

Homework #4

*** Answer the following questions. Be sure to clearly show the procedures to solve the problems.**

A chemostat having $V = 2,000 \text{ m}^3$ receives a flow rate of $Q = 1,000 \text{ m}^3/\text{d}$ of wastewater containing $S^0 = 500 \text{ mg BOD}_L/\text{L}$. Also included in the wastewater is the inert biomass $X_i^0 = 50 \text{ mg VSS}/\text{L}$. The following parameters are found for aerobic biodegradation:

$$\hat{q} = 20 \text{ g BOD}_L/\text{g VSS}_a - \text{d}$$

$$Y = 0.42 \text{ g VSS}_a/\text{g BOD}_L$$

$$K = 20 \text{ mg BOD}_L/\text{L}$$

$$b = 0.15/\text{d}$$

$$f_d = 0.8$$

$$k_1 = 0.12 \text{ g COD}_p/\text{g BOD}_L$$

$$k_2 = 0.09 \text{ g COD}_p/\text{g VSS}_a - \text{d}$$

$$\hat{q}_{UAP} = 1.8 \text{ g COD}_p/\text{g VSS}_a - \text{d}$$

$$K_{UAP} = 100 \text{ mg COD}_p/\text{L}$$

$$\hat{q}_{BAP} = 0.1 \text{ g COD}_p/\text{g VSS}_a - \text{d}$$

$$K_{BAP} = 85 \text{ mg COD}_p/\text{L}$$

1. Calculate S_{\min} , θ_x^{\min} and θ_x of the chemostat. (10 points)
2. Calculate effluent VSS, COD and BOD_L . (30 points)
3. Calculate the effluent N and P concentrations when influent concentrations are $50 \text{ mg NH}_4\text{-N}/\text{L}$ and $10 \text{ mg PO}_4\text{-P}/\text{L}$, respectively. (20 points)
4. Calculate the amount of O_2 that should be supplied to the reactor when influent and effluent DO are 6 and 2 mg/L, respectively. (20 points)
5. Assuming that the influent also contains biodegradable particulate organic matter with a concentration of 100 mg COD_L and the hydrolysis rate coefficient is $k_{\text{hyd}} = 0.2/\text{d}$, recalculate the effluent VSS, COD, and BOD_L . (30 points)

Hints:

- Effluent VSS should include both active and inert biomass, and particulate organic matter supplied from the influent (if there is any)
($X_v = X_a + X_i + S_p$ in VSS)
(for COD→VSS conversion of S_p , assume S_p has a chemical formula similar to biomass)
 - Effluent COD should include COD of the substrate, SMP, and VSS
(eff. COD = substrate COD + SMP COD + VSS COD)
 - Conversion needed for VSS: recall 1.42 g COD/g VSS for biomass ($C_5H_7O_2N$)
 - BOD_L stands for "ultimate BOD", the oxygen demand for all biodegradable organic matter
 - S^0 is given as " BOD_L/L ", so substrate is assumed to be fully biodegradable
 - SMP is fully biodegradable
 - active biomass is partially biodegradable (biodegradable fraction = f_d)
 - inert biomass is non-biodegradable
- So: eff. BOD_L
- $$= \text{substrate } BOD_L (=COD) + \text{SMP } BOD_L (=COD) + f_d \times \text{active biomass COD}$$