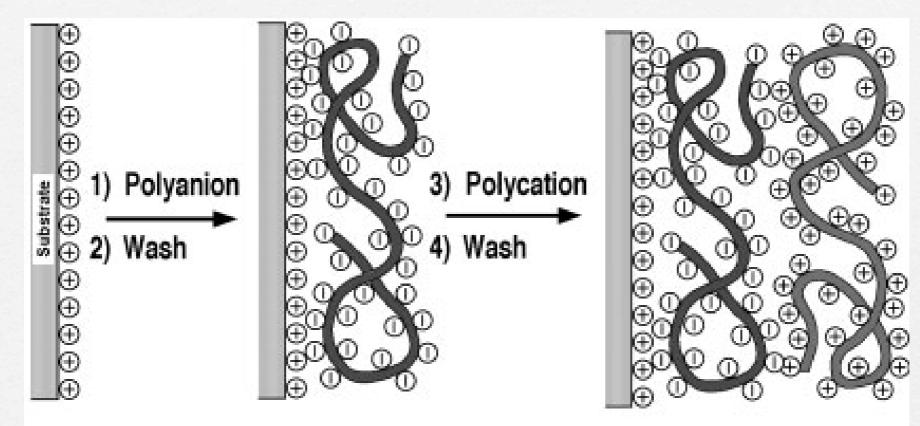
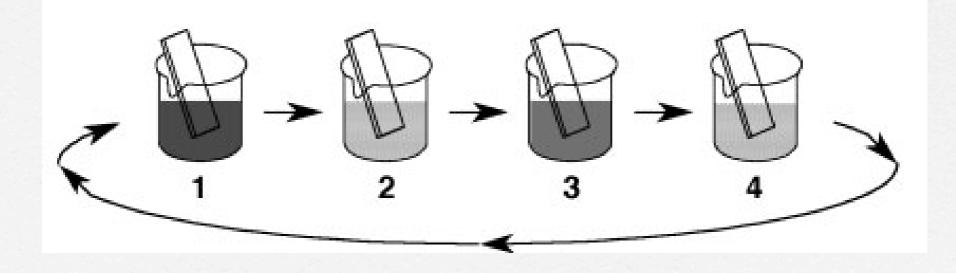
#### polymer multilayers

# layer-by-layer (LbL) adsorption





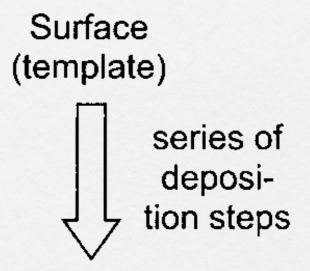
# layer-by-layer (LbL) adsorption

#### **Classic Synthesis**

Reagent(s) (atoms, synthons)

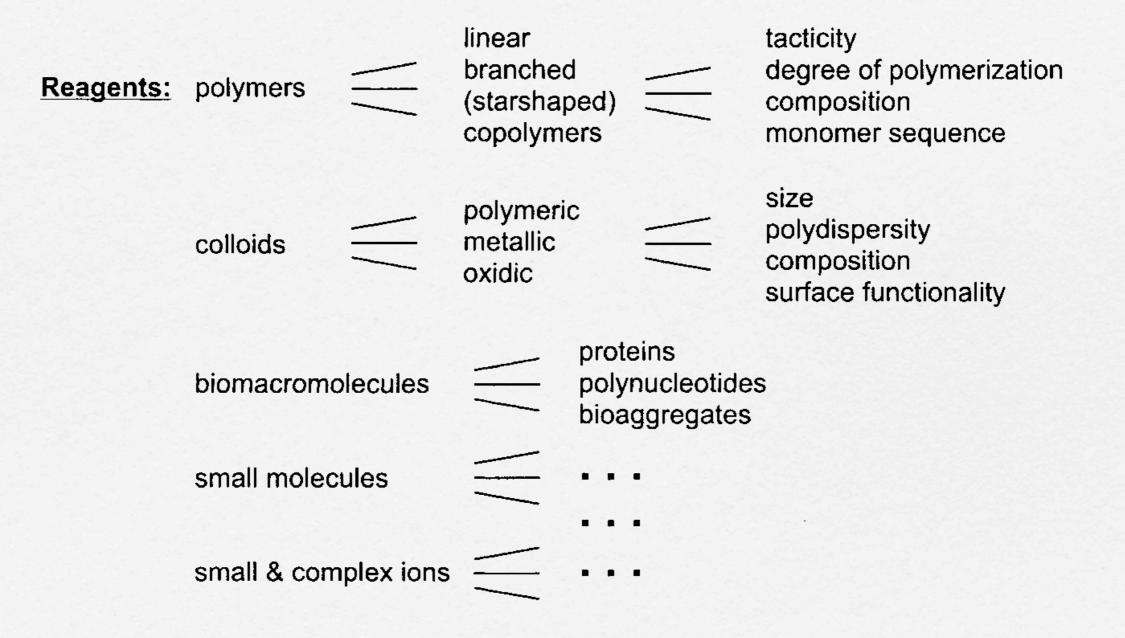
> series of reaction steps

Product(s) (typically single species) **LbL - Deposition** 

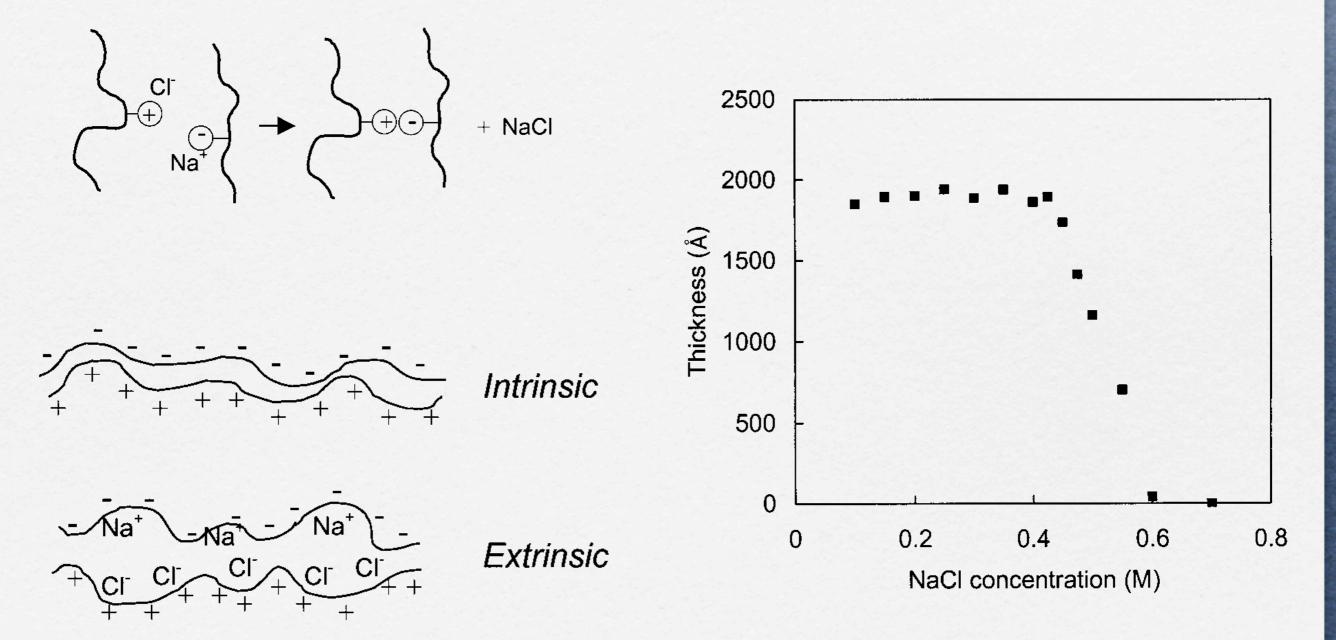


Multilayer Film (defined layer sequence)

