

2009 spring

***Microstructural Characterization
of
Materials***

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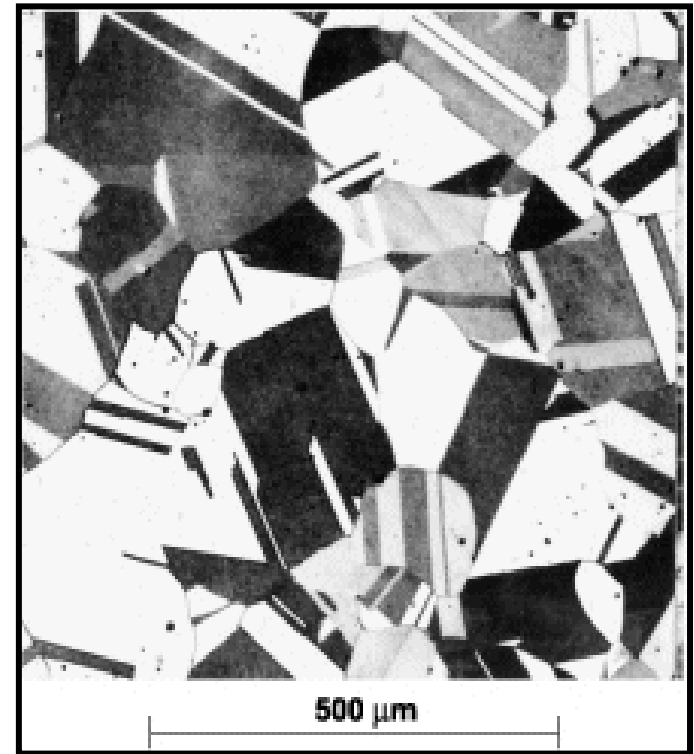
Contents for previous class

- **Presentation for various optical microscopes**
 - Phase contrast microscope
 - Differential interference contrast microscope
 - Fluorescence Microscopy
 - Multi-photon laser scanning microscope
 - 벡터장 나노 현미경
 - Digital Holographic Microscope

Sample Preparation

What is Metallography??

- Metallography is the science and art of preparing a surface for analysis by **grinding**, **polishing**, and **etching** to show microstructural component. (mainly metallic materials)
- Metallography does not only apply to metallic materials, such technology can also be used examine ceramics, polymers and semiconductors.



General Sample Preparation

- 1. Sectioning – typically abrasive cutting
- 2. Mounting (optional)
- 3. Coarse grinding
- 4. Fine grinding
 - 3. and 4. Grinding sequence – 120, 240, 320, 400, 600 grit SiC rotating 45 or 90 degrees between steps (center to edge)
- 5. Coarse polishing – 6 micron diamond paste or 6 micron and 1 micron diamond. Rotate sample counter to the wheel rotation. Wash after each step under running water and rinse with alcohol
- 6. Fine polishing – 0.3 and 0.05 micron alumina slurry

(1) Sectioning

Why sectioning?

1. Size limitation of specimen to be examined under optical microscope.
2. Locate area needs to be selected from a large sample.

Sectioning a sample – must be careful not to significantly alter the microstructure during sectioning.

- **Fracturing**
- **Shearing**: substantial damage which must be subsequently ground off
- **Sawing**: HB<350 can typically be cut by sawing. Produces roughness and heat and must be subsequently ground to remove damage.
- **Abrasive Cutting**: (most common) thin rotating disk of a suitable abrasive in a supporting media is used.

Sample Preparation

- Abrasive Cutting is the most common sectioning method.



