

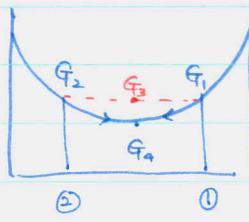
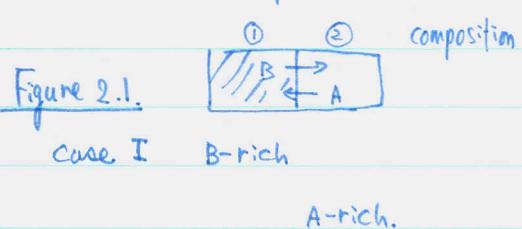
Chap. 2. Diffusion.

In Chap. 1. → talked about the stability or equil.^m arrangement of atoms in alloys.

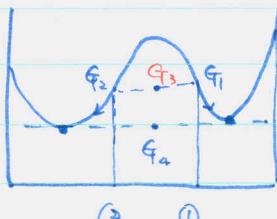
Phase transf → concerns about timing factor to reach the equil.^m.

- Any rxns in alloys ^{occur} to accomplish intermediate or final equil.^m.
- In the same context, diffusion occurs to reduce G of a sys.

* Two examples.



* general case
diffusion occurs to reduce and eliminate the conc. grad.
"down-hill diffusion"

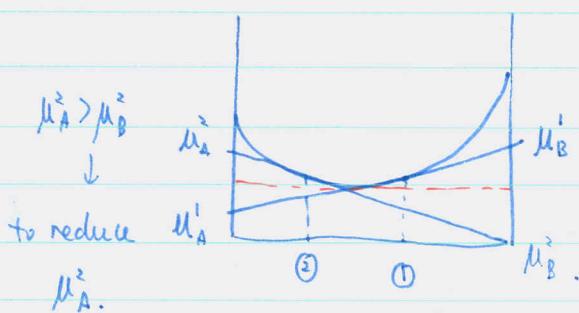


* miscibility gap.
A.B. diffuse towards the regions of high conc.
"up-hill diffusion"

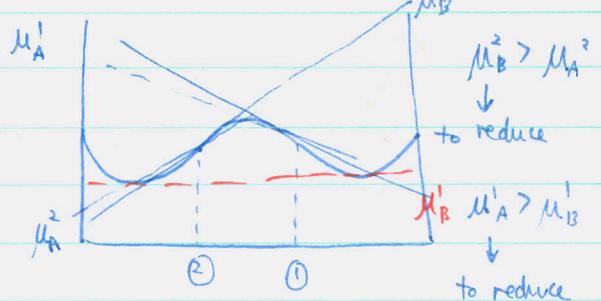
both to reduce "G_{total}" !!

from where the chem. pot. of A.B is high

to where " " is low.



$\mu_B^1 > \mu_A^1$
↓
to reduce μ_B^1



$\mu_B^2 > \mu_A^2$
↓
to reduce μ_B^2

∴ better to express the driving force for diffusion
in terms of a chem. pot. gradient.

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Use this approach. → but case I is more common, easy to measure conc. grad. compared to μ.
∴ diffusion - related to conc. grad.

2.1. Atomic Mechanisms of Diffusion.

Two common mechanisms.

- Substitutional atoms by a vacancy mech.
- interstitial atoms (small) by forcing their way betn the larger atoms.

Substitutional.

- oscillates w/ vibrational E. of $3kT$. $E \propto T(K)$.
assume freq. of the vib. is const. \therefore amplitude \uparrow as $T \uparrow \rightarrow$ ET
on next to
- If an atom oscillates violently in the vicinity of a vacancy, it can jump to the vac. site.
 \therefore Probability (to jump into a vac. site) $\approx P$ (to acquire suff. vib. E)
 \therefore migration rate \rightarrow related to jump.freq. \rightarrow conc. of vac. = $f(T)$.