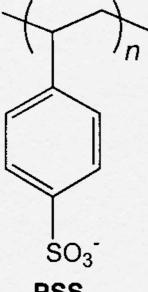
# reagents for LbL adsorption

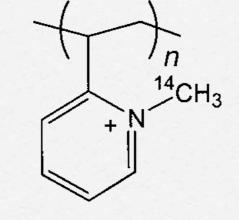


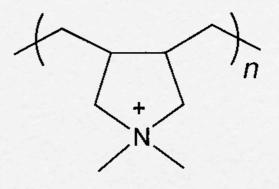
# driving force for LbL



# polyelectrolytes for LbL



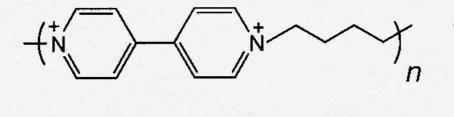


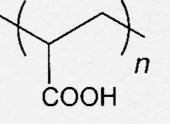


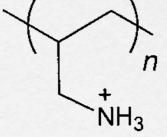
PSS



**PDADMA** 





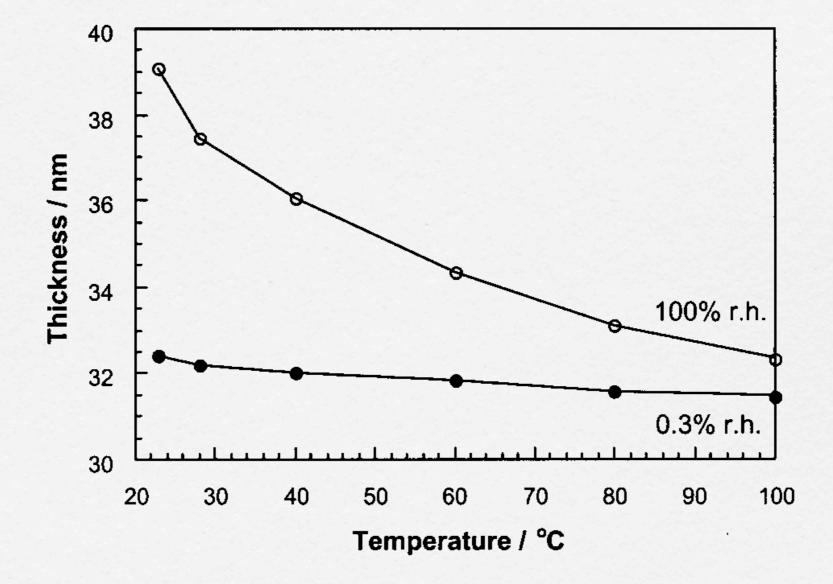


**PBV** 

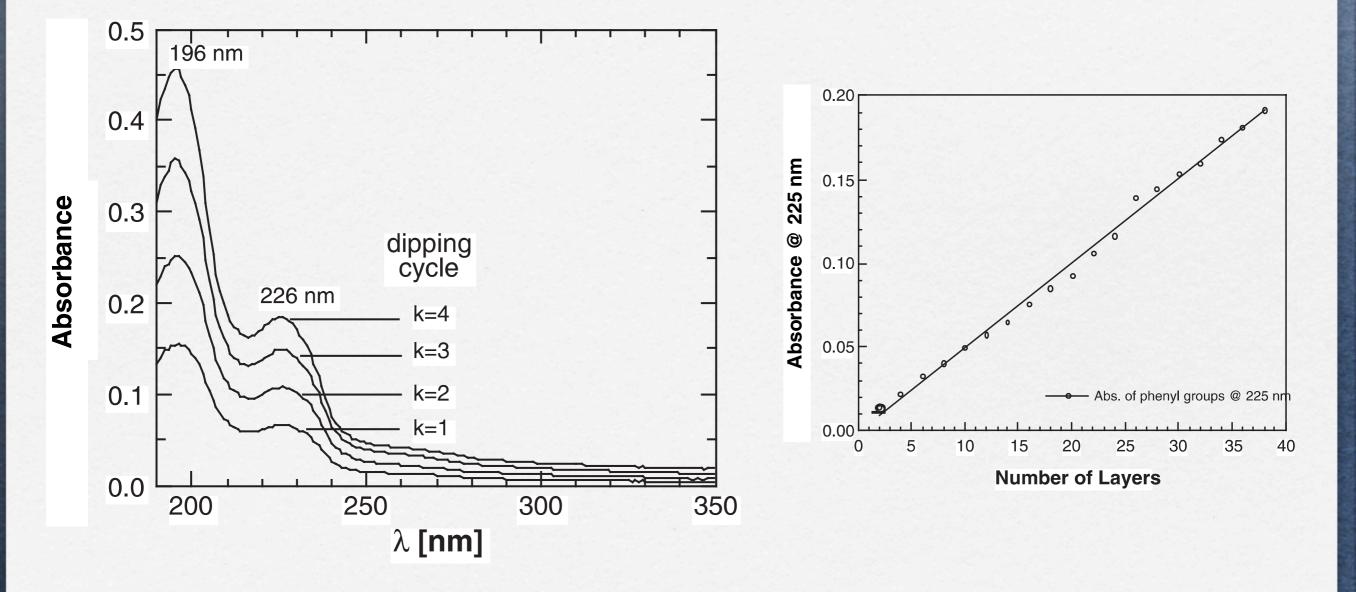
PAA

PAH

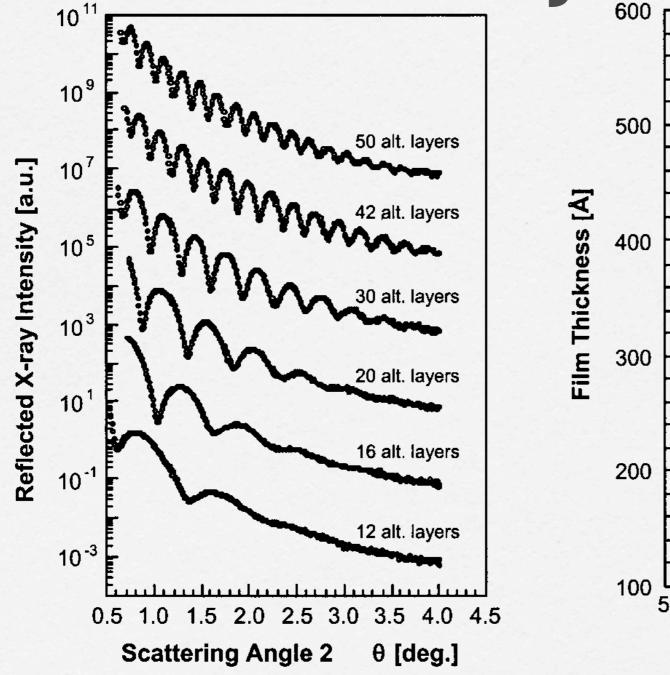
# temperature & humidity

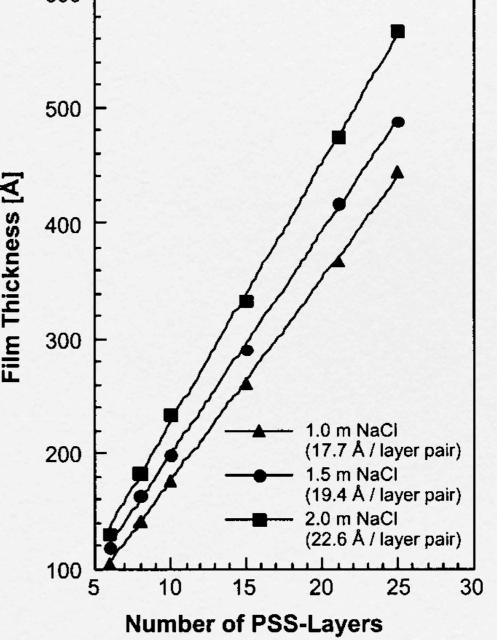