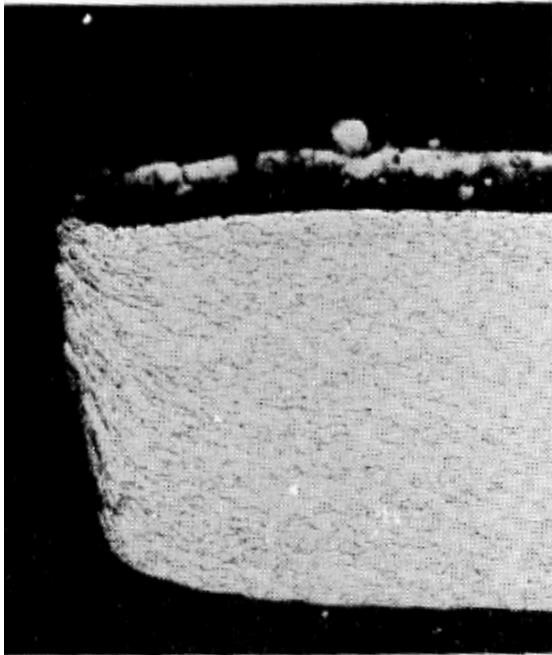
Band saw



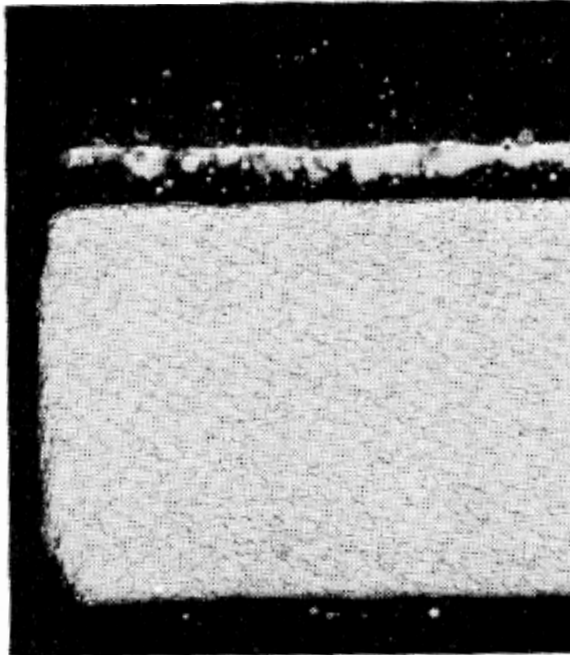
Abrasive cutter



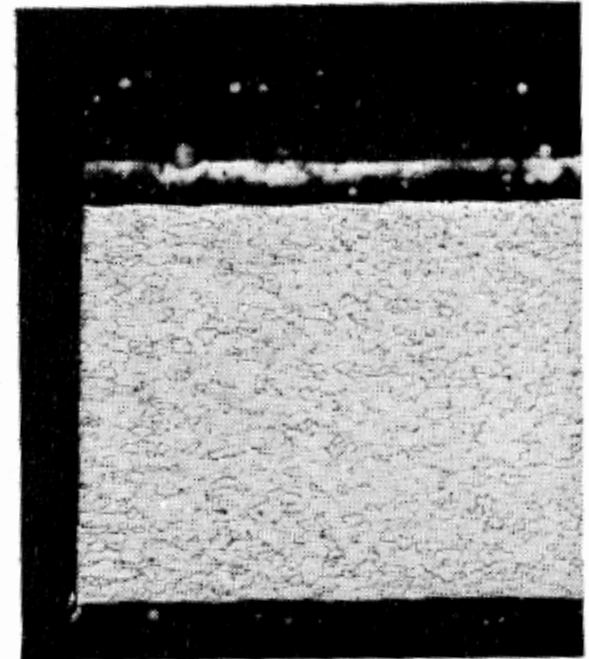
Sectioning Artifacts



shear



Band saw



Abrasive cutoff

Influence of cutting procedure on deformation and damage to porcelain-enameled Steel, 45x, 2% nital, polarized light. (courtesy of A. O. Bescoter, Bethlehem Steel Corp.)

Sample Preparation

Electric Discharge Machining (EDM)

- Electrically conductive materials can be EDMd.
- Cutting is accomplished by an electric discharge between an electrode and the sample submerged in a dielectric fluid.



<http://drm.kist.re.kr/cerapedia/process/jgl007.html>

Sample Preparation

Microtomy:

- Useful for preparing soft materials such as polymer samples.
- Steel, glass or diamond knives in a microtome can cut samples into very thin sections



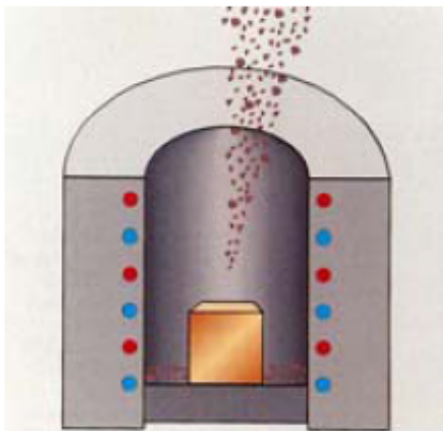
Sample Preparation

(1) Mounting

Required when (1) the sample is small or too oddly shaped to be handled. (2) The sample edge area needs to be examined

Thermal Mounting:

The sample is encased in thermosetting plastics at high temperatures and pressure



Sample Preparation Techniques


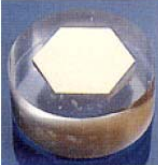



Cold Mounting:

The sample is encased in epoxy type materials




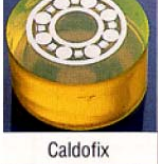



Sample Preparation Techniques

Hot Mounting Materials

Name	Features		Material
	Application	Specific property	
Resin 1 	Electropolishing	Electrically conductive Low shrinkage	Acrylic Thermoplastic
Resin 3 	Transparent mounts Porous material	Transparent Low shrinkage	Acrylic Thermoplastic
Resin 5 	Edge retention Planeness For highest requirements	Hard Good adhesion Wear resistant No shrinkage	Epoxy Thermosetting
Resin 6 	Serial mounting	Medium shrinkage	Bakelite Thermosetting
Pre-Mounts 	Serial mounting of un-complicated shapes	Easy to handle Medium shrinkage	Bakelite Thermosetting

