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Estimation Methods of GHG Emissions from Waste Sectors



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BACKGROUND

Waste Sector Categories

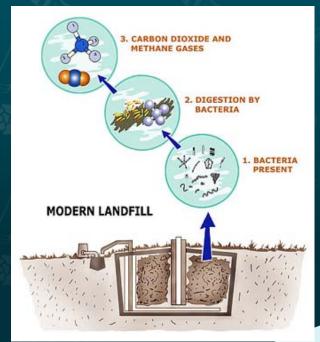
Waste Sector is composed of Solid Waste Disposal on Land(SWDL), Wastewater Handling(WH), Waste Incineration(WI).

Recently, Biological
 Treatment of Solid Waste
 (ex, composting) is added.



Background on Waste Sector

- Solid waste disposal on land(SWDL)
 - The final disposal of Solid Waste(SW) by placing it in a controlled fashion in a place intended to be permanent. This term is used for both controlled dumps and sanitary landfills.
 - Secure Landfill?
 - SWDL is divided into managed waste disposal on land, and unmanaged waste disposal sites.

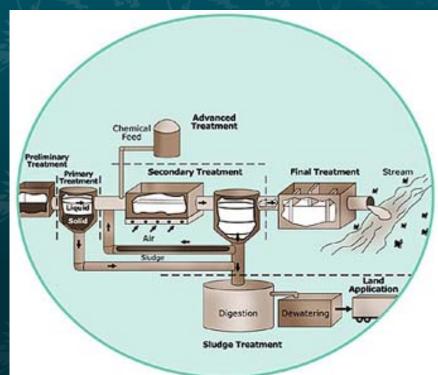


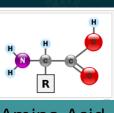
Background on Waste Sector

Wastewater handling(WH)

- The process of wastewater treatment under controlled conditions to reduce its concentrations of contaminants (BOD, SS, etc.)
- WH is divided into <u>industrial</u> <u>wastewater</u>, and <u>domestic and</u> <u>commercial wastewater</u>.

 N₂O emissions are made from the human sewage component of domestic wastewater: Protein





Amino Acid

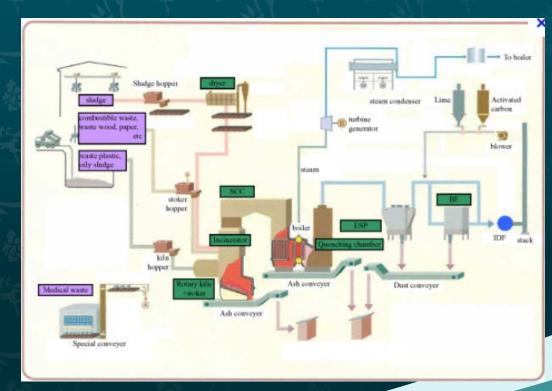
Background on Waste Sector

Waste incineration(WI)

 The process of burning SW under controlled conditions to reduce its weight and volume, and often to produce energy

Incinerated wastes?

 Waste incinerated is divided into wastes of biogenic origin, and those of non-biogenic origin(e.g. plastics, rubber, certain textiles, waste oil, etc.).



Major Processes Related to GHG Emissions

Emission producing Processes	SWDL	WWH	WI
Methanogenesis	CH₄	CH₄	
Nitrification and Denitrification		N ₂ O	
Incineration			CO ₂ , CH ₄ , N ₂ O
General Decomposition	CO ₂	CO ₂	

Methanogenesis

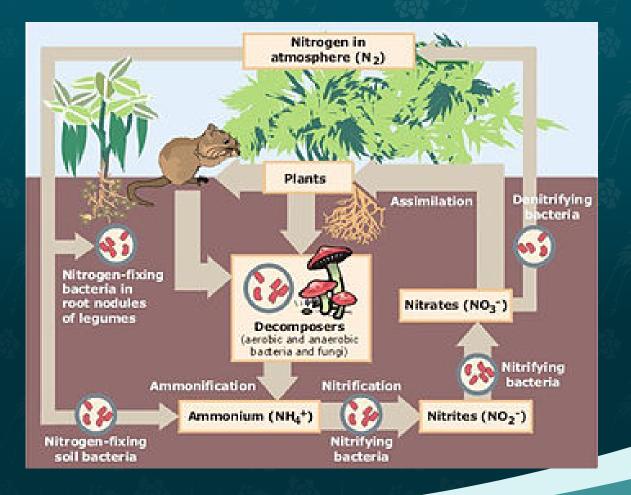
- The process by which methanogenic bacteria breakdown waste to produce methane. The process only occurs under <u>anaerobic</u> <u>conditions</u>
- The amount of methane produced through methanogenesis is related to the balance between aerobic and anaerobic processes and is therefore influences by waste management practices, waste composition, and physical factors(e.g., moisture, temperature, and pH)

Waste subsector: SWDL and WH

- Methanogenesis occurs in land disposal sites for solid waste where near optimal anaerobic conditions are often created.
- Methane is also produced in the process of <u>wastewater handling</u> when the wastewater undergoes anaerobic or partially anaerobic treatments.

Nitrification/Denitrification

- The biological process of nitrification and denitrification produces N₂O
 - Nitrification : Oxidation of ammonia to nitrite(NO₂⁻) and then nitrate(NO₃⁻) by microorganisms; NH₃→N₂O→NO⁻₂→NO⁻₃
 - Denitrification : Reduction of nitrate to nitrogen(N₂) by microorganisms; $NO_3 \rightarrow NO_2 \rightarrow N_2O \rightarrow N_2$
- The process can occur during WH and it especially important to consider when waste flows have relatively high nitrogen contents.



Incineration

The characteristics of the incineration(combustion) process, the technologies used, and composition of the waste itself are important parameters determining emissions.

Contents

(Session1) Solid Waste Disposal on Land
 (Session2) Wastewater Incineration

(Session 1) Solid Waste Disposal on Land

(Session 1) 1.1. Landfill Processes

Landfill Process

LFG



Solid Wastes



Transfer Station



Electricity Generation Facility



Landfill Gas Collection and Transfer



Landfills

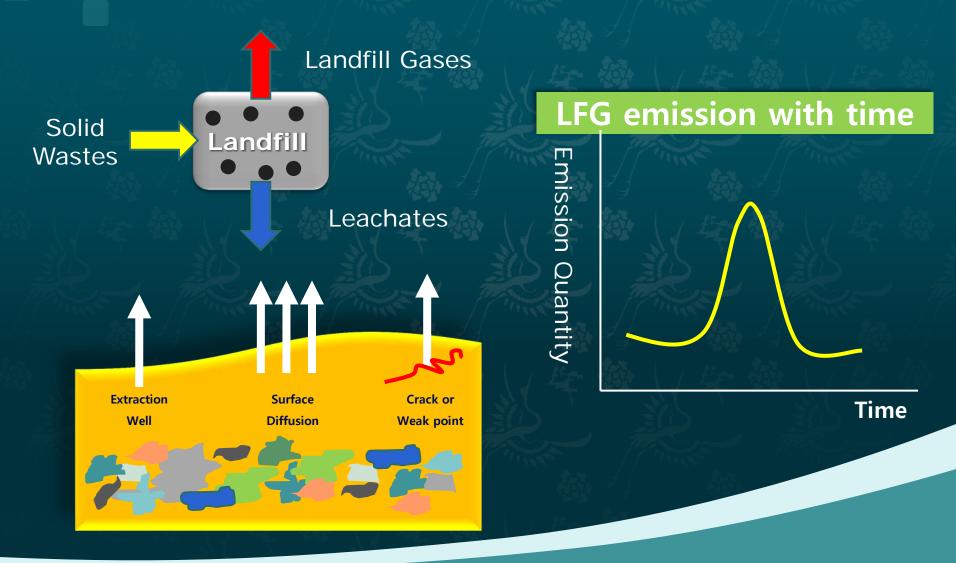
Leachates

Landfill Process

Process		Explanation	
Inlet processes of SW	[1] Transfer Station	 Pre-treatment processes performing separation of recyclables and compaction of SW before transferring to landfill 	
	[2] Weighing Station	 Weighing station of SW for landfills 	
Landfilling	[3] Landfill Site	 Landfilling of SW, non-combustibles that cannot be recycled, inorganic residues from incineration process 	
Leachate Treatment	[4] Leachate Treatment	 Treatment process of leachates (<i>in situ</i> primary treatment) Transferring to domestic and commercial wastewater treatment plant 	
LFG Treatment	[5] Collection and Transfer of LFG	 Collection of landfill gas(LFG) and transfer to energy recovery system 	
	[6] Electricity Generation	 Generation of electricity and heat energy from LFG 	

(Session 1) 1.2. Methane Emission from SWDL

♦ Landfills are large anthropogenic sources of methane(CH_4) emissions which result from the decomposition of organic landfill materials such as paper, food scraps, and yard trimmings.



- The decomposition process primarily results from biological activity through which microorganisms derive energy. After being placed in a landfill, organic waste is initially digested by aerobic bacteria, which break down organic matter into substances such as <u>cellulose</u>, <u>amino acids</u>, <u>and sugars</u>.
- These substances are further broken down through fermentation into gases and short-chain organic compounds that form the substrates for the growth of methanogenic bacteria.
- Methane-producing anaerobic bacteria convert these fermentation products into stabilized organic materials and biogas consisting of approximately 50% CO₂ and 50% CH₄ by volume(less than 1% of NMVOCs).
 - The percentage of CO₂ in biogas may be smaller because some CO₂ dissolves in leachates.
- Methane production typically begins one or two years after waste disposal in a modern landfills and may last from 10 to 60 years.

Degradation Processes of Organic Wastes in Landfill

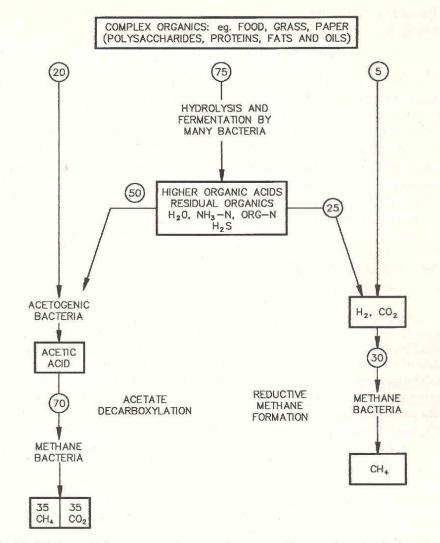


Fig. 4.3 Biodegradation pattern for organic matter. The numbers in circles give the percent of carbon.

Phases of LFG and Leachates Generation

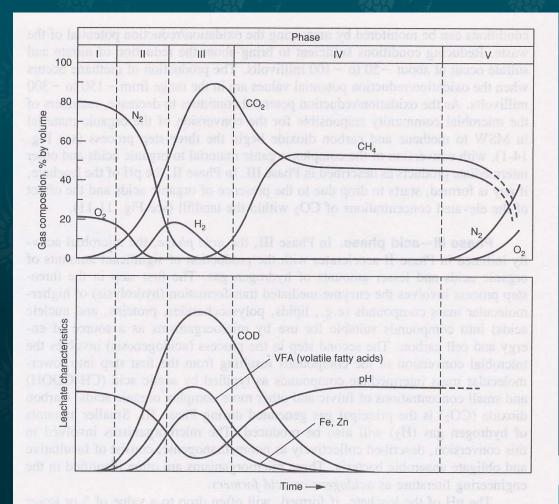


FIGURE 11-11

Generalized phases in the generation of landfill gases (I = initial adjustment, II = transition phase, III = acid phase, IV = methane fermentation, and V = maturation phase). (Adapted from Refs. 13, 34, 37, and 38.)

Phase I : Initial Adjustment

 Organic biodegradable components in MSW(Municipal Solid Waste) undergo microbial decomposition under <u>aerobic</u> <u>conditions</u> because of air trapped within landfill.

Phase II : Transition Phase

Oxygen is depleted and anaerobic conditions begin to develop.
 Organic materials are converted into organic acids and other intermediate products.

Phase III : Acid Phase

 Microbial activities accelerate the biological degradations with the production of significant amounts of organic acids and lesser amounts of hydrogen gas.

- 1st step : Breakdown of high molecular weight compounds into smaller ones suitable for a source of energy and cell carbon
- 2nd step : Further breakdown process to generate acetic acid and other smaller molecular weight compounds
- 3rd step : CO₂ and H₂ would be generated from acetic ac

Phase IV : Methane Fermentation Phase

Anaerobic microorganisms convert the acetic acid and hydrogen gas formed in acid phase to CH₄ and CO₂.

Both methane and acid formation proceed simultaneously, although the rate of acid formation is considerably reduced.

♦ pH will rise to more neutral values in the range of 6.8 to 8. $4H_2 + CO_2 \rightarrow CH_4 + 2H_2O$

CH₃COOH- CO_{2}

Phase V : Maturation Phase

- Maturation phase occurs after the readily available biodegradable organic material had been converted to CH₄ and CO₂.
- The rate of LFG generation diminishes significantly because most of the available nutrients have been removed with the leachate.
- The principal LFG evolved in this phase are CH_4 and CO_2 .
- The leachate will often remain humic and fulvic acids, which are difficult to process further biologically.

Methane emissions from landfills are a function of several factors, including:

- the total amount of MSW in landfills, which is related to total MSW landfilled annually for <u>the last 50 years</u>;
- the amount of methane that is recovered and either flared or used for energy purpose; and
- the amount of methane oxidized in landfills instead of being released into the atmosphere.

Key parameters for estimating methane emissions from solid waste disposal on land are 1) the quantity of carbon contained in the waste, 2) the process, and 3) the moisture content. Most solid waste contains a mixture of biogenic and non-biogenic components, and the carbon in the nonbiogenic waste (ex; plastics) is generally not able to be decomposed by anaerobic bacteria.

Treatment and Utilization of LFG

♦ Treatment : Flaring



Treatment and Utilization of LFG

 \diamond Measuring CH₄ flow rate and collection samples



Treatment and Utilization of LFG

Utilization

- Electricity Generation
 - Gas Engine, Gas Turbine, Steam Turbine
- Medium Quality Gas: Direct use of LFG as a fuel after minimum pre-treatments (Removing dust and siloxane)
- High Quality Gas : CLG, Fuel Cell
 - CH_4 concentration > 95%
 - CH_4 in LFG : 50%
 - LFG Heating value : 4,000~5,000 kcal/Nm³ (Half of LNG)
 - Separation and Purification Process are necessary to obtain high-quality gas

(Session 1) 1.3. Estimation of Methane Emission from SWDL

Estimation Methodologies

GHG Emission Rate(tCO₂-eq/yr) =

Activity Data X Emission Factor

Activity Data

 Data on the <u>magnitude of human activity</u> resulting in emissions or removals taking place during a given period of time.

