

Chapter 1.

Semiconductors: A General Introduction

Sung June Kim

kimsj@snu.ac.kr

<http://helios.snu.ac.kr>



Syllabus



Contents

- ❑ General Material Properties
- ❑ Crystal Structure



□ General Material Properties

- ✓ Examine the general nature of semiconductor materials
- ✓ Conductivity : insulator → semiconductor → metal

• Composition

- ✓ Elemental semiconductor: group IV (Si, Ge)
- ✓ Compound semiconductor: group III & V (or II & VI) (GaAs, ZnSe, and alloys like $\text{Al}_x\text{Ga}_{1-x}\text{As}$)
- ✓ Si : the most important of the semiconductors → major ICs including CPU
- ✓ GaAs : superior electron transport properties and special optical properties → laser diode, LED, and high speed ICs
- ✓ Binary : two elements
- ✓ Ternary : three elements

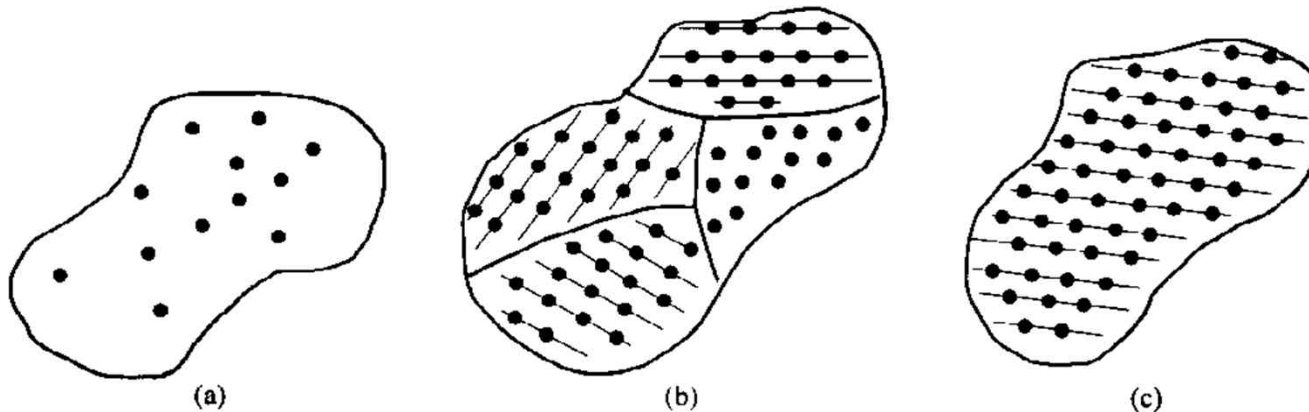


- Purity

- ✓ Dopants can have a drastic effect on the electrical properties of semiconductors
- ✓ Typically, dopant atoms at levels ranging from one part per 10^8 to one impurity atom per 10^3 semiconductor atoms will be purposely added to the semiconductor to control its electrical properties