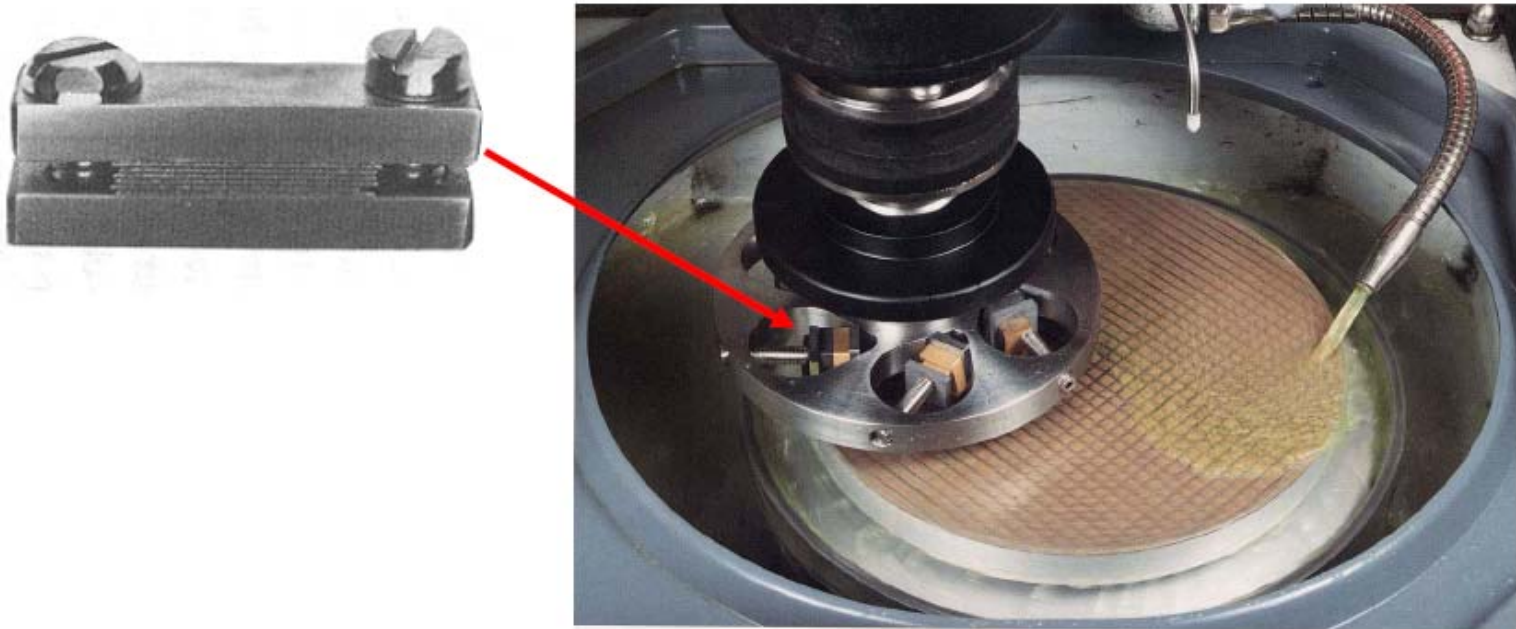
Cold Mounting Materials

Name	Features		Material/filler	Moulds
	Application	Specific property	Curing time	
Citofix 	Serial mounting Irregularly shaped specimens	Low viscosity Good adhesion Translucent Low shrinkage	Acrylic 7 - 10 min	Epoform Flexiform Seriform Monoform
Durofix 	Serial mounting Edge retention Irregularly shaped specimens	Low viscosity Hard Wear resistant Low shrinkage	Acrylic Mineral fillers 15 min	Epoform Flexiform Seriform Monoform
Triofix-2 	Edge retention Planeness	Good adhesion Very hard Wear resistant Very low shrinkage	Polyester/ Acrylic/ Mineral filler 15 - 18 min	Epoform Flexiform Monoform
Epofix 	Vacuum impregnation Porous samples Mineralogical samples	Low vapour-pressure Transparent Good adhesion Low viscosity No shrinkage	Epoxy 8 h	Epoform
Caldofix 	Vacuum impregnation Porous samples Mineralogical samples	Low vapour-pressure Transparent Good adhesion Low viscosity Very low shrinkage	Epoxy 1 h at 80°C	Epoform

Sample Preparation Techniques

Adhesive Mounting: The sample is glued to a piece of a large holder

Clamps: The sample is fixed in mechanical clamping devices



Sample Preparation

(3) Grinding

1. Grinding removes the damage from the surface produced by sectioning.
2. Grinding also produced damage which must be minimized so that subsequent grinding with finer abrasives.
3. At the end of grinding phase, the only grinding damage present must be from the last grinding step.
4. Such damage will be removed by polishing.

Grinding Materials: Abrasive paper (covered with silicon carbide grit).
Commonly a series of abrasive papers are used from coarse to fine.

Grit Sequence: 120-, 240-, 320-, 400-, 600-, 1200, 2400, and etc.

What is grit?

- When talking about sandpaper, "**grit**" is a reference to the number of abrasive particles per inch of sandpaper.
- The lower the grit the rougher the sandpaper and conversely, the higher the grit number the smoother the sandpaper.
- This make sense if you imagine how small the particles on an 800-grit sandpaper would need to be to fit into a 1" square.
- Sandpaper is referred to by the size of its grit (i.e. 150-grit sandpaper).



