

Mechanical properties of metallic glasses

Current Status of Structural Materials

2018.05.02

안혜상

Introduction of Metallic glass

Atomic structure of crystalline and amorphous materials





Crystalline materials



Unlike crystalline materials, absence of long-range order

 \rightarrow Unique mechanical properties, corrosion resistance induced by no defect, homogeneous distribution



Properties of metallic glasses

Mechanical properties of Metallic glass



High fracture strength over 5 GPa in Fe-based BMGs A.L. Greer, E. Ma, MRS Bulletin, 2007; 32: 612.

High strength of metallic glasses



Large elastic limit of metallic glasses

Corrosion resistance

JMR Volume 22, Issue 2 September 2006, pp. 302-313



Wear resistance Materials Science and Engineering A 386 (2004) 326–330







Zr₄₁Ti₁₄Cu_{12.5}Ni₁₀Be_{22.5} BMG

100um KYKY-2800 54

• BMG shows better wear resistance than GCr15 steel as a bearing roller



Characteristic temperature of Metallic glasses

Representative DSC curve of metallic glasses



Application of metallic glasses



Thermoplastic forming & Joining technique



Micro-forming of Pt-BMG fabricated by hot embossing on an etched Si wafer and hot cutting

Nano-rod of Pt-BMG formed by embossing on porous alumina



Drawback of metallic glasses



Although BMGs possess very high strength compared to their crystalline

counterparts, they generally suffer from low ductility



Overcoming the drawback of metallic glasses

Improvement of ductility through alloy design





P. Jia et al. / Scripta Materialia 54 (2006) 2165–2168

Improvement in the tensile ductility of BMG composites



Dramatic improvement in the tensile ductility of titanium–zirconium-based BMG composites containing a ~50 vol% dendritic phase



Atomistic models for plastic deformation in metallic glasses



Formation of shear band in metallic glass

Materials Science and Engineering: R: Reports Volume 74, Issue 4, April 2013, Pages 71-132







Preferential etching of shear bands on a polished cross-section of deformed Pd77.5Cu6Sil6.5glass.

Fracture in glassy metals proceeds by highly localized shear deformations which contrasts with the brittle fracture commonly observed in non-metallic glasses.



Deformation mechanism of metallic glasses

1. Shear transformation zones (STZs)_A.S.Argon



Diffuse and rearrangement

2. Free volume model_F.Spaepen



"atomic jumps" into free volume spaces

Ect

- Modified Spaepen model_Steif

by including additional free volume change due to pressure

- Directional structural relaxation model_Khonik

suggesting that each rearrangement event can be interpreted as a thermally-activated shear due to local atomic structures and subsequently nearly athermal viscous flow by external stress

- Cooperative shear model_Johnson and Samwer

yielding of metallic glasses displays a (T/Tg) 2/3 temperature dependence

Spaepen's and Argon's models remain most popular for describing deformation of metallic glasses



- Local topological fluctuation_T.Egami

propose an alternative approach based upon the exchange and fluctuation of atomic bonds, described in terms of the atomic level stresses.

Free-volume theory \implies However, volume **responds only to pressure**, not to shear stress.



A more realistic approach is to consider deformation from the point of view of atomic bond rearrangement.

If the structure is defined by the topology of **atomic connectivity**, deformation should involve **changes in the bond arrangement**.

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total number of bonds → conserved during the rearrangement
deformation → by bond-exchange
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well describe glass transition, structural relaxation, glass formation and mechanical deformation



Analysis of shear band_bending test



J. Appl. Phys., Vol. 94, No. 2, 15 July 2003





(b)

FIG. 1. (a) SEM micrograph showing shear bands in a 0.5 mm thick meltspun ribbon of Vitreloy 106, bent over a mandrel with a radius of 1 mm. The shear displacements associated with the shear bands are easily visible and secondary shear bands are also observed. The shear bands on the tension side of the sample extend farther into the sample than do those on the compression side. Conner *et al.*¹¹ (b) SEM micrograph showing shear bands in a 0.58 mm thick melt-spun ribbon of Vitreloy 106, bent over a mandrel with a radius of 1 mm. Some of the shear bands on the tension side of the sample appear to have developed as cracks. Conner *et al.*¹¹



FIG. 10. Calculated shear band spacing vs plate thickness at the point of fracture for Vitreloy I for both symmetric and nonsymmetric bending for two different values of the critical shear displacement, Δu^* , 3 μ m, and 10 μ m. A fracture toughness of $K_{I_c} = 20 \text{ MPa} \sqrt{m}$ was used for these calculations. The shear band spacing at the point of fracture is observed to increase with increasing plate thickness; the relationship is similar to the experimental results shown in Fig. 3.



Analysis of shear band_In-situ TEM

100 nm 100 nm b Diamond punch t=0.00 s Diamond punch t=4.47 s 100 nm d 100 nm Diamond punch t=5.78 s Diamond punch t=8.28 s 100 nm 100 nm t-9,78 s Diamond punch Diamond punch t=20.0 s

FIG. 3. In situ dark-field TEM observation of the formation and evolution of a major shear. The individual still frames [(a)–(f)] are extracted from a dynamic video sequence. The growing shear offset is indicated by the white arrow shown in (b)–(f) (see video in Ref. 25 for the jerky advancement of the shear offset and the flow of the MG outside the shear band region). PHYSICAL REVIEW B 77, 155419 (2008)



FIG. 4. Load vs displacement curve for the displacementcontrolled compression test. The various stages corresponding to those shown in Fig. 3(b)–3(e) are marked with letters. The load drops in the curve are observed to synchronize with the jerky advancement of the shear step seen during the compression test (see Ref. 25).



Analysis of shear band_Nano-indentation

Scripta Materialia, Volume 45, Issue 8, 29 October 2001, Pages 947-952



amorphous

partially crystalline



Localized plastic flow around a Berkovich indent on the surface of bulk amorphous Pd40Cu30Ni10P20.

Distinct shear band can clearly be observed in the amorphous region of the sample.



pop-ins can lead to nucleation and propagation of shear bands inside of the material as well



ESPark Research Group

S. Vincent et al. / Materials and Design 65 (2015) 98–103

Analysis of shear band_Nano-indentation





Analysis of shear band_Nano-indentation_Example 1



TABLE I. Parameters for the five metallic glasses. ρ_m is mass density; *E* is elastic modulus; *G* is shear modulus; *H* is hardness.

Metallic glasses	$\rho_{\rm m}$ (g/cm ³)	E (GPa)	G (GPa)	H (GPa)	
Co-based	9.285	293	111.53	16.4	"Hard" metallic glass
Fe-based	7.904	256	97.78	13.8	5
Zr-based	6.125	90	32.81	5.6	
Mg-based	3.794	60	22.85	2.9	"Soft" metallic glass
Ce-based	6.752	45	16.94	2.8	



Analysis of shear band_Nano-indentation_Example 1

 Loosely packed atom
 Densely packed atom
 (1) Local atom rearrangements upon loading
 (2) Expansion and coalescence of individual regions (Self-organized deformation events confined by elastic matrix)
 (3) Formation of an apparent shear band (Chaotic deformation events cuting through the elastic matrix)

Co-based metallic glass "Hard" metallic glass

Mg-based metallic glass "Soft" metallic glass



Analysis of shear band_Nano-indentation_Example 2



Mechanical response of local region from Statistical analysis by nano-indentation $Ta \rightarrow Nb \rightarrow Ti \rightarrow Zr$ decrease larger strain burst, increase smaller strain burst