- Structure

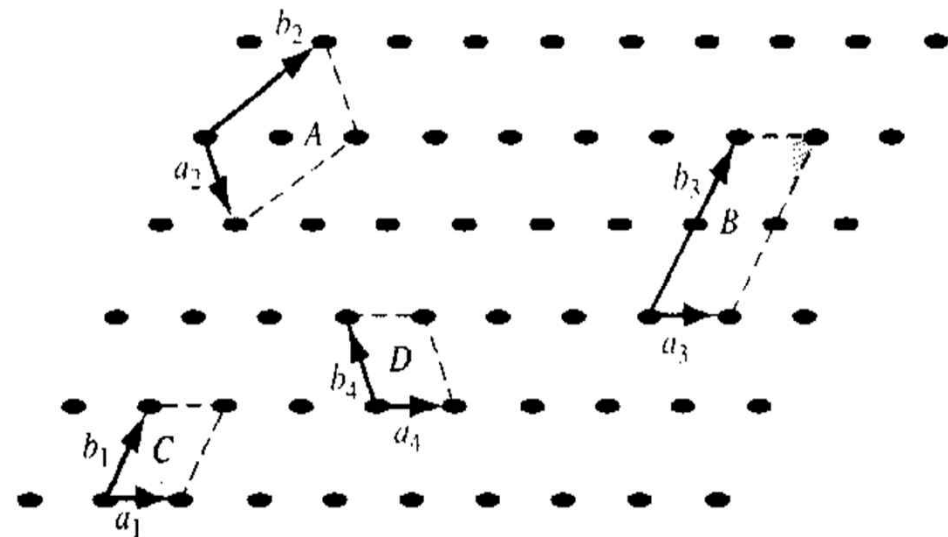
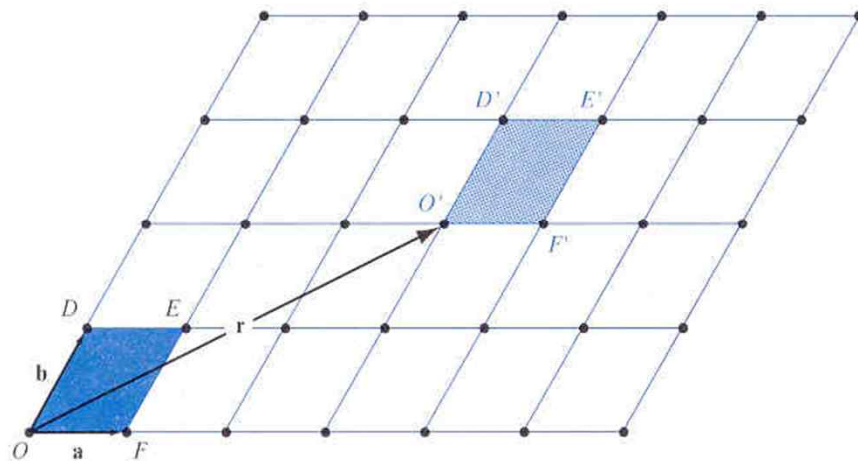
- ✓ Amorphous
- ✓ Polycrystalline : grain, grain boundary
- ✓ Single crystal : electrical properties are superior



□ Crystal Structure

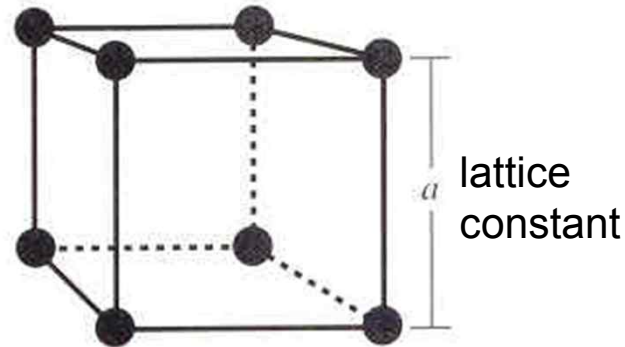
• The Unit Cell Concept

- ✓ Lattice = periodic arrangement of atoms in the crystal
- ✓ Lattice point = a dot which represents a particular atomic array
- ✓ Unit Cell = a small volume of the crystal that can be used to reproduce the entire crystal
- ✓ Primitive cell = the smallest unit cell that can be repeated to form the lattice

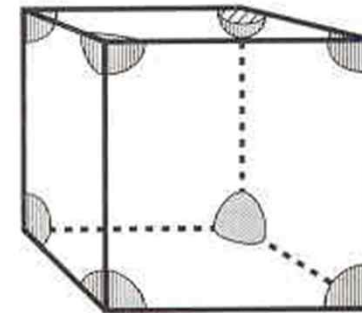


- Simple 3-D Unit Cells

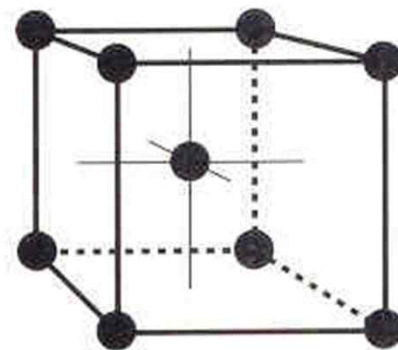
✓ simple cubic = sc, body-centered cubic = bcc, face-centered cubic = fcc