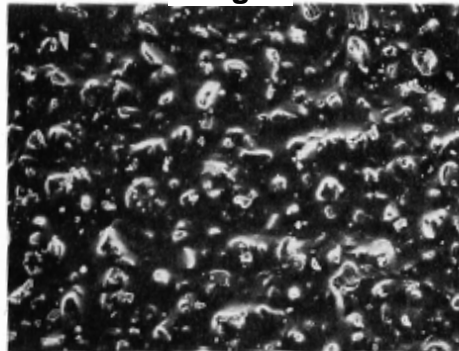
Grinding Media – SiC paper



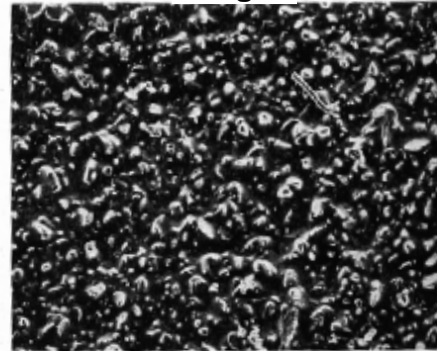
120 grit



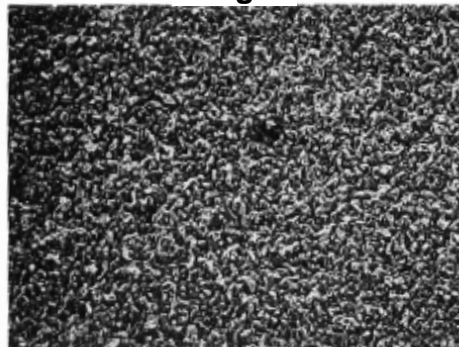
240 grit



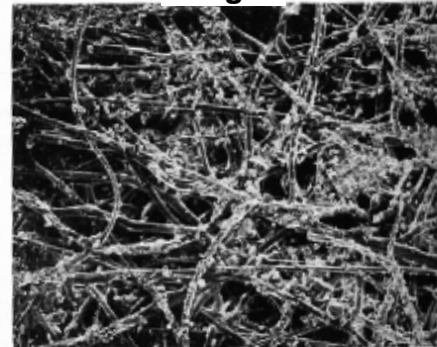
320 grit



400 grit



600 grit



600 soft

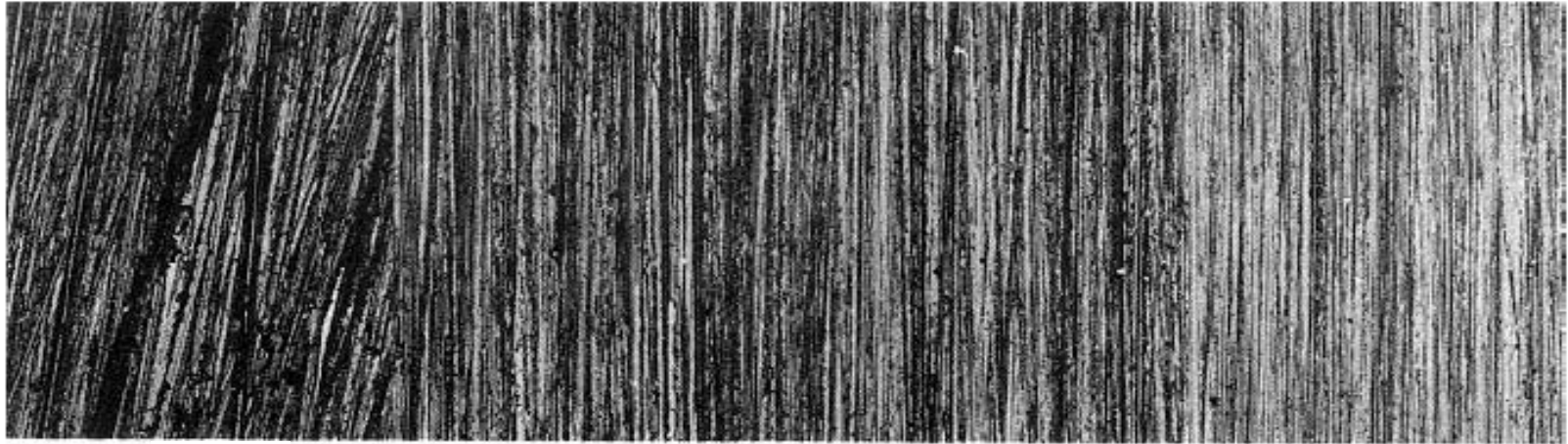
Appearance of the surface of silicon carbide grinding paper (Buchler Carbiment). 60 x

Sample Preparation Techniques

- Grinding procedure – each step typically 1-2 minutes. Rotate specimen 45-90 degrees after each polishing step. Rinse between steps to remove previous grinding media
- Grinding media
 - Silicon carbide SiC (mohs hardness 9.5) most common (high hardness and low cost)
 - Alumina (Al₂O₃) (mohs 9.1)
 - Emory (Al₂O₃ and iron oxide) (mohs 8.0) smoother – only good for dry grinding – not used much because wet grinding is preferred.
- Equipment
 - Polishing wheels – 8-12 inch diameter, moderate pressure moving sample center to edge. Rotation speed ~ 300-600rpm
 - Automatic grinding wheels – more reproducible



