# 재료상변태

# **Phase Transformation of Materials**

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# **Contents for previous class**

Diffusion in multiphase binary system

**Chapter 3 Crystal Interfaces and Microstructure** 

- Interfacial Free Energy
- $\rightarrow$  The Gibbs free energy of a system containing an interface of area A

 $\rightarrow \mathbf{G}_{\text{bulk}} + \mathbf{G}_{\text{interface}} \qquad \begin{array}{c} \text{vapor} \\ \text{solid} \end{array} \qquad \rightarrow \mathbf{G} = \mathbf{G}_0 + \gamma \mathbf{A}$ 

Solid/Vapor Interfaces

Origin of the surface free energy?  $\rightarrow$  Broken Bonds Equilibrium shape  $\rightarrow$  Wulff surface

## Surface energy for high or irrational {hkl} index

A crystal plane at an angle  $\theta$  to the close-packed plane will contain broken bonds in excess of the close-packed plane due to the atoms at the steps.



(cosθ/a)(1/a) : broken bonds from the atoms on the steps
 (sin|θ|/a)(1/a): additional broken bonds from the atoms on the steps

# **Contents for today's class**

• Boundaries in Single-Phase Solids

(a) Low-Angle and High-Angle Boundaries

(b) Special High-Angle Grain Boundaries

(c) Equilibrium in Polycrystalline Materials

## **Boundaries in Single-Phase Solids**

## grain boundary





(hkl)

 $\rightarrow$ 



1) misorientation of lattice in two grains

2) orientation of grain boundary

Single phase Poly grain





## **3.3 Boundaries in Single-Phase Solids**



symmetric tilt or twist boundary non-symmetric tilt or twist boundary

# 3.3.1 Low-Angle and High-Angle Boundaries

## **Low-Angle Boundaries**

(a)

## Symmetrical low-angle tilt boundary



## Symmetrical low-angle twist boundary



Fig. 3. 7 (a) Low-angle tilt boundary, (b) low-angle twist boundary: ○ atoms in crystal below, • atoms in crystal above boundary. (After W.T. Read Jr., *Dislocations in crystals*, McGraw-Hill, New York, 1953.)

#### tilt Boundaries



Figure 1 - 23° symmetric tilt boundary about a <001> axis. The circles with dashed lines represent one layer and the circles with solid lines the other layer of the AB....stacked {001} planes. The atoms labelled A and B denote the structural unit.

Figure 2 - 23° symmetric tilt boundary about a <001> axis.  $\Delta$ represent one layer and 0 represent the other layer of the AB.... stacked {001} planes. The ledge like character of the boundary is shown by the dashed lines.



## Dislocations



#### twist Boundaries





**Figure 2.** A screw dislocation; note the screw-like 'slip' of atoms in the upper part of the lattice



### **Screw dislocation**



### **Growth of Screw dislocation**



#### twist Boundaries





45°, (010) symmetric twist GB



Asymmetric GB with tilt and twist components

#### **Non-symmetric Tilt Boundary**



Fig. 3.8 An unsymmetric tilt boundary, Dislocations with two different Burgers vectors are present. (After W.T. Read Jr., Dislocations in Crystals, McGraw-Hill, New York, 1953.)

### If the boundary is unsymmetrical, dislocations with different Burgers vectors are required to accommodate the misfit.

In general boundaries of a mixture of the tilt and twist type,  $\rightarrow$  several sets of different edges and screw dislocations.

3.3.1 Low-Angle and High-Angle Boundaries

Low-Angle tilt Boundaries



→ around edge dislocation : strain ↑
but, LATB ~ almost perfect matching



(a)

3.3.1 Low-Angle and High-Angle Boundaries



- **Low-Angle tilt Boundaries**  $\rightarrow$  around edge dislocation : strain  $\uparrow$ but, LATB ~ almost perfect matching
  - $\rightarrow$  g.b. energy :  $\gamma_{g.b.} \rightarrow$  E /unit area (energy induced from dis.)
    - \* Relation between D and  $\gamma$ ?

 $\sin\theta = b/D$ , at low angle

- $\rightarrow$  D=b/ $\theta$   $\rightarrow$   $\gamma_{q.b.}$  is proportional to 1/D
- $\rightarrow$  low angle tilt boundary
- $\rightarrow$  Density of edge dis.

(cf. low angle twist boundary  $\rightarrow$  screw dis.)

(a)

#### **Low-Angle tilt Boundaries**

 $\implies$  As  $\theta$  increases,  $\gamma_{g.b.} \uparrow$ 



 $\rightarrow \gamma_{g.b.}$  increases and the increasing rate of  $\gamma$  (=d  $\gamma$ /d  $\theta$ ) decreases.

 $\rightarrow$  if  $\theta$  increases further, it is impossible to physically identify the individual dislocations



#### Soap Bubble Model



Fig. 3.11 Rafts of soap bubbles showing several grains of varying misorientation. Note that the boundary with the smallest misorientation is made up of a row of dislocations, whereas the high-angle boundaries have a disordered structure in which individual dislocations cannot be identified. (After P.G. Shewmon, *Transformations in Metals*, McGraw-Hill, New York, 1969, from C.S. Smith.)

#### **High Angle Grain Boundary**



Fig. 3.10 Disordered grain boundary structure (schematic).