Interstitial

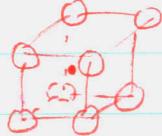
- when the size of a solute atom \ll that of the solvent.
- fcc octahedral site - in the middle of the unit cell
or mid way along the cubic edges.
bec. edge-centering or face-centering positions.
- conc. of interstitial atom is low \therefore small fraction of available sites
 \rightarrow each interstitial often surrounded by vac. \rightarrow jump to overcome the strain energy barrier.
interstitial-vac pairs = Frenkel defects

$$+ \text{charge vac} - \text{charge vac} = \text{Schottky defect}$$

2.2. Interstitial Diffusion (ID).

1. I.D. as a Random Jump Process.

how the I.D. diffusion is related to the Random Jump.



parent atoms
on a simple cubic lattice

sites

- Fig. 2.5. - equal prob. of the atom jumping to every one of the 6 adj.
- if the plane (1) contains n_1 (B-atoms) / m^2 .

$$\left| \begin{array}{l} \text{1. } \text{2.} \\ \text{1. } \text{1.} \end{array} \right. \quad (2.1) \quad \overrightarrow{J_B} = \frac{1}{6} \Gamma_B n_1 \text{ (atoms/m}^2\cdot\text{sec).} \quad (1) \rightarrow (2)$$

$$\left. \begin{array}{l} \text{1. } \text{2.} \\ \text{1. } \text{1.} \end{array} \right. \quad (2.1) \quad \overleftarrow{J_B} = \frac{1}{6} \Gamma_B n_2 \text{ (..)} \quad (2) \rightarrow (1)$$

(success. atom)
Jump rate
Jump / sec

$$(2.2) \quad \therefore \overrightarrow{J_B} = \frac{1}{6} \Gamma_B (n_1 - n_2) \quad (\text{net})$$

when α = the separation dist. of planes (1) and (2).

$$C_B(1) = n_1 / \alpha \text{ (atoms/m}^3), \quad C_B(2) = n_2 / \alpha \text{ (atoms/m}^3)$$

$$\therefore n_1 - n_2 = \alpha (C_B^{(1)} - C_B^{(2)}).$$

을 때까지 농도는 $C_B^{(1)} - C_B^{(2)}$.

$$\text{and Fig 2.5} \rightarrow C_B^{(1)} - C_B^{(2)} = -\alpha \cdot \frac{\partial C_B}{\partial x}.$$

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$$(2.3) \quad (\text{net flow}) \quad \therefore \overrightarrow{J_B} = - \left(\frac{1}{6} \Gamma_B \alpha^2 \right) \frac{\partial C_B}{\partial x} \quad \text{atoms/m}^2 \cdot \text{sec.}$$

$$(2.4) \quad \text{Fick's first law of Diffusion.} \quad = - D_B \frac{\partial C_B}{\partial x} \quad \text{where } \frac{1}{6} \Gamma_B \alpha^2 = D_B.$$

$$\text{unit [atoms/m}^2\cdot\text{s}] = [\text{m}^2/\text{sec}][\text{atoms/m}^3]$$

\downarrow
Intrinsic diffusivity
on diff. coeff.

(2.4) based on a simple cubic lattice, $\Gamma_B = f(C)\alpha^2$ good for cubic.
random diff. in

In non-cubic $D_B = f(\text{direction-Xtallgraphic})$ hcp. mmoclinic etc.

- $D = f(\text{Comp.} = \text{conc.})$. in real alloys

ex. $\gamma\text{-Fe}$ 1000°C $2.5 \times 10^{-11} \text{ m}^2/\text{sec}$ at 0.15 wt% C.

$7.7 \times 10^{-11} \text{ m}^2/\text{sec}$ at 1.4 wt.% C.

to release the strain E induced by C int.

ex. $\gamma\text{-Fe}$ 1000°C lattice para ($= a$) = 0.37 nm

$$\therefore \alpha = 0.37/\sqrt{2} = 0.26 \text{ nm} \quad \text{for Fe.} \quad D_{13} = \frac{1}{6} F_B \alpha^2.$$

using $2.5 \times 10^{-11} \text{ m}^2/\text{sec}$ as D , $\therefore \Gamma_c = 2 \times 10^9 \text{ jumps/sec.}$

If the vib. freq. of C $\sim 10^{13}$, then $\frac{1}{10^4}$ success rate $\sim 10^{13} \text{ freq.}$

- random walk - the direction of each new jump - indep. of that of previous jump.