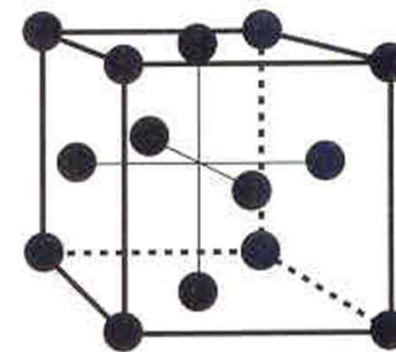
(a) Simple cubic



(b) Pedantically correct simple cubic



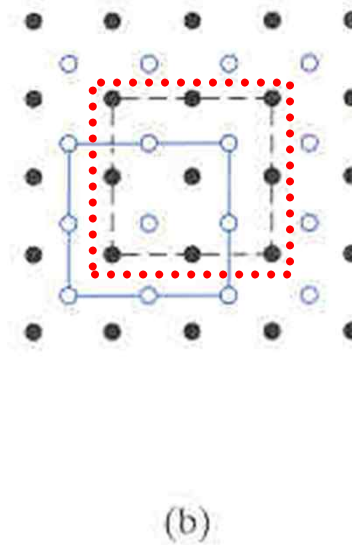
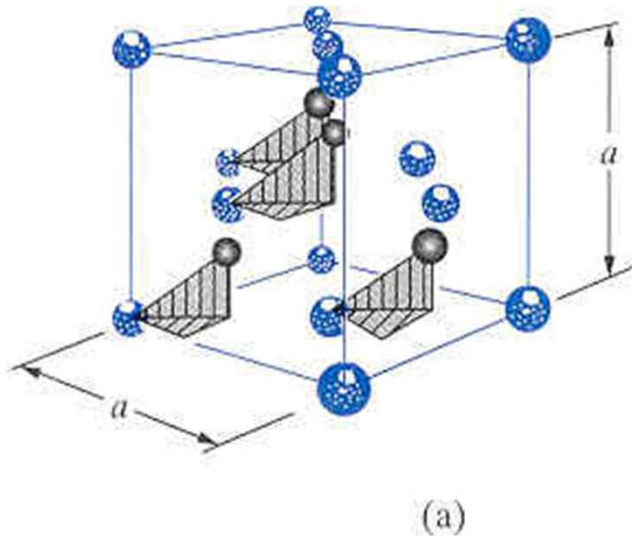
(c) bcc