Sample Preparation Techniques

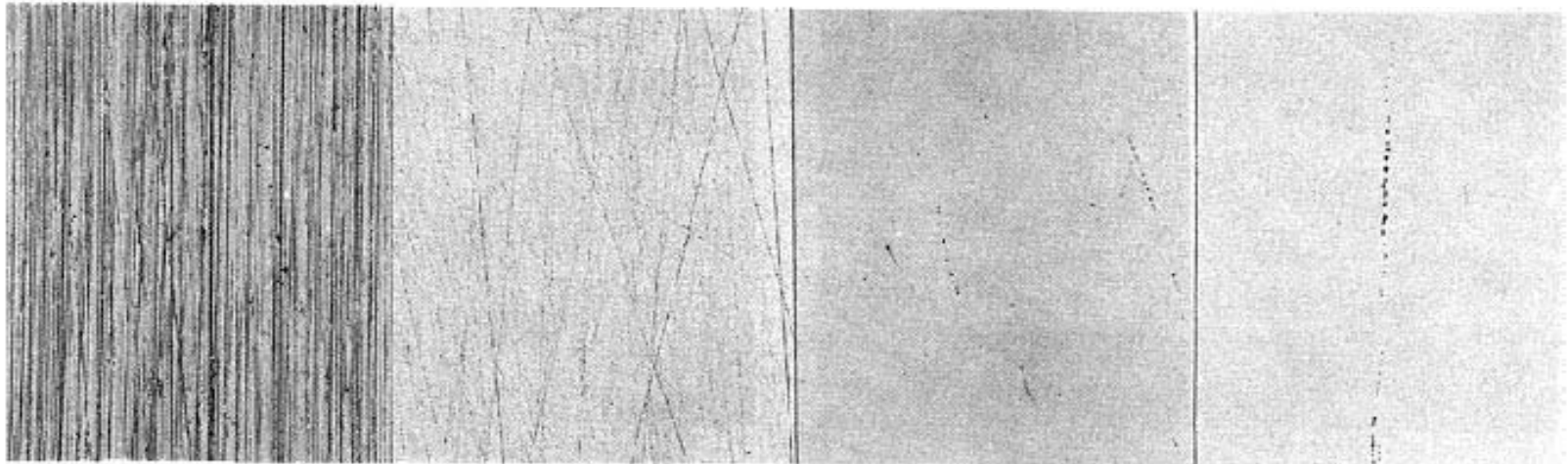


120 grit

240 grit

320 grit

400 grit



600 grit

6- μm diamond

1- μm diamond

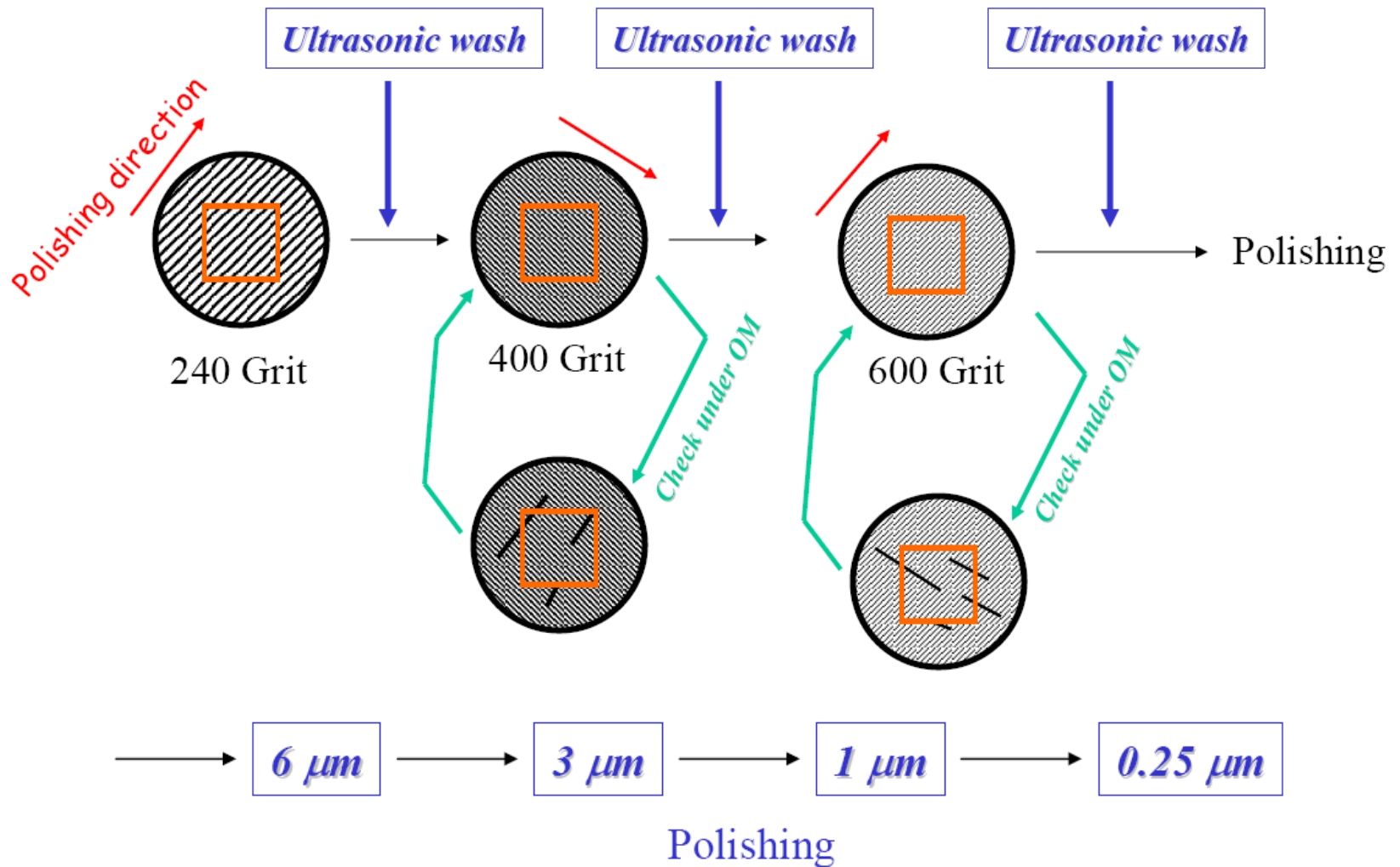
colloidal silica

Sample Preparation Techniques

(4) Polishing

- Lapping – preparing surface using a disc surface impregnated with abrasive particles. Rotate sample in circular pattern counter to the wheel rotation
- Polishing
 - Coarse – 30 to 3 micron abrasives
 - Fine – typically < 1 micron
 - Procedure – careful cleaning between polish steps is critical to minimize carryover of larger abrasive particles to smaller abrasives
 - Sample orientation should not be held constant – continuously change moving sample from center to edge in a circular pattern counter to the wheel rotation
 - Rough polish – 150-600rpm ~ 6micron diamond
 - Finle polish – 1 micron diamond then 0.3 and 0.05 micron alumina

Sample Preparation Techniques