• Random walk in 3-D

after n steps w/ jump dist. α

증명은 쉬워!!

avg. displacement of an atom = $\alpha\sqrt{n}$ from its org. posit

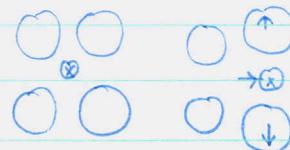
($\because n = \Gamma t$)

$$(2.5) \quad \therefore \text{after a time } t, \quad r = \alpha\sqrt{n} = \alpha\sqrt{\Gamma t} \quad \text{where } D = \frac{1}{6} \Gamma \alpha^2$$

$$(2.6) \quad = 2.4 \sqrt{Dt}$$

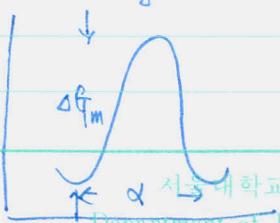
$$\therefore r \propto \sqrt{Dt}.$$

2. Effect of Temp - Thermal Activation.



- occasional violent oscillation of an int. atom - on some coincidence of the matrix & int. atom movement → results in an advance of an atom.

- factors controlling F , the effect of Temp.



$$\Delta G_m = \text{activation } E.$$

- the fraction of atoms w/ $\Delta G = \exp(-\Delta G/RT)$
(or w/ ΔG more than ΔG_m)

- for the int. atoms v freq. in x dir. successful atoms = $v \exp(-\frac{\Delta G_m}{kT})$

성장되는 next site의 출현 빈도 (jump freq.)

- for an atom in 3-D w/ Z sites to jump in.

$$(2.7) \quad \text{the jump freq. } \Gamma_B = Zv \exp\left(-\frac{\Delta G_m}{RT}\right) \text{ migration.}$$

jump freq.

$$\text{where } \Delta G_m = \Delta H_m - T\Delta S_m.$$

$$(2.8) \quad D_B = \left[\frac{1}{2} \alpha^2 Zv \exp\left(\frac{\Delta S_m}{R}\right) \right] \exp\left(-\frac{\Delta H_m}{RT}\right)$$

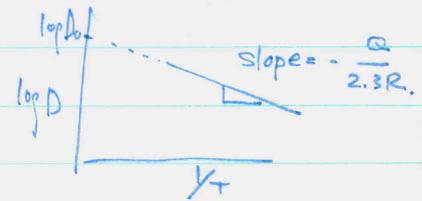
$$(2.9) \quad = D_{B0} \exp\left(-\frac{Q_{1D}}{RT}\right) \quad \text{where } \Delta H_m = Q_{1D}.$$

$$D_0 \neq f(T)$$

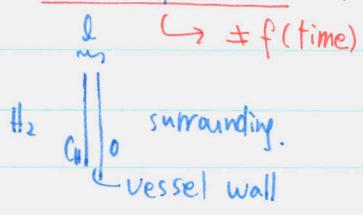
D_{B0} agree w/ exp. measurements of diff. coeff. in subst. and interstitial

- Graphic representation.

$$\log D = \log D_0 - \frac{Q}{2.3R} \left(\frac{1}{T} \right)$$



3. Steady-State Diffusion.



$$\frac{dC}{dx} = \frac{C_H - C_L}{l}$$

flux thru the wall

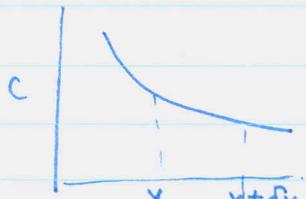
$$\therefore J_H = \frac{D_H C_H}{l}$$

$$A_H \neq f(\text{conc.})$$

S.S. is reached when the conc. everywhere reaches a const. value.

4. Nonsteady-state Diffusion

Where the conc. changes w/ time and distance, 1st law won't work.



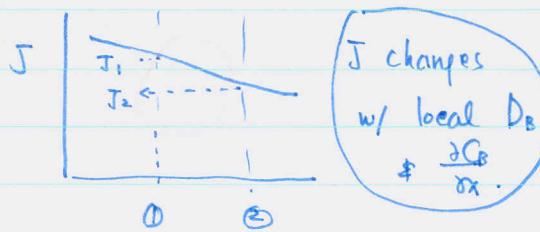
i) Mass balance at ①

in a small time Δt

[atom/cm² sec] $J_1 \Delta t$

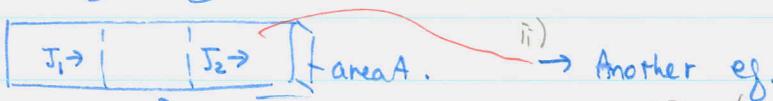
at ② $J_2 \Delta t$.

$J_1 > J_2$.



