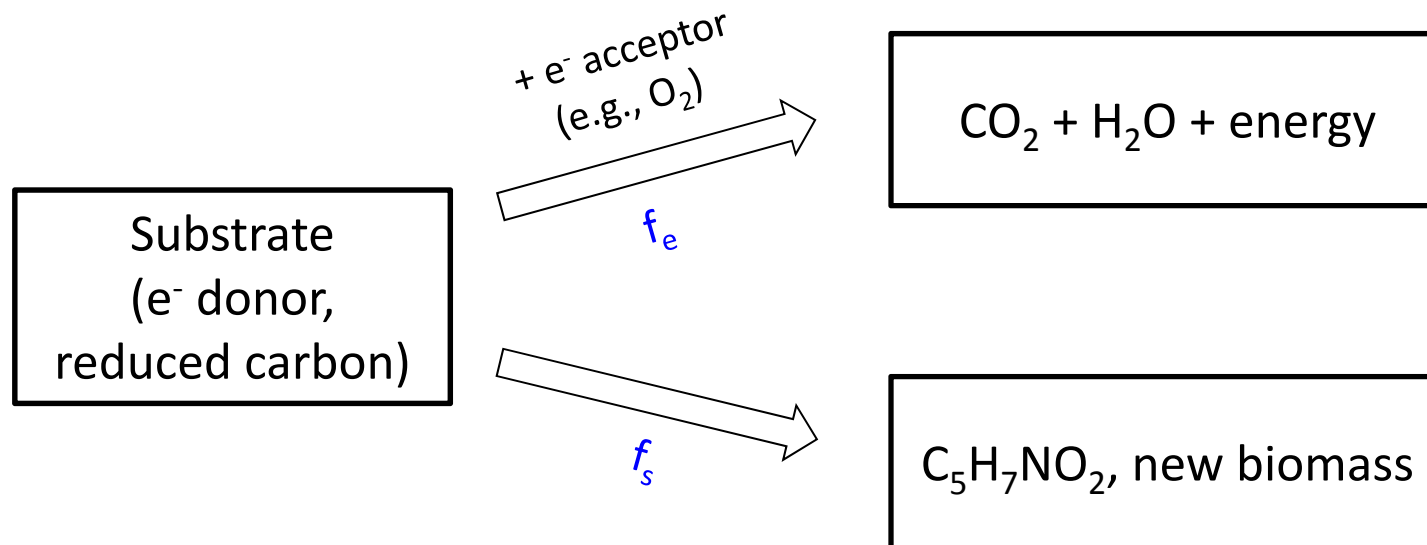


Fundamentals of biological treatment II

Oxygen uptake rate

- How much oxygen is needed for the bacterial growth?
- **Electron partitioning**



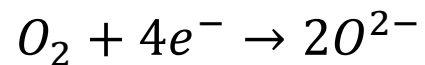
f_e: fraction of e⁻ donor electron used for energy generation

f_s: fraction of e⁻ donor electron used for cell synthesis

$$f_e + f_s = 1$$

Oxygen uptake rate

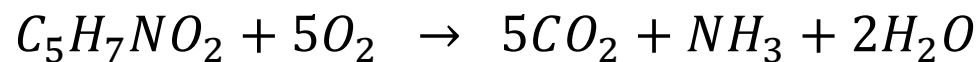
- COD of a substrate is directly proportional to “how much the carbon is reduced compared to its fully oxidized state (CO₂)”



32 g O₂ per 4 moles of electrons

electrons are provided by the e⁻ donor

- Biomass COD (C₅H₇NO₂ – representative cell formula)



$$\frac{5 \times 32 \text{ g COD/mole } C_5H_7NO_2}{113 \text{ g } C_5H_7NO_2/\text{mole } C_5H_7NO_2} = \mathbf{1.42 \text{ g COD/g } C_5H_7NO_2}$$