# UV/Vis characterization

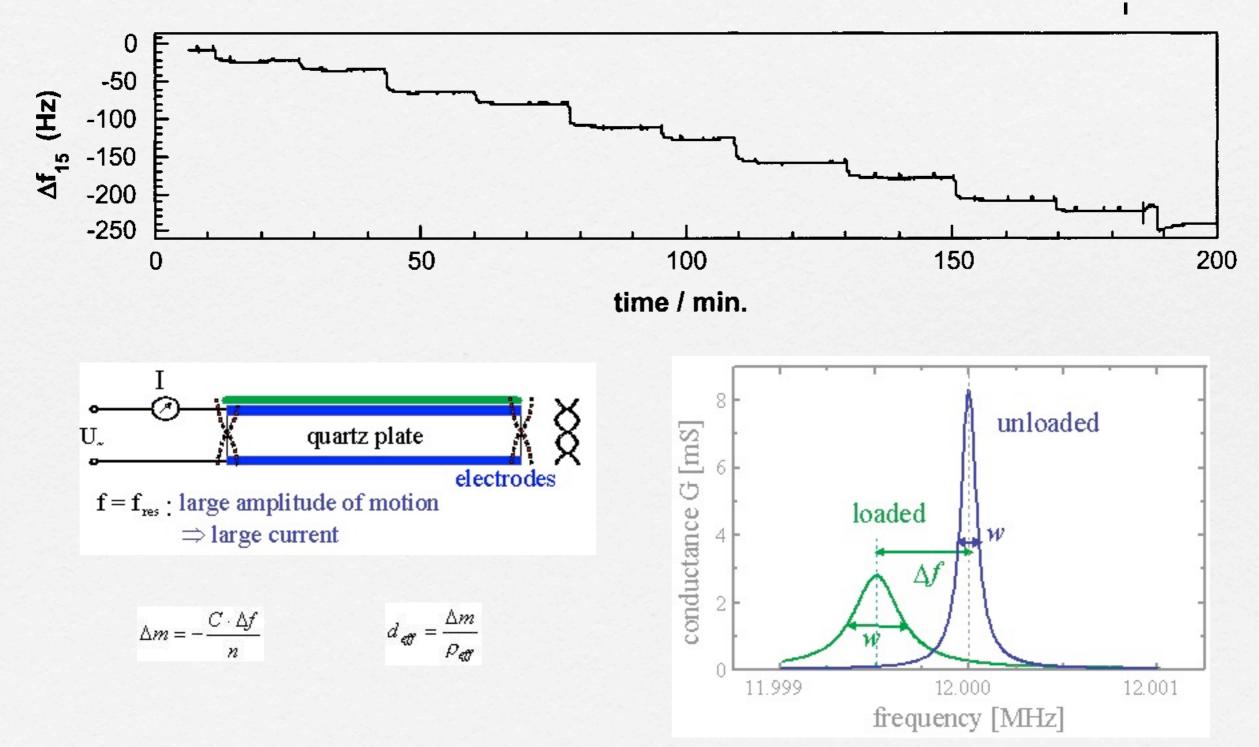


#### X-ray reflectometry of multilayer film

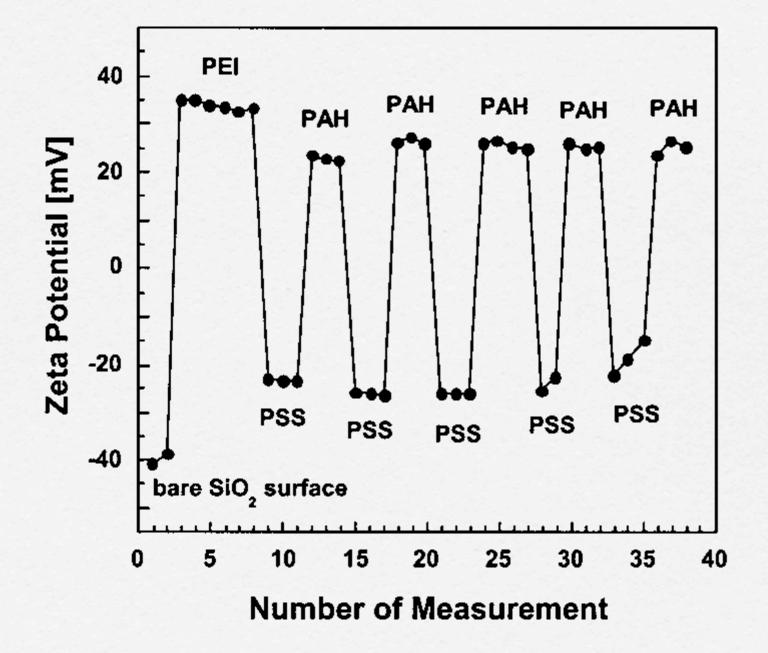




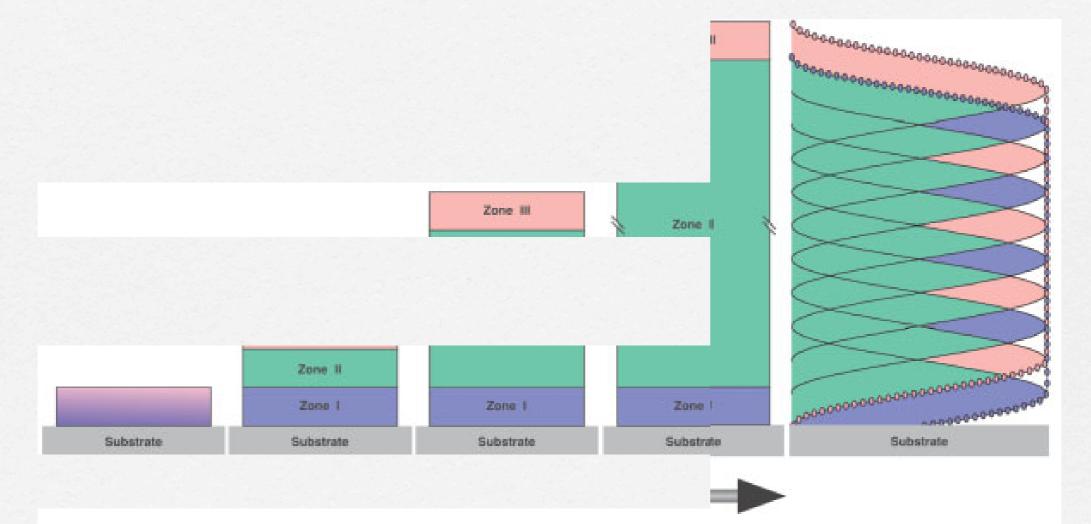
# QCM characterization



# Zeta potential measurement



# zone model for LbL multilayers



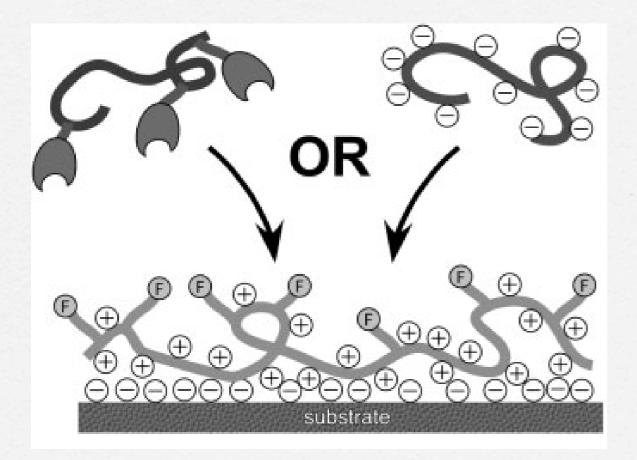
small gradients of

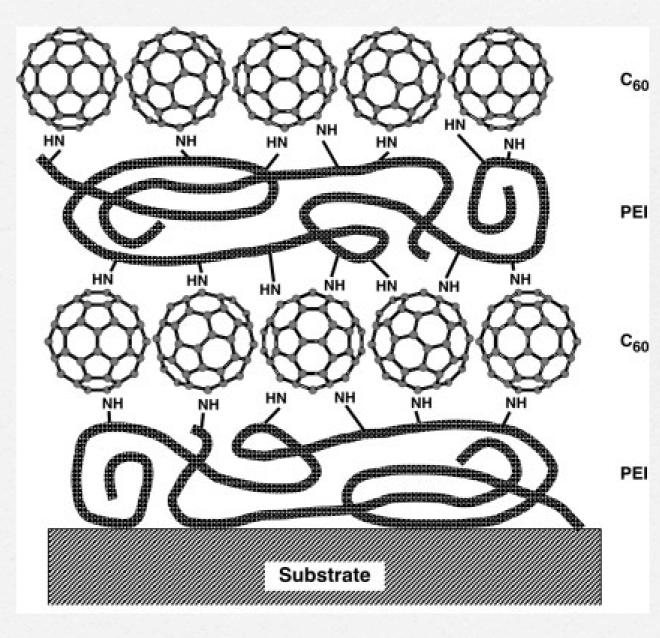
excess charge

charge neutrality in Zone II essentially charge compensated

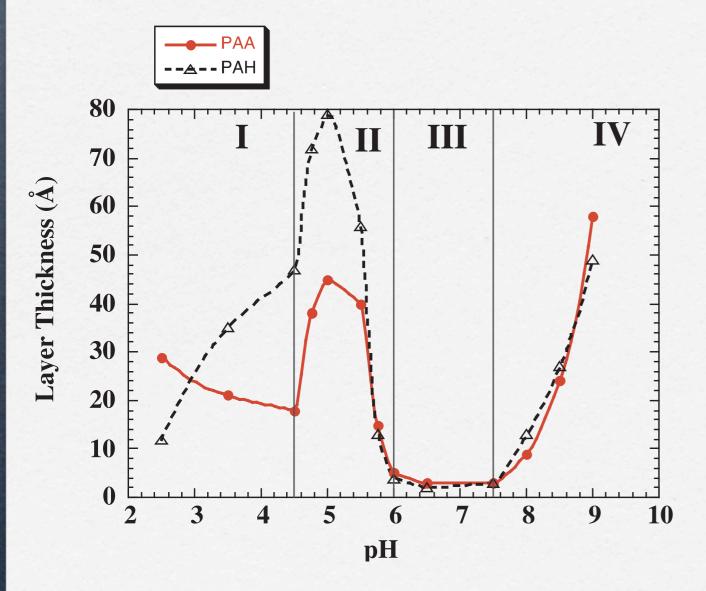
increasing numbers of layers

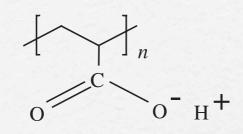