∴ the increase in conc. within the slice

$$\frac{\partial C_B}{\partial x} = \frac{(J_1 - J_2) \Delta t}{A \Delta x} \quad (2.14)$$



$$\Delta J = (J_2 - J_1) = \frac{\partial J}{\partial x} \cdot \frac{\partial x}{\partial x}$$

$$J_2 = J_1 + \frac{\partial J}{\partial x} \cdot \Delta x \quad (2.15)$$

Fig. 2.8. 2nd law.

$$\therefore \text{iii) } (2.14) \rightarrow J_2 = J_1 + \frac{\partial C_B}{\partial t} \Delta x = J_1 + \frac{\partial C_B}{\partial t} \Delta x \text{ as } \Delta t \rightarrow 0.$$

$$(2.16), (2.17) \quad \therefore \frac{\partial C_B}{\partial t} = - \frac{\partial J_B}{\partial x} = \frac{\partial}{\partial x} (D_B \frac{\partial C_B}{\partial x})$$

②/10

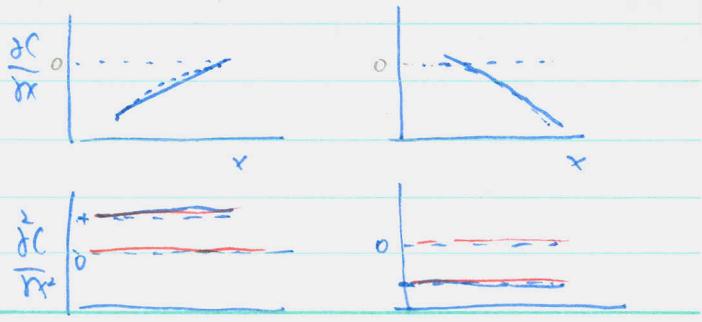
Fick's 2nd Law.

if $D_B \neq f(\text{conc.})$

$$(2.18) \quad \frac{\partial C_B}{\partial t} = D_B \frac{\partial^2 C_B}{\partial x^2}$$



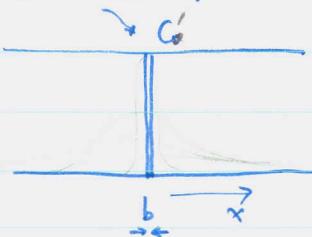
Ex. thin film or Homogenization model.



(Shewmon p15 -)

I. Diffusion in an infinite sys.

1. Thin film soln.



application of thin film \rightarrow welded \rightarrow diffuse far t .
A general soln for $\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$ is $c(x,t) = \frac{C_0}{b} \exp\left(-\frac{x^2}{4Dt}\right)$

$$c(x,t) = \frac{bC_0}{2\sqrt{\pi Dt}} \exp\left[-\frac{x^2}{4Dt}\right] \text{ 경계! (1.14)}$$

where $\sqrt{Dt} > b$.

Is it correct soln? Then show.

① satisfy $\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$?

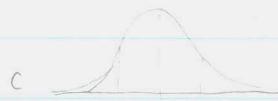
② " b.c. ?

for $|x| > 0$ $c \rightarrow 0$ as $t \rightarrow \infty$ for $x = 0$ $c \rightarrow \infty$ as $t \rightarrow 0$.

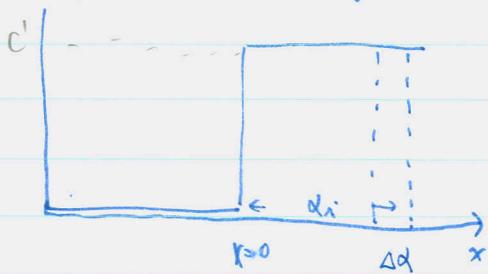
③ Yet the total quantity of solute

Charac of this soln.

$\int_{-\infty}^{\infty} c(x,t) dx = bC_0$ unit area
solute/cm³

 $x = \sqrt{4Dt}$ of the $\exp\left(-\frac{x^2}{4Dt}\right)$ c decrease as $\frac{1}{\sqrt{t}}$ from eq (1.14)dist betw $x=0$, $\frac{c}{\sqrt{t}}$ increases as \sqrt{t} .

3' Soln for a pair of semi-infi. solids. plane A & B

 $c = 0$ for $x < 0$ at $t = 0$. $c = c'$ for $x > R_0$ at $t = 0$.