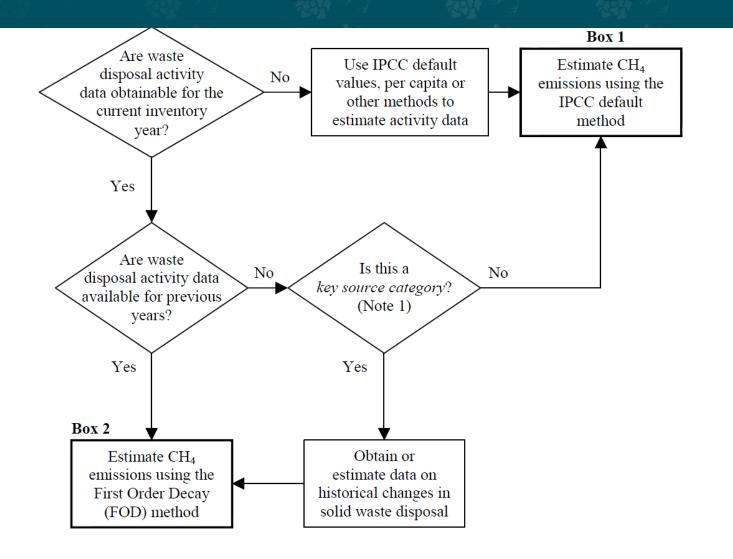
Emission Factor

 A coefficient that relates the activity data to the amount of chemical compound which is the source of later emissions

Emission quantity per unit activity data

Estimation Methodologies ♦ IPCC(1996) guideline showed two methodologies; ◆ Tier 1 : Default Method Tier 2 : First Order Decay Method ◆Tier: Level of Complexity of Estimation Methods; higher tier method is likely to be more accurate but does not guarantee the accurate r

Decision Tree for Choosing Methodology



Tier 1; Default Method

$$Q_{CH_4} = \left(MSW_T \times MSW_F \times MCF \times DOC \times DOC_F \times F \times 16/12 - R\right) \times (1 - OX)$$

- where Q_{CH_4} = Methane Generation(Gg/yr),
- $MSW_T = Total MSW generated (Gg/yr),$
- MSW_F = Fraction of MSW disposed to landfills (Fraction)
- MCF = Methane Correction Factor (Fraction)
- DOC = Degradable Organic Carbon (Fraction)
- DOC_F = Fraction DOC dissimilated
- F = Fraction of methane in landfill gas
- R = Recovered methane (Gg/yr)
- OX = Oxidation Factor (Fraction)

MSW_T and MSW_F

♦ MSW_T

 Total MSW(MSW_T) can be calculated from population and per capita annual generation rate of MSW.

♦MSW_F

 Fraction of landfill treatment can be calculated from population and per capita annual landfill rate of MSW

MSW Generation Data

TABLE 2.1 MSW GENERATION AND TREATMENT DATA - REGIONAL DEFAULTS							
Region	MSW Generation Rate ^{1, 2, 3} (tonnes/cap/yr)	Fraction of MSW disposed to SWDS	Fraction of MSW incinerated	Fraction of MSW composted	Fraction of other MSW management, unspecified ⁴		
Asia							
Eastern Asia	0.37	0.55	0.26	0.01	0.18		
South-Central Asia	0.21	0.74	-	0.05	0.21		
South-East Asia	0.27	0.59	0.09	0.05	0.27		
Africa ⁵	0.29	0.69	-	-	0.31		
Europe							
Eastern Europe	0.38	0.90	0.04	0.01	0.02		
Northern Europe	0.64	0.47	0.24	0.08	0.20		
Southern Europe	0.52	0.85	0.05	0.05	0.05		
Western Europe	0.56	0.47	0.22	0.15	0.15		
America	America						
Caribbean	0.49	0.83	0.02	-	0.15		
Central America	0.21	0.50	-	-	0.50		
South America	0.26	0.54	0.01	0.003	0.46		
North America	0.65	0.58	0.06	0.06	0.29		
Oceania ⁶	0.69	0.85	-	-	0.15		

¹ Data are based on weight of wet waste.

² To obtain the total waste generation in the country, the per-capita values should be multiplied with the population whose waste is collected. In many countries, especially developing countries, this encompasses only urban population.

³The data are default data for the year 2000, although for some countries the year for which the data are applicable was not given in the reference, or data for the year 2000 were not available. The year for which the data are collected, where available, is given in the Annex 2A.1.

Other, unspecified, includes data on recycling for some countries.

⁵ A regional average is given for the whole of Africa as data are not available for more detailed regions within Africa.

6 Data for Oceania are based only on data from Australia and New Zealand.

MCF

(Methane Correction Factor)

MCF reflects the way in which MSW is managed and the effect of management practices on CH_4 generation.

TABLE 3.1 SWDS CLASSIFICATION AND METHANE CORRECTION FACTORS (MCF)						
Type of Site	Methane Correction Factor (MCF) Default Values					
Managed – anaerobic ¹	1.0					
Managed – semi-aerobic ²	0.5					
Unmanaged ³ – deep (>5 m waste) and /or high water table	0.8					
Unmanaged ⁴ – shallow (<5 m waste)	0.4					
Uncategorised SWDS 5	0.6					

Anaerobic managed solid waste disposal sites: These must have controlled placement of waste (i.e., waste directed to specific deposition areas, a degree of control of scavenging and a degree of control of fires) and will include at least one of the following: (i) cover material; (ii) mechanical compacting; or (iii) levelling of the waste.

² Semi-aerobic managed solid waste disposal sites: These must have controlled placement of waste and will include all of the following structures for introducing air to waste layer: (i) permeable cover material; (ii) leachate drainage system; (iii) regulating pondage; and (iv) gas ventilation system.

³ Unmanaged solid waste disposal sites – deep and/or with high water table: All SWDS not meeting the criteria of managed SWDS and which have depths of greater than or equal to 5 metres and/or high water table at near ground level. Latter situation corresponds to filling inland water, such as pond, river or wetland, by waste.

⁴ Unmanaged shallow solid waste disposal sites; All SWDS not meeting the criteria of managed SWDS and which have depths of less than 5 metres.

⁵ Uncategorised solid waste disposal sites: Only if countries cannot categorise their SWDS into above four categories of managed and unmanaged SWDS, the MCF for this category can be used.

Sources: IPCC (2000); Matsufuji et al. (1996)

Degradable Organic Carbon (DOC)

♦ DOC is the organic carbon in waste that is accessible to biochemical decomposition, and should be expressed as Gg C per Gg waste. The DOC in bulk waste is estimated based on the composition of waste and can be calculated from a weighted average of the degradable carbon content of various components of the waste stream.

DOC

	TABLE 2.4					
DEFAULT DRY MATTER CONTENT, DOC CONTENT	F, TOTAL CARBON CONTENT AND FOSSIL CARBON FRACTION OF					
DIFFERENT MSW COMPONENTS						

	-								
MSW component	Dry matter content in % of wet weight ¹	DOC content in % of wet waste		DOC content in % of dry waste		Total carbon content in % of dry weight		Fossil carbon fraction in % of total carbon	
	Default	Default	Range	Default	Range ²	Default	Range	Default	Range
Paper/cardboard	90	40	36 - 45	44	40 - 50	46	42 - 50	1	0 - 5
Textiles ³	80	24	20 - 40	30	25 - 50	50	25 - 50	20	0 - 50
Food waste	40	15	8 - 20	38	20 - 50	38	20 - 50	-	-
Wood	85 ⁴	43	39 - 46	50	46 - 54	50	46 - 54	-	-
Garden and Park waste	40	20	18 - 22	49	45 - 55	49	45 - 55	0	0
Nappies	40	24	18 - 32	60	44 - 80	70	54 - 90	10	10
Rubber and Leather	84	(39) 5	(39) ⁵	(47) 5	(47) ⁵	67	67	20	20
Plastics	100	-	-	-	-	75	67 - 85	100	95 - 100
Metal ⁶	100	-	-	-	-	NA	NA	NA	NA
Glass ⁶	100	-	-	-	-	NA	NA	NA	NA
Other, inert waste	90	-	-	-	-	3	0 - 5	100	50 - 100

The moisture content given here applies to the specific waste types before they enter the collection and treatment. In samples taken from collected waste or from e.g., SWDS the moisture content of each waste type will vary by moisture of co-existing waste and weather during handling.

² The range refers to the minimum and maximum data reported by Dehoust *et al.*, 2002; Gangdonggu, 1997; Guendehou, 2004; JESC, 2001; Jager and Blok, 1993; Würdinger *et al.*, 1997; and Zeschmar-Lahl, 2002.

⁹ 40 percent of textile are assumed to be synthetic (default). Expert judgement by the authors.

⁴ This value is for wood products at the end of life. Typical dry matter content of wood at the time of harvest (that is for garden and park waste) is 40 percent. Expert judgement by the authors.

⁵ Natural rubbers would likely not degrade under anaerobic condition at SWDS (Tsuchii et al., 1985; Rose and Steinbüchel, 2005).

⁶ Metal and glass contain some carbon of fossil origin. Combustion of significant amounts of glass or metal is not common.

DOC_F (DOC Dissimilated)

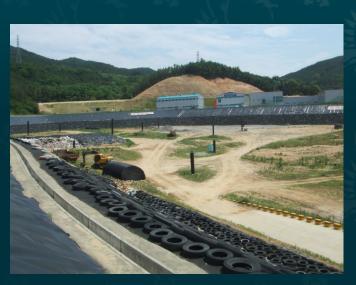
Fraction of degradable organic carbon which decomposes is an estimate of the fraction of carbon that is ultimately degraded and released from SWDS, and reflects the fact that some degradable organic carbon does not degrade, or degrades very slowly, under anaerobic conditions.

Default DOC_F is 0.5 which depends on many factors like temperature, moisture, pH, composition of waste, etc.

F(Fraction of CH_4 in LFG) Most waste in SWDS generated a gas with approximately 50 percent CH_{4} . Only material including substantial amounts of fat or oil can generate gas with substantially more than 50%. The IPCC default value is 0.5.

R(Recovered Amount of CH_{4})

LFG can be collected through LFG extraction well.





OX(Oxidation Factor)

- The OX reflects the amount of CH₄ from SWDS that is oxidized in the soil or other material covering the waste.
- CH₄ oxidation is achieved by methanotrophic microorganisms in cover soils and can range from negligible to 100% of internally produced CH₄.
 Sanitary, well-managed SWDS tend to have higher oxidation rates than unmanaged dump sites.

TABLE 3.2 OXIDATION FACTOR (OX) FOR SWDS						
Type of Site	Oxidation Factor (OX) Default Values					
Managed ¹ , unmanaged and uncategorised SWDS	0					
Managed covered with CH ₄ oxidising material ²	0.1					
¹ Managed but not covered with aerated material ² Examples: soil, compost						

Tier 2 : First Order Decay (FOD) Method

$$G_{CH_4}(t) = \sum_{x} \left\{ \left(A \times k \times MSW_T(x) \times MSW_F(x) \times L_0(x) \times e^{-k(t-x)} \right) \right\}$$
$$E_{CH_4}(t) = \sum_{x} \left(Q_{CH_4} - R(t) \right) \bullet \left(1 - OX \right)$$

where t = year of inventory

x = years for which input data should be added

A = $(1-e^{-k})/k$; normalization factor which corrects the summation k = methane generation rate constant(yr⁻¹)

- MSW_T = Total municipal solid waste generated in year x (Gg/yr)
- MSW_F = Fraction of MSW disposed at SWDS in year x
- $L_0(x) = Methane generation potential (MCF(x) \cdot DOC(x) \cdot DOC_F \cdot F \cdot 16/12)$

Tier 2 : FOD Method

$$Q_{CH_4}(t) = M_0 L_0 \left(e^{-k(t-1)} - e^{-kt} \right)$$

where

- t = year of inventory
- $M_0(x)$ = Landfilled amount at a year x
- $L_0(x) = Methane generation potential (MCF · DOC · DOC_F · F · 16/12)$ landfilled at a year x

Tier 2 : FOD Method

Decomposition kinetics of MSW:

$$\frac{dM_{w}}{dt} = -kM_{w}$$

• Generation kinetics of CH_4 : $\frac{dM_{CH_4}}{dt}$

$$- = -L_0 \frac{\partial M_w}{\partial t} = kL_0 M_w = kL_0 M_0 \exp(-k \cdot t)$$

✤ Total generation quantity until year t:

 $\mathcal{M}_{C\mathcal{H}_4}(t) = \mathcal{M}_0 \mathcal{L}_0(1 - e^{-k \cdot t})$

✤ Generation quantity during one year from *t*-1 to *t*:

$$Q_{CH_4}(t) = M_0 L_0 \left(e^{-k(t-1)} - e^{-kt} \right)$$

Tier 2 : FOD Method

TABLE 3.3

Recommended default methane generation rate (k) values under Tier 1

(Derived from k values obtained in experimental measurements, calculated by models, or used in greenhouse gas inventories and other studies)

Type of Waste		Climate Zone*								
		Boreal and Temperate (MAT ≤ 20°C)				Tropical ¹ (MAT > 20°C)				
		Dry (MAP/PET < 1)		Wet (MAP/PET > 1)		Dry (MAP < 1000 mm)		Moist and Wet (MAP ≥ 1000 mm)		
		Default	Range ²	Default	Range ²	Default Range ²		Default	Range ²	
Slowly	Paper/textiles waste	0.04	0.03 ^{3,5} – 0.05 ^{3,4}	0.06	0.05 - 0.07 ^{3,5}	0.045	0.04 - 0.06	0.07	0.06 - 0.085	
degrading waste	Wood/ straw waste	0.02	0.01 ^{3,4} – 0.03 ^{6,7}	0.03	0.02 - 0.04	0.025	0.02 - 0.04	0.035	0.03 - 0.05	
Moderately degrading waste	Other (non – food) organic putrescible/ Garden and park waste	0.05	0.04 – 0.06	0.1	0.06 - 0.1 ⁸	0.065	0.05 - 0.08	0.17	0.15 - 0.2	
Rapidly degrading waste	Food waste/Sewage sludge	0.06	0.05 - 0.08	0.185 ⁴	0.1 ^{3.4} - 0.2 ⁹	0.085	0.07 – 0.1	0.4	0.17 – 0.7 ¹⁰	
Bulk Waste		0.05	0.04 - 0.06	0.09	0.08 ⁸⁻ 0.1	0.065	0.05 - 0.08	0.17	0.15 ¹¹ – 0.2	

Uncertainties of the default parameters in the IPCC Methods

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Parameter	Uncertainty Range ^b				
	Country-specific:				
Total Municipal Solid Waste (MSW _T) and Fraction of MSW sent to SWDS (MSW _F)	> $\pm 10\%$ (<-10%, >+10%. The absolute value of the uncertainty range is greater than 10%.) for countries with high quality data (e.g. weighing at all SWDS)				
	For countries with poor quality data: more than a factor of two.				
Degradable Organic Carbon (DOC) = 0.21 (maximal default value in the <i>IPCC Guidelines</i>)	-50%, +20%				
Fraction of Degradable Organic Carbon Dissimilated $(DOC_F) = 0.77$	-30%, +0%				
Methane Correction Factor (MCF)					
= 1	-10%, +0%				
= 0.4	-30%, +30%				
= 0.6	-50%, +60%				
Fraction of CH_4 in Landfill Gas (F) = 0.5	-0%, +20%				
Methane Recovery (R)	The uncertainty range will depend on how the amounts of CH ₄ recovered and flared or utilised are estimated, but the uncertainty is likely to be relatively small compared to other uncertainties if metering is in place.				
Oxidation Factor (OX)	Include OX in the uncertainty analysis if a value other than zero has been used for OX itself. In this case the justification for a non-zero value should include consideration of uncertainties, as specified in Section 5.1.1.2, Choice of Emission Factors and Activity Data.				
Methane Generation Rate Constant (k) = 0.05	-40%, +300%				
* The estimates are valid only for the default values given in the <i>IPCC Guidelines</i> or in the table, and are based on expert judgement.					

^a The estimates are valid only for the default values given in the *IPCC Guidelines* or in the table, and are based on expert judgement.
^b If the evaluation of additional data on the parameters provides data for the revision of the default values, the uncertainty range should also be changed. When country-specific values are used, they should be accompanied with appropriate uncertainty values.

Source: Judgement by Expert Group (see Co-chairs, Editors and Experts; CH4 Emissions from Solid Waste Disposal).

(Session 1) 1.4. Quality Assurance, Quality Control

Estimate of the emissions using different approaches

If the emissions are estimated with the FOD method, inventory agencies should also estimate them with the IPCC default method.

 The results can be useful for cross-comparison with other countries.

 Inventory agencies should record the results of such comparisons for internal documentation, and investigate any discrepancies

Review of emission factors

 Inventory agencies should cross-check country-specific values for estimation with the available IPCC values.

The intent of this comparison is to see whether the national parameters used are considered reasonable relative to the IPCC default values, given similarities or differences between the national source category and the emission sources represented by the default.