# multimaterials films

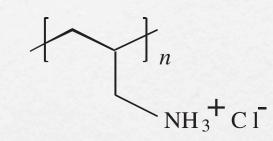




#### pH control using weak polyelectrolytes

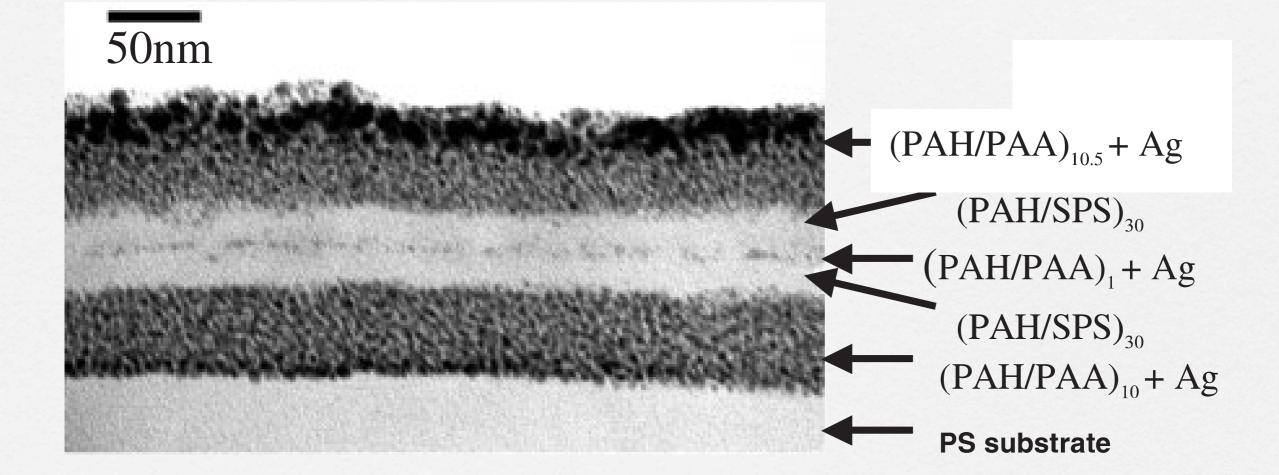






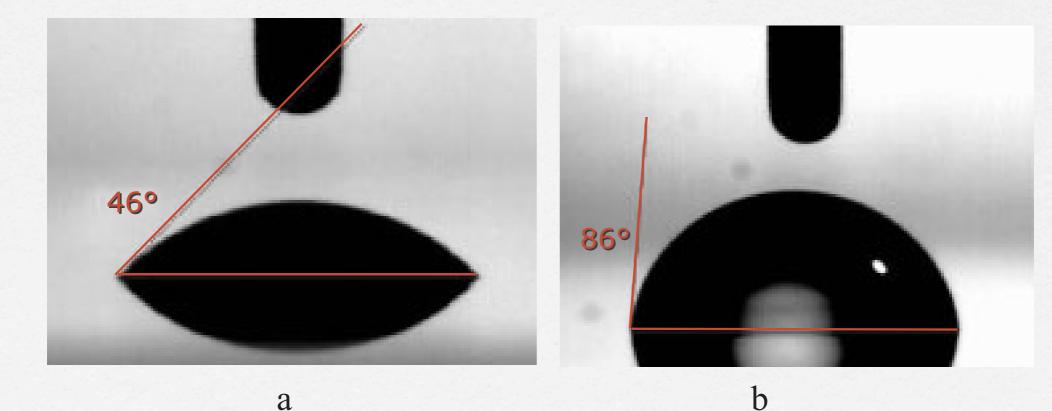
average incremental thickness contributed by a PAA and PAH adsorbed layer as a function of solution pH.

# particle incorporation in films

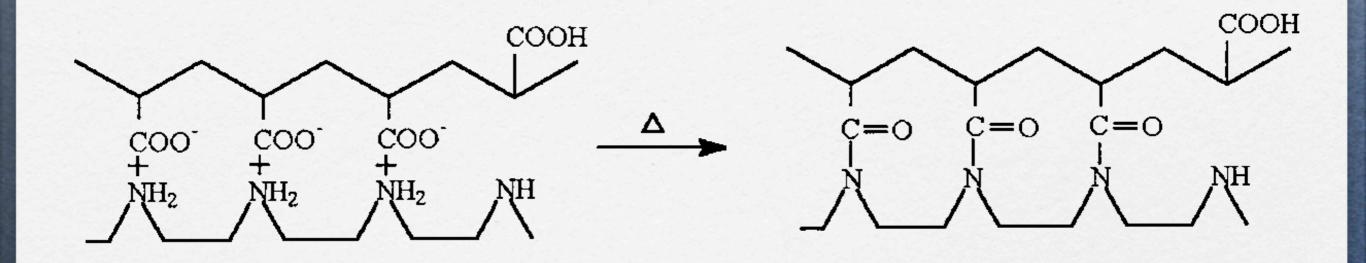


## surface functionalization

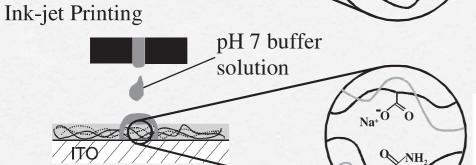
Images of water droplets on 3.5/7.5 (PAA/PAH)7.5 multi- layer films: (a) asprepared and (b) after immersing the multilayer film into a PS–PAA block copolymer solution.



