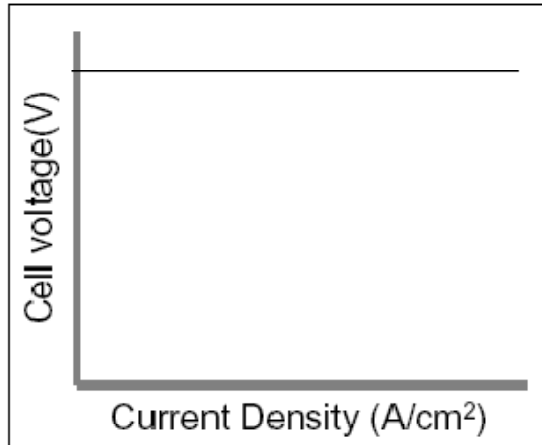
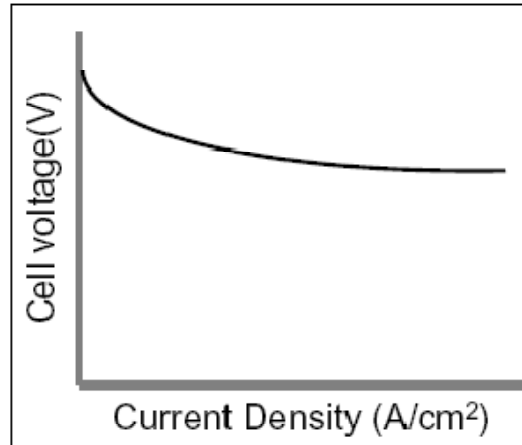


Losses in Fuel Cells

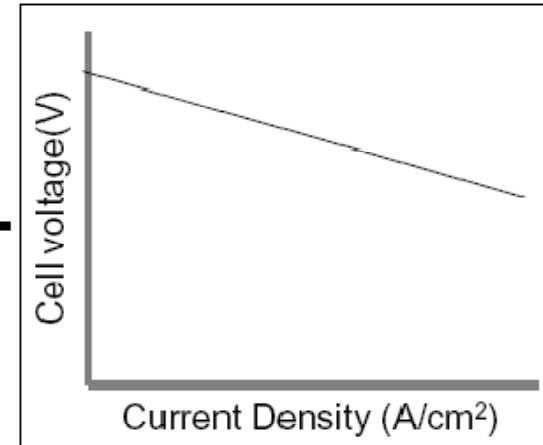
Reversible Voltage (Chapter 2)



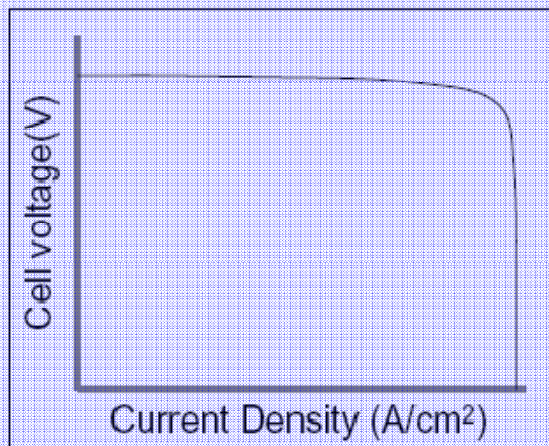
Rxn. Loss (Chapter 3)



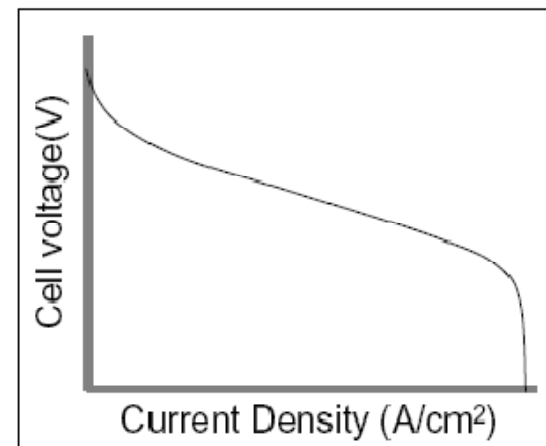
Ohmic Loss (Chapter 4)



Concentration Loss (Chapter 5)