Review of activity data

- Inventory agencies should compare country-specific data to IPCC default values for the following activity level parameters: MSW_T, MSW_F, and DOC.
- They should determine whether the national parameters are reasonable and ensure that errors in calculations have not occurred.
- If the values are very different, inventory agencies should characterize municipal solid waste separately from industrial solid waste.

- Where survey and sampling data are used to compile national values for solid waste AD,
 QC procedures should include:
 - Reviewing survey data collection methods, and checking the data to ensure they were collected and aggregated correctly. Inventory agencies should cross-check the data with previous years to ensure the data are reasonable.
 - Evaluating secondary data sources and referring QA/QC activities associated with the secondary data preparation. This is particularly important for solid waste data, since most of these data are originally prepared for purposes <u>other than GHG inventories</u>

Involvement of industry and government experts in review

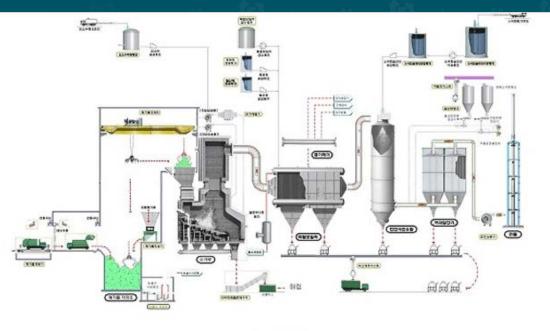
Inventory agencies should provide the opportunity for experts to review input parameters. For example, individuals with expertise in the country's solid waste management practices should review the characteristics of the solid waste stream and its disposal. Other experts should review the methane correction factors.

Verification of emissions

- Inventory agencies should compare national emission rates with those of similar countries that have comparable demographic and economic attributes.
- This comparison should be made with countries whose inventory agencies use the same landfill CH₄ emissions method.
- Inventory agencies should study significant discrepancies to determine if they represent errors in the calculation or actual differences.

(Session 2) Waste Incineration

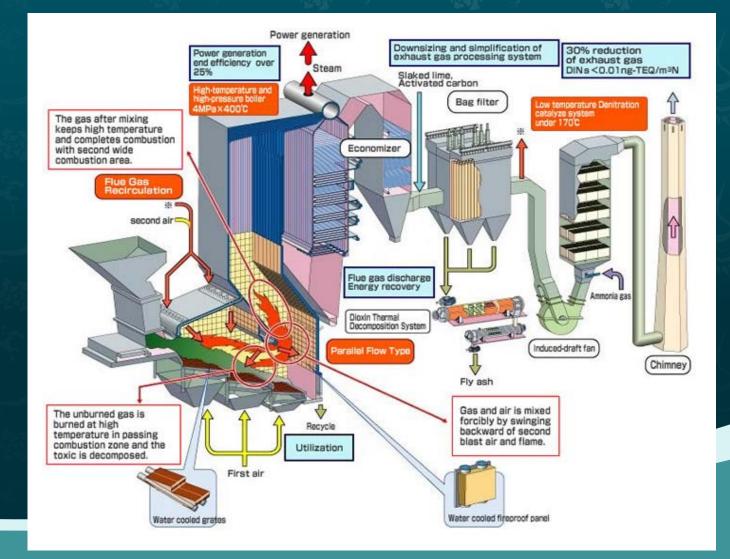
Stoker Incinerator



<처리공정도>



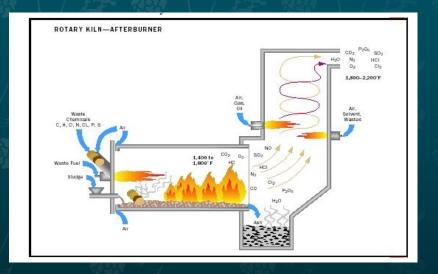
Stoker Incinerator



Rotary Kiln Incinerator

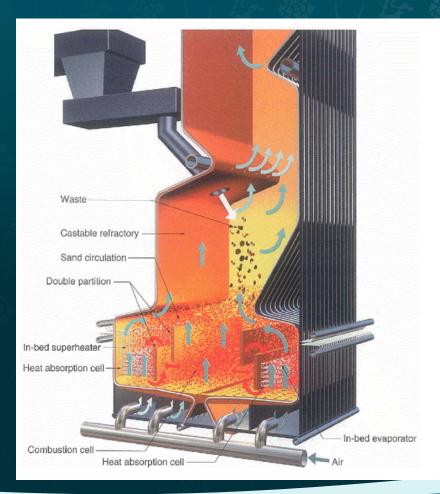


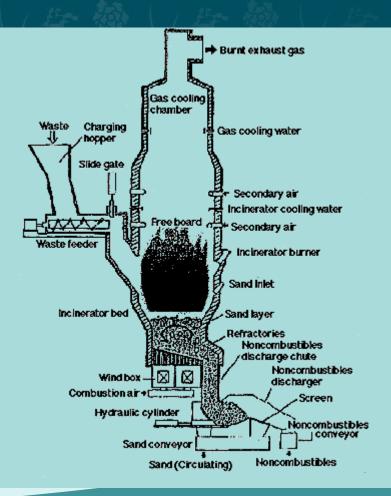
Rotary Kiln Waste Incinerator Plant with Vertical Thermo-Reactor





Fluidized Bed Incinerator







Application of biogas plant to recycle organic wastes

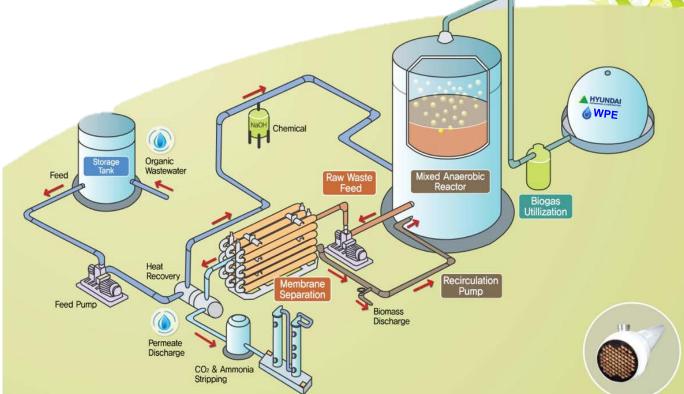
2013.09.16

YOUNG-O KIM



INOVATION for Eco-SUSTAINABILITY

"Harvesting Clean Energy & Clean Water from Wastewater"







Talk outline

- 1. Value of Organic Wastes
- 2. <u>Anaerobic Digestion</u> description.
- 3. <u>Stage</u> of Biogradation
- 4. Anaerobic Membrane Bioreactor (AnMBR)
- 5. See Hyundai Biogas Plant (HAnDs) Results
- 6. <u>Opportunity</u> for Application

Where do organic wastes come from?

Industrialization



Urbanization

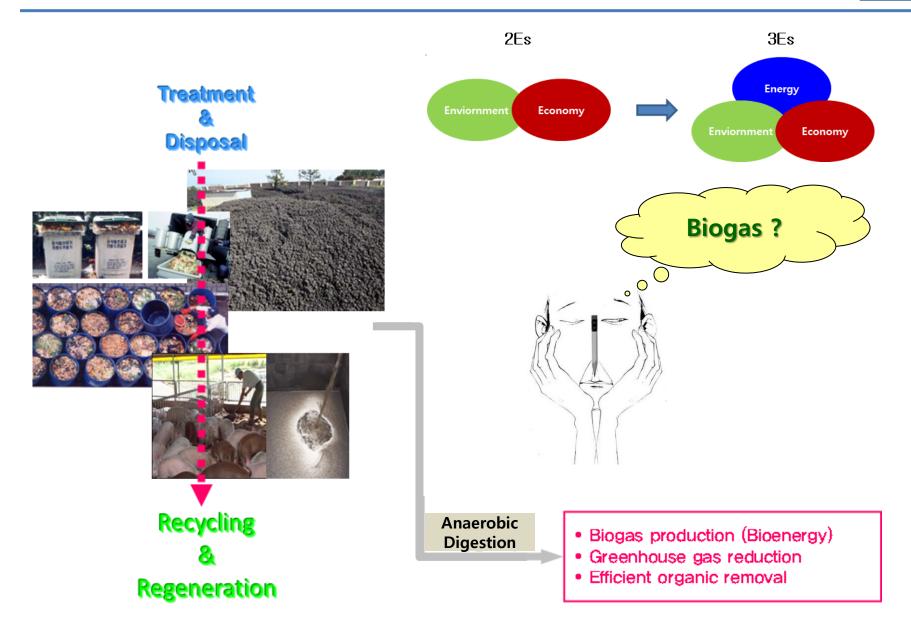


Population

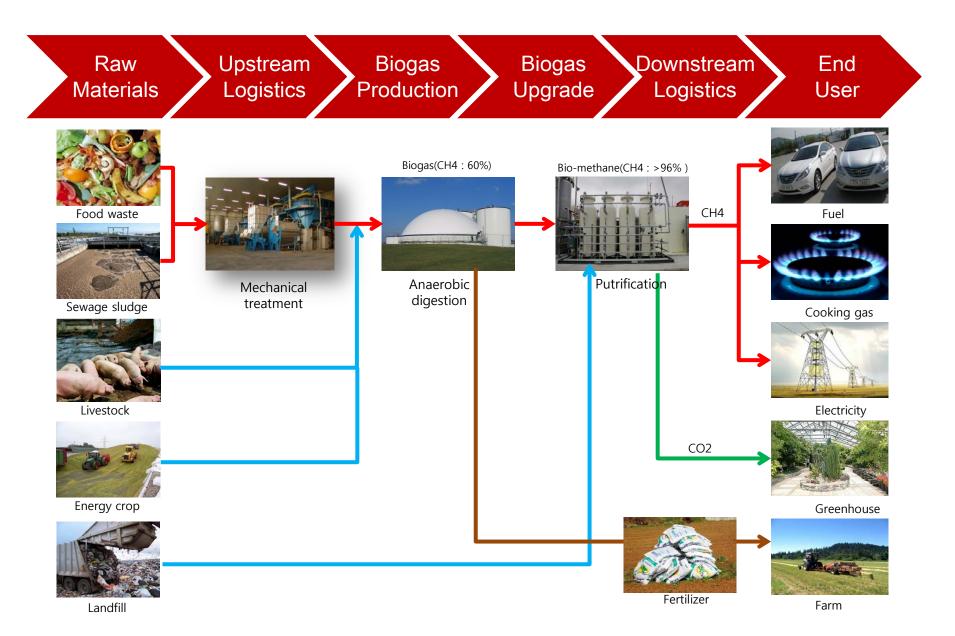




Paradigm of organic wastes treatment



Value Chain





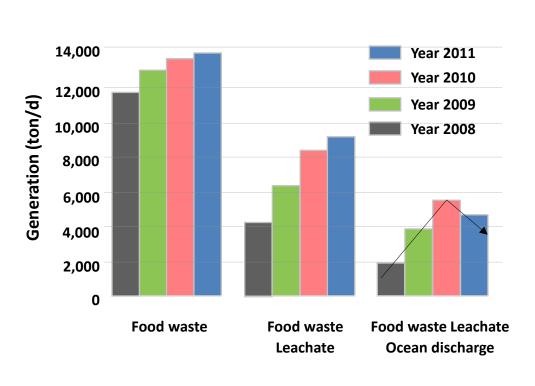
High Volatile Solid Content

High Moisture Contents Readily biodegradable
 → Easy corruption
 Renewable energy source

Hard to apply physical and chemical treatment such as incineration etc.
High cost for management

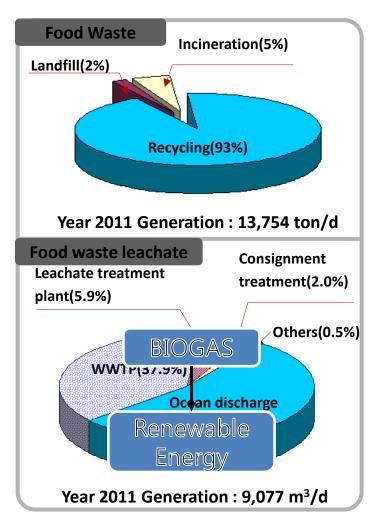
Heterogeneous Bacteria/ Pathogens

- Necessity of pretreatment
 Hard to apply microbial
 /enzyme treatment techniques
 - → due to contamination or competition between microorganisms

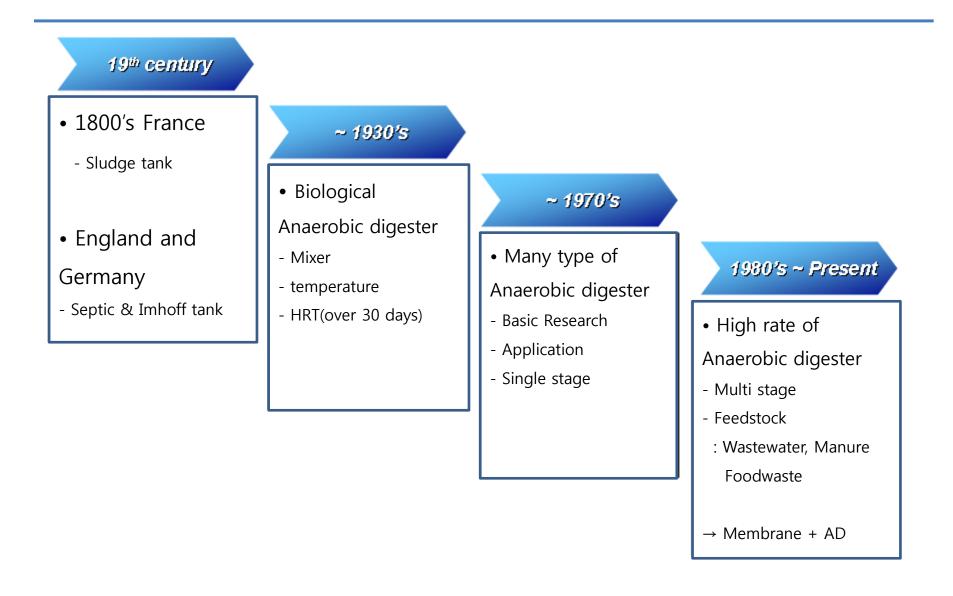


Year's Generation and Discharge





Anaerobic Digestion : Biogas History



What is Biogas Digestion

- Reduce
 - Smell
 - Greenhouse gas
 - Pathogen level
- Produce biogas
- Improve fertilizer value of manure
- Protect water resources
- Biogas Digestion is the process of taking biogas to produce electricity, heat, or hot water
- Biogas means a gas formed by carbon dioxide and methane from breakdown of organic materials such as manure.

• Basic of digestion

Substrates must be degradable

Substrates must/should be available at a constant mass/volume flow

Substrates should have a nearly constant composition

Concentration of organic dry matter should be higher than 2 %

Substrates should be a liquid slurry

Digester volume should be more than about 100m³

What is a Digester?