Superposition of thin film soln. for many slices

$$c(x,t) \approx \frac{c'}{2\sqrt{\pi Dt}} \sum_{i=1}^n \Delta x_i \exp\left[-\frac{(x-x_i)^2}{4Dt}\right]$$

 c' [solute/cm³].unit cross section
 $\therefore c' \Delta x_i = \# \text{ of solute}$

(1) $c(x,t) = \frac{c'}{2\sqrt{\pi Dt}} \int_0^{\infty} \exp\left[-\frac{(x-u)^2}{4Dt}\right] du$, let $\frac{R-d}{\sqrt{Dt}} = u$

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(2) $= \frac{c'}{\sqrt{\pi}} \int_{-\infty}^{\infty} \exp(-u^2) du$.

⑧ B.

Date . . .

$$(1.18) \quad \text{erf}(z) = \frac{2}{\sqrt{\pi}} \int_0^z \exp(-u^2) du.$$

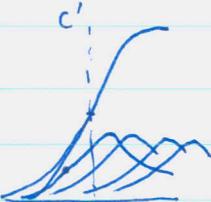
$$\text{erf}(0) = 1 \quad \text{and} \quad \text{erf}(-z) = -\text{erf}(z).$$

$$\text{erf}(-z) = -\text{erf}(z), \quad \text{erf}(\frac{1}{2}) \approx \frac{1}{2}$$

$$(1.19) \quad \therefore C(x,t) = \frac{c'}{2} \left[1 + \text{erf}\left(\frac{x}{2\sqrt{Dt}}\right) \right]$$

2) Decarburization.

i) From (1.19) $x=0$ $C = \frac{c'}{2} = \text{const}$ regardless of any t



then (1.19) is the soln for a homo alloy of solute c' held in an atmosphere which reduces the surface conc. to $\frac{c'}{2}$ and keeps it there for all $t > 0$.

B.C.

$$C = \frac{c'}{2} \quad \text{for } x=0 \quad \text{at } t>0.$$

$$C = c' \quad \text{for } x>0 \quad \text{at } t=0.$$

$$\begin{aligned} C &= A + B \text{erf}\left(\frac{x}{2\sqrt{Dt}}\right) \\ &\text{satisfies } \frac{\partial C}{\partial t} = \frac{c'}{\sqrt{\pi}} \text{ erf}\left(\frac{x}{2\sqrt{Dt}}\right) \end{aligned}$$

ii) if the surface conc. is held at $c=0$ for all $t > 0$.



B.C.

$$C = 0 \quad \text{for } x=0 \quad \text{at } t>0$$

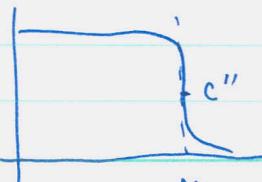
$$C = c' \quad \text{for } x>0 \quad \text{at } t=0$$

$$C = A + B \text{erf}(0)$$

$$C = A + B \text{erf}(0) \quad \therefore B = c' \quad \therefore C = c' \text{erf}(0)$$

$$\text{soln} \quad C(x,t) = c' \text{erf}\left(\frac{x}{2\sqrt{Dt}}\right)$$

iii) If the surface conc. of an initially solute-free specimen is maintained at c'' for all $t > 0$.



B.C.

$$C(x,t) = c'' \left[1 - \text{erf}\left(\frac{x}{2\sqrt{Dt}}\right) \right] \quad C'' = A + B \text{erf}(0) \quad ; C'' = c''$$

$$C = c'' \quad \text{for } x=0 \quad \text{at } t>0$$

$$C = 0 \quad \text{for } x>0 \quad \text{at } t=0.$$

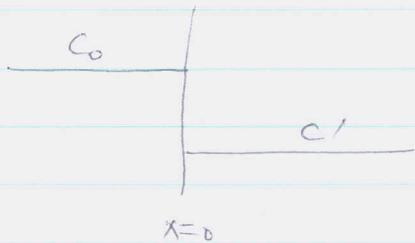
$$C = A + B \text{erf}(0)$$

IV(4) For a given boundary condition.

$$C = C_0 \quad \text{for } x > 0 \quad \text{at } t = 0$$

$$C = C' \quad \text{for } x < 0 \quad \text{at } t = 0$$

then the solution $C(x,t) - C_0 = \frac{C' - C_0}{2} \left[1 - \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right) \right]$.