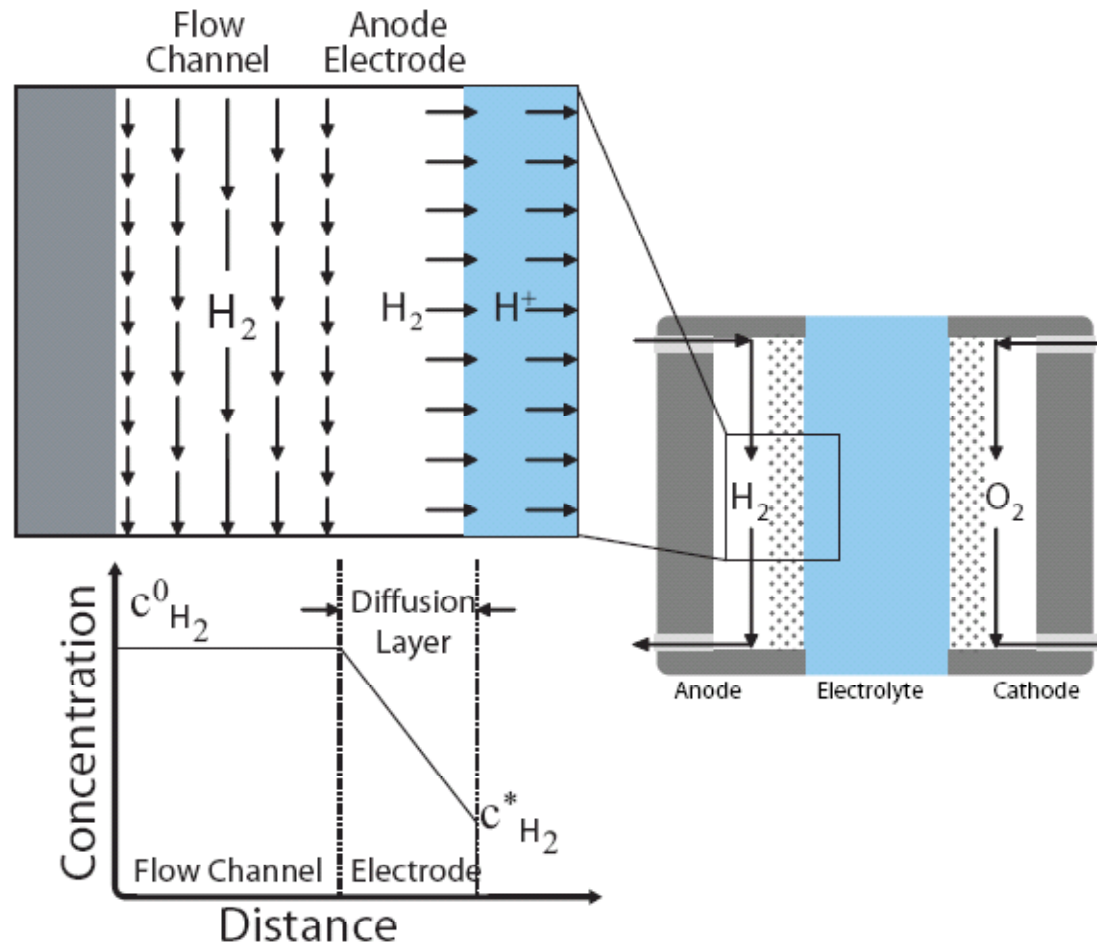
Net Fuel Cell Performance



$$V = E_{thermo} - \eta_{act} - \eta_{ohmic} - \eta_{conc}$$

Fuel Cell Charge Transport

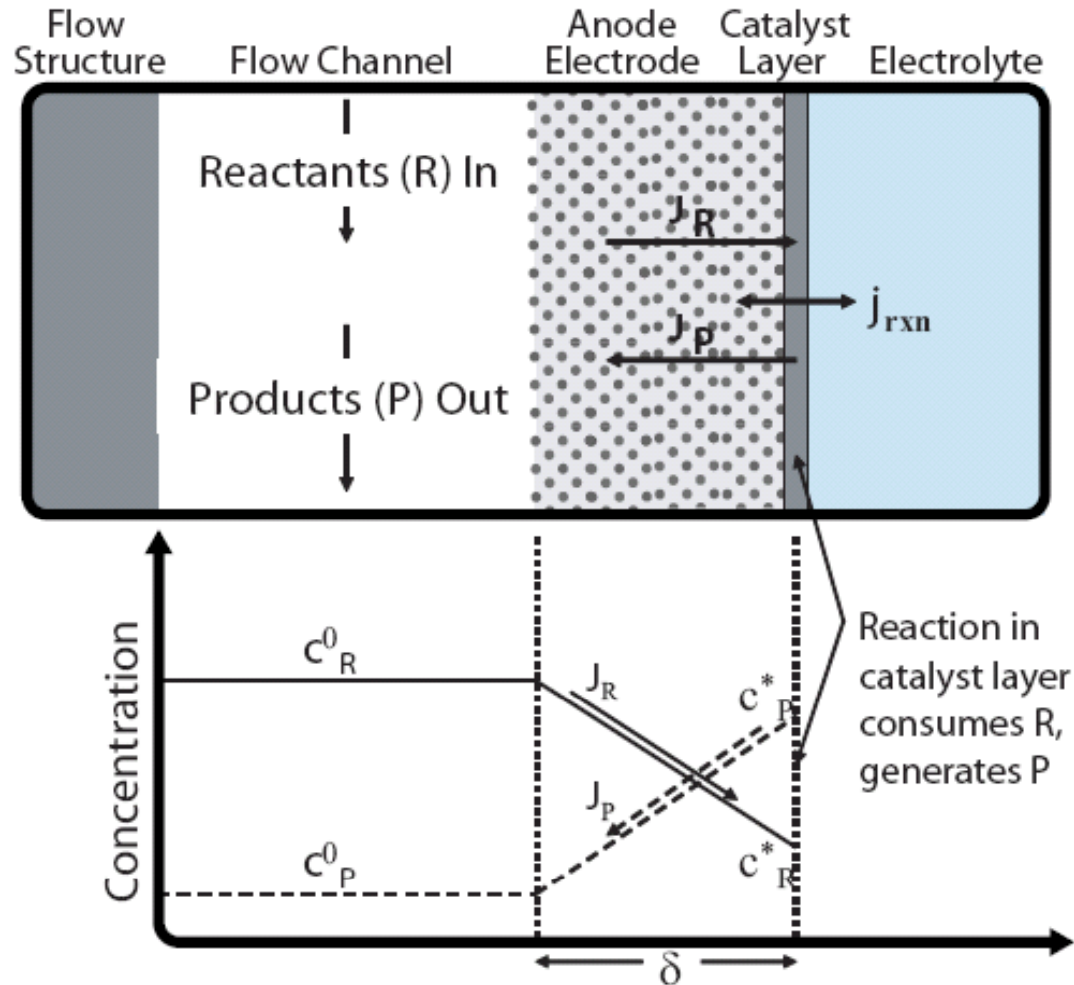
Mass Transport



Convection dominant in flow channels

Diffusion dominant in electrode

Concentration Drop



$$E = E^0 - \frac{RT}{nF} \ln \frac{\prod p_{\text{Product}}^{v_i}}{\prod p_{\text{Reactants}}^{v_i}}$$

$$j = j_0 \left(\frac{C_R}{C_R^0} e^{\left(\frac{\alpha n F \eta}{RT}\right)} - \frac{C_P}{C_P^0} e^{-\left(\frac{(1-\alpha) n F \eta}{RT}\right)} \right)$$

Diffusion In Electrode

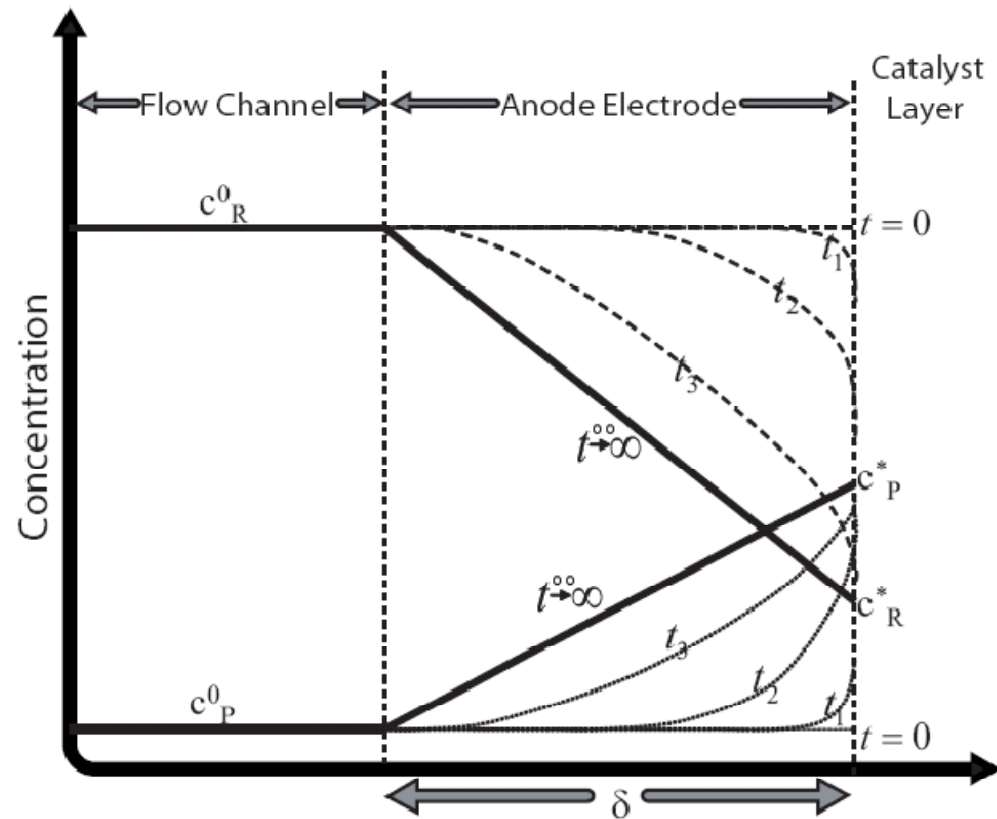
$$J_{diff} = -D \frac{dc}{dx}$$

$$J_{diff} = -D^{eff} \frac{c_R^* - c_R^0}{\delta}$$

$$j = nF J_{diff}$$

$$j = -nFD^{eff} \frac{c_R^* - c_R^0}{\delta}$$

$$c_R^* = c_R^0 - \frac{j\delta}{nFD^{eff}}$$



Limiting Current Density

$$j_L = nFD^{eff} \frac{c_R^0}{\delta}$$

Typical values for δ are around $100-300\mu m$ and D^{eff} are around $10^{-2}cm^2/s$. Therefore typical limiting current densities are on order of $1-10A/cm^2$. This

– To increase j_L

1. Ensuring a high c_R^0 (by designing good flow structures that evenly distribute reactants).
2. Ensuring that D^{eff} is large and δ is small (by carefully optimizing fuel cell operating conditions, electrode structure, and diffusion layer thickness).

Q: Are 1 & 2 are easy?

Q: Does anode has larger j_L ?

Diffusivity & Effective Diffusivity

Diffusivity from kinetic theory

$$p \cdot D_{ij} = a \left(\frac{T}{\sqrt{T_{ci}T_{cj}}} \right)^b (p_{ci}p_{cj})^{1/3} (T_{ci}T_{cj})^{5/12} \left(\frac{1}{M_i} + \frac{1}{M_j} \right)^{1/2}$$

Substance	Molecular Weight, M	$T_c(K)$	$p_c(\text{atm})$
H ₂	2.016	33.3	12.80
Air	28.964	132.4	37.0
N ₂	28.013	126.2	33.5
O ₂	31.999	154.4	49.7
CO	28.010	132.9	34.5
CO ₂	44.010	304.2	72.8
H ₂ O	18.015	647.3	217.5

Effective diffusivity

$$D_{ij}^{eff} = \epsilon^{1.5} D_{ij}$$

$$D_{ij}^{eff} = \epsilon^\tau D_{ij}$$

$$D_{ij}^{eff} = D_{ij} \frac{\epsilon}{\tau}$$

Nernst Effect

$$E = E^0 - \frac{RT}{nF} \ln \frac{\prod a_{PRODUCTS}^{v_i}}{\prod a_{REACTANTS}^{v_i}}$$

$$\eta_{conc} = E_{Nernst}^0 - E_{Nernst}^*$$

$$\eta_{conc} = \left(E^0 - \frac{RT}{nF} \ln \frac{1}{c_R^0} \right) - \left(E^0 - \frac{RT}{nF} \ln \frac{1}{c_R^*} \right)$$

$$\eta_{conc} = \frac{RT}{nF} \ln \frac{c_R^0}{c_R^*}$$

$$c_R^* = c_R^0 - \frac{j\delta}{nFD^{eff}}$$

$$c_R^* = \frac{j_L\delta}{nFD^{eff}} - \frac{j\delta}{nFD^{eff}}$$

$$\frac{c_R^0}{c_R^*} = \frac{\frac{j_L\delta}{nFD^{eff}}}{\frac{j_L\delta}{nFD^{eff}} - \frac{j\delta}{nFD^{eff}}}$$

$$\frac{c_R^0}{c_R^*} = \frac{j_L}{j_L - j}$$

$$\eta_{conc} = \frac{RT}{nF} \ln \frac{j_L}{j_L - j}$$

Concentration Effect

B-V for high current density

$$j = j_0^0 \left(\frac{c_R^*}{c_R^{0*}} e^{\left(\frac{\alpha n F \eta_{act}}{RT} \right)} \right)$$

$$\eta_{act} = \frac{RT}{\alpha n F} \ln \frac{j c_R^{0*}}{j_0^0 c_R^*}$$

$$\eta_{conc} = \eta_{act}^* - \eta_{act}^0$$

$$\eta_{conc} = \left(\frac{RT}{\alpha n F} \ln \frac{j c_R^{0*}}{j_0^0 c_R^*} \right) - \left(\frac{RT}{\alpha n F} \ln \frac{j c_R^0}{j_0^0 c_R^0} \right)$$