Oxygen uptake rate

- When substrates are utilized by microorganisms,
 - Some fraction of substrate COD is converted to biomass COD
(Some fraction of electrons is transferred to biomass)
 - The other fraction of substrate COD is consumed
(The other fraction of electrons is donated to e⁻ acceptor)
- So:
(substrate COD utilization rate)
= (biomass COD production rate) + (oxygen consumption rate)

$$r_o = r_{su} - 1.42r_x$$

r_o = oxygen uptake rate (g O₂/m³-d)

r_{su} = substrate utilization rate (g bsCOD/m³-d)

r_x = net biomass growth rate (g VSS/m³-d)

Temperature effect

- Recall the modified van't Hoff-Arrhenius relationship:

$$k_T = k_{20}\theta^{(T-20)}$$

*θ = temperature correction factor,
range from 1.02 to 1.25*

Total VSS and active biomass

- Note that total VSS includes not only active biomass but also:
 - Cell debris resulting from endogenous decay
 - Non-biodegradable VSS (nbVSS) in the influent wastewater
- During cell death, some portion dissolves into the liquid for consumption by other bacteria, and the other portion remains as non-biodegradable material
 - 10~15% of original cell weight is converted to nbVSS
 - This is referred to as cell debris
- **Rate of production of cell debris**

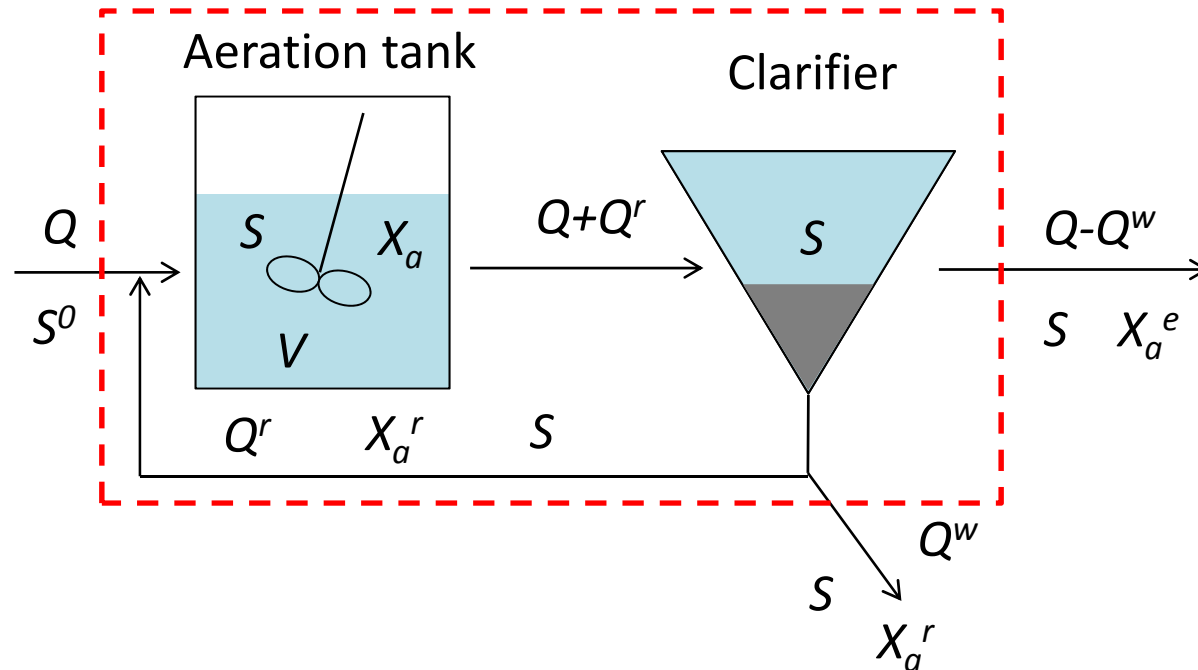
$$r_{X,i} = f_d bX$$

$r_{X,i}$ = rate of cell debris production (g VSS/m³-d)