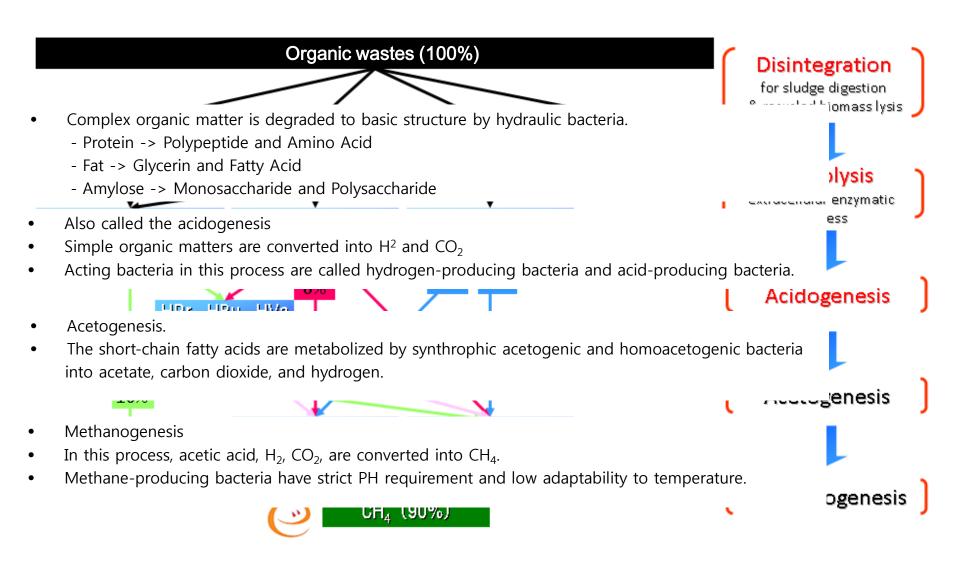
- Digester is a vessel or container where the biogas process takes place. Bacteria breaks down waste products to create biogas. Products may be fed into the chamber such as manure and food waste or the container could be used to cover a place that is already giving off biogas such as a swamp or a landfill.
- Biodigester is a system that promotes decomposition of organic matter.
- It produces biogas, generated through the process of anaerobic digestion.
- Biogas generated can be used for cooking, heating, electricity generation, and running a vehicle.

Basic Designs of Digester

- Continuous-fed
- Batch-fed

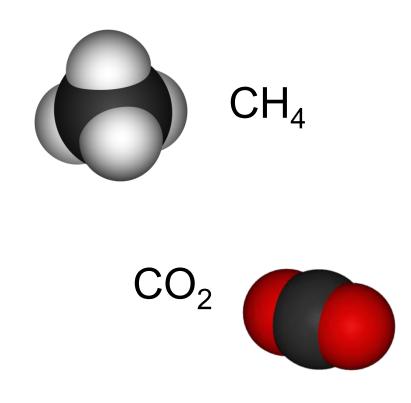


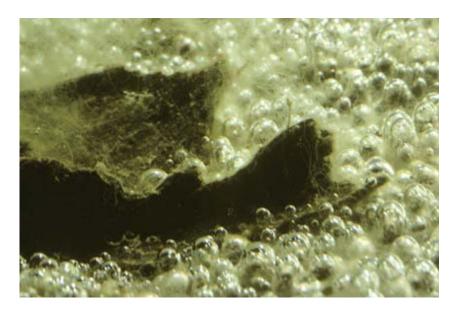




What is Biogas ?

• A mixture of methane and carbon dioxide





• Methane or 'swamp gas', prod**uceat mathral**ly in swampy ponds

What is it used for ?

• Biogas is a fuel used as an energy source for light, heat or movement





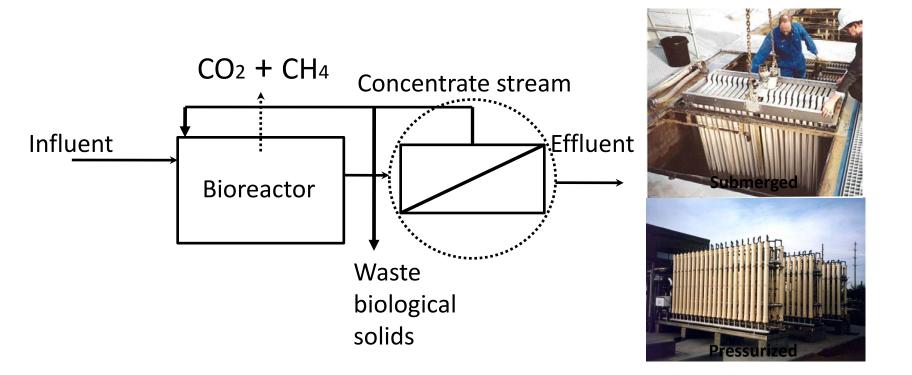


Potential of Biogas

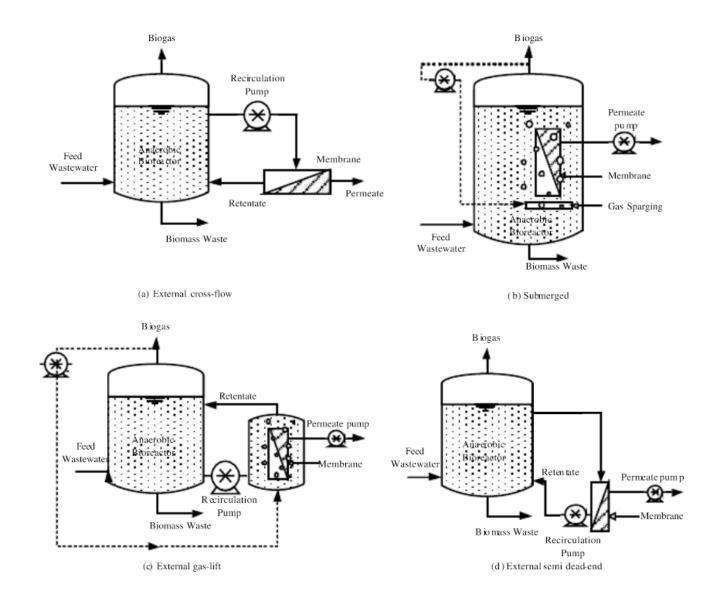
61	C	
	L	7

Biogas potential:	total organic solids (%)	m ³ CH ₄ /m ³ substrate	
Waste water, municipal	0.05	0.15	
Waste water, food industry	0.15	0.5	
Sewage sludge	2	5 to 10	
Cow manure	8	20 to 30	
Pig manure	6 to 8	30 to 50	
Food waste	15 to 20	100 to 120	
Food waste leachate	6 to 14	30 to 60	

Anaerobic Membrane Bioreactor(AnMBR)



- Providing long SRT needed while operating at short HRT as required to reduce reactor size
- Pressurized type use more since membrane cleaning is easier to perform



Wastewater	Volume (mํ)	Temp. (℃)	MLSS (g/L)	OLR (g/㎡/d)	Removal (%)	Reference
Sweet factory	0.09	35~37	-	8~9	97~99	Defour et al., 1994
Feed industry	0.4	37	6~8	4.5	81~94	He et al., 2005
Soybean processing	2	30	-	2.5	71	Yushina and Hasegawa, 19 94
Brewery	0.12	36	> 50	>28.5	97~99	Ince et al., 2000, 2001
Potato starch bleaching	4	-	15~100	1.5~5	65~85	Brockmann and seyfried, 19 96, 1997
Palm oil mill	0.05	35	50~57	14.2~21.7	91.7~94.2	Fakhru'l-Razi and Noor, 199 9
Kraft pulp mill	5	53	8	9~28	86~89	Imasaka et al., 1993
Sewage	0.018	24~25	16~22	0.4~11	60~95	Wen et al., 1999
Primary sludge	0.12	35	-	0.4~0.68	25~57	Ghyoot and Verstraete, 199 7
Heat treated sewage	0.2	35~38	10~20	4~16	79~83	Kayawake et al., 1991
Swine manure	0.006	37	20~40	1~3	-	Padmasiri et al., 2007

Operating Conditions of Membrane system

Туре					Membrane		linear			
of		Temp.		Pore	area	TMP	velocity	Initial flux	Final flux	
wastewater	Scale ^a	(°C)	Material	size ^c	(m²)	(kPa)	(m. s ⁻¹)	(L. m ⁻² . h ⁻¹)	(L. m ⁻² . h ⁻¹)	Reference
Wheat starch	Ρ	40	-	18,000 D	144	690	_d	-	14~25	Butcher et al, 1989 Choate et al, 1983
Brewery effluent	Р	35	Poly-ethersulfone	40,000 D	0.44	140~340	1.5~2.6	-	7~50	Strohwald et al, 1992
Maize processing	Ρ	-	Poly-ethersulfone	20,000~80,000	668	450	1.6	-	8~37	Ross et al, 1992
				D						
Wool scouring	Р	40~47	Poly-acrylonitrile	13,000 D	3.1	2~2.2 ^e	-	30~45	17~25	Hogetsu et al, 1992
Glucose, peptone	L	35~38	Ceramic	0.2	0.4	30~200	0.5~4	-	12.5~125	Shimizu et al, 1992
Kraft mill effluent	Ρ	48.4	Ceramic,	0.16	1×24	60	1.75	50	27	Imasaka et al, 1993
			aluminum oxide							
Acetate	L	35	Ceramic	0.2	0.20	25~150	0~3.5	-	18~127	Beaubien et al. 1996
Sewage sludge	L	30~35	Poly-ether sulfone	60,000 D	0.3	375	0.75	31	19	Ghyoot et al, 1997
Molasses ^f	L	20	Polypropylene	10	0.051	-	-	100~160	10~80	Hernandez et al, 2002
Sewage	Р	10~28	Ceramic	13,000 D	13.6	1~2 ^e	2	-	15~20	Tanaka et al, 1987
Heat-treated liquor from sewage sludge ^f	Ρ	35~38	Ceramic	0.1	1.06	200 ^g	0.2~0.3	8~13	3~8	Kayawake et al, 1991
Food waste leachate	Р	55	Polyvinylidene	0.04	13.1	100~300	1~3	-	15	Kim et al, 2011
			fluoride							

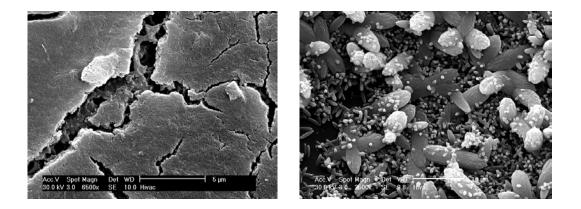
^aAll membranes were external cross-flow unless otherwise noted. ^bL = laboratory/bench scale, P = pilot scale.

^cD = Daltons (molecular weight cutoff).^d-Indicates value not reported.

^ePressure reported as kg/cm².^fSubmerged membrane.

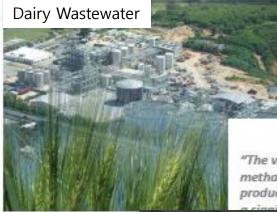
g-Indicates value not reported

- Membrane fouling is an inevitable and complex phenomena
- Biogas sparging, fluidized media for submerged system
- TMP, cross-flow velocity for pressurized system Effectiveness varies depending upon foulant materials (e.g., particle size distribution) and module design (e.g., channel height etc)



Application of AnMBR

New: Memthane®, for industrial high-strength wastewater



"The valuable methane-rich biogas produced can cover

Biothane's anaerobic biological

wastewater treatment and a Pentair's X-Flow Ultra Filtration membrane separation process. Influent is fed to the anaerobic bioreactor where the organic components are converted into energy-rich biogas. Next, the anaerobic effluent is processed through the UF membrane unit, separating the 'clean' permeate from the biomass. The biomass is

> returned to the bioreactor, while the ultraclean filtrate is

discharged as

particle-free, low BOD/COD

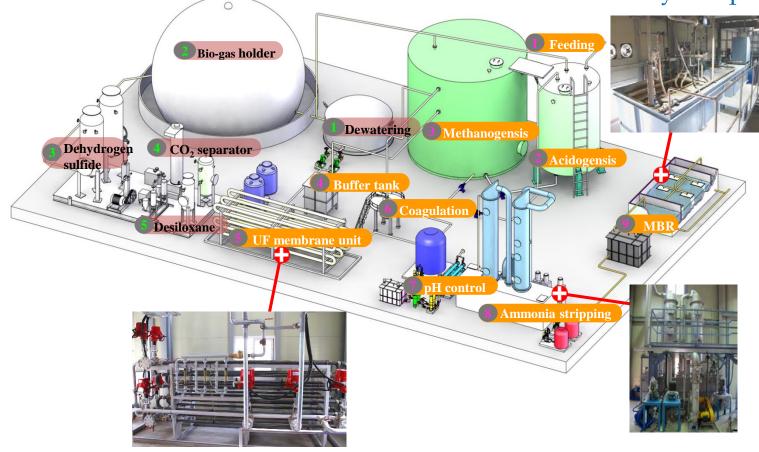
effluent, often at levels low

Membhane*, Veolia's Anaerobic Membrane Bio-Reactor (AnMBR), delivers high-energy efficiency and





- Maximizes biogas energy
- Excellent effluent quality
- Easy nutrient recovery as fertilizer
- Odor-free
- Vastly reduces biosolid disposal costs
- Reduces the facility footprint



Specifications of Membrane Unit



	Shape	Tubular	
	Material	PVDF	
Spec.	Diameter	11 mm	
	MWCO	100 k Dalton	
	Max. working Temp.	90 °C	
	Total membrane area	13.1 m ²	
Operating condition	Operating pressure	1~3 kgf/cm ² (In-Out)	
	Cross-flow velocity	1~2 m/sec	

Why Crossflow Membrane ?

- Flat-sheet membranes can only be installed in a submerged tank, which client pays to build
- For membrane cleaning, the entire membrane tank must be drained, halting treatment of wastewater
- Crossflow membranes are installed on skids with minimal footprints, and require no storage tank of any kind
- Much easier cake-fouling control just by adjusting the crossflow velocity
- Possible clean-in-place, meaning any individual membrane can be bypassed and removed from the system for cleaning without even pausing treatment

Picture of Plant



Properties of food waste leachate



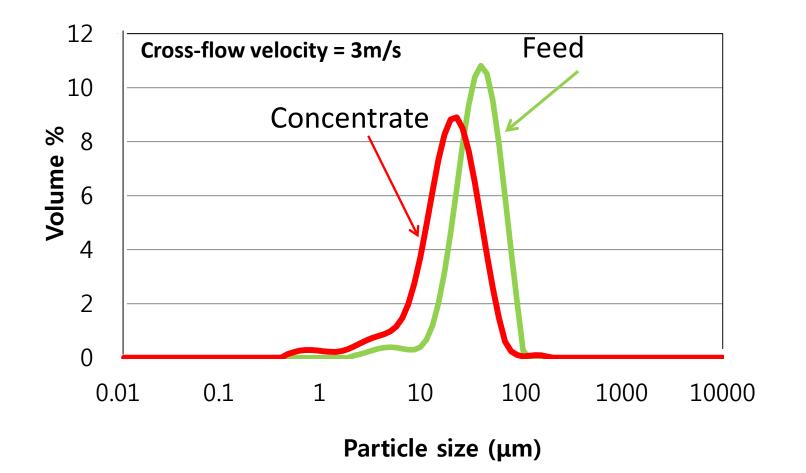
Qualities of food waste leachate



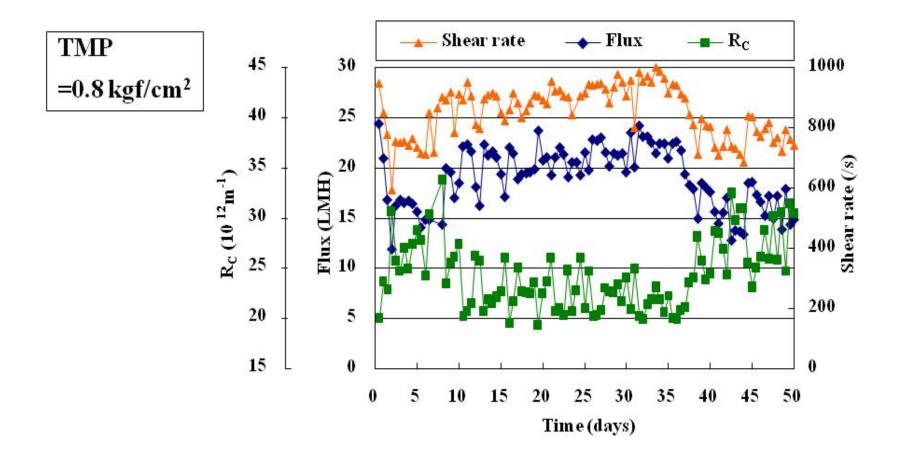
X Source:

SUDOKWON Landfill Site Management Corp. (2008), "The feasibility study of biogas production with organic waste"

Change of Floc Size Distribution in Membrane Concentrate Stream

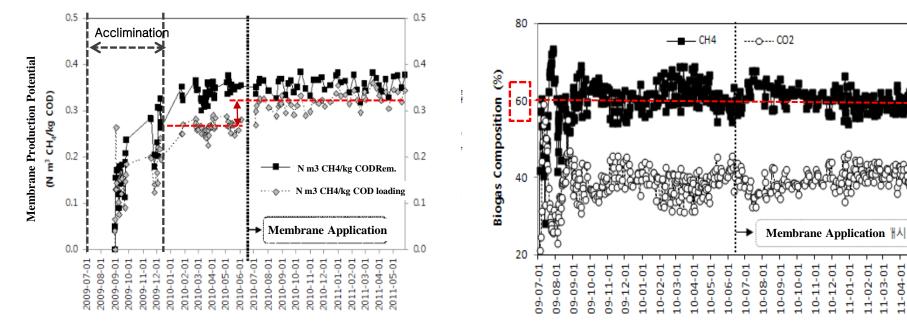






Shear rate played critical role in controlling membrane fouling

Biogas Production Potential and Compostion



(단위	: Nm³	CH₄/kg	COD)
-----	-------	--------	------

	Before	After	Variation
CH ₄ production/	0.28	0.32	∆ 22.1%
CODloading	(0.25~0.29)	(0.29~0.36)	
CH ₄ production/	0.34	0.36	∆6.0%
COD _{Rem.}	(0.30~0.38)	(0.32~0.38)	

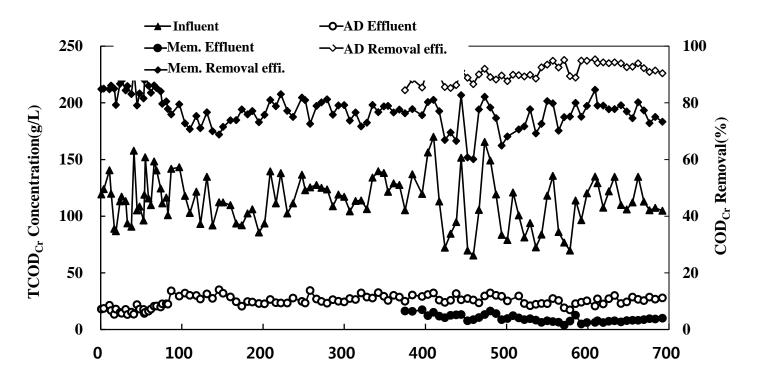
Compositio	CH ₄	CO ₂	H ₂ S
n	(%)	(%)	(ppm)
Content	63	37	1,582
	(56~68)	(32~44)	(1,100~2,360)

1. Biogas Production : 567 Nm^3/ton_COD_{Rem}

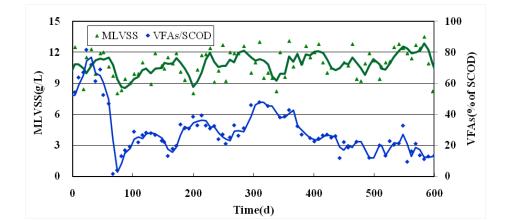
2. Methane Yield : 359 $N\,m^{\!*}\!/ton_COD_{Rem}$

11-05-01

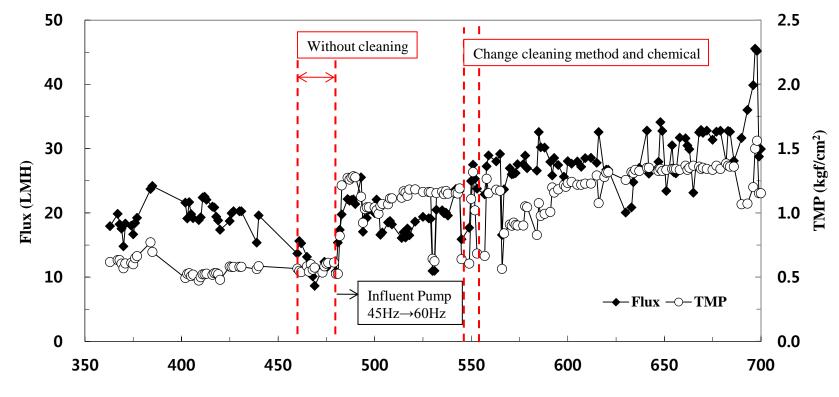
COD Removal Efficiency



Operating Time(days)

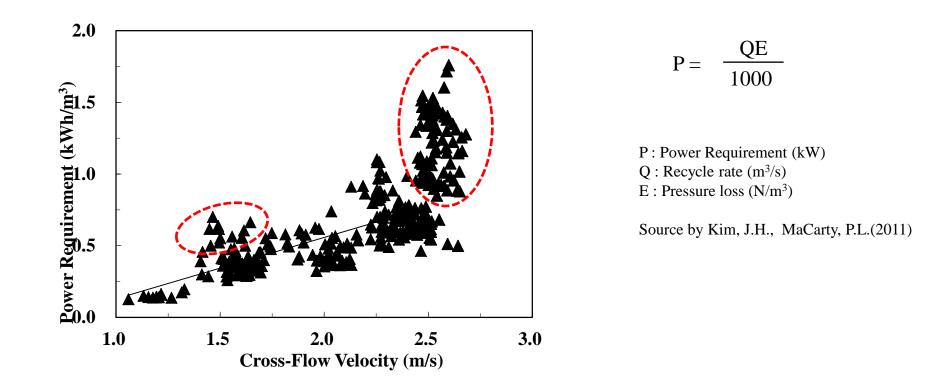


MLVSS has been increasing continuously since the digester was integrated with UF membrane



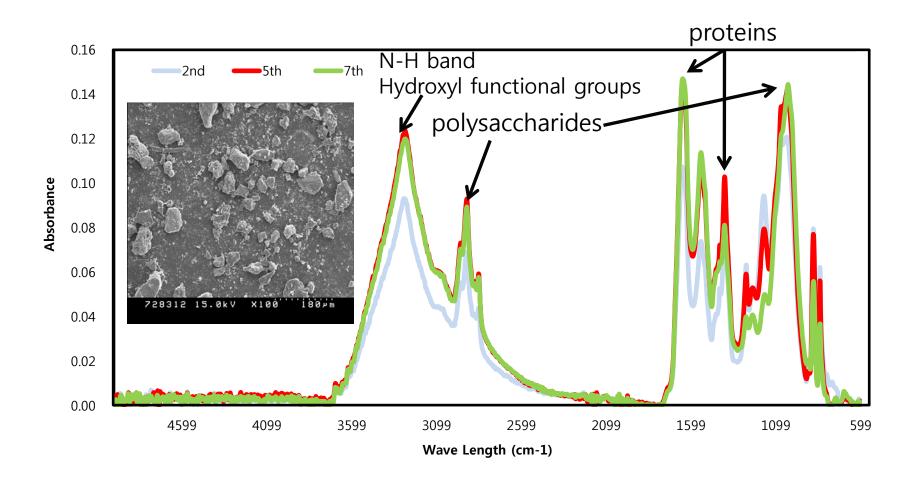
Operating time (days)

Relationship between Cross-flow velocity and Power Requirement in AnMBR

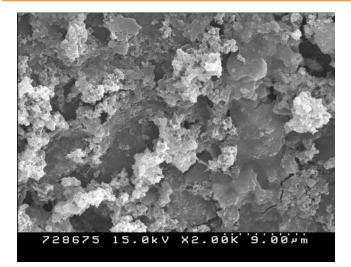


• The more fouling progressed, the more required electrical power to get the constant flux

Membrane Autopsy Works-FTIR Observation on Membrane Surface



Cake Layer after chemical cleaning

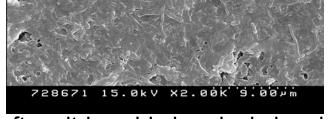


after NaOCI chemical cleaning

	Foodwaste Leachate	Methanogenic Sludge			
Ca ²⁺ (mg/L)	>1,000	> 800			
рН	< 4.0	> 7.5			
Alkalinity (mg CaCO ₃ /L)	-	>10,000			
LSI	< - 4.0	>+4.0			

after NaOH chemical cleaning

- At an LSI value greater than zero, the concentrate stream is supersaturated with calcium carbonate and would likely scale membrane surface as cake layer formation
- Strong binding and solidification can lead to pronounced cake resistance

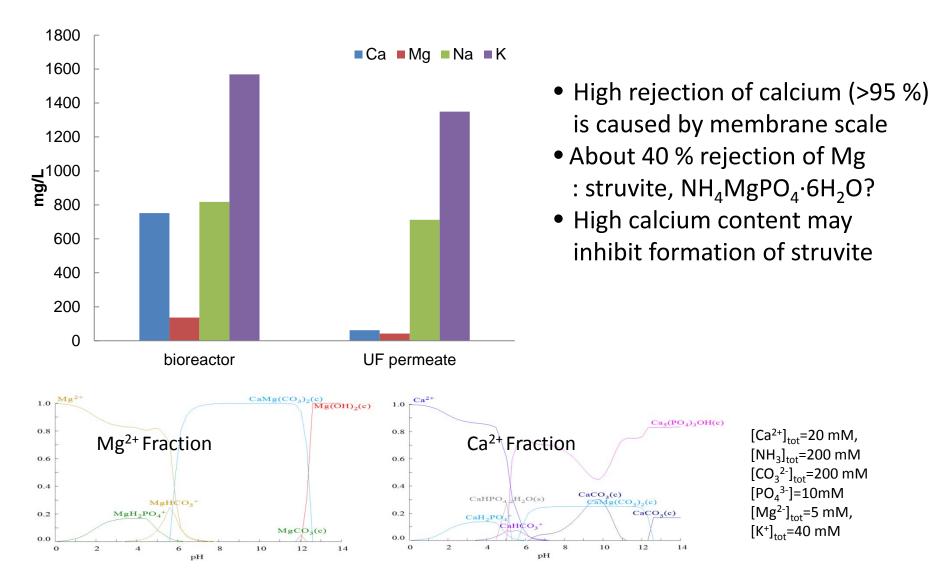


after citric acid chemical cleaning

rate?

Cross-flow velocity = 3 m/s TMP= 0.8 bar

Membrane Scale and Rejection

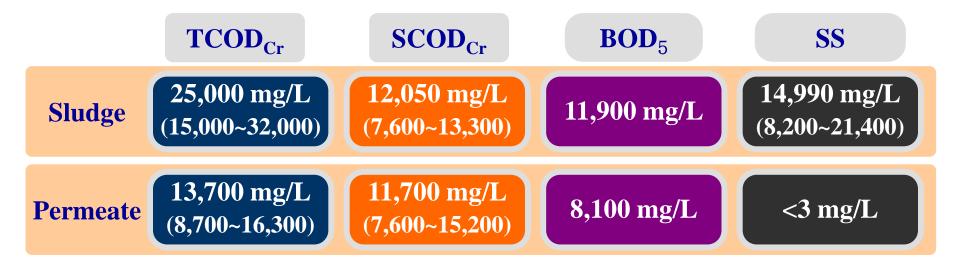


Water Quality of HAnDS[®]

Sludge



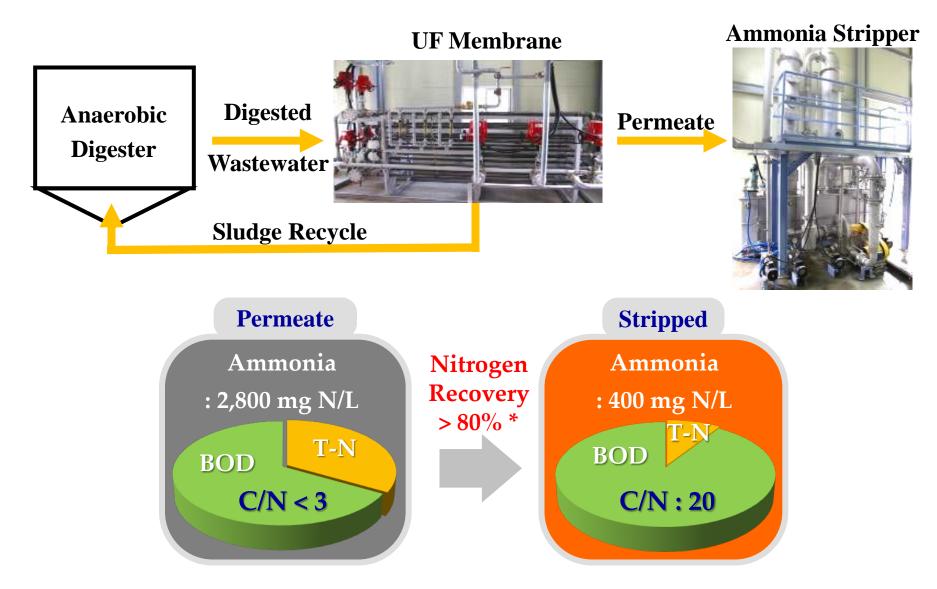
Permeate



(Unit : mg/L)	Food Waste Leachate	Anaerobic Digestion	Ultrafiltration	Aerobic MBR*
BOD ₅	51,000	9,000	6,000	< 3.0
COD _{Cr}	120,000	25,000	10,000	300
TN	3,000	4,000	2,000	<60
TS (g/L)	65	25	<10	-
n-Hexane	11,000	380	350	-

(*After ammonia stripping process)

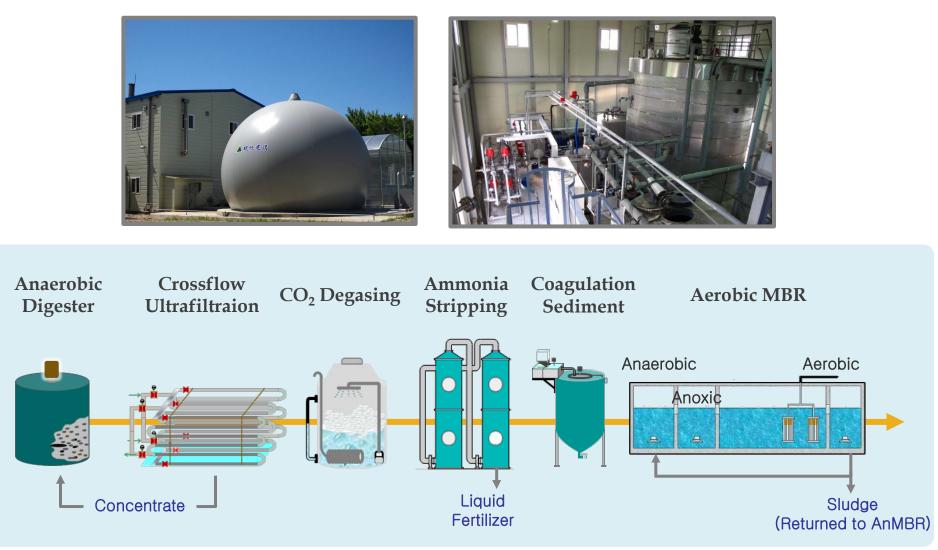
Addendum : Ammonia Stripping

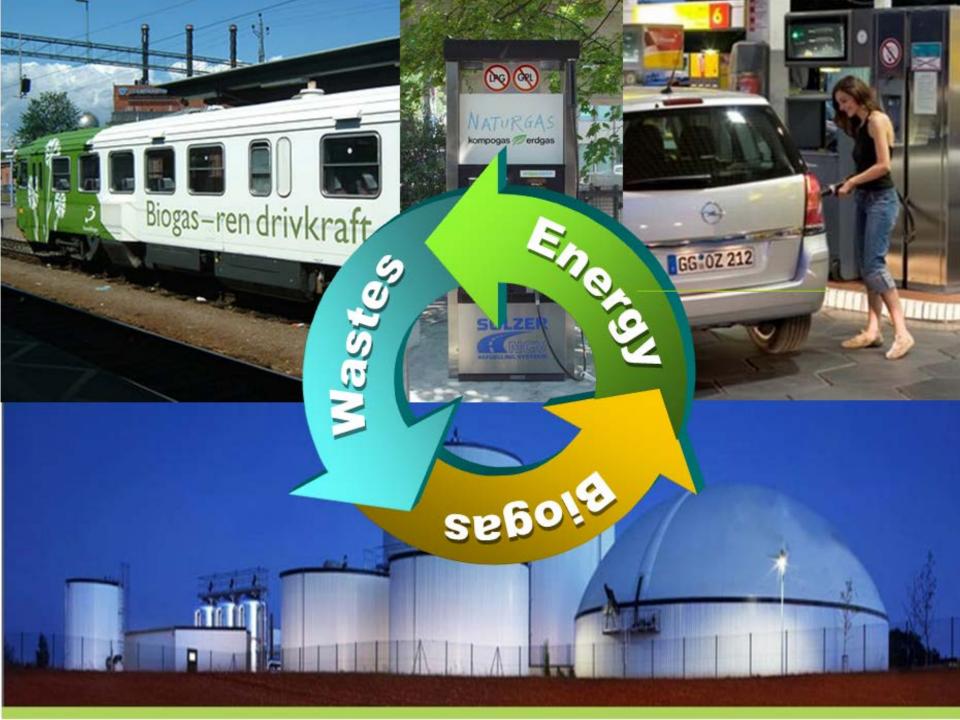


*Recovery can be controlled by the operating conditions: temperature, pH and air volume

How to apply ?