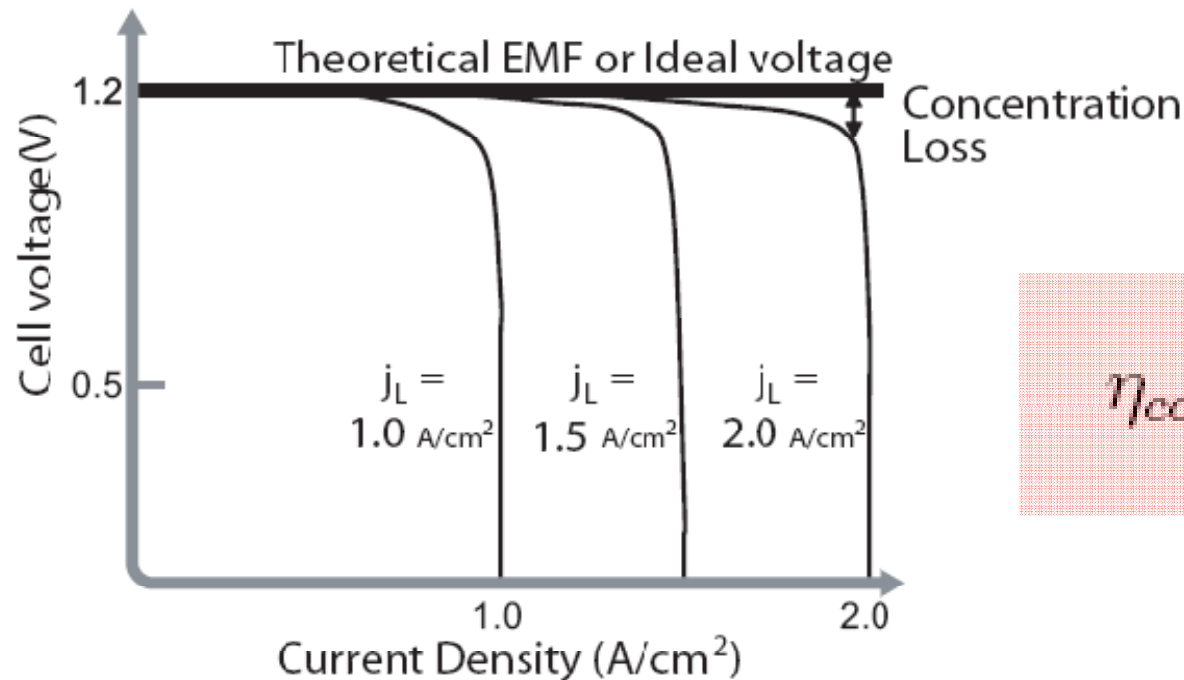
$$\eta_{conc} = \frac{RT}{\alpha n F} \ln \frac{c_R^0}{c_R^*}$$

$$\eta_{conc} = \frac{RT}{\alpha n F} \ln \frac{j_L}{j_L - j}$$

Mass Transportation Loss

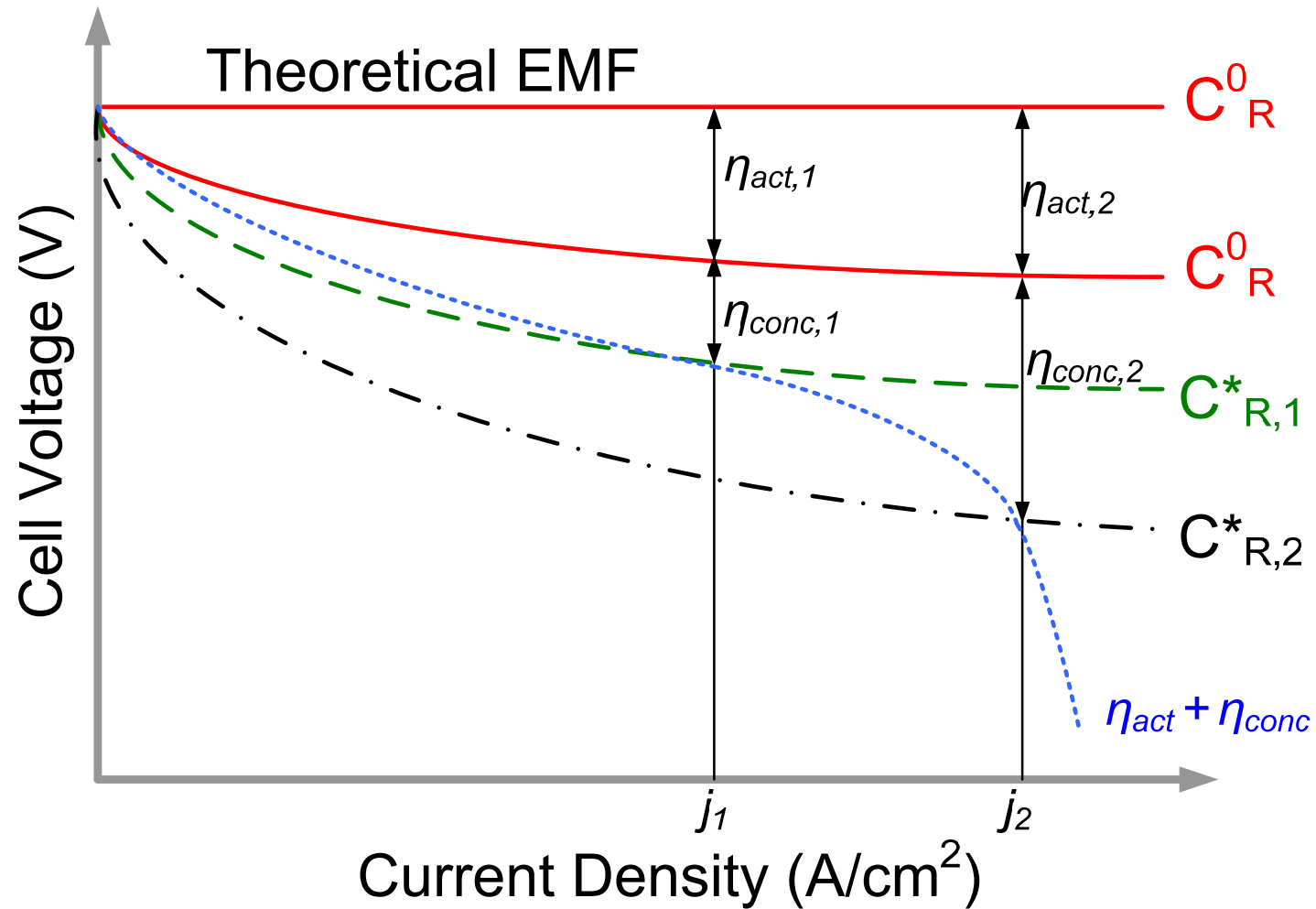
Adding two losses

$$\eta_{conc} = \frac{RT}{nF} \ln \frac{j_L}{j_L - j} + \frac{RT}{\alpha nF} \ln \frac{j_L}{j_L - j}$$
$$\eta_{conc} = \left(\frac{RT}{nF}\right) \left(1 + \frac{1}{\alpha}\right) \ln \frac{j_L}{j_L - j}$$