f_d = fraction of biomass that remains as cell debris,

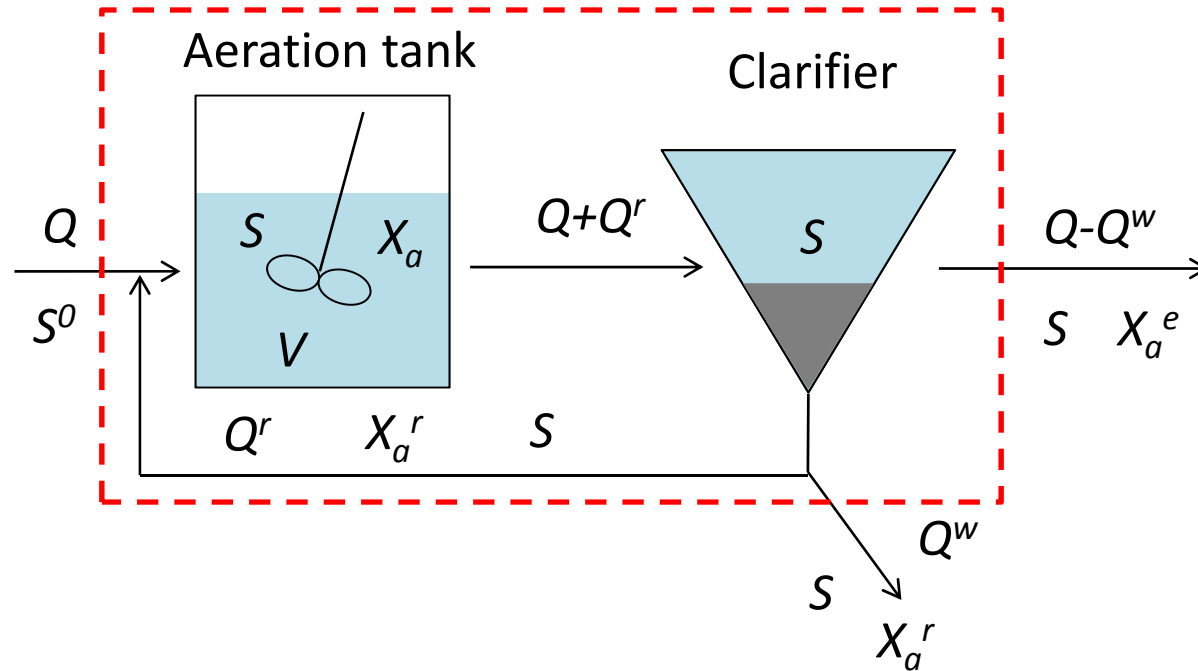
0.10-0.15 g VSS/g biomass VSS depleted by decay

Activated sludge - schematic diagram



- The majority of the settled sludge in the secondary clarifier is returned to the aeration tank to obtain high biomass concentration
- Some portion of the settled sludge (due to net growth of biomass) is removed from the system for steady state operation

Activated sludge - schematic diagram



- General assumptions:
 - Biodegradation of substrate occurs in the aeration tank only, not in the clarifier
 - No active biomass in the influent

Key variable - SRT

- Solids Retention Time (or mean cell residence time)
- The average time the activated sludge solids are in the system
- So SRT can be defined as:
(Amount of solids in the system) / (rate of solids exiting the system)

Assuming that the amount of solids in the clarifier is negligible compared to that in the aeration tank,

$$SRT = \frac{VX_a}{(Q - Q^w)X_a^e + Q^wX_a^r}$$

SRT = solids retention time (d)

V = aeration tank volume (m³)

Q = influent flowrate (m³/d)

X_a = active biomass concentration in the aeration tank (g VSS/m³)

Q^w = waste sludge flowrate (m³/d)

X_a^e = active biomass concentration in the effluent (g VSS/m³)

X_a^r = active biomass concentration in the return activated sludge line (g VSS/m³)

Modeling suspended growth processes

- Use the mass balance technique
- Use the kinetic expressions we have discussed
- We can set mass balance for two substances in the activated sludge system:

1) Biomass mass balance

$$V \frac{dX_a}{dt} = 0 - [(Q - Q^w)X_a^e] - Q^w X_a^r + r_x V$$

2) Substrate mass balance

$$V \frac{dS}{dt} = QS^0 - QS - r_{su} V$$

Modeling suspended growth processes

- Solving the two mass balance equations, we get:

$$X_a = \left(\frac{SRT}{\tau} \right) \left[\frac{Y(S^0 - S)}{1 + b(SRT)} \right]$$

$$S = \frac{K_s [1 + b(SRT)]}{SRT(Yk - b) - 1}$$

- The effluent substrate (=our target!) concentration is a function of SRT and growth kinetic parameters
- SRT is the only controllable variable
- The effluent substrate concentration is **NOT** a function of influent concentration (but S^0 affects X)

Solids production

- Remember:
 - VSS = active biomass + α
 - TSS = active biomass + α + β
- **Daily production of total sludge** from the system
 - At steady state, the mixed liquor can be assumed as a homogeneous mixture of active biomass and other solids (\rightarrow same SRT applies to VSS and TSS!)

$$P_{X,VSS} = \frac{X_{VSS}V}{SRT}$$

$P_{X,VSS}$ = daily production of total sludge as VSS (g VSS/d)

X_{VSS} = total MLVSS concentration in aeration tank (g VSS/m³)

$$P_{X,TSS} = \frac{X_{TSS}V}{SRT}$$

$P_{X,TSS}$ = daily production of total sludge as TSS (g TSS/d)

X_{TSS} = total MLSS concentration in aeration tank (g TSS/m³)

Modeling solids production

- Total MLVSS in the aeration tank, X_{VSS}

$$X_{VSS} = X_a + X_i$$

$X_i = nbVSS$ concentration in aeration tank (g VSS/m³)

→ Additional mass balance needed for nbVSS

$$V \frac{dX_i}{dt} = QX_i^0 - \frac{X_i V}{SRT} + r_{X,i} V$$

$X_i^0 = nbVSS$ concentration in influent (g VSS/m³)

$r_{X,i} =$ rate of nbVSS production from cell debris (g/m³-d)

At steady state:

$$X_i = \frac{X_i^0 (SRT)}{\tau} + (f_d)(b)(X_a)(SRT)$$

Modeling solids production

Therefore,

$$X_{VSS} = \left(\frac{SRT}{\tau}\right) \left[\frac{Y(S^0 - S)}{1 + b(SRT)} \right] + (f_d)(b)(X_a)(SRT) + \frac{X_i^0(SRT)}{\tau}$$