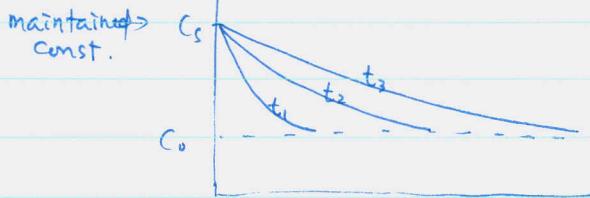
$$\begin{aligned} C(x,t) &= \frac{C' - C_0}{2} + C_0 - \frac{C' - C_0}{2} \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right) \\ &= \frac{C' + C_0}{2} - \frac{C' - C_0}{2} \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right) \end{aligned}$$

The Carburization of Steel.

1) carburization for a harder, wear-resistant surface.

using CH_4 or CO/CO_2 for $\gamma\text{-Fe}$.

• equil^m bet'n a gas mixture and surface C conc.



Analytical expression using Fick's 2nd law.

w/ B.C. $C_B = \dots$ at $x=0$

$C_B = C_0$ at $x=\infty$.

↳ the original conc.

assuming $D \neq f(\text{conc.})$ and infinitely long bar.

(2.23)

$$C = C_s - (C_s - C_0) \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right) \quad \text{where } \operatorname{erf}(z) = \frac{2}{\sqrt{\pi}} \int_0^z \exp(-y^2) dy$$

when $\operatorname{erf}\left(\frac{1}{2}\right) \approx \frac{1}{2}$ i.e. $x = \sqrt{Dt}$, $C = \frac{1}{2}(C_s + C_0)$

(2.24) i.e. $C = \frac{1}{2}(C_s + C_0)$ where $\frac{x}{2\sqrt{Dt}} \approx \frac{1}{2} \rightarrow x \approx \sqrt{Dt}$

2) Decarburization.

(2.25)

$$C = C_0 \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right)$$

why? see back

3) two semi-infinite specimens.



: if $C_1 = 0$, $C = \frac{C_2}{2} [1 + \operatorname{erf}(z)]$

$$C = \left(\frac{C_1 + C_2}{2} \right) - \left(\frac{C_1 - C_2}{2} \right) \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right)$$

why? see back

Various Cases (Shewmon p.7-19)

of 11 (iv) et 3/2

A. Thin-film soln.

II. Diffusion in Finite System - Complete Homogenization.

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2}$$

separation of variables

$$C(x,t) = X(x) T(t)$$

substitute \rightarrow Fick's 2nd law. $X \frac{dT}{dt} = DT \frac{d^2X}{dx^2}$
this into

rearrange $\frac{1}{DT} \frac{dT}{dt} = \frac{1}{X} \frac{d^2X}{dx^2} = -\lambda^2$ T, X function of λ^2
 \downarrow time fn only \downarrow to be equal λ^2 of dist.
 λ^2 only

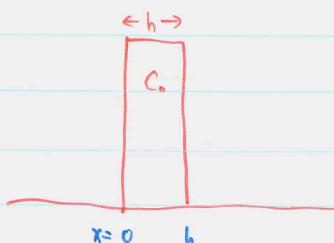
i) then $\frac{1}{T} \frac{dT}{dt} = -\lambda^2 D \quad \therefore T = T_0 \exp(-\lambda^2 D t) \quad T \rightarrow 0 \text{ as } t \rightarrow \infty$

ii) similarly $\frac{d^2X}{dx^2} + \lambda^2 X = 0. \quad \text{solut} \quad X(x) = A' \sin \lambda x + B' \cos \lambda x$

$$\therefore \text{solut} \quad C(x,t) = (A \sin \lambda x + B \cos \lambda x) \exp(-\lambda^2 D t)$$

(1-27) \therefore a pen'l sol. $C(x,t) = C_0 + \sum_{n=1}^{\infty} (A_n \sin \lambda_n x + B_n \cos \lambda_n x) \exp(-\lambda_n^2 D t)$
 $\lambda_n = \frac{\pi}{l_n}$ \Rightarrow periodicity

Diffusion out of a slab. $x=l, 2l \quad C=\bar{C} \quad \lambda_n = \frac{\pi}{l_n}$
 $\therefore C(x,t) = \bar{C} + \sum A_n \sin\left(\frac{\pi}{l_n} x\right) \exp\left(-\frac{\pi^2}{l_n^2} D t\right)$



B.C. $C=C_0 \quad \text{for } 0 < x < h \quad \text{at } t=0$

$C=0 \quad \text{for } x=h, t>0 \quad \text{at } t>0$

Eventually the conc. in the slab $\rightarrow 0 \quad \therefore A_0 = 0 \quad (1-27)$