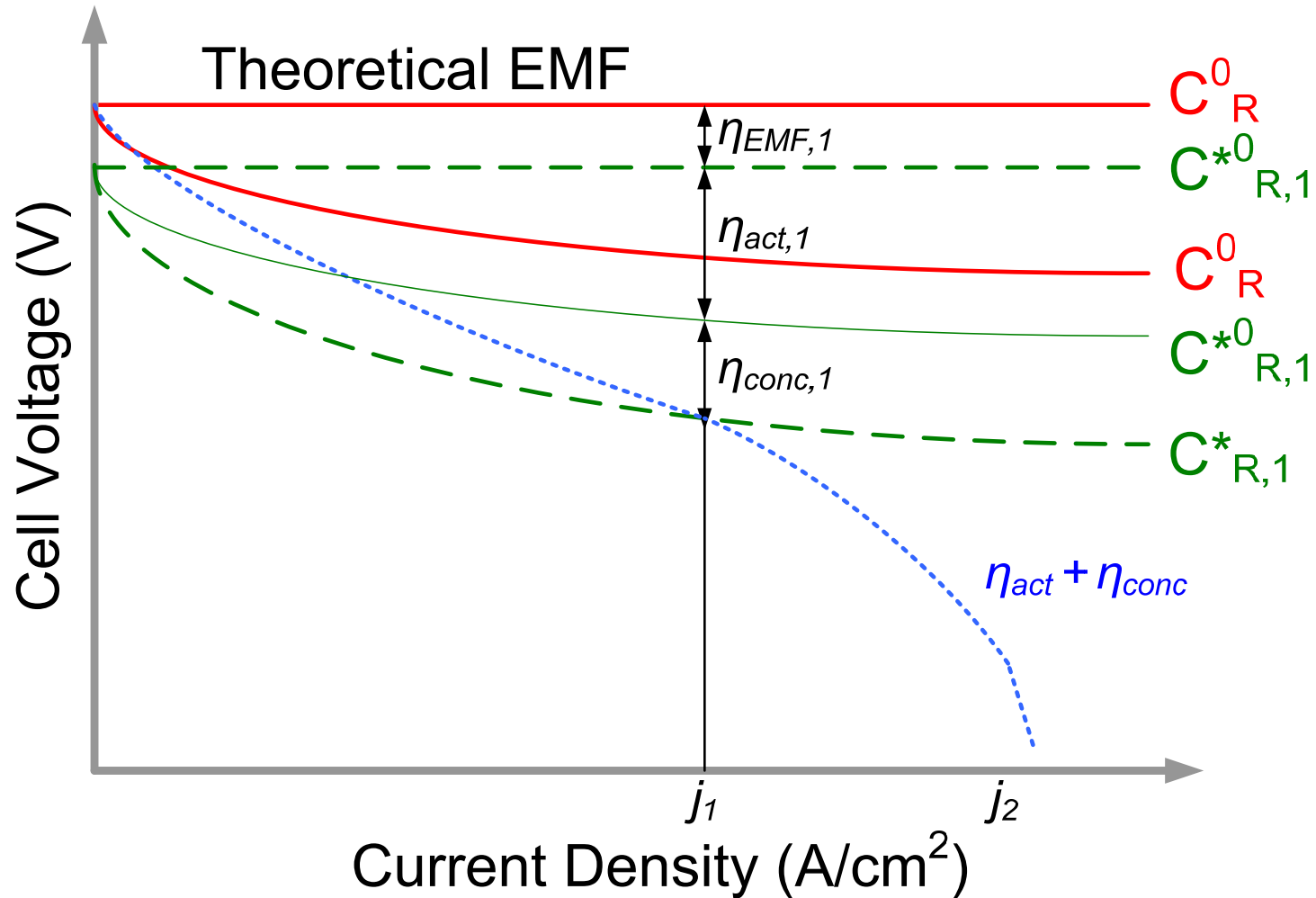
$$\eta_{conc} = c \ln \frac{j_L}{j_L - j}$$

Pictorial View



$$j = j_0^0 \left(\frac{C_R^*}{C_R^{0*}} e^{\left(\frac{\alpha n F \eta_{act}}{RT} \right)} \right)$$

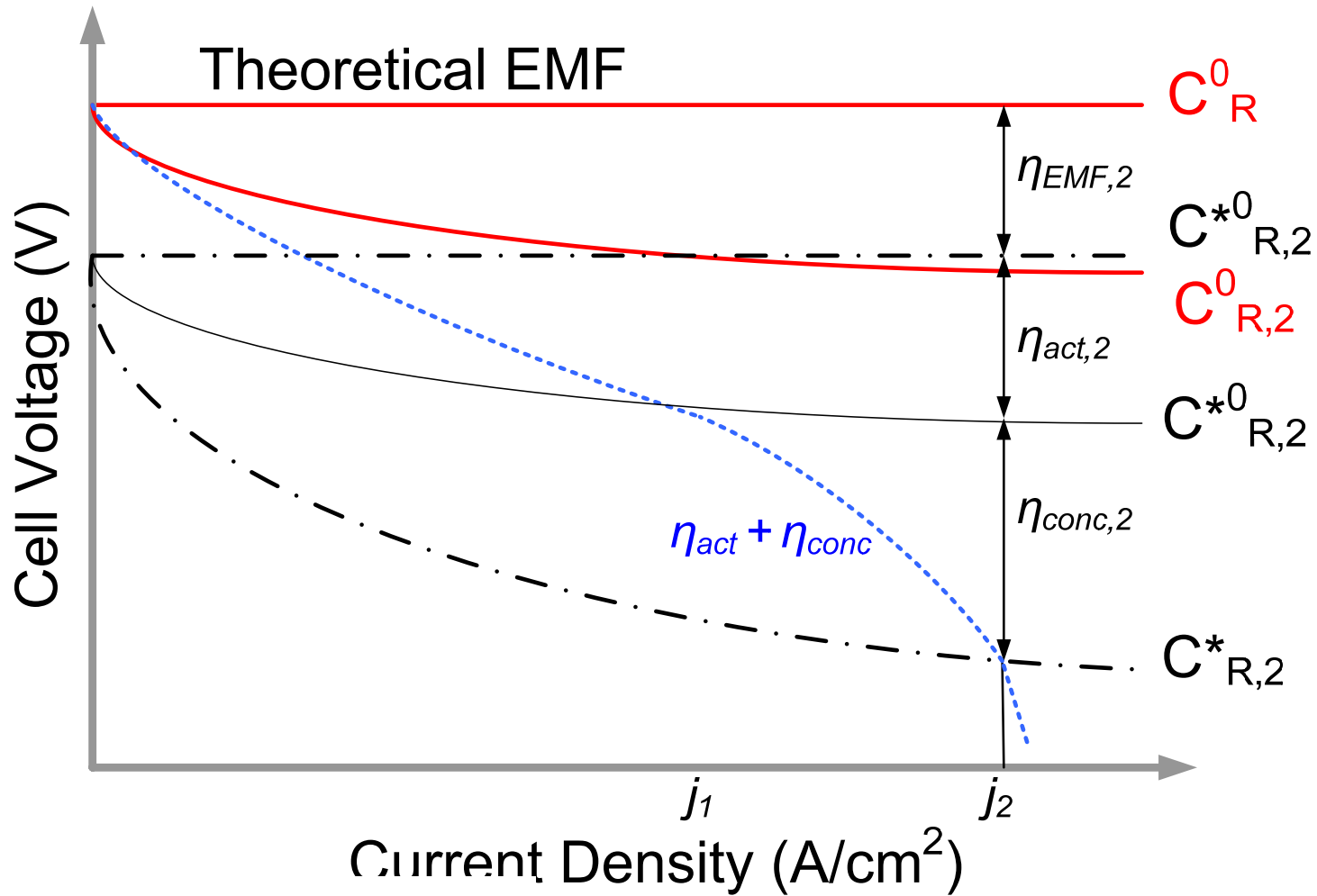
Pictorial View



$$j = j_0^0 \left(\frac{C_R^*}{C_R^{0*}} e^{\left(\frac{\alpha n F \eta_{act}}{RT} \right)} \right)$$

$$E = E^0 - \frac{RT}{nF} \ln \frac{\prod a_{PRODUCTS}^{v_i}}{\prod a_{REACTANTS}^{v_i}}$$

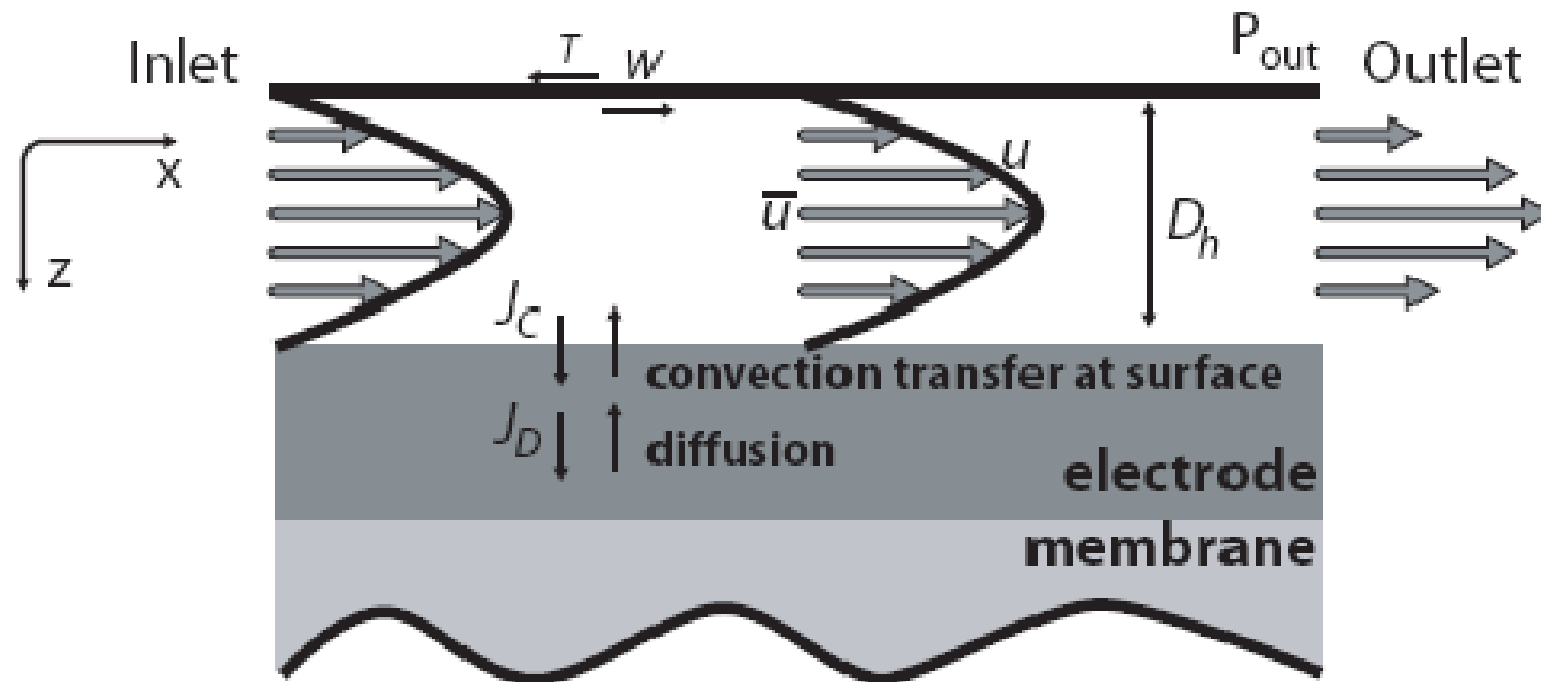
Pictorial View



$$j = j_0^0 \left(\frac{C^*_R}{C^0_{R^*}} e^{\left(\frac{\alpha n F \eta_{act}}{RT} \right)} \right)$$

$$E = E^0 - \frac{RT}{nF} \ln \frac{\prod a_{PRODUCTS}^{v_i}}{\prod a_{REACTANTS}^{v_i}}$$