Anaerobic Membrane Bioreactor (AnMBR, Korean NET No.352)





Seoul National University Seminar

Anaerobic Digestion & Biogas Plants - From Organic wastes to Energy -

KOLON GLOBAL CORPORATION

R&BD Center Oct. 14. 2013

Kyu-Jung Chae (채규정)

Work Experiences



Chae et al., Water Sci. Technol., 49(5-6), 2004 Chae et al., Water Sci. Technol., 50(6), 2004 Chae et al., J. Environmental Management, 88(4), 2008 Chae et al., Chemosphere, 71(5), 2008 Chae et al., Bioprocess and Biosystem Engineering, 2012 Chae et al., Biores. Technol., 99, 2008 Chae et al., Water Sci. Technol., 49(5-6), 2004 Chae et al., J. of KOWREC, 9(4), 2001 Chae et al., J. of KOWREC, 9(3), 2001 Chae et al., Biores. Technol., 101, 2010 Chae et al., Environ. Sci. Technol., 43(24), 2009 Chae et al., Biores. Technol., 100 (14), 2009 Chae et al., Int. J. of Hydrogen Energy, 2009 Chae et al., Int. J. of Hydrogen Energy, 33, 2008 Chae et al., Energy & Fuels, 22(1), 2008 Kim et al., Environ. Eng. Res., 13(2), 2008 Ajayi et al., Int. J. of Hydrogen Energy, 2008 etc. Choi et al., Biores. Technol., 102, 2011 Kim et al., Biores. Technol., 102, 2011



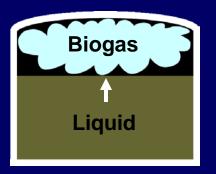
Anaerobic Digestion

How Are Biofuels Produced?

Anaerobic Digestion

Biogas:

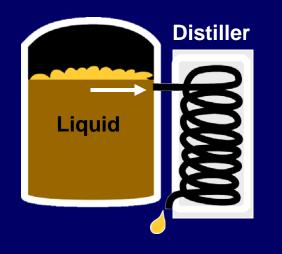
- Manure
- Microbes



Fermentation and Distillation

Bioethanol:

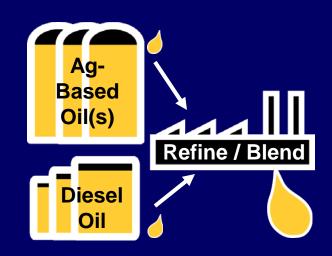
- Yeast / Sugar
- Alcohol



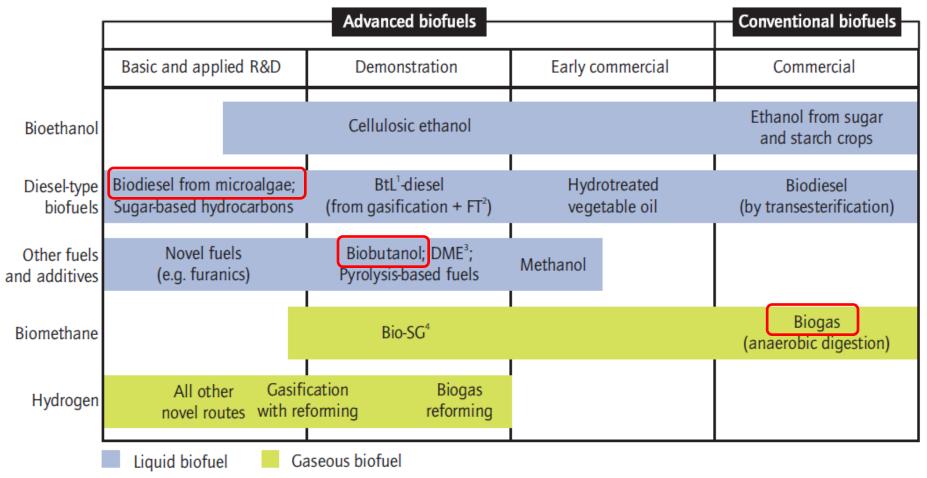
Extraction, Blending, and Refining

Biodiesel:

- Refined Diesel Fuel
- Refined Ag-Based Oils



Commercialization status of main biofuel technologies



1. Biomass-to-liquids; 2. Fischer-Tropsch; 3. Dimethylether; 4. Bio-synthetic gas.

<Ref.: Technology Roadmap, Biofuels for Transport, IEA 2011>

What's Biogas?

Biogas is <u>NOT</u> pure methane (natural gas).

Methane (60%) CO2 (40%) – with trace amounts of H2S and water vapor

Typical Composition of Biogas

CH ₄	50–75%
CO ₂	25–50%
N ₂	0–10%
H ₂	0–1%
H ₂ S	0–3%
O ₂	0–2%

What's AD?

Anaerobic digestion (AD) is a series of processes in which <u>microorganisms</u> break down <u>biodegradable</u> material in the absence of <u>oxygen</u>, used for industrial or domestic purposes to manage waste and/or to release energy.

e.g., $C_6H_{12}O_6 \rightarrow 3CO_2 + 3CH_4$

www.daviddarling.info/. ./M/methanogen.html

Why anaerobic?

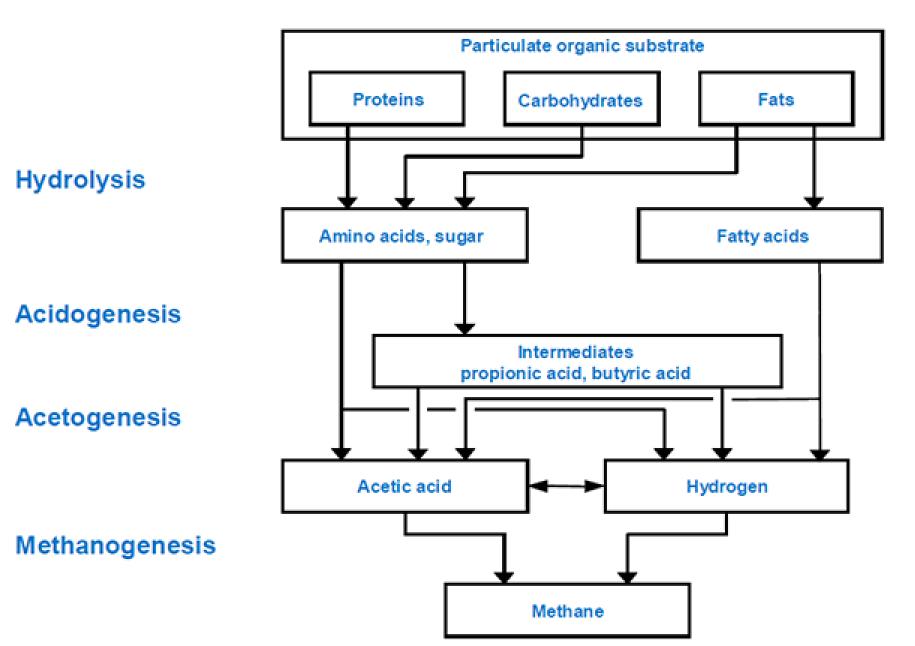
• Advantages:

- Low energy input
- Biogas production: net energy production
- Minimal sludge production
- Fertilizer quality sludge
- Pathogen destruction
- Odour reduction

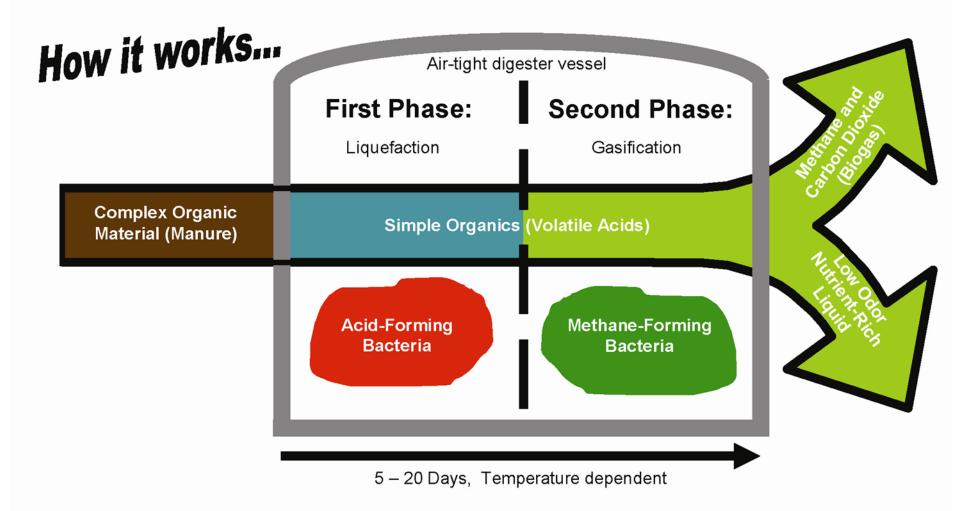
• Disadvantages:

- High investment
- Ammonia & Phosphate production

Anaerobic biodegradation



Anaerobic digestion uses microbes in an oxygen-free tank to break down manure into Biogas and a nutrient-rich liquid.



Anaerobic Bacterial Granule: A Community of Bacterial Species for Anaerobic Digestion of Biomass

Anaerobic Ba

rial Granule

WD29.8mm 20.0kV x30

Anaerobic Digestion depends on consortia of Hydrolytic & Acidogenic Bacteria working with Methane-producing bacteria (methanogens) growing in STRUCTURED COLONIES or FILMS for structural support and metabolic interchange. Structured

Structured Microbial Colonies

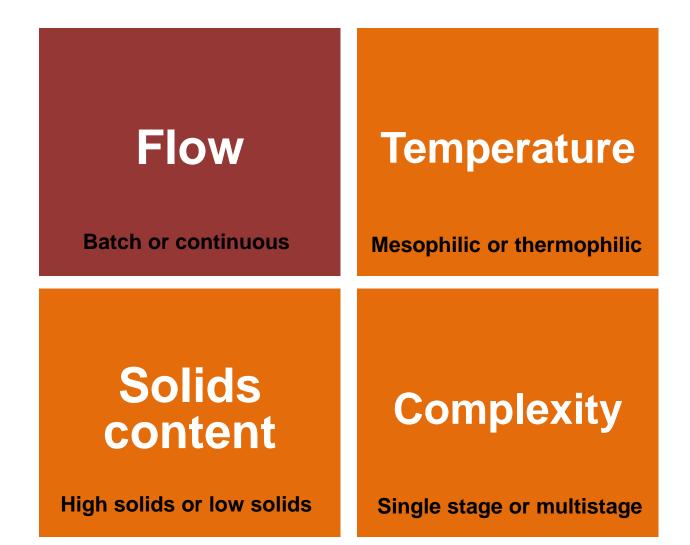
Methanosarcina mazei

Methanobrevibacter smithii

Methanogens

Source: Prof. Chang D.J.

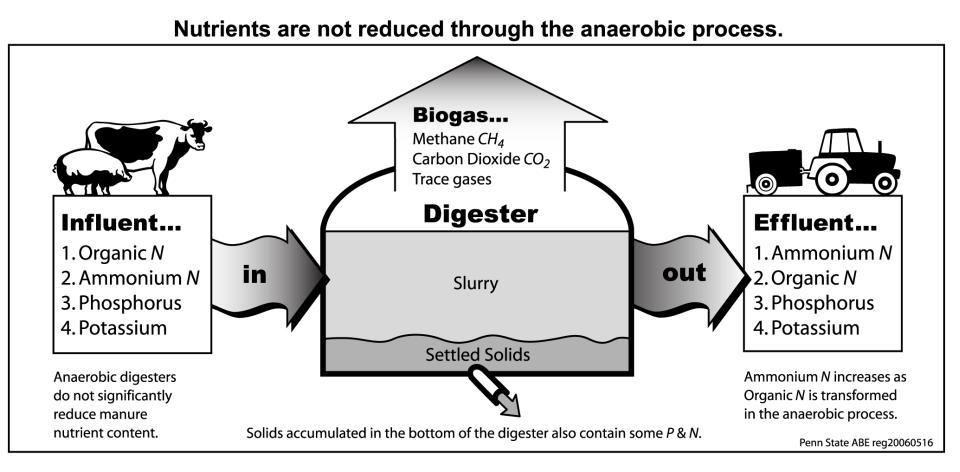
AD Configuration



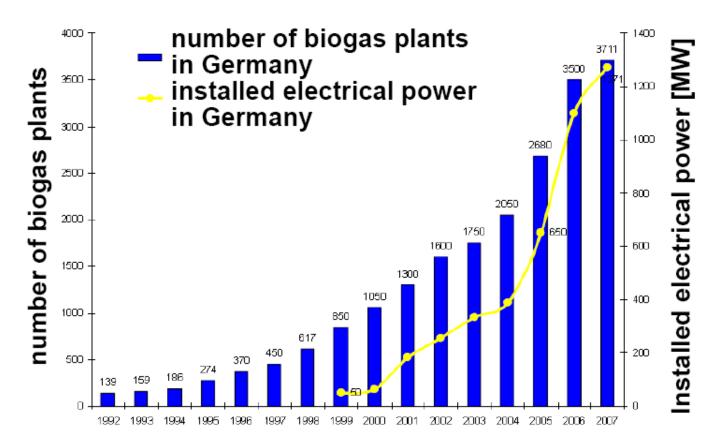


- Fundamentals & Practices -

How Do Anaerobic Digesters Work?

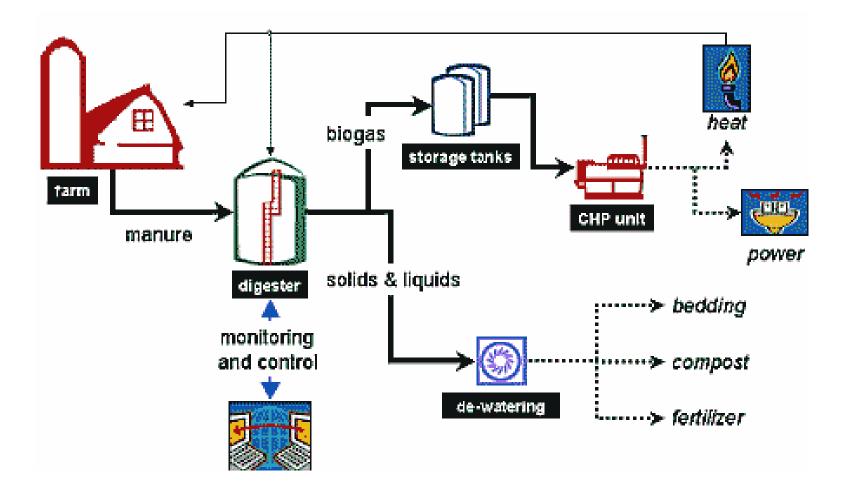






Source: Torsten Fischer (Krieg & Fischer Ingenieure GmbH)

Biogas plant



Nutrients are not reduced through the AD process...