# chemistry within films



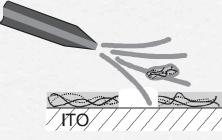
#### (PAA / PAAm), self-assembled multilayer TYO H-bonds



pH 7 buffer ionizes carboxylic acid groups

Drying and Crosslinking

Washing



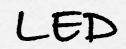
Only the intact film is crosslinked

Non-crosslinked polymer is washed off with water

Patterned Film

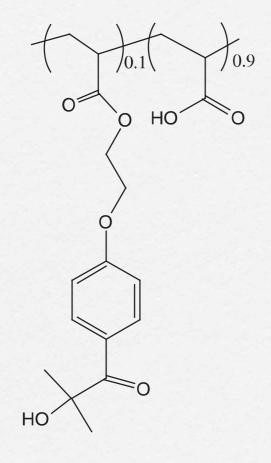


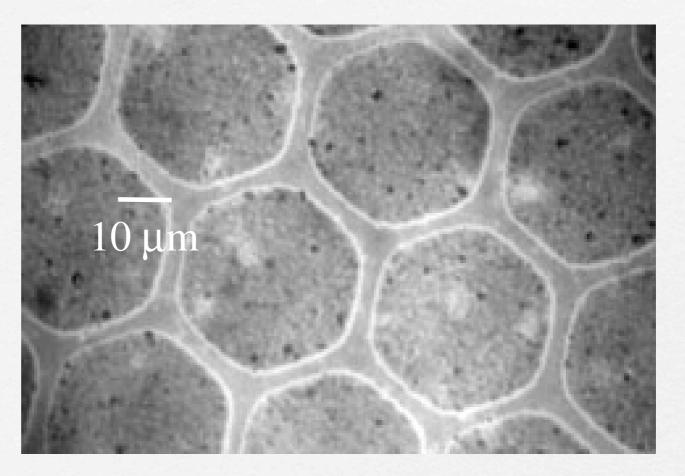
Patterned area can be filled with EL material