Convection



$$\tau_{xy} = 2\mu\dot{\epsilon}_{xy} = 2\mu \cdot \frac{1}{2} \left(\frac{\partial u}{\partial y} + \frac{\partial v}{\partial x} \right)$$

Viscosity

Temperature effect

$$\frac{\mu}{\mu_0} \approx \left(\frac{T}{T_0}\right)^n \quad \text{or}$$

$$\frac{\mu}{\mu_0} \approx \left(\frac{T}{T_0}\right)^{1.5} \frac{T_0 + S}{T + S}$$

Gas	$\mu_0(10^{-6}(\text{kg/m}\cdot\text{s}))$	$T_0(\text{K})$	n	S
Air	17.16	273	0.666	111
CO ₂	13.7	273	0.79	222
CO	16.57	273	0.71	136
N ₂	16.63	273	0.67	107
O ₂	19.19	273	0.69	139
H ₂	8.411	273	0.68	97
H ₂ O(vapor)	11.2	350	1.15	1064

Mixture

$$\mu_{mix} = \sum_{i=1}^N \frac{x_i \mu_i}{\sum_{j=1}^N x_j \Phi_{ij}}$$

$$\Phi_{ij} = \frac{1}{\sqrt{8}} \left(1 + \frac{M_i}{M_j}\right)^{-1/2} \left[1 + \left(\frac{\mu_i}{\mu_j}\right)^{1/2} \left(\frac{M_i}{M_j}\right)^{1/4}\right]^2$$

Viscosity: Example

Example 5.1 Consider a fuel cell operating at 80°C . In the cathode, humidified air at 1 atm is supplied with a water vapor mole fraction of 0.2. If the fuel cell employs circular channels with a diameter of 1 mm, find the maximum tolerable air velocity that still ensures laminar flow.

$$\mu_{N_2}|_{80^\circ\text{C}} = \mu_0 \left(\frac{T}{T_0}\right)^n = 16.63 \left(\frac{353.15}{273}\right)^{0.67} = 19.76 \times 10^{-6} \text{ kg/m} \cdot \text{s} \quad (5.37)$$

Similarly, we can obtain $\mu_{O_2}|_{80^\circ\text{C}} = 22.92 \times 10^{-6} \text{ kg/m} \cdot \text{s}$ and $\mu_{H_2O}|_{80^\circ\text{C}} = 11.32 \times 10^{-6} \text{ kg/m} \cdot \text{s}$.

Species	Mole fraction, x_i	Molecular Weight, M_i	Viscosity, $\mu_i(10^{-6} \text{ kg/m} \cdot \text{s})$
1. N_2	$0.8 \times 0.79 = 0.632$	28.02	19.76
2. O_2	$0.8 \times 0.21 = 0.168$	32.00	22.92
3. H_2O	0.200	18.02	11.32

Viscosity: Example

Species i	Species j	M_i/M_j	μ_i/μ_j	Φ_{ij}	$x_j\Phi_{ij}$	$\sum_{j=1}^3 x_j\Phi_{ij}$
1. N ₂	1. N ₂	1.000	1.000	1.000	0.632	1.059
	2. O ₂	0.876	0.862	0.930	0.156	
	3. H ₂ O	1.555	1.746	1.356	0.271	
2. O ₂	1. N ₂	1.142	1.160	1.079	0.682	1.146
	2. O ₂	1.000	1.000	1.000	0.168	
	3. H ₂ O	1.776	2.025	1.482	0.296	
3. H ₂ O	1. N ₂	0.643	0.573	0.776	0.491	0.814
	2. O ₂	0.563	0.494	0.732	0.123	
	3. H ₂ O	1.000	1.000	1.000	0.200	

$$\begin{aligned} \mu_{mix} &= \left(\frac{0.632 \times 19.76}{1.059} + \frac{0.168 \times 22.92}{1.146} + \frac{0.200 \times 11.32}{0.814} \right) \times 10^{-6} \\ &= 17.93 \times 10^{-6} \text{ kg/m} \cdot \text{s} \end{aligned}$$

Viscosity: Example

$$M_{mix} = \sum_{i=1}^N x_i M_i = 0.632 \times 28.02 + 0.168 \times 32.00 + 0.200 \times 18.02 = 26.69 \text{ g/mol}$$

$$\rho = \frac{p}{\frac{R}{M_{mix}}} = \frac{101325 \text{ Pa}}{\frac{8.314 \text{ J/mol}\cdot\text{K}}{0.02669 \text{ kg/mol}} (273.15 + 80) \text{ T}} = 0.921 \text{ kg/m}^3$$

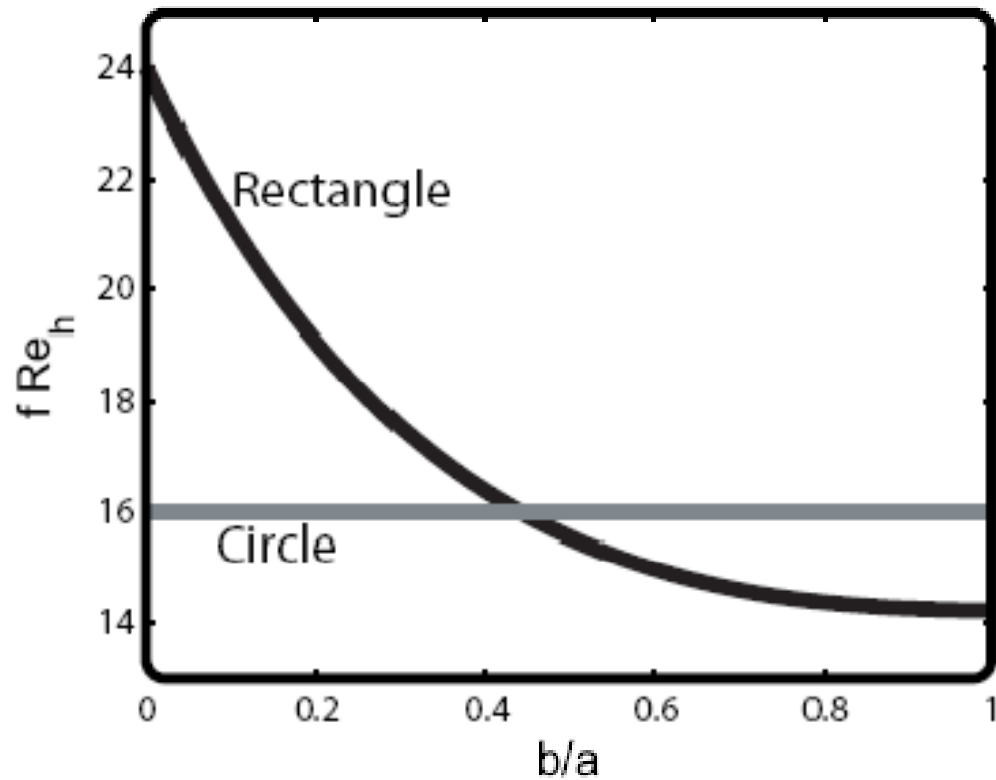
Roughly, laminar flow holds for $Re \sim 2000$, thus:

$$V_{max} = \frac{Re^{max} \mu_{mix}}{\rho L} = \frac{2000 \times (17.93 \times 10^{-6} \text{ kg/m}\cdot\text{s})}{(0.921 \text{ kg/m}^3) \times (0.001 \text{ m})} = 38.03 \text{ m/s} \quad (5.39)$$

This is very fast flow considering the channel is only 1 mm in diameter.

In general, flow in fuel cell is laminar.

