_{\ell}^2)^{1/2}}$	(11.30c)
<i>n</i> shell passes $(2n, 4n, \ldots$ tube passes)	Use Equations 11.30b and 11.30c with	
	$\varepsilon_1 = \frac{F-1}{F-C_r}$ $F = \left(\frac{\varepsilon C_r - 1}{\varepsilon - 1}\right)^{1/n}$ NTU = n (NTU) ₁	(11.31b, c, d)
Cross-ow (single pass)		
C_{max} (mixed), C_{\min} (unmixed)	$NTU = -\ln\left[1 + \left(\frac{1}{C_r}\right)\ln(1 - \varepsilon C_r)\right]$	(11.33b)
C _{min} (mixed), C _{max} (unmixed)	$NTU = -\left(\frac{1}{C_r}\right) \ln[C_r \ln(1-\varepsilon) + 1]$	(11.34b)
All exchangers ($C_r = 0$)	$\mathrm{NTU} = -\ln(1-\varepsilon)$	(11.35b)

Table. ε-NTU relation (for NTU)

5.6 ε-NTU method for several cases

Flow Arrangement	Relation		
Parallel ow	$\varepsilon = \frac{1 - \exp\left[-\text{NTU}(1 + C_r)\right]}{1 + C_r}$		(11.28a)
Counterow	$\varepsilon = \frac{1 - \exp[-\text{NTU}(1 - C_r)]}{1 - C_r \exp[-\text{NTU}(1 - C_r)]}$	$(C_r < 1)$	
	$\varepsilon = \frac{\text{NTU}}{1 + \text{NTU}}$	$(C_r = 1)$	(11.29a)
Shell-and-tube			
One shell pass (2, 4, tube passes)	$\varepsilon_1 = 2 \left\{ 1 + C_r + (1 + C_r^2)^{1/2} \times \frac{1 + ex}{1 - ex} \right\}$	$\frac{p[-(NTU)_{1}(1+C_{r}^{2})^{1/2}]}{p[-(NTU)_{1}(1+C_{r}^{2})^{1/2}]}\right\}^{-1}$	(11.30a)
n shell passes (2n, 4n, tube passes)	$\varepsilon = \left[\left(\frac{1 - \varepsilon_1 C_r}{1 - \varepsilon_1} \right)^n - 1 \right] \left[\left(\frac{1 - \varepsilon_1 C_r}{1 - \varepsilon_1} \right)^n \right]$	$-C_r$	(11.31a)
Cross-ow (single pass)			
Both fluids unmixed	$\varepsilon = 1 - \exp\left[\left(\frac{1}{C_{t}}\right)(\mathrm{NTU})^{0.22} \left\{\exp\left[-C_{t}\right]\right\}\right]$	$C_r(NTU)^{0.78}] = 1$	(11.32)
C_{\max} (mixed), C_{\min} (unmixed)	$\varepsilon = \left(\frac{1}{C_r}\right)(1 - \exp\left\{-C_r\left[1 - \exp\left(-N\right)\right]\right)$	FU)]})	(11.33a)
C _{min} (mixed), C _{max} (unmixed)	$\varepsilon = 1 - \exp(-C_r^{-1} \{1 - \exp[-C_r(NTU)]\}$	D]})	(11.34a)
All exchangers $(C_r = 0)$	$\varepsilon = 1 - \exp(-\text{NTU})$		(11.35a)

Table. ϵ -NTU relation (for ϵ)

5.10 Binary solutions

- Mass fraction of A :
$$x_A = \frac{m_A}{m_A + m_B}$$

- Mole fraction of A :
$$y_A = \frac{N_A}{N_A + N_B} = \frac{m_A/M_A}{m_A/M_A + m_B/M_B}$$



Fig. Typical binary solution

5.11 Temperature-concentration-pressure characteristics





Fig. Temperature-concentration diagram at a constant pressure

Fig. Temperature-concentration diagram for two different pressure

5.12 Develiping a T vs x diagram

- There exist three tools to develop the binary properties

(Saturation pressure-temperature relation)



5.13 Condensation of a binary mixture

- A pure substance condenses at constant pressure, the temperature remains constant
- On the other hand, temperature of a binary mixture changes progressively



Fig. Condensation of a binary mixture

5.14 Single stage distillation



Fig. Single-stage still

Fig. Some possible outlet conditions

5.15 Rectification



Fig. A rectification column

Fig. States of binary system in rectification column

5.16 Enthalpy

- Enthalpy values of binary solutions and mixtures of vapor are necessary
- For system simulation, the enthalpy data would be most convenient in equation form
- More frequently the enthalpy data appear in graphic form as shown in the Figure below



Fig. Form of an enthalpy-concentration diagram

5.17 Pressure drop and pumping

- Pressure drop of an incompressible fluid :

(C : constant, w : mass of flow, n : $1.8 \sim 2.0$)

$$\Delta \boldsymbol{p} = \boldsymbol{C}(\boldsymbol{w}^n)$$

- Power required incompressible fluid :

$$Power = \eta_{pump} T \varpi = \Delta p Q = C(w^n) \cdot \frac{w}{\rho} = \frac{C}{\rho} w^{n+1}$$

5.18 Turbomachinery

- Dimensional form :
$$f(p_1, c_p, T_1, \varpi, w, D) = p_2$$

- Non-dimensional form : $f\left(\frac{w\sqrt{c_pT_1}}{D^2\rho_1}, \frac{\varpi D}{\sqrt{c_pT_1}}\right) = \frac{p_2}{p_1}$ using Pi theorem
 $\Pi_1 = \frac{w\sqrt{c_pT_1}}{D^2\rho_1}, \Pi_2 = \frac{\varpi D}{\sqrt{c_pT_1}}, \Pi_3 = \frac{p_2}{p_1}$