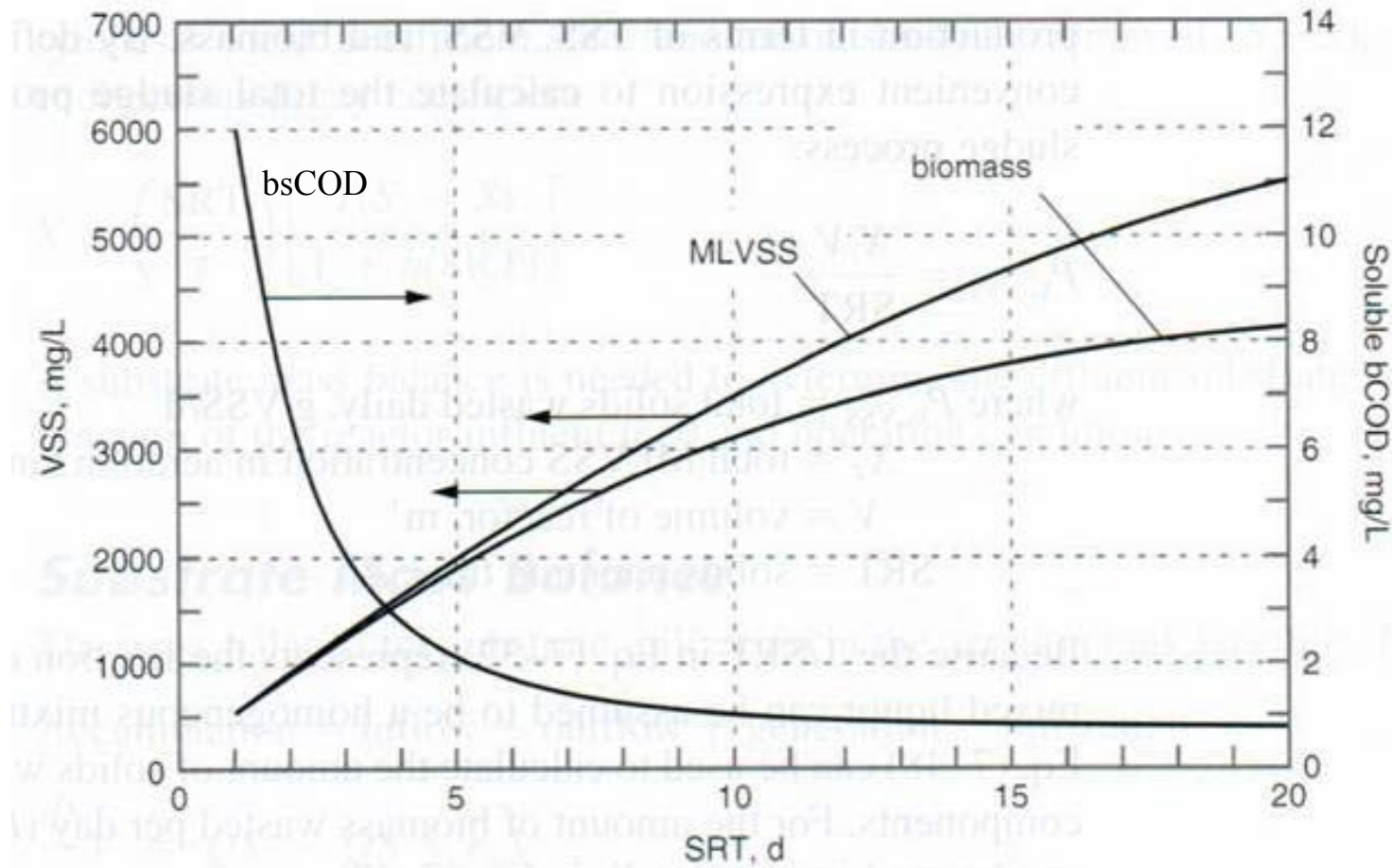
Active biomass *Cell debris* *nbVSS in influent*

The daily total VSS production (=wasted) rate, $P_{X,VSS}$ (g VSS/d):

$$P_{X,VSS} = \frac{QY(S^0 - S)}{1 + b(SRT)} + \frac{(f_d)(b)YQ(S^0 - S)SRT}{1 + b(SRT)} + QX_i^0$$

(A) **(B)** **(C)**

Effect of SRT on bsCOD, biomass, and MLVSS



Modeling solids production

- **The daily MLSS (total dry solids) wasted**

Note: TSS = VSS + FSS (inorganics)

- Inorganic solids originate from influent and the biomass
- Biomass contains 10-15% inorganic solids by dry weight
- Use a VSS/TSS ratio of 0.85 for a typical biomass

$$P_{X,TSS} = \frac{(A)}{0.85} + \frac{(B)}{0.85} + C + Q(X_{TSS}^0 - X_{VSS}^0)$$

$P_{X,TSS}$ = daily MLSS produced per day (g TSS/d)

X_{TSS}^0 = influent wastewater TSS concentration (g/m³)

X_{VSS}^0 = influent wastewater VSS concentration (g/m³)

Oxygen requirements

- Additional matter of interest: **how much oxygen should be provided** to support the aerobic biodegradation?
- Recall that by degradation of substrates:
 - some portion of the biodegradable COD (bCOD) is combined with O_2 to be mineralized or converted to oxidized organic compounds
- Some O_2 is also consumed for endogenous respiration
- Consider the COD mass balance of the system:
Oxygen used = (bCOD removed) – (COD of waste sludge)

$$R_o = Q(S_o - S) - 1.42P_{X,bio}$$

R_o = daily oxygen requirement (g/d)