Pressure Drop In Flow Channels



$$\frac{dp}{dx} = \frac{4}{D} \bar{\tau}_w$$

$$f = \frac{\bar{\tau}_w}{1/2 \rho \bar{u}^2}$$


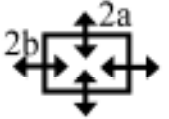

$$Re_h = \frac{\rho \bar{u} D_h}{\mu}$$

$$D_h = \frac{4A}{P} = \frac{4 \times \text{cross section area}}{\text{perimeter}}$$

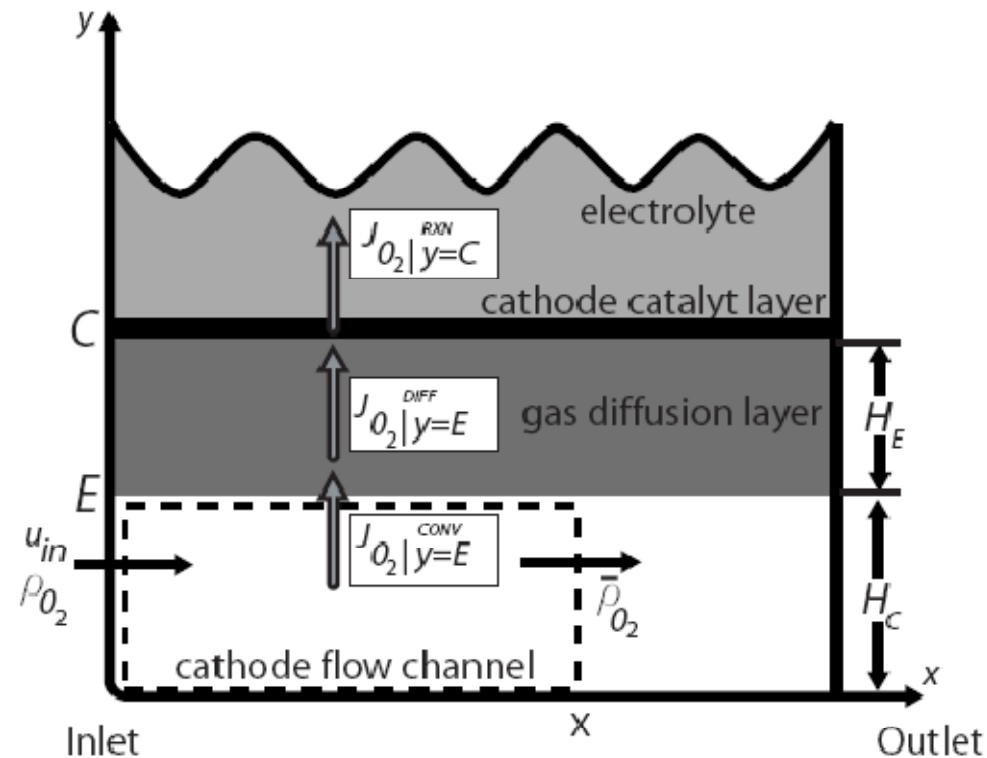
Convective Mass Transport

$$J_{C,i} = h_m (\rho_{i,s} - \bar{\rho}_i)$$

$$h_m = Sh \frac{D_{ij}}{D_h}$$

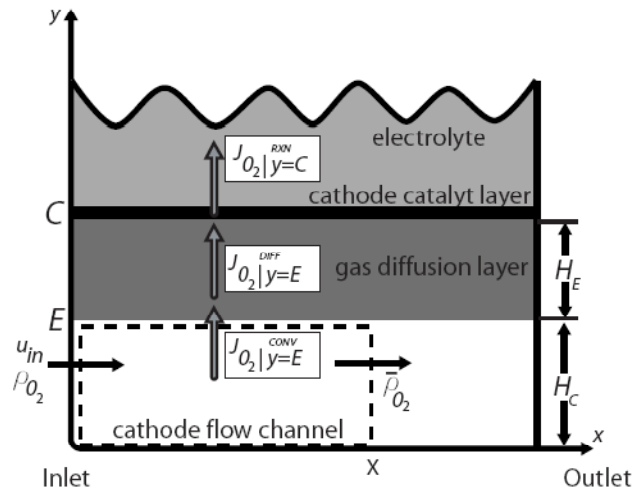
Cross section	α^*								
	0.2	0.4	0.7	1.0	2.0	2.5	5.0	10.0	
	Sh_D	4.36							
	Sh_F	3.66							
	Sh_D	4.80	3.67	3.08	2.97	3.38	3.67	4.80	5.86
	Sh_F	5.74	4.47	3.75	3.61	4.12	4.47	5.74	6.79
	Sh_D	0.83	1.42	2.02	2.44	3.19	3.39	3.91	4.27
	Sh_F	0.96	1.60	2.26	2.71	3.54	3.78	4.41	4.85

Convective Mass Transport in PEMFC



1. The catalyst layer is infinitely thin.¹
2. Water exists only in the vapor form.
3. Diffusive mass transport dominates in the diffusion layer. Furthermore, only y-direction diffusion is considered.
4. Convection dominates in the flow channel.
5. Flow velocity in the channel is constant.

Convective Mass Transport in PEMFC



$$\hat{J}_{O_2}^{rxn}|_{x=X,y=C} = M_{O_2} \frac{j(X)}{4F}$$

$$\hat{J}_{O_2}^{diff}|_{x=X,y=E} = -D_{O_2}^{eff} \frac{\rho_{O_2}|_{x=X,y=C} - \rho_{O_2}|_{x=X,y=E}}{H_E}$$

$$\hat{J}_{O_2}^{conv}|_{x=X,y=E} = -h_m (\rho_{O_2}|_{x=X,y=E} - \bar{\rho}_{O_2}|_{x=X,y=channel})$$

$$\hat{J}_{O_2}^{rxn}|_{x=X,y=C} = \hat{J}_{O_2}^{diff}|_{x=X,y=E} = \hat{J}_{O_2}^{conv}|_{x=X,y=E}$$

$$\hat{J}_{O_2}^{conv}|_{x=X,y=E} = M_{O_2} \frac{j(X)}{4F}$$

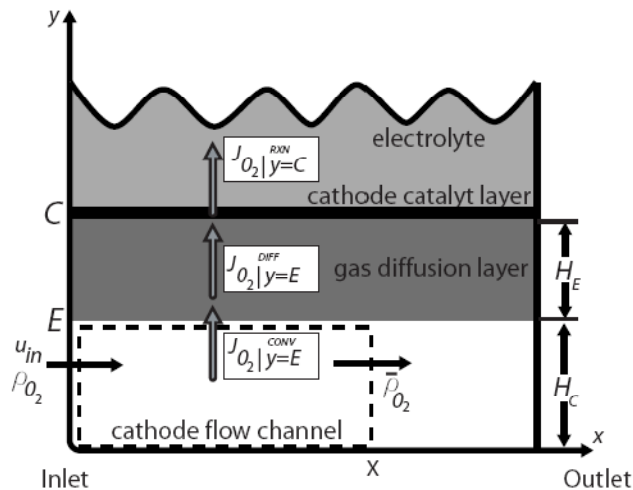
$$\rho_{O_2}|_{x=X,y=C} = \rho_{O_2}|_{x=X,y=E} - M_{O_2} \frac{j(X)}{4F} \frac{H_E}{D_{O_2}^{eff}}$$

$$\rho_{O_2}|_{x=X,y=E} = \bar{\rho}_{O_2}|_{x=X,y=channel} - M_{O_2} \frac{j(X)}{4F} \frac{1}{h_m}$$

Convective Mass Transport in PEMFC

$$u_{in} H_C \bar{\rho}_{O_2} |_{x=0, y=channel} - u_{in} H_C \bar{\rho}_{O_2} |_{x=X, y=channel} = \int_0^X \left(\hat{J}_{O_2} |_{y=E}^{conv} \right) dx \quad (5.6)$$

Amount of gas entering from left
Amount of gas leaving from right
Amount of gas leaving out the top

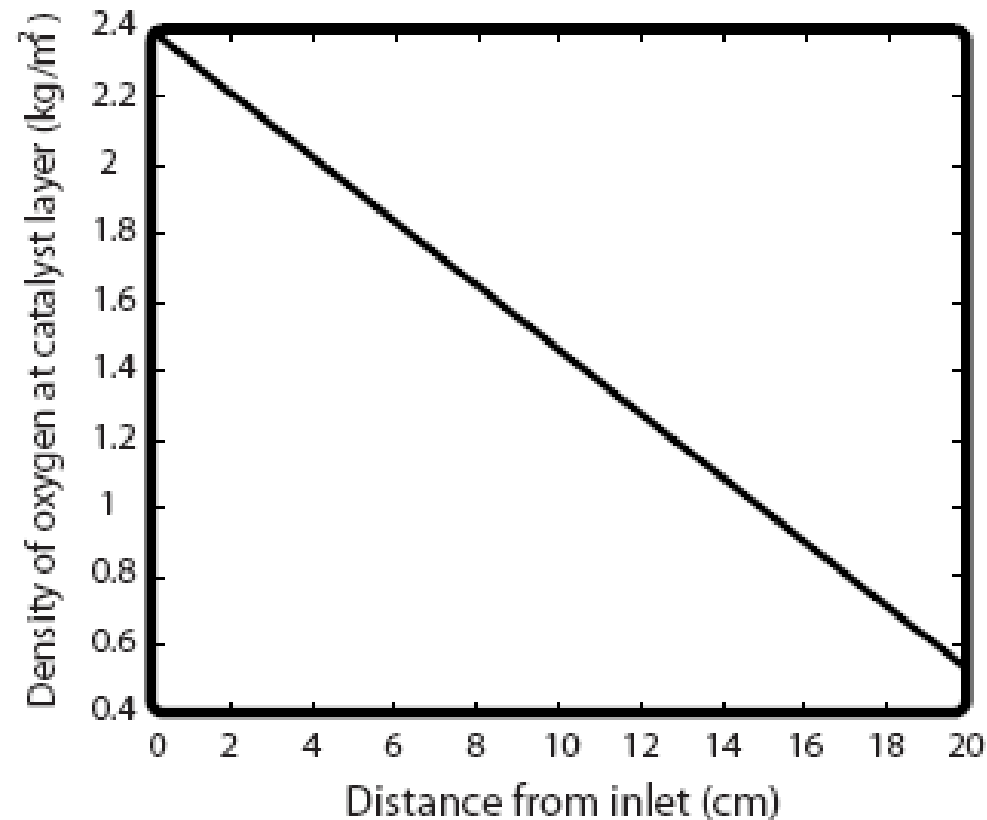


$$\int_0^X \left(\hat{J}_{O_2} |_{y=E}^{conv} \right) dx = \int_0^X \frac{M_{O_2} j(x)}{4F} dx$$