Biogas Plants: Key Equipment

Pretreatment Screen/ Hygienization tank	Anaerobic digester	Gas hold	er
CHP unit or boiler	Biogas purification system	Safety device	Mixer
		Pump	Heating system



Source: German biogas association



Screening/sorting

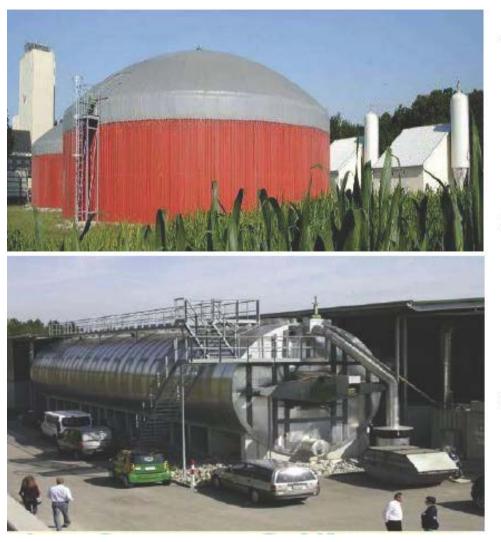
Feeding







Digester Types



Courtesy: Sustainable Energy Ireland

- Vertical digester
 - Different heights, diameter
 - Several digesters/stages possible
- Plug flow digester
 - Horizontal shaft
 - High organic loading rate
 - Dry digestion
- Combination plug flow vertical digester
 - Two (several) stages





Top mounted agitator for complete digester mixing

(~up to 20m height)



Low energy Low mechanical problem

Heat Exchangers



Double-pipe heat exchangers

Serial wound HE Plate HE.... We need a specialized HE for particle-rich waste(waters).

Pump Types



Courtesy: Sustainable Energy Ireland

Biogas Utilization

Local heating & electricity production



Microturbine

CHP unit

Boiler

Upgrade Biomethane sold to public grid or grid owner



Car fuel

Fuel cell

Biogas

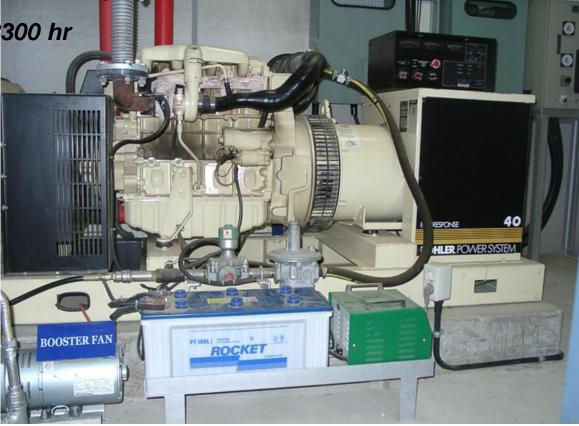
CHP unit for Biogas

Combined Heat and Power

Modified Gas- or Diesel-engines Dual fuel engines (diesel~10% + biogas 90%)

- 50~2000kWp
- Annual operating 7500~8300 hr
- H2S, NH3 damage
- **Efficiency rates:**
- Electrical 30~42%
- Thermal 25~50%
- Overall up to 85%





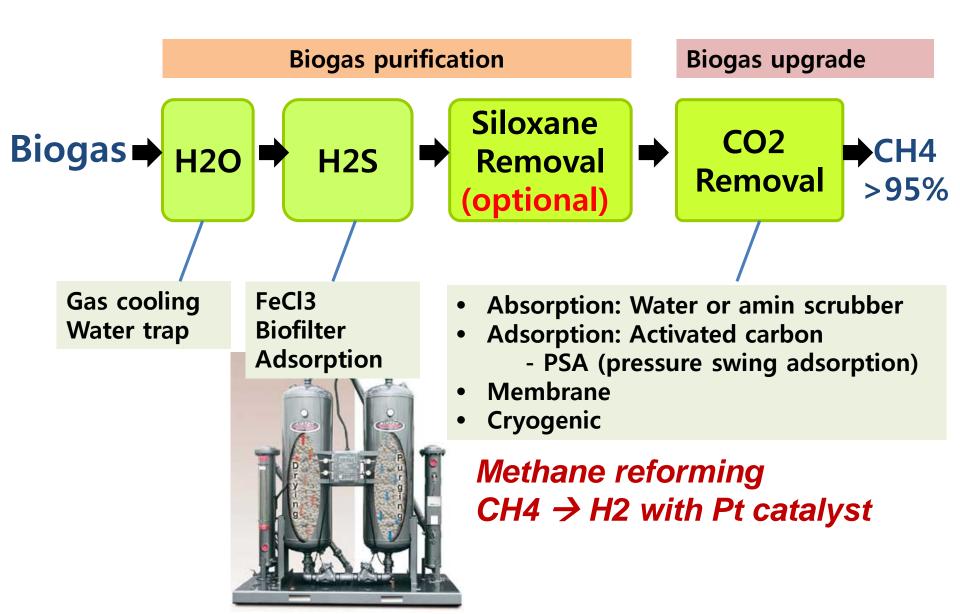
Containerized CHP unit for farm-scale biogas plant

Interconnection with the electricity grid.





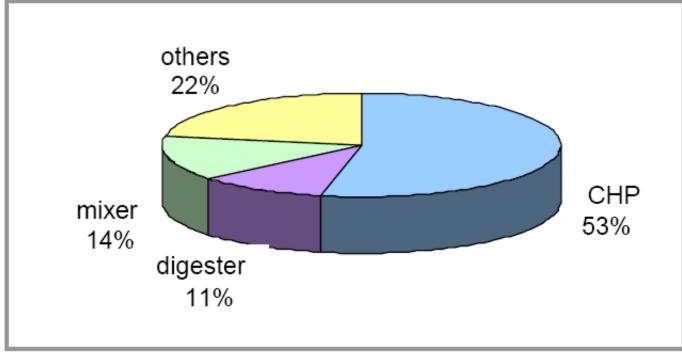
Biogas Upgrading



What is the most problematic?

Damages at different parts of the plant

payments related to different plant parts





Experience of the insurance company Source: Torsten Fischer



Case Study Lessons from biogas plant experiences

For Success of AD Operation

Proper Heating Proper Mixing Proper pH Proper SRT

In some cases, mechanical aspect is more important than biological features for the successful operation of biogas plants.

Biogas plant (for swine manure)



- Feedstock : swine manure
- run : 1999-2002
- Digester Vol. : 200 m³
- AD + CHP(dual-fuel gas engine 37kW)
- New Excellent Technology(NET) certified



Design & Construction: Rural Development Institute / Livestock Research Institute / Kolon Global Corp.

Design Parameters (for swine manure)

Gas Yield (Biogas m ³ /kg VS _{added})	0.4-0.7 (0.47)
CH ₄ (%)	60-75 (65)
VS/TS (%)	> 60
TS (%)	4-10 (6)
HRT (day)	15-20
OLR (kg VS/m ³ -d)	0.5 - 6.0
Organic reduction rate (%)	50-70 (60)
Temperature(°C)	30-35
Digester type	Flat bottomed CSTR single stage AD
Heating	External heat exchanger
Mixing	Gas mixing + external hydraulic circulation
Biogas use	CHP (Dual-fuel type)
H2S removal	FeCl ₃
Free ammonia (mg/L)	< 100

Anaerobic Digester Selection (Mesophilic CSTR digester)

WHY Completely mixed?

- suitable for swine manure having high solids contents
- resistant to toxicant inhibitors (NH3, disinfectant etc.)
- enhanced substrate-microbe contact

WHY Mesophilic?

- reduced heating requirement for winter season
- reduced ammonia toxicity

Biogas Yield

Mesophilic vs. Thermophilic ?

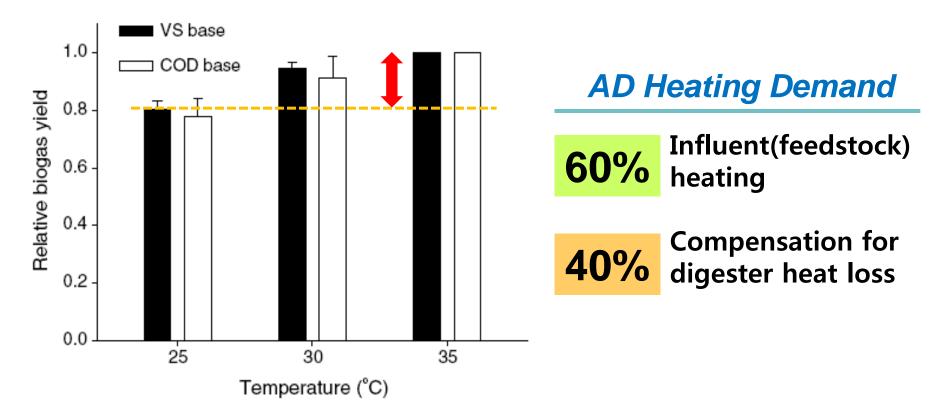
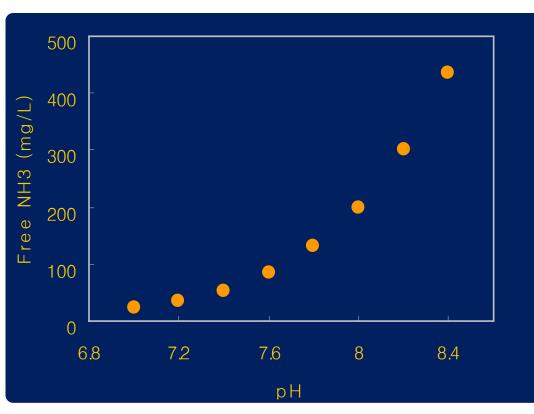


Fig. 3. Relative biogas yields (% of gas production at 35 °C, means of four feed loads) for various digester operating temperatures. The data are expressed as the mean \pm SD (n = 4).

Chae et al., Biores. Technol., 2007

Ammonia Inhibition



Free ammonia (NH3) vs pH

@ total ammonia concentration = 2000 mg/L, Temp. = 35 °C

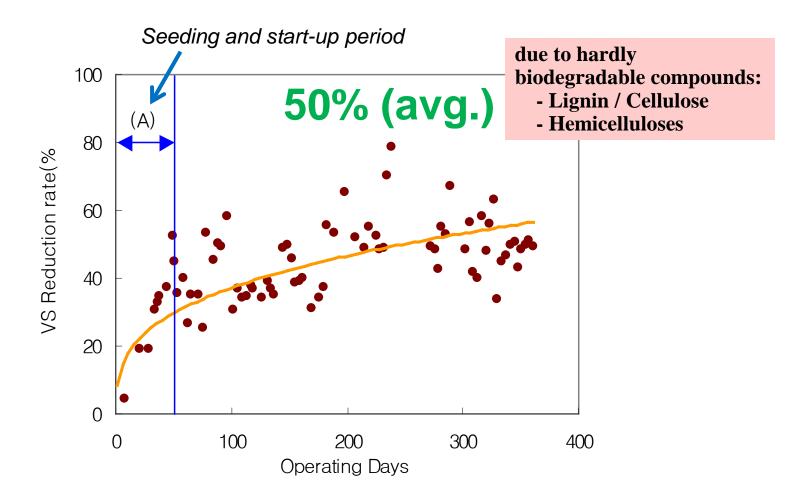
Temp (°C)	Free NH ₃ (mg/L)
25	28
30	39
35	54
55	177

at T- NH4 = 2000 mg/L, pH = 7.4

T-NH4	Free NH3 (mg/L)
1000	27
2000	54
3000	82
4000	109

at Temp. = 35 °C, pH = 7.4

VS reduction



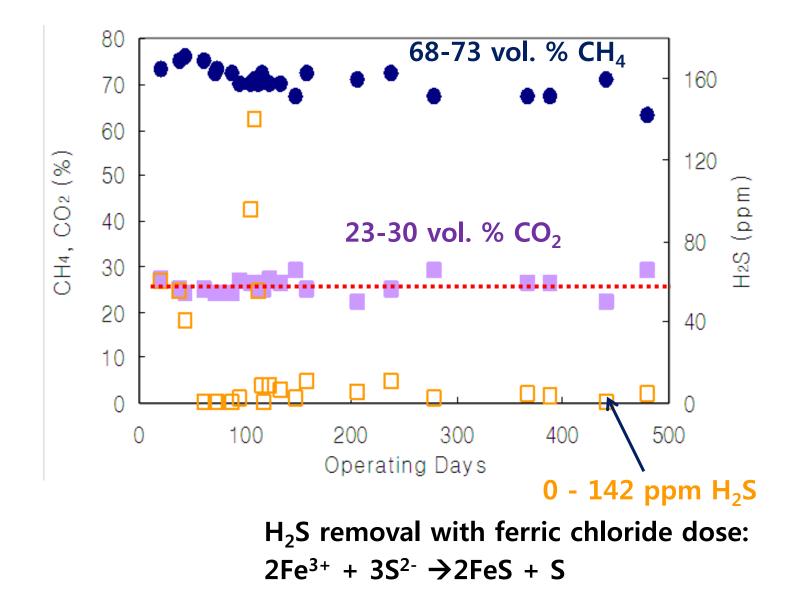
VS reduction of farm scale CSTR Biogas plant digesting swine manure

Characteristics of Influent Pig Slurry and Digester Effluent

Compounds	Raw Pig Slurry	Digester Effluent ^{a)}
рН	7.15-8.55	6.32-8.54
TS	3,750 – 40,800	750-1,500
VS	2,620-29,115	487-975
COD	4,530-44,800	764-3,740
SCOD	1,750-34,580	580-2,200
TN	500-3,561	-
NH4 ⁺ -N	290-1,250	715-1,500
ТР	120-580	-
PO ₄ ³⁻ -P	60.5-480	15-152
Alk. as CaCO ₃	1,742-7,882	2,966-6,606

All units in mg/L except pH

Biogas Composition



Inconvenient Truth of Digestate (liquid fertilizer)



In Germany, All year round utilization

In Korea, Only spring and fall (2 times)

Temperature Shock Effect

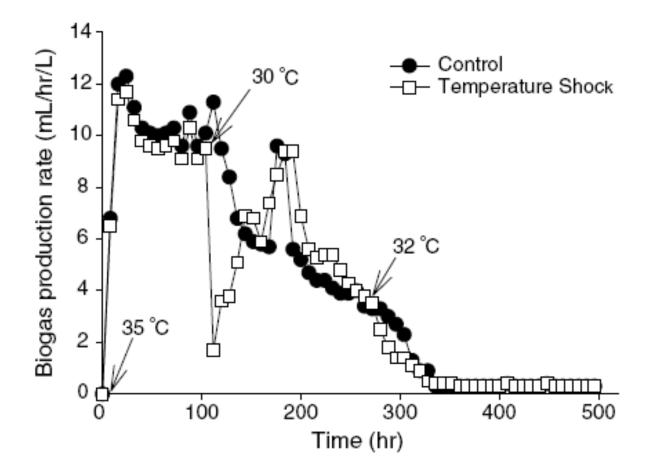


Fig. 5. Influence of temperature shock on the digestion of swine manure at a 20% (v/v) feed load.

