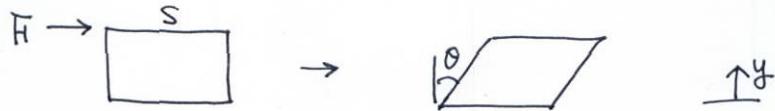


Chap. 2 What is a fluid and how much so?

Three phases : solid / liquid / $\underbrace{\text{gas}}_{\text{fluid}}$



solids : resist shear deformation

$$\frac{F}{S} = G \dot{\theta} \quad : \text{Hookean solid}$$

G : shear modulus

fluids : cannot resist shear deformation

$$\frac{F}{S} = \mu \dot{\theta} \quad : \text{Newtonian fluid}$$

$$\dot{\theta} = \frac{du}{dx} \quad : \text{shear rate } [1/s] \quad (= \frac{du}{dy})$$

μ : (dynamic) viscosity [Pa·s]

* Deborah number

$$De = \frac{\text{observation time}}{\text{relaxation time}} \\ = \frac{\text{time a process takes}}{\text{time for significant plastic deformation}}$$

$\ll 1$ ~ solid

$\gg 1$ ~ fluid

In this course we shall assume (mostly) fluids are Newtonian, continua, incompressible flows are steady.

§ Inverse viscosity - temperature relationship



- sinking rate \uparrow as $T \uparrow$: shape change
to reduce sinking rate

Fig. 2.5 (water flea)

- as $T \uparrow - \mu \downarrow - \text{drag} \downarrow$: whale, tuna
release body heat ?

internal circulation