(d) fcc



- Semiconductor Lattices
 - ✓ The diamond structure

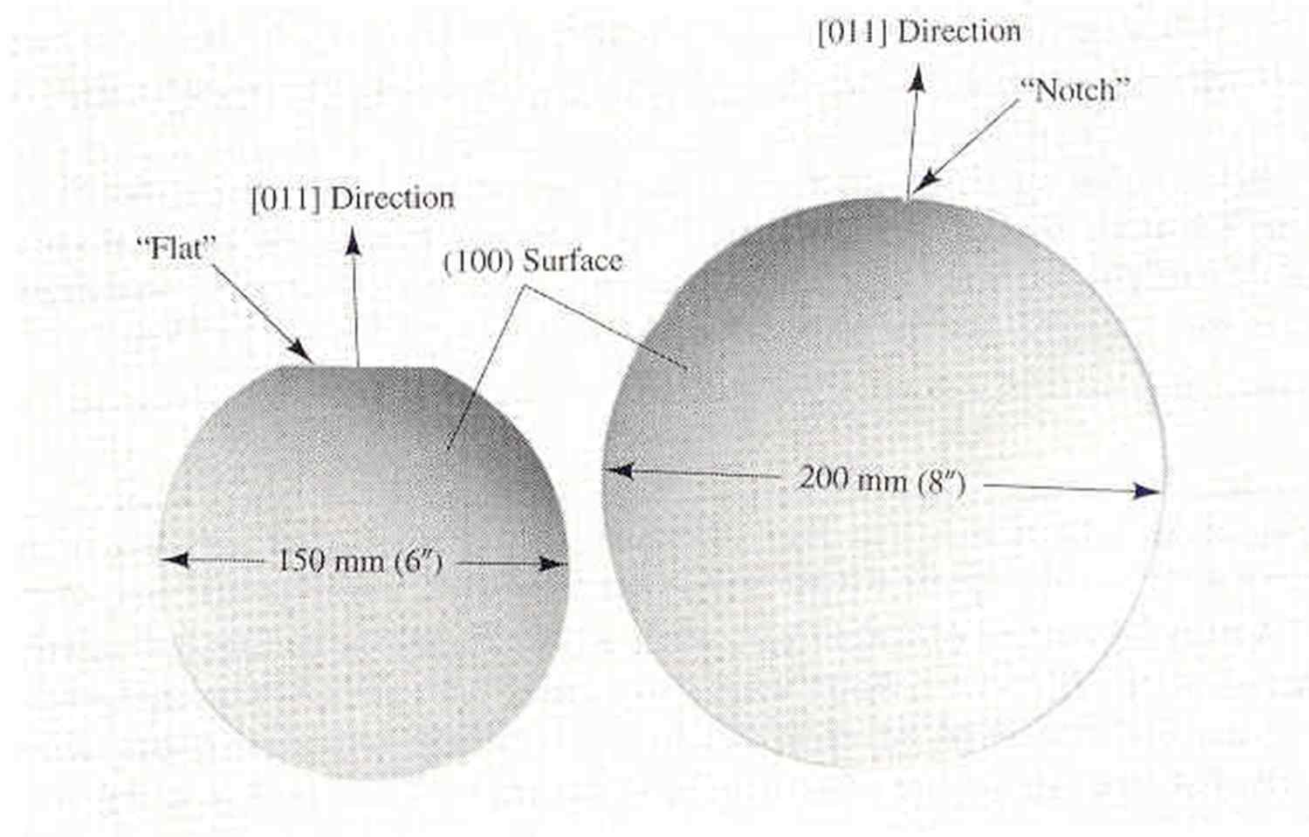


For Si,
 8 atoms in a cell
 $a = 5.43 \text{ \AA}$
 $(1 \text{ \AA} = 10^{-8} \text{ cm})$
 $8/a^3 = 5.22 \times 10^{22} \text{ cm}^{-3}$

(a) A unit cell of the diamond lattice constructed by placing atoms $\frac{1}{4}, \frac{1}{4}, \frac{1}{4}$ from each atom in an fcc; (b) top view (along any $\langle 100 \rangle$ direction) of an extended diamond lattice. The colored circles indicate one fcc sublattice and the black circles indicate the interpenetrating fcc.



- Miller Indices



Single crystal silicon wafers (6" and 8") typical of the starting substrates



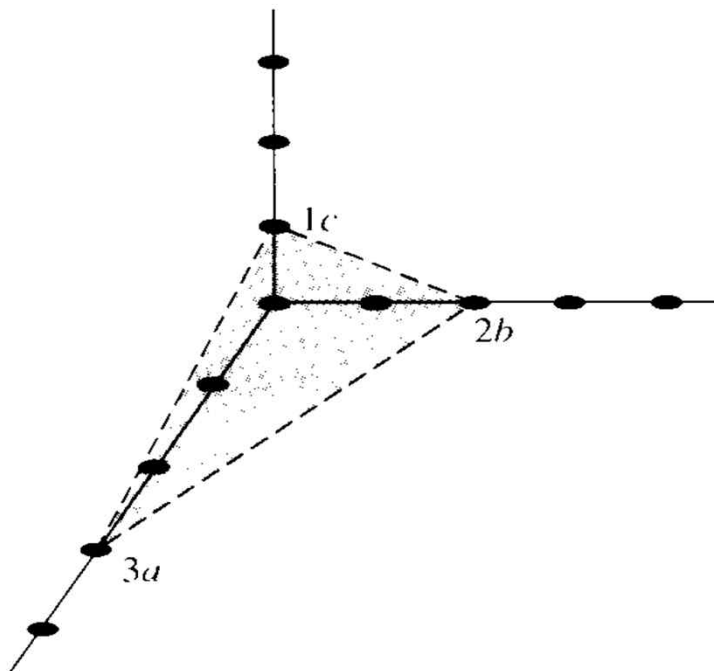
✓ Indexing procedure:

Intercept set in the order x, y, z

Invert the intercept values

Convert the 1/intercept set to the smallest possible set of whole numbers

Enclose the whole number set in curvilinear brackets

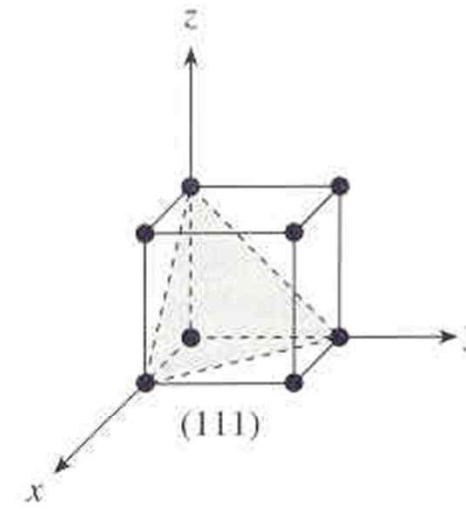
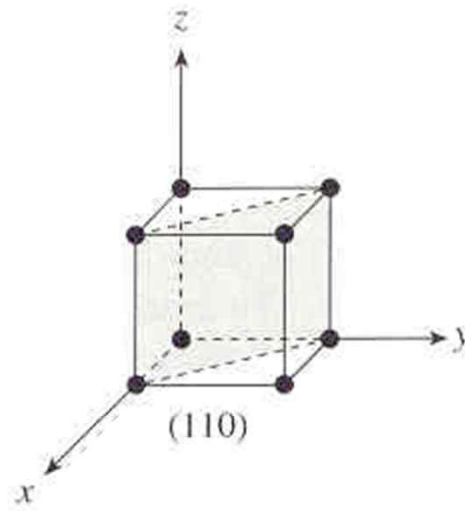
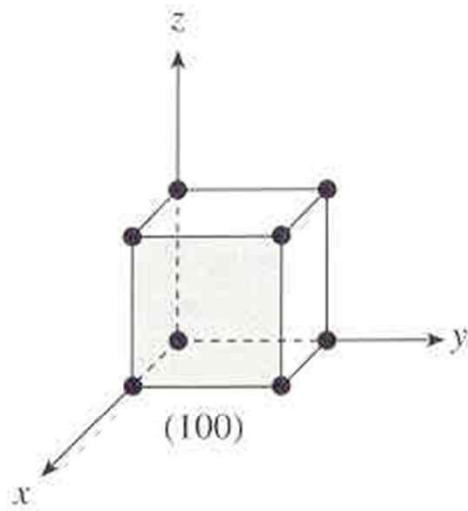


$$p=3, q=2, s=1$$

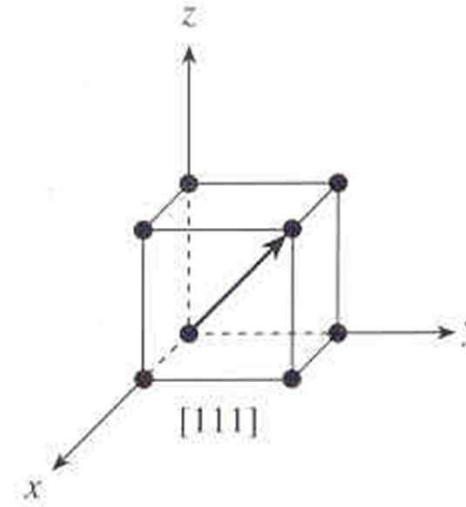
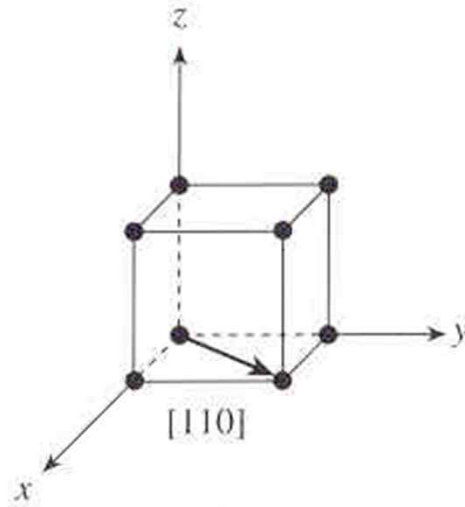
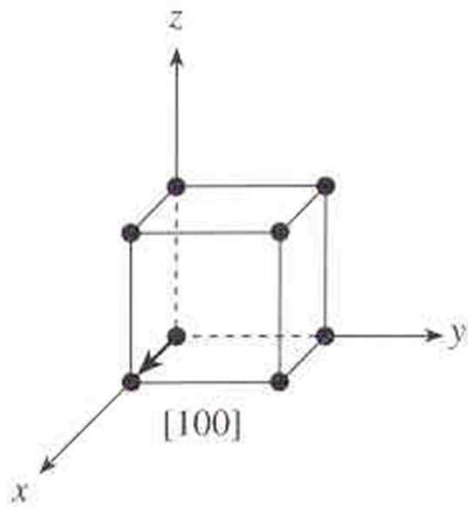
$$\left(\frac{1}{3}, \frac{1}{2}, 1\right) \rightarrow (236) \text{ plane}$$

Miller indices





(a)



(b)

Summary