$$\rho_{O_2} |_{x=X, y=C} = \bar{\rho}_{O_2} |_{x=0, y=channel} - \frac{M_{O_2}}{4F} \left(\frac{j(X)}{h_m} + \frac{H_E j(X)}{D_{O_2}^{eff}} + \int_0^X \frac{j(x)}{u_{in} H_C} dx \right)$$

$$\rho_{O_2} |_{x=X, y=C} = \bar{\rho}_{O_2} |_{x=0, y=channel} - M_{O_2} \frac{j}{4F} \left(\frac{1}{h_m} + \frac{H_E}{D_{O_2}^{eff}} + \frac{X}{u_{in} H_C} \right)$$

Convective Mass Transport in PEMFC

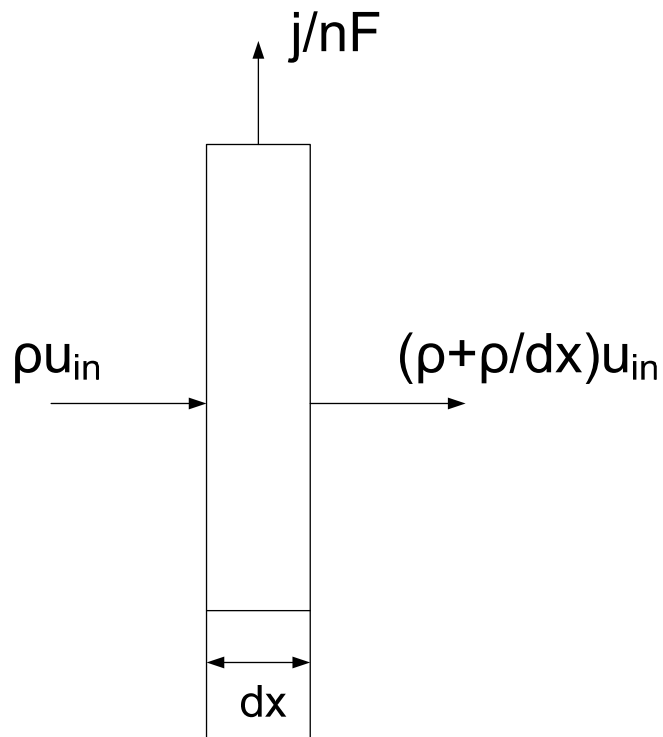


$$\rho_{O_2}|_{x=X,y=C} = \bar{\rho}_{O_2}|_{x=0,y=channel} - M_{O_2} \frac{j}{4F} \left(\frac{H_C}{Sh_F D_{O_2}} + \frac{H_E}{D_{O_2}^{eff}} + \frac{X}{u_{in} H_C} \right)$$

$$h_m = \frac{Sh_F D_{O_2}}{H_C} \quad N_{total} = u_{in} H_C = constant$$

A More Realistic Alternative Solution

$$\rho_{O_2}|_{x=X,y=C} = \bar{\rho}_{O_2}|_{x=0,y=channel} - \frac{M_{O_2}}{4F} \left(\frac{j(X)}{h_m} + \frac{H_E j(X)}{D_{O_2}^{eff}} + \int_0^X \frac{j(x)}{u_{in} H_C} dx \right)$$



$$u_{in} \rho - u_{in} \left(\rho + \frac{\rho}{dx} \right) = \frac{j(x)}{nF} dx$$

$$\frac{d\rho}{dx} = - \frac{j(x)}{u_{in} nF}$$

$$\frac{dC(x)}{dx} = -Bj(x)$$

A More Realistic Alternative Solution

If constant Voltage instead of constant current is assumed

$$j = j_0^0 \frac{c_R^*}{c_R^{0*}} e^{\alpha n F \eta / RT}$$

$$\frac{dc(x)}{dx} = -Bj(x) = -Bj_0^0 \frac{c(x)}{c^0} e^{\alpha n F \eta / RT} = ac(x)$$

$$c(x) = Ae^{-ax}$$

$$c(0) = c_{in} = A$$

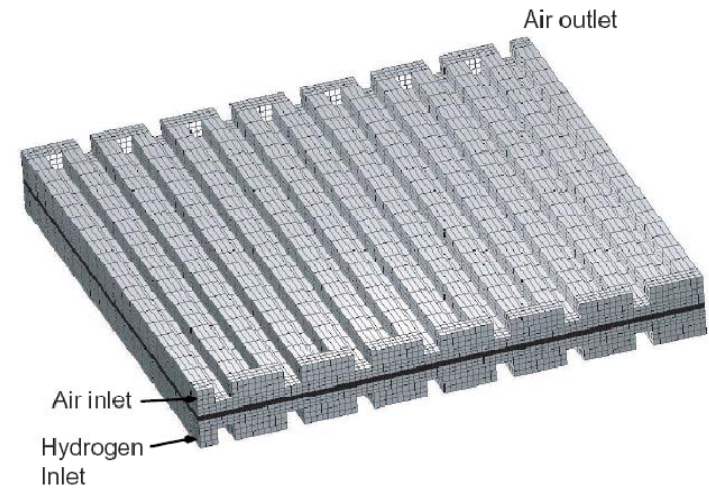
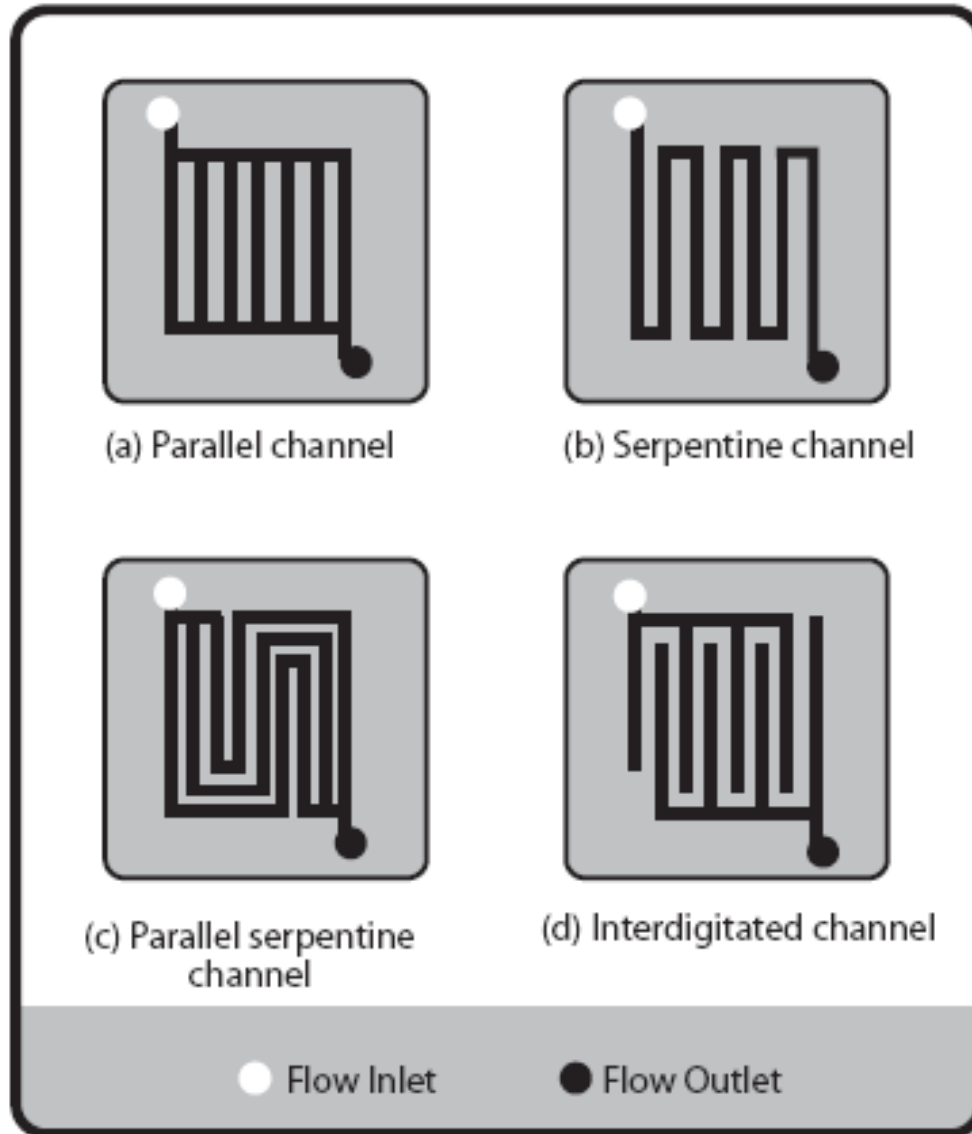
$$j(x) = Ke^{-ax}$$