*$P_{X,bio}$ = biomass as VSS wasted per day, **(A) + (B)** (g/d)*

Example question

Q: A complete-mix suspended growth activated sludge process with recycle is used to treat municipal wastewater after primary sedimentation. The characteristics of the primary effluent are: flow = 1000 m³/d, bsCOD = 192 g/m³, nbVSS = 30 g/m³, and inert inorganics = 10 g/m³. The aeration tank MLVSS is 2500 g/m³. Using these data and the kinetics coefficients given below, design a system with a 6-d SRT and determine the following:

- 1) The effluent bsCOD concentration
- 2) Hydraulic retention time required
- 3) Daily sludge production (in kg/d as VSS and TSS)
- 4) Fraction of active biomass in the MLVSS
- 5) Oxygen requirement (in kg/d)

$$k = 12.5 \text{ g COD/g VSS} \cdot \text{d}$$

$$Y = 0.40 \text{ g VSS/g COD}$$

$$b = 0.10 \text{ /d}$$

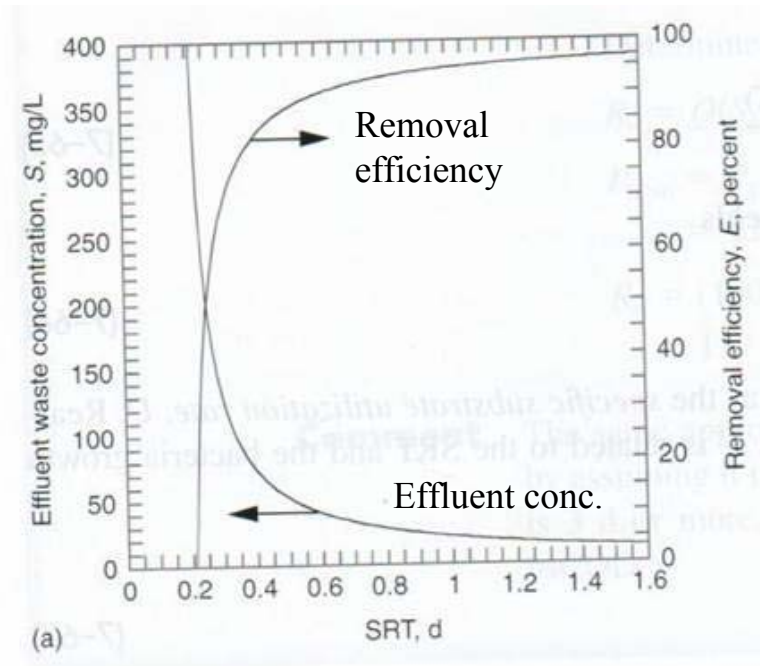
$$K_s = 10 \text{ g COD/m}^3$$

$$f_d = 0.15$$

$$\text{Biomass VSS/TSS} = 0.85$$

Design & operating parameters

- **SRT: key variable**
 - When kinetic coefficients are fixed, the effluent concentration is solely a function of SRT
- Effluent concentration as a function of SRT



For CSTR with recycle

Design & operating parameters

- **The minimum solids retention residence time, SRT_{min}**
 - The SRT at which the cells are washed out from the system faster than they can reproduce

$$\frac{1}{SRT_{min}} = \frac{YkS^0}{K_s + S^0} - b$$

- In many situations, $K_s \ll S^0$, so:

$$\frac{1}{SRT_{min}} \approx Yk - b = \mu_m - b$$

- Process safety factor, SF

$$SF = \frac{SRT_{des}}{SRT_{min}} \quad SRT_{des} = \text{design SRT (d)}$$

Design & operating parameters

- **F/M ratio (food to microorganism ratio)**

$$F/M = \frac{QS^0}{VX} = \frac{S^0}{\tau X}$$

F/M = food to microorganism ratio (g bsCOD/g VSS-d)

- High F/M \Rightarrow low steady-state SRT

- **Volumetric organic loading rate**

- The amount of BOD or COD applied to the aeration tank volume per day

$$L_{org} = \frac{QS^0}{V \cdot (10^3 \text{ g/kg})}$$

*L_{org} = volumetric organic loading rate
(kg bsCOD/m³-d)*