Sample Preparation Techniques

Polishing Continued

- Polishing cloths – must hold abrasives and must not contain foreign particles
- Grinding and polishing theory – hard abrasive particles scratch grooves, allow metal removal, and produce a plastically deformed surface region
 - Both grinding and polishing produce these three artifacts, however the extent is different depending on the pressure and particle size
 - The main difference between grinding and polishing is the rigidity of the grinding abrasive for grinding relative to the elasticity of the polishing media. This results in a lower contact pressure for polishing

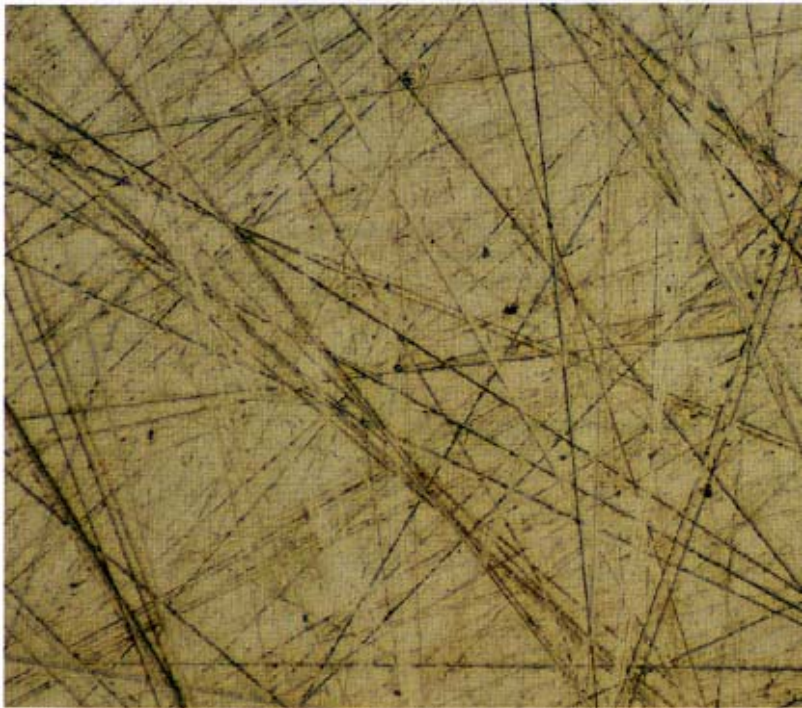
Sample Preparation Techniques



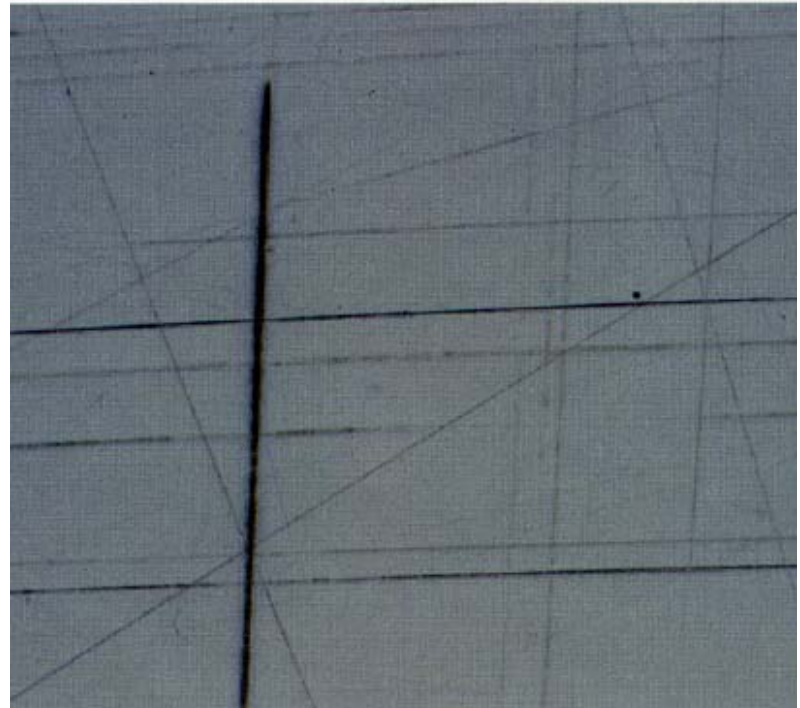
Sample Preparation Techniques

Artifact structure from improper grinding

Surface deformation from improper grinding should be avoided, otherwise the microstructure may be obscured as shown below.