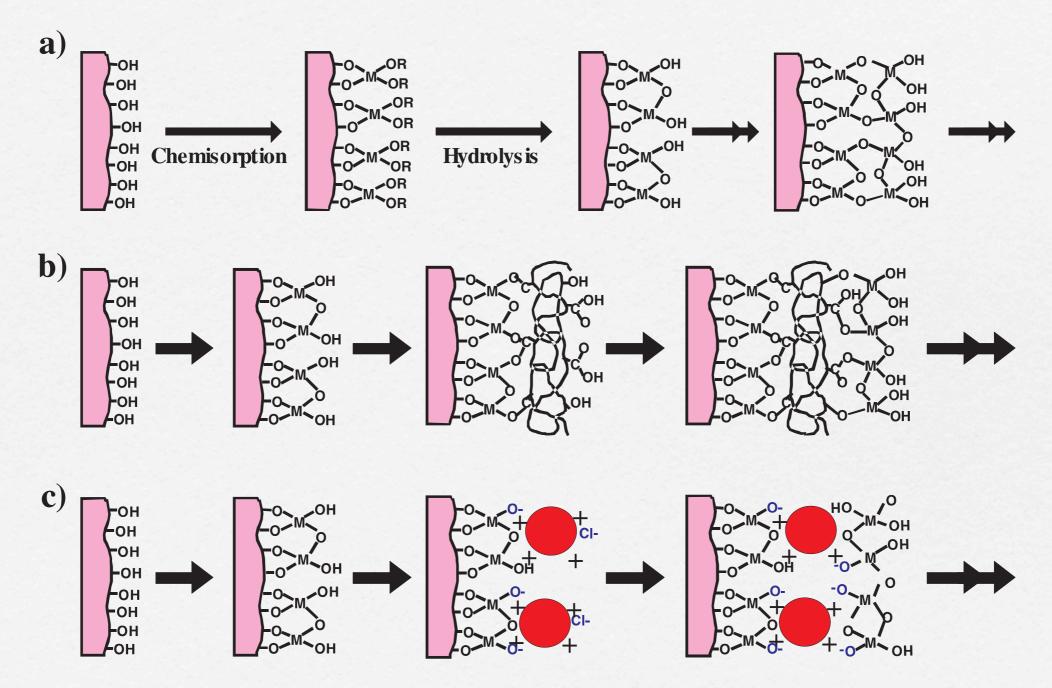


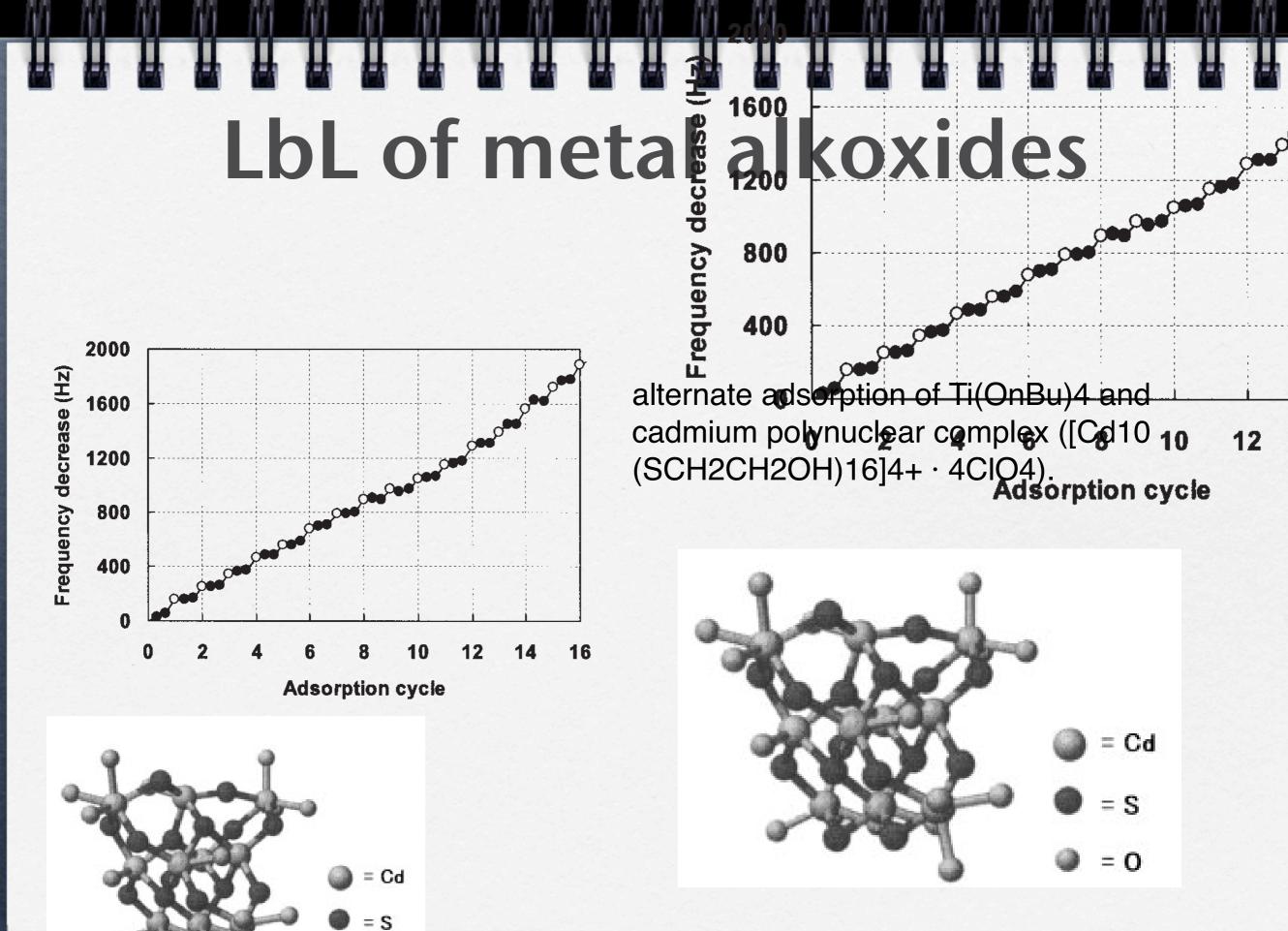
# patterning of films



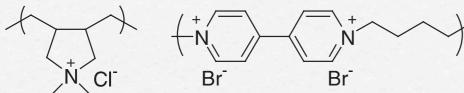


## LbL of metal alkoxides



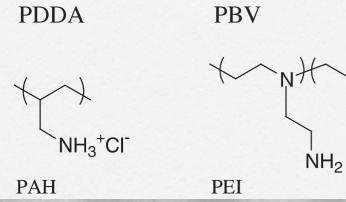


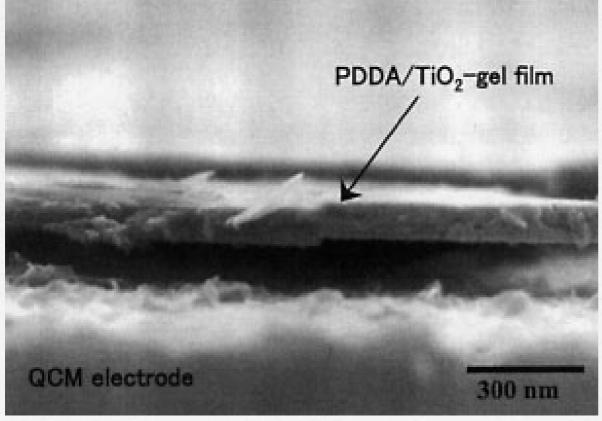
# LbL of metal alkoxides

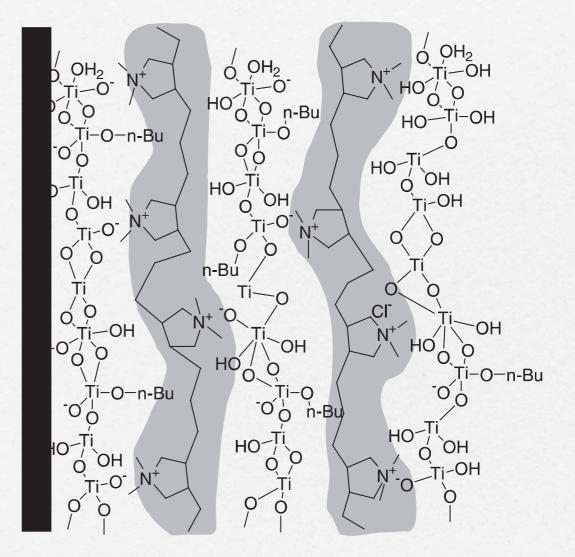


H

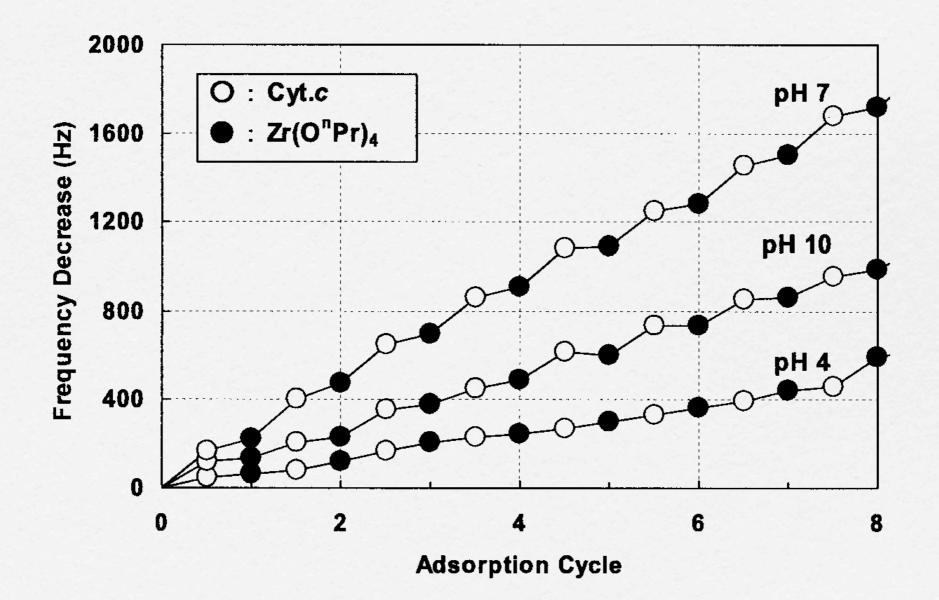
PDDA



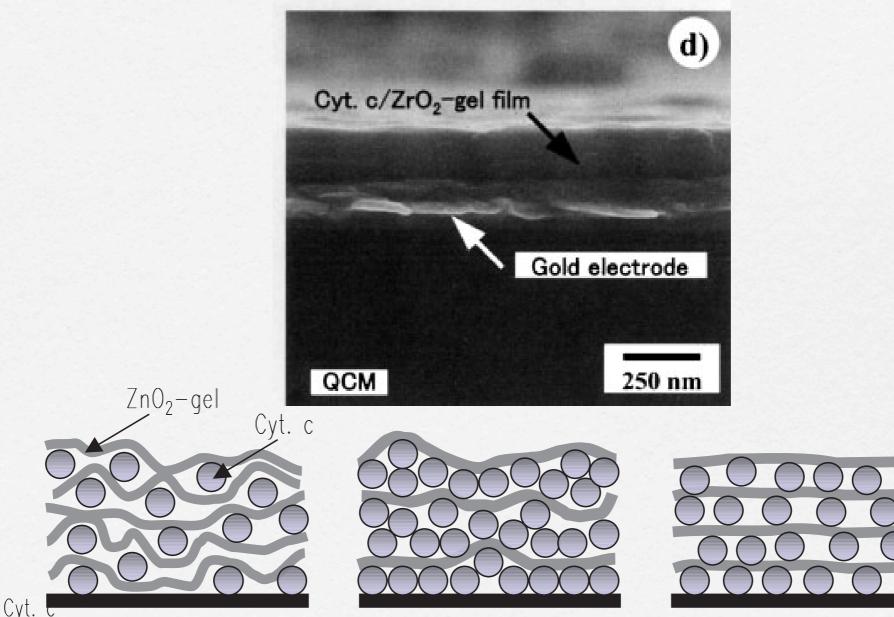




#### Multilayer Assembly of Metal Oxides and Proteins



#### **Multilayer Assembly of Metal Oxides and Proteins**

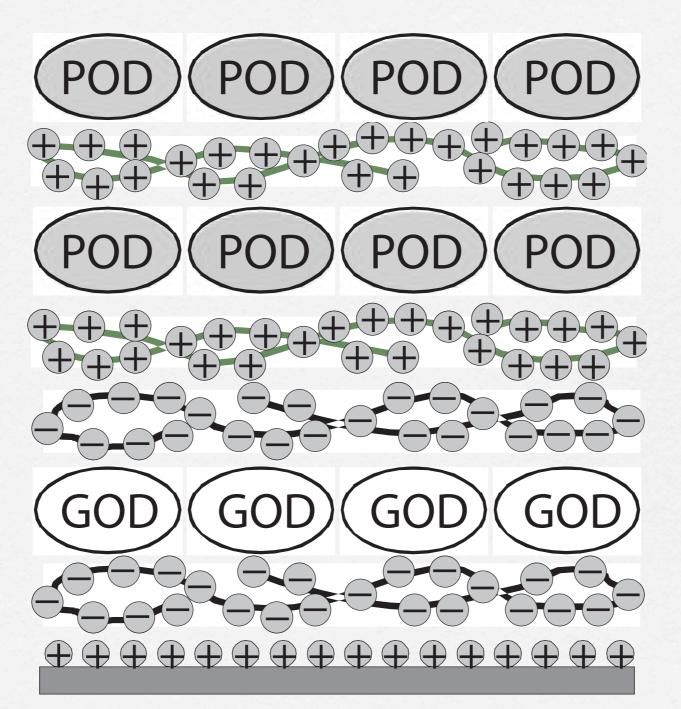


pH 10

pH 4

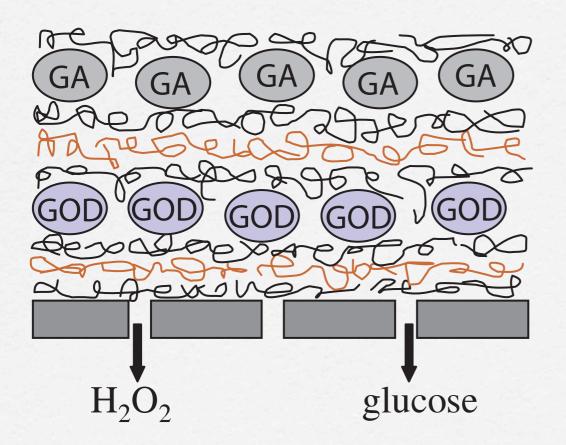
ZnO<sub>2</sub>-gel

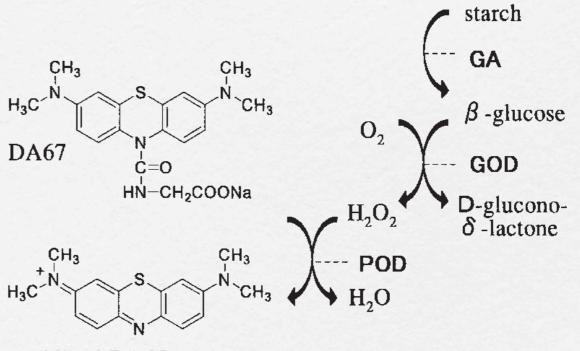