- Current drop Exponentially
- Concentration also drop exponentially
- HW 5.10

Flow Channel Design

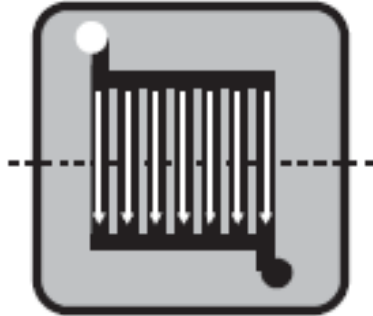
- High electrical conductivity
- High corrosion resistance
- High chemical compatibility
- High thermal conductivity
- High gas tightness
- High mechanical strength
- Low weight and volume
- Ease of manufacturability
- Cost effectiveness

PEMFC Flow Channel Design



PEMFC Flow Channel Design

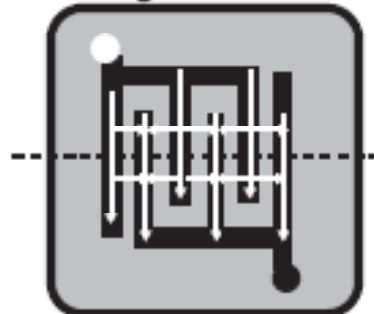
(a) Parallel channel



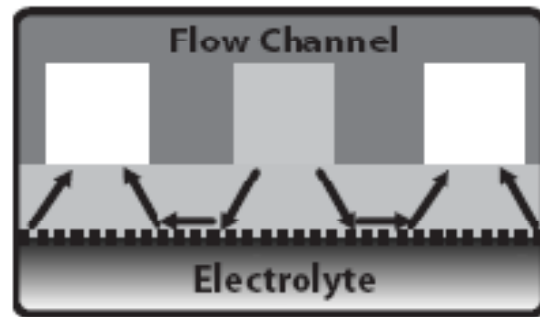
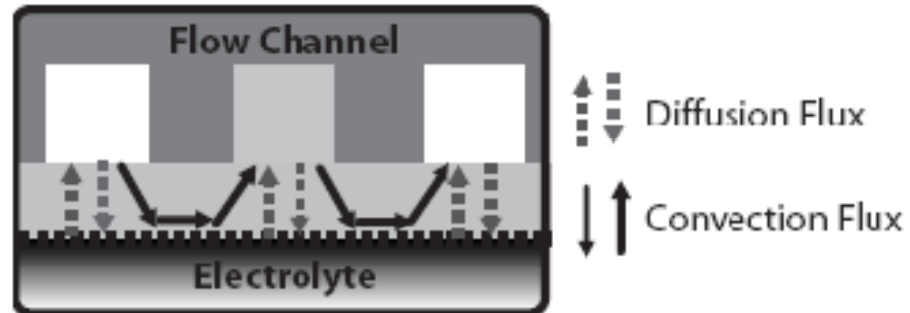
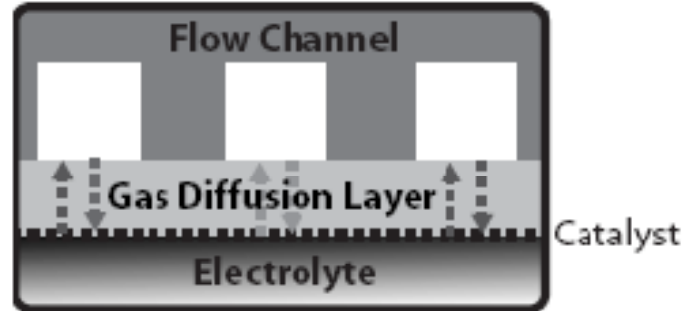
Serpentine channel



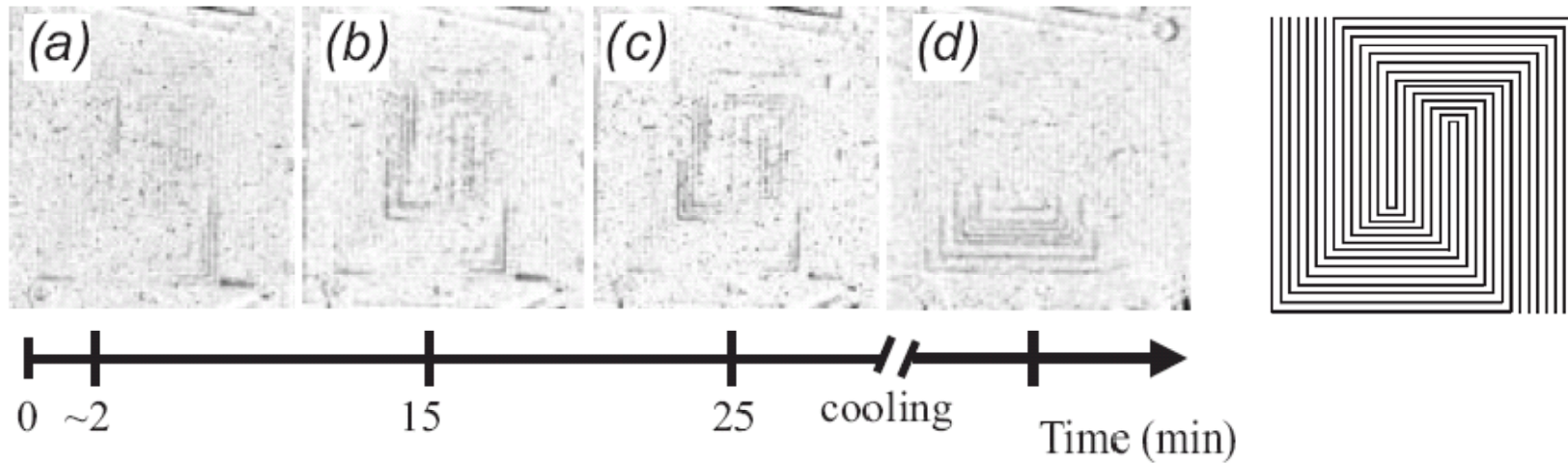
Interdigitated channel



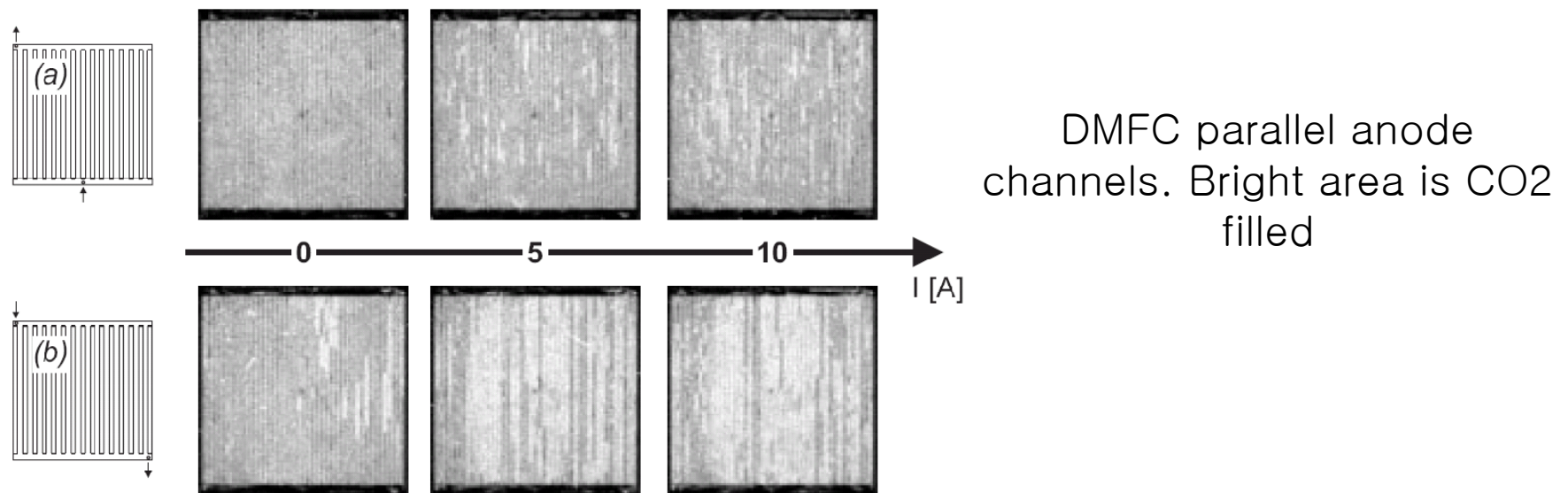
(b)



Flow Visualization (Neutron Radiography)



PEMFC serpentine cathode channels. Dark area is liquid filled.



Flow Visualization