After FG, scratches from PG are still visible.
Mag: 200x



After diamond polishing, scratches from FG are still remain.
The very deep vertical scratch might even be left over from PG.
Mag: 200x

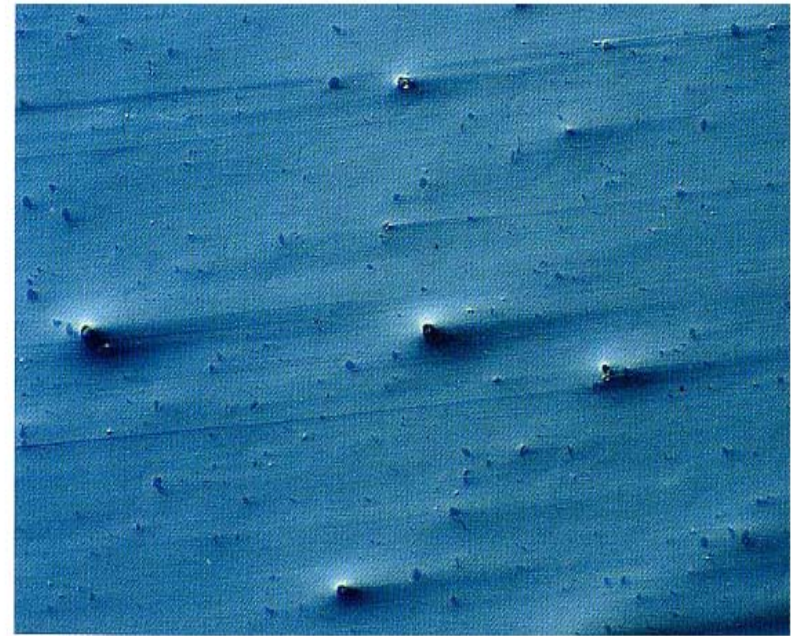
Sample Preparation Techniques

Artifact structure from improper polishing

Polishing should produce a scratch-free surface. Excessive pressure may cause artifact of second phase particles as shown below.



20 x



200 x

Polishing Continued

- **Electromechanical polishing** – add an electrolytic etch to the mechanical polish (requires dc or ac power)
- **Attack polishing** – add a dilute chemical etchant to the polishing media to facilitate the mechanical abrasive with a chemical etch.
 - Can reduce or eliminate surface damage because it is a much gentler process.
 - Etch chemicals can damage equipment and person must take necessary precautions when handling the solutions
- **Chemical polishing** – purely chemical etch “a controlled corrosion process”
 - Put sample in a corrosive media and stir rigorously for uniform material removal

Sample Preparation Techniques

- **Electro polishing** – grind to 600 grit or mesh then the sample is made the anode of a electrolytic cell (+)
 - The recipe must have an appropriate (electrolyte, temperature, current, voltage, and time.
 - Advantages – easy to minimize surface damage
 - Disadvantage – dangerous chemicals, some phases in multiphase materials preferentially electrochemically etch so non-uniform polish results.