# protein/polymer multilayer



#### 

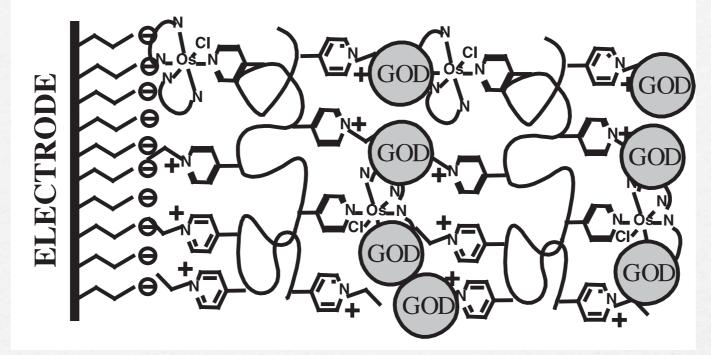
starch

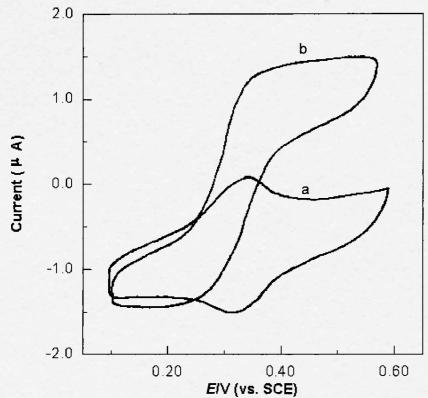




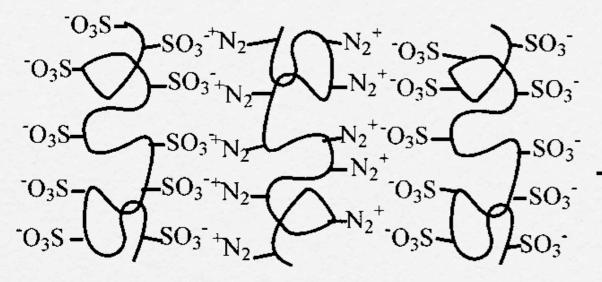
oxidized DA67

## Glucose oxidase films

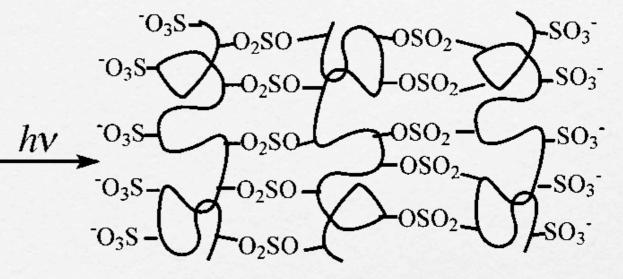




(a) without glucose(b) in the presence of glucose

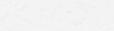


DAR



DAR

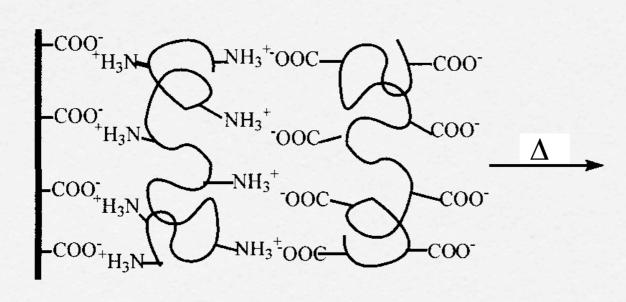
PSS

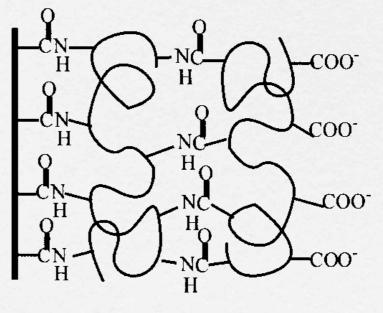


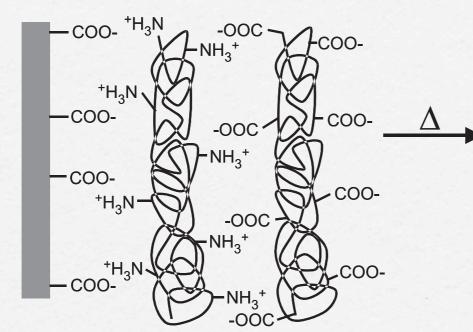
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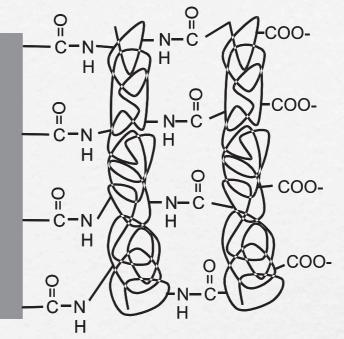
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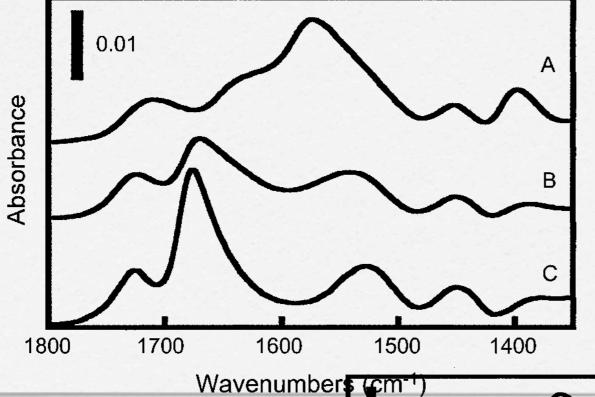
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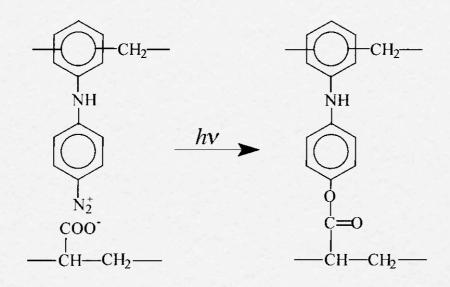