Chae et al., Biores. Technol., 2007

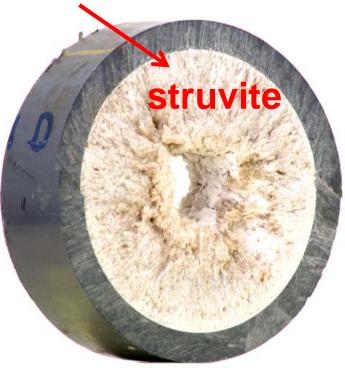
Think Differently!!! (Diverse choice for pipe materials)



(Courtesy Krieg & Fischer GmbH)

What Happen if you do not control struvite?

- Reduce flow capacity, foul pumps, and damage valves
- Occur after point of turbulence or pressure drop



(Courtesy enorca.blogspot.com)



Struvite encrusted roller



Struvite

- Magnesium ammonium phosphate MgNH₄PO₄ · 6H2O
- White, yellowish (or brownish) white in color
- FW = 245.41
- Specific density = 1.7
- Very insoluble in water, pK_{so} = 12.6 13.15 at 25°C

$MgNH_4PO_4.6H_2O \Leftrightarrow Mg^{2+} + NH_4^+ + PO_4^{3-} + 6H_2O \qquad pK=12.6$

Struvite formation occurs when the conditions are such that the concentration product exceeds the struvite conditional solubility product

Struvite Control

Prevent formation by reducing reactant ion(s) Mg⁺²+ NH4⁺+ PO4⁻³+ 6 H2O \rightarrow MgNH4PO4•6H2O

Minimize build-up by eliminating turbulence and/or using smooth pipe materials

Mechanical cleaning: hydro-jetting / mechanical grinding

BUT Questionable Effectiveness...

Problems with Current Struvite Control Techniques

Reduce PO₄³⁻with metallic salts

- Ferrous/ferric chloride and alum most common
- Requires large chemical dose to be effective
- Increases inorganic fraction in biosolids
- Increases risk of forming other deposits
 - Ferrous phosphate (vivianite) Fe₃(PO₄)₂ · 8H₂O
 - Aluminum silicates

Phosphate recovery from ferric phosphate salt(s) is nearly impossible

Lower pH with acid

- Large dose required for any significant change
- Increases risk of corrosion
- Requires handling of hazardous material

Struvite: Slow-release fertilizer



Recovered struvite



gure 1. Stuvite crystallizer used for 1.4 gal/min. flow rate of vered digester liquid. A) Add chemicals for Mg and pH inease at bottom of crystalizer cone or in inflow line; B) Wasteater inflow; C) Struvite crystals in bottom of cone

Design & Construction Failures

Overfilling of hydrolysis tank (source: Krieg & Fischer GmbH)

Our experience (leakage)



...until the cover of the hydrolysis tank had been pushed up fluid pressure and blasted from the tank.

Design & Construction Failures

Y City AD for food waste leachate



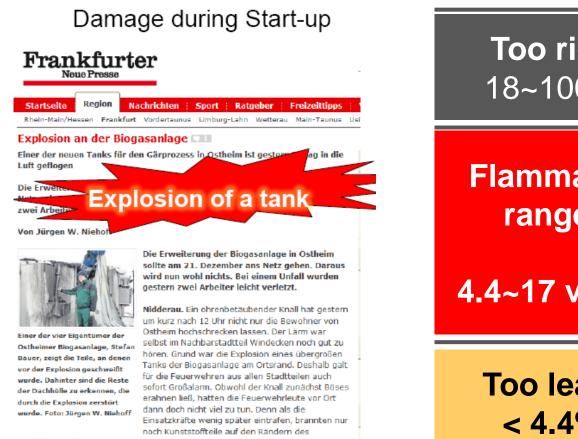
Tubular UF membrane for solid separation



Strainer blockage

Explosion & Fire Damage

During start-up explosive methane/oxygen mixture exists. Operator is not yet experienced with his biogas plant.



Methane explosive limit (%)

Too rich 18~100%	100%
Flammable range:	upper explosive limit:15-17%
4.4~17 vol.%	lower explosive limit: 4.4-5%
Too lean: < 4.4%	

Explosion of J City biogas holder

Gas holder 2,000 m³

Anaerobic digester 3,000 m³×2

Suggested biogas use (CHP)

Engine-generator

420 kWe

Gas consumption

Expected benefits

190 m³/hr

1.5 times greater energy recovery
Peak cut of electric demand

Summary: AD Process for Korea

Feedstock

- Mono-giestion (single feedstock)
- Co-digestion (food waste + farm waste)

Temperature

- Mesophilic (30~40 °C)
- Thermophilic (50~65 °C)

Digestion

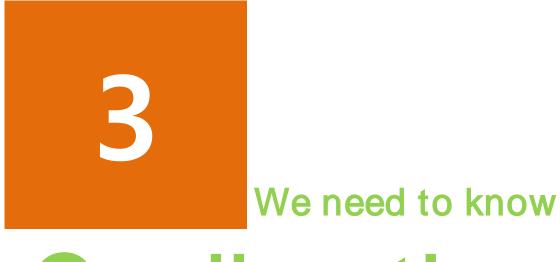
- Dry (>25% DM)
- Wet (normal)

Biogas utilization

- Boiler vs. CHP
- Upgrading

Plant design / Stage

- Batch
- Continuous (CSTR)
- Single- vs. Multi-stage



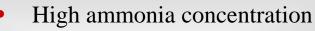
Co-digestion

Anaerobic Co-digestion



Piggery wastewater

Food waste



- High pH value (about 8.0)
- Energy depleted waste

Co-digestion

High organic strength Low pH after hydrolysis and acidogenesis

- Supplement micronutrients
- Improve buffering capacity
- Reduce ammonia inhibition
- Improve the organic strength

Source: Prof. Chang D.J.

Feedstock Characteristics

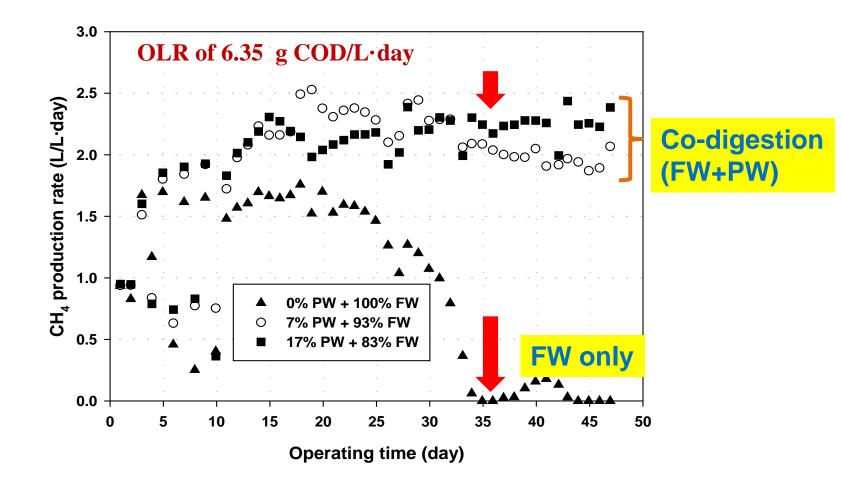
		Food wast	e	Piggery wastewater				
Parameter	Han and Shin (2004)	Zhang et al. (2007)	This study	Ahn et al. (2006)	Hansen et al. (1998, 1999)	This study		
pН				6.37 ± 0.10	7.62 ± 0.02	6.64		
TS (%)	20.5	30.90 ± 0.07	18.1 ± 0.62	6.18 ± 0.04		5.95		
VS (%)	19.5	26.35 ± 0.14	17.1 ± 0.58	4.45 ± 0.02	4.5 ± 0.1	3.89		
VS/TS	0.95	85.30 ± 0.65	0.94 ± 0.0004	0.72		0.65		
Total COD (g/L)			238.5 ± 3.83	130.8 ± 3.0		94.2		
Soluble COD (g/L)			106.6 ± 5.28	59.7 ± 0.9		54.2		
Carbohydrate (g/L)			151.7 ± 22.2					
Lipid (g/L)			23.3 ± 0.45	20.1 ± 0.1	4.86	2.30		
Carbon, C (% of TS)	51.4	46.78 ± 1.15	46.6738					
Hydrogen, H (% of TS)	6.1		6.3894					
Oxygen, O (% of TS)	38.9		36.3919					
Nitrogen, N (% of TS)	3.5	3.16 ± 0.22	3.5392					
Sulfur, S (% of TS)	0.1		0.3299					
TKN (N g/L)			5.42 ± 0.26	7.3 ± 0.1	6.6	7.6		
TP(g/L)			1.49 ± 0.09					
Ammonia-N (g/L)				4.8 ± 0.1	5.3 ± 0.1	4.95		
Total protein (g/L)				15.8 ± 0.9	8.13	16.6		
Alkalinity (CaCO ₃ g/L								
C/N	14.7	14.6	13.2	6.72		4.35		
Source: Prof. Chang D.J.								

Metal Element Levels in Food Wastes and Piggery Wastewaters

Property		Food wast	te	Piggery wastewater			
(mg/L)	Zhu et al. (2008)	Zhang et al. (2007)	This study		Creamer et al. (2010)	This study	
Sodium (Na)	143 mg/L		3547.65	900 ± 520	155	606.65	
Magnesium (Mg)	12.5	453±32	144.92	144.92	551	672.15	
Aluminium (Al)		1202 ± 396	10.01			41.28	
Potassium (K)	160	2913±356	3389		501	3956.82	
Calcium (Ca)	38	2160 ± 290	274.20			1775.03	
Chromium (Cr)		3±1	0.403	1.10 ± 1.15		0.169	
Manganese (Mn)	0.12	60 ± 30	2.2294	25 ± 32	45.1	24.9328	
<u>Iron (Fe)</u>	<u>1.35</u>	<u>766±402</u>	<u>7.36</u>	<u>127 ± 160</u>	<u>177.6</u>	<u>98.91</u>	
Cobalt (Co)		<u></u>	<u>nd</u>	<u>0.14 ± 0.16</u>	=	<u>0.1188</u>	
<u>Nickel (Ni)</u>		<u>2±1</u>	<u>0.4417</u>	<u>0.94 ± 0.94</u>	<u>0.6</u>	<u>0.4542</u>	
Copper (Cu)	0.17	31±1	7.0927	42 ± 51	13.7	39.1763	
Zinc (Zn)	0.36	76 ± 22	19.1965	172 ± 176	133.2	154.5396	
<u>Molybdenum(Mo)</u>	<u>0.01</u>		<u>0.0585</u>	==		<u>0.418.7</u>	
Cadmium (Cd)		<1	0.0531	0.10 ± 0.09	0.1	0.0142	
Lead (Pb)		4±3	0.4073	0.65 ± 0.55		0.3348	

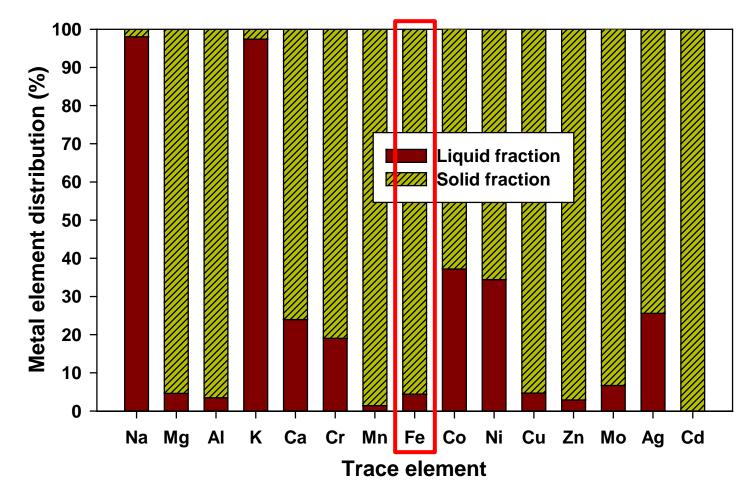
Source: Prof. Chang D.J.

Anaerobic Co-digestion of Food Waste and Piggery Wastewater



- The addition of piggery wastewater increased the stability of anaerobic digestion of food waste.
- Some substances from piggery wastewater help the process performance.

Metal Elements Distribution in Solid Fraction and Liquid Fraction of Piggery Wastewater



It's highly possible that trace elements in the solid fraction contributed the enhanced anaerobic digestion of food waste

Thanks to Mr. Torsten Fischer

for sharing his valuable knowledge and materials. Krieg & Fischer Ingenieure GmbH



Thanks to Prof. Chang D.J. (MyoungJ Univ.) for sharing his valuable materials for co-digestion.



For energy self-sufficiency of WWTPs

Renewable Energy Technologies

0.5% 395,121 toe/yr

0.8%

SEOUL

-EN de

Energy Self-sufficiency

ENERGY SAVINGS	RENEWABLE ENERGIES
10-20% (fine bubble controlled aeration, energy efficient motors and pumps)	5-10% (wind power, photovoltaic, solar thermal power, geothermal power)
ENERGIES FROM SEWAGE FLOWS	SLUDGES
2 – 10% (hydro-turbines, heat pumps, in-sewer heat exchangers)	40 → 60-80% (anaerobic sludge digestion, pre-treatment to increase digestability)

Figure 6: Major components of the 'positive energy plant' – zones where the energy efficiency of wastewater treatment can be improved.

Concluding remarks

The current configuration of urban water management systems means that significant quantities of water and energy are consumed and nutrients are inefficiently managed. Historically, water and energy have been managed independently, but in the City of the Future the whole water cycle should



Water21, April 2012, p16, Lazarova, Choo, and Cornel

HOW?

You will see more details during the field trip next class...



건설사업

차별화된 기술력으로 끊임없는 혁신을 추진

KOLON GLOBAL

Thank you very much! ckj@kolon.com





Status of organic waste to energy technologies

2013. 10. 28

DOHWA Engineering Co., LTD.





1. Policy and Trend of Renewable Energy

2. Treatment Technology of Organic Waste

3. Characteristic of Organic Waste

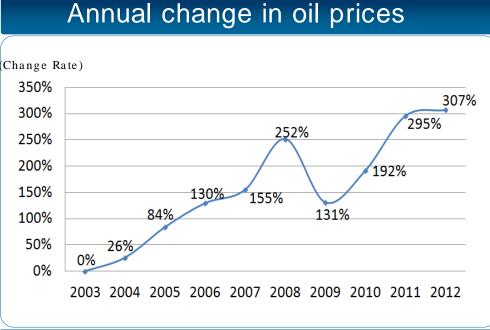
4. Technology for Organic Waste Energization

5. Cases in Domestic of Food Water Energization

1. Policy and Trend of Renewable Energy



Background of Renewable Energy



International oil price rapidly increase recently 3 years

Year	Cost (\$/bbl)
2003	26.80
2004	33.77
2005	49.37
2006	61.55
2007	68.43
2008	94.29
2009	61.92
2010	78.13
2011	105.98
2012	109.03

International oil price is 107.93 \$/bbl(2013.09)

10th largest energy consumer of the world

relies on imports for 97%



EU	The goal of renewable energy supply in 2020 is 20 % of total energy 34 % of generation , 10 % of transportation fuel								
Japan	(Rec	The goal of renewable energy supply in 2020 is 20 % of total energy (MOE, ' 10.1) (Reopen to give solar energy subsidy(' 09.1) Mandatory for purchase remain solar energy (' 09.11)							
USA	Provide renewable energy which is 25% of eletric power in 2025 (Announcement of Obama Government)								
China	The goal of renewable energy supply in 2020 is 15% of Primary Energy (300GW of Water, 30GW of Wind, 1.8GW of Solar, 30GW of Biomass) Develop and supply plan of Wind, solar, water etc.								
Germany	ermany The gold of renewable energy supply in 2020 is 18% of Final Energy (30% of generation amount)								
Div.	USA Japan Germany