Since $C=0$ at $t>0, x=0$. $B_n \Rightarrow 0$

$$C=0 \quad x=h \quad \sin \lambda_n h = 0 \quad \therefore \lambda_n h = n\pi$$

$$\therefore \lambda_n = \frac{n\pi}{h} \quad n \text{ any + integer.}$$

(1-28) $\therefore C_0 = \sum_{n=1}^{\infty} A_n \sin \frac{n\pi x}{h}$

multiple both sides w/ $\sin\left(\frac{p\pi x}{h}\right)$ & integrate x over the range $0 < x < h$.

$$\int_0^h C_0 \sin\left(\frac{p\pi x}{h}\right) dx = \sum A_n \int_0^h \sin\left(\frac{p\pi x}{h}\right) \sin\left(\frac{n\pi x}{h}\right) dx$$

orthogonal fn.

$$= -C_0 \frac{h}{p\pi} \cos\left(\frac{p\pi x}{h}\right) \Big|_0^h$$

$$= -C_0 \frac{h}{p\pi} [\cos(p\pi) - \cos 0] \quad \text{if } p \neq n$$

$$= \frac{h}{2} \quad \text{if } p = n$$

$$\therefore A_n = \frac{2}{h} \int_0^h C_0 \sin(nx\pi/h) dx = 0 \quad \text{for all even values of } n$$

$$= \left(\frac{2}{h}\right) \left(-\frac{hC_0}{n\pi}\right) [\cos(n\pi) - \cos 0] = \frac{4C_0}{n\pi} \quad \text{for all odd } n$$

* $\therefore A_n = A_j = \frac{4C_0}{(2j+1)\pi} \quad j = 0, 1, 2, \dots$

The soln $(x,t) = \frac{4C_0}{\pi} \sum_0^\infty \frac{1}{(2j+1)} \sin \frac{(2j+1)\pi x}{h} \exp\left(-\left[\frac{(2j+1)\pi}{h}\right]^2 dt\right)$

in a short time, only few terms useful $\because \exp(-t)$
 \rightarrow applies to the decarburization of a thin sheet of steel.

3/4

Date .

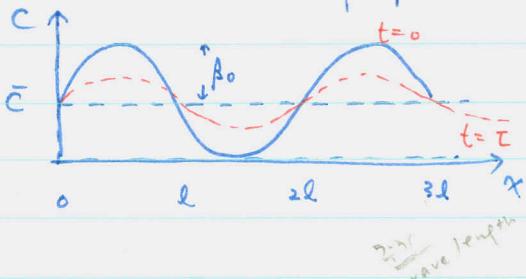
3

Homogenization

S. Solutions to the Diff. Eq. (w/ Practical Examples)

Homogenization

ex. elimination of segregation in castings.



$0 < x < l$ negative C_B decrease
 $l < x < 2l$ positive curvature
 C_B increase

$$(2.19) \quad C = \bar{C} + \beta_0 \sin \frac{\pi}{l} x \quad \text{at } t=0.$$

mean comp — the amplitude of the initial conc.

Assume $D_B \neq f(\text{conc})$.

$$(2.20) \quad \text{at } t, \quad C(t) = \bar{C} + \beta_0 \sin \frac{\pi x}{l} \exp\left(-\frac{t}{\tau}\right)$$

$$(2.21) \quad \text{where } \tau \text{ (=relaxation time)} = \frac{l^2}{\pi^2 D_B} \quad \text{see back!} \quad \leftarrow \text{find out where it's from!!}$$

at $x = \frac{l}{2}$, at time t the amplitude of the conc. profile

$$(2.22) \quad \beta = \beta_0 \exp\left(-\frac{t}{\tau}\right) \rightarrow \text{exponentially decrease.}$$

$\Leftrightarrow \text{at } t \rightarrow \infty \quad \beta \rightarrow 0 \approx \bar{C}$

- the Speed to reach \bar{C} is determined by τ

at $t = \tau$ $\beta = \beta_0/e = \beta_0/2.72$ 30% remaining
 $\Leftrightarrow t = \tau$

at $t = 2\tau$ $\beta = \beta_0/e^2 \approx \beta_0/10$ 10% "

from (2.21) the rate of homogen. ↑ rapidly as δl (wavelength of fluctuation)

→ In practice, the conc. profile will not be sinusoidal & but considered as a

Sum of infinite series of sine waves. → short wavelength decay soon.

Department of Inorganic Materials Engineering

Small influence

τ of long wavelength determine homogenization