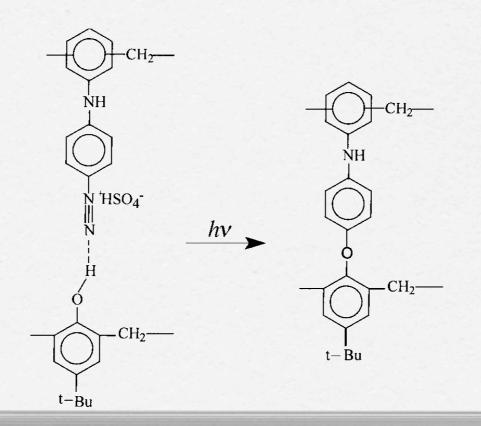


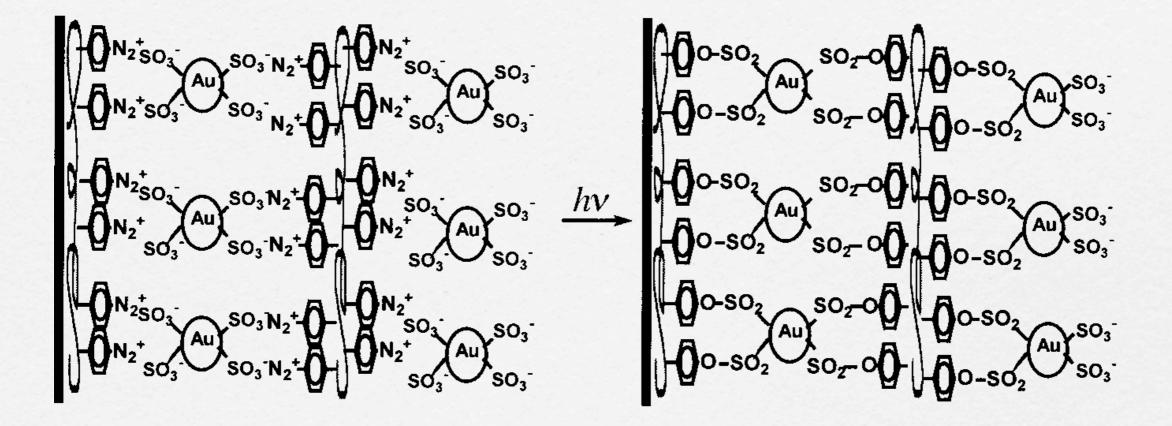


Reflectance FTIR spectra of ninebilayer PAH/PAA films before heating (A) and after heating for 2 h at 130°C (B) or 215°C (C).

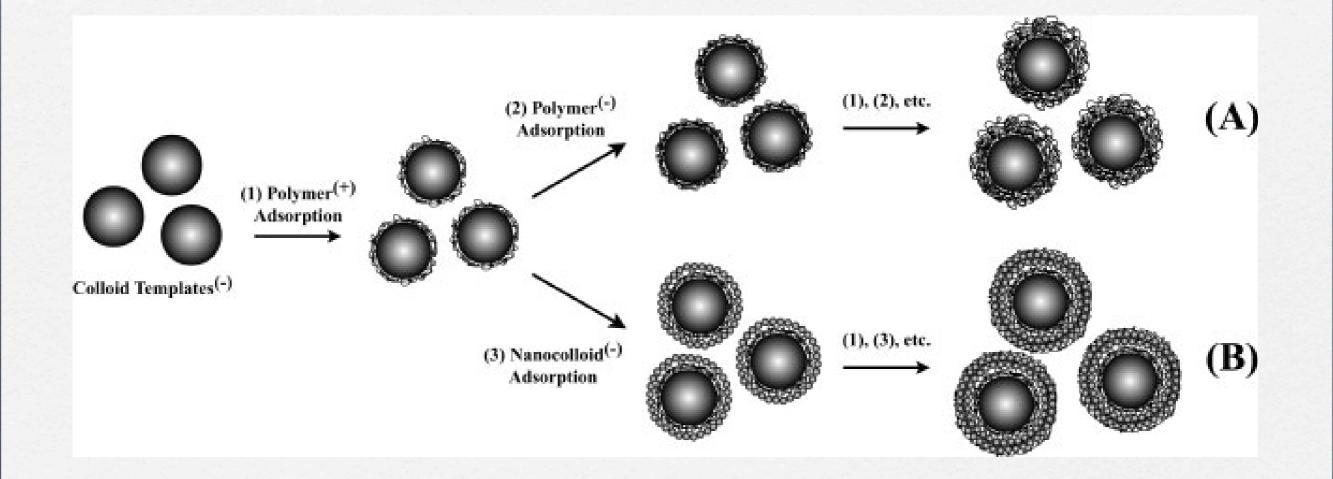


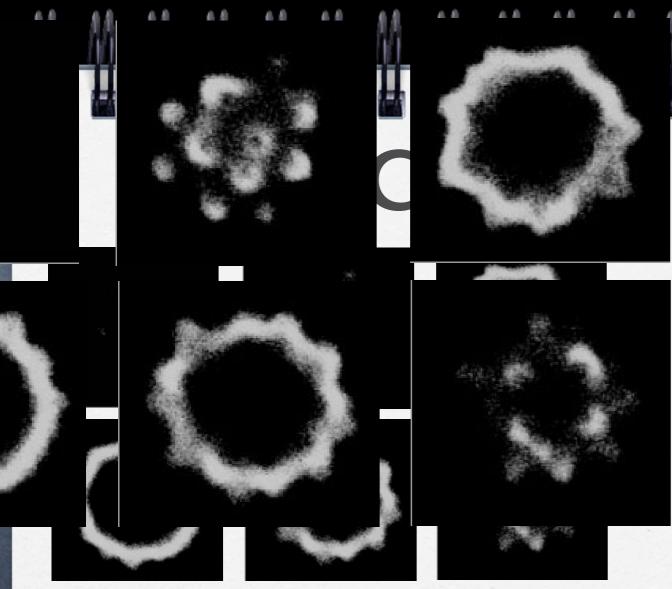
В



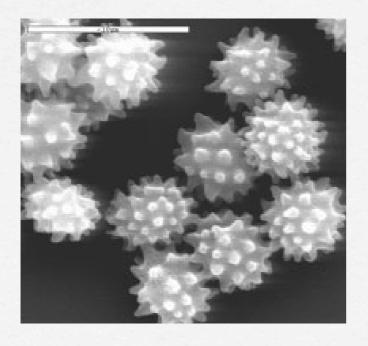


# Coated Colloids



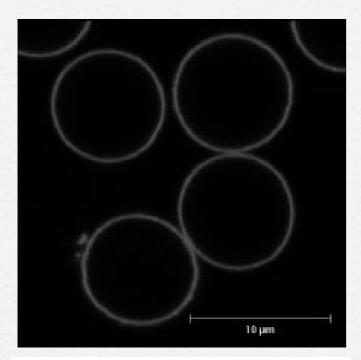


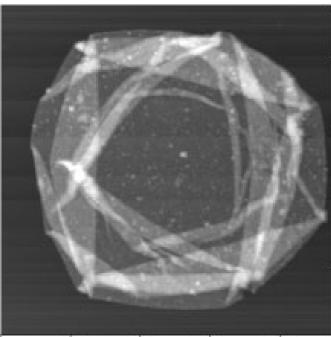
### Colloids

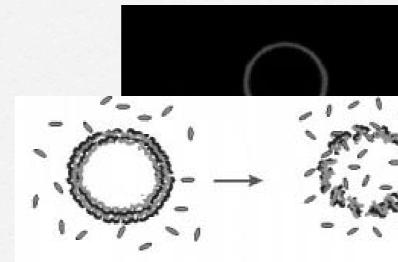


polyelectrolyte shell consisting of 11 layers of PSS/PAH templated on an echinocyte. The outer layer is FITC labelled PAH.

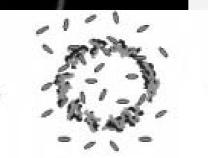
# hollow capsules







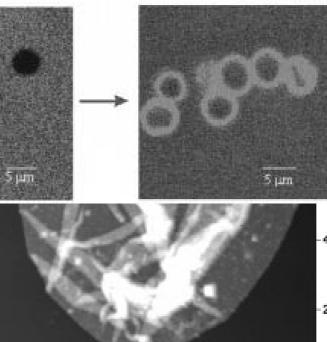
In Water, closed



Ethanol/Water 1:1, open



In Water. Encapsulated



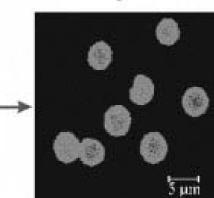
4.00

2.00

ò

6.00

500 nm



4.00

-2.00

-0

Įни

8.00

## 

#### Multilayer Thin Films

Sequential Assembly of Nanocomposite Materials

Edited by Gero Decher and Joseph B. Schlenoff Foreword by Jean-Marie Lehn