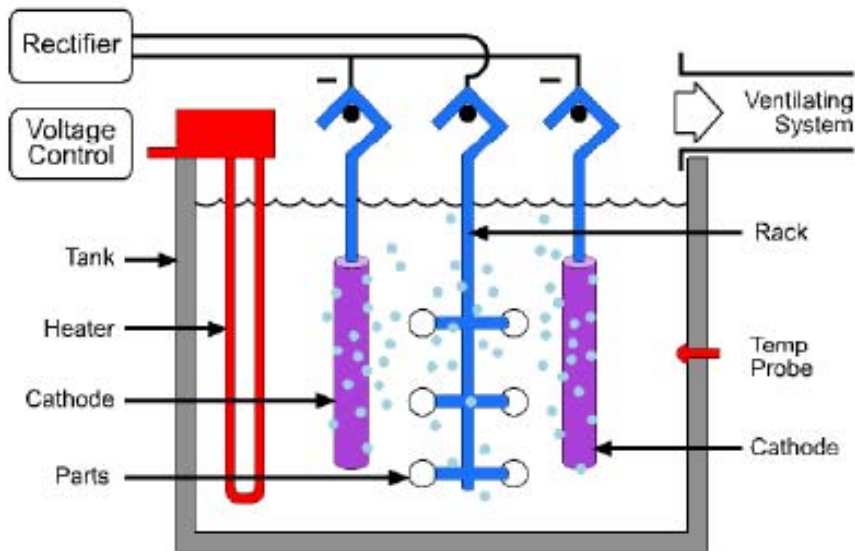


Figure 1: Typical Electropolishing Installation

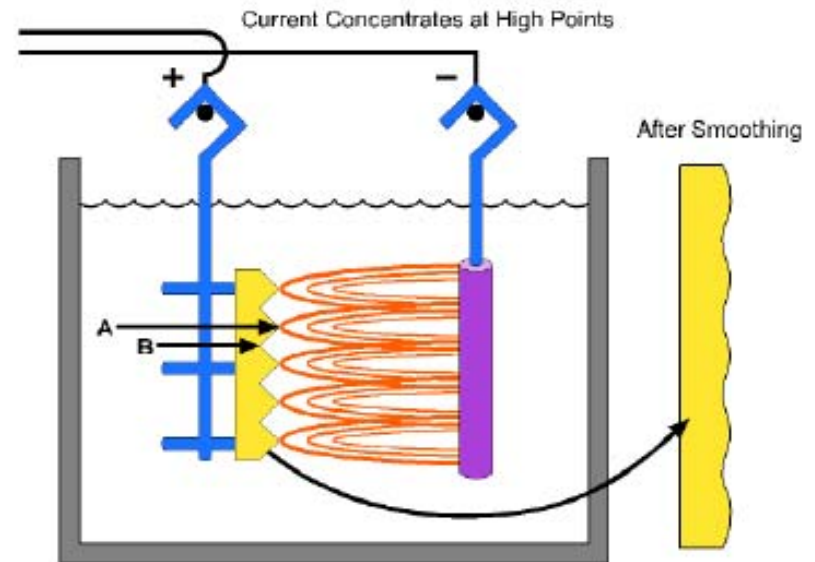


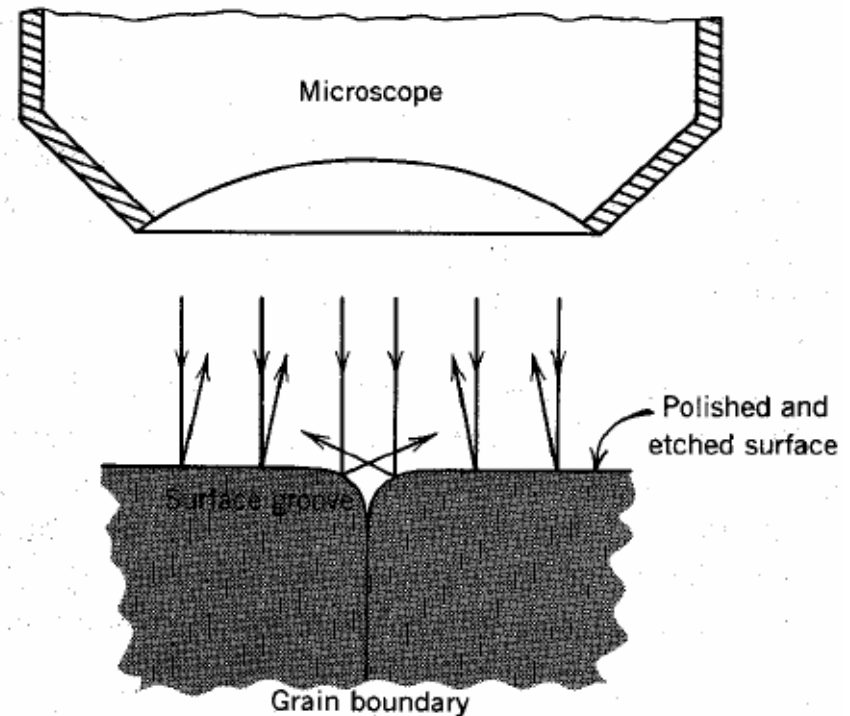
Figure 2: Smoothing by Electropolishing

Sample Preparation Techniques

(5) Etching

Using chemical to dissolve selectively the surface of materials in order to reveal the inhomogeneous nature in microscopic scale.

For example the grain boundaries of polycrystalline metal



Sample Preparation Techniques

etching

Etching is basically a controlled corrosion process resulting from electrolytic action between surface area of different potential.

Electrolytic activity results from local physical or chemical heterogeneities which render some features anodic and others cathodic under the specific etching conditions.

Chemical Etchants produce contrast by

- Crystal faceting
- Selective phase dissolution.

Common chemical etchants have three components:

- A corrosive agent (acids)
- A modifier (alcohol, glycerin...)
- An oxidizer (hydrogen peroxide, Fe^{3+} , Cu^{2+} ...)

Sample Preparation Techniques

Common etchants for metals (for example)

Keller's reagent

2.5 ml HNO₃/1.5 ml HCl

1.0 ml HF, 95 ml water

for Al and alloys

Nital

1-10 ml HNO₃ in 90-99 ml methanol

for Fe and steel

Picral

4 -10 g picric acid, 100 ml ethanol

for Fe and steel

10 ml HF/5 ml HNO₃ 85 ml water

for Ti and alloys

NH₄OH/3% H₂O₂

for Cu and alloys

Sample Preparation Techniques

Etchants for polymers (for example)

Aqueous solution of CrO_3 for bulk polypropylene (PP)
Aqueous solution of $\text{H}_2\text{SO}_4/\text{H}_3\text{PO}_4/\text{CrO}_3$ for ABS, HIPS and PPO

Etchants for minerals (for example)

Concentrated HCl for CaO, or MgO.
Aqueous solution of H_3PO_4 for Al_2O_3
Aqueous solution of HNO_3/HF for CeO_2 , SrTiO_3 , Al_2O_3 , and ZrO-ZrC.

Sample Preparation Techniques

Safety Instructions

- Optical Metallography involves the use of etchants (standard solutions containing a variety of chemicals such as strong acids and solvents) which can be very corrosive and poisonous.



Sample Preparation Techniques

Safety



- You must wear gloves and goggles and handle the chemicals with extreme care.
- **You must not have direct skin or eye contact with the etchants. Etchants must be used in well ventilated area.**
- **Do not try to smell the etchants**
▪