High angle boundaries contain large areas of poor fit and have a relatively open structure.

 $\rightarrow$  high energy, high diffusivity, high mobility (?) (cf. segregated gb)

#### **High Angle Grain Boundary**

Low angle boundary

 $\rightarrow$  almost perfect matching (except dislocation part)

High angle boundary (almost)

 $\rightarrow$  open structure, large free volume

\* low and high angle boundary

high angle  $\gamma_{g.b.} \approx 1/3 \gamma_{S/V.} \rightarrow Broken Bonds$ 

Crystal	$\gamma_b/mJ\ m^{-2}$	T/°C	$\gamma_b/\gamma_{sv}$	
Sn	164	223	0.24	
Al	324	450	0.30	
Ag	375	950	0.33	
Au	378	1000	0.27	
Cu	625	925	0.36	
γ-Fe	756	1350	0.40	
δ-Fe	468	1450	0.23	
Pt	660	1300	0.29	
W	1080	2000	0.41	

#### Measured high-angle grain boundary energies

#### **Special High-Angle Grain Boundaries**

: high angle boundary but with low  $\gamma_{\text{g.b.}}$ 



## **Twin boundary**





## **Twin boundary**



## **Twin boundary**

#### **Special High-Angle Grain Boundaries**

(c) Twin boundary energy as a function of the grain boundary orientation



 Table 3.3 Measured Boundary Free Energies for Crystals in Twin Relationships

 (Units mJ/m<sup>2</sup>)

Crystal	Coherent twin boundary energy	Incoherent twin boundary energy	Grain boundary energy
Cu Ag Fe-Cr-Ni (stainless steel type 304)	21 8 19	<< 498 126 209	< 623 377 835

#### **Special High-Angle Grain Boundaries**



Fig. 3.13 Measured grain boundary energies for symmetric tilt boundaries in Al (a) When the rotation axis is parallel to <100>, (b) when the rotation axis is parallel to <110>. (After G. Hasson and C. Goux, Scripta Metallurgica, 5 (1971) 889.)

#### Why are there cusps in Fig. 3.13 (b)?



Fig. 3. 14 Special grain boundary. (After H. Gleiter, Physica Status Solidi (b) 45 (1971) 9.)

#### **Equilibrium in Polycrystalline Materials**



Fig. 3.15 Microstructure of an annealed crystal of austenitic stainless steel. (After P.G. Shewmon, Transformations in Metals, McGraw-Hill, New York, 1969)

#### **Poly grain material**



Fig. 3.15 Microstructure of an annealed crystal of austenitic stainless steel. (After P.G. Shewmon, *Transformations in Metals*, McGraw-Hill, New York, 1969.)

- →  $\gamma_{g.b.}$  → metastable equilibrium (equil. (no g.b.))
- $\rightarrow \gamma_{\text{g.b}}~$  is minimum
- $\rightarrow$  force acting on junction of

g.b. segment



P is moved at a small distance( $\delta y$ )

- A. work done by  $: F_v \delta y$
- B. increase boundary energy caused

by the change in orientation  $\delta\theta \sim l (d\gamma/d\theta) \delta\theta$ 

 $F_y \delta y = l (dγ/dθ) \delta θ$ →  $F_y = dγ/dθ$  torque force

 $\rightarrow$  segment of g.b. moves to low energy position





\* general high angle boundary :  $d\gamma/d\theta \approx 0$ 

 $\rightarrow$  consider more simply



Measurement of grain-boundary energy using g.b groove profile annealing a specimen at a high temperature and then measure the angle at the intersection of the surface with the boundary

If the solid-vapor energy  $(\gamma_{S/V})$  is the same for both grains,

$$2\gamma_{SV}\cos\frac{\theta}{2} = \gamma_b$$



Fig. 3. 18 The balance of surface and grain boundary tensions at the intersection of a grain boundary with a free surface.

#### **3.3.4. Thermally Activated Migration of Grain Boundaries**

If the boundary is curved in the shape of cylinder, Fig. 3.20a, it is acted on by a force of magnitude  $\gamma$ /r towards its center of curvature.

Therefore, the only way the boundary tension forces can balance in three dimensions is if the boundary is planar ( $r = \infty$ ) or if it is curved with equal radii in opposite directions, Fig. 3.20b and c.



Fig. 3.20 (a) A cylindrical boundary with a radius of curvature r is acted on by a force  $\gamma/r$ . (b) A planar boundary with no net force. (c) A doubly curved boundary with no net force.

#### **Direction of Grain Boundary Migration during Grain Growth**

For isotropic grain boundary energy in two dimensions, Equilibrium angle at each boundary junction?  $\rightarrow 120^{\circ}$ 



Fig. 3.21 Two-dimensional grain boundary configurations. The arrows in

#### Equilibrium angle at each boundary junction in $3D? \rightarrow 109^{\circ}28'$

#### Grain Growth (Soap Bubble Model)



Fig. 3.22 Two-dimensional cells of a soap solution illustration the process of grain growth. Numbers are time in minutes. (After C.S. Smith, Metal Interfaces, American Society for Metals, 1952, p. 81.)

**Example of Grain Growth simulation in 3D** 



결실의 계절 가을에 중간고사 준비 열심히 해서 모두 좋은 결과 있기를 ...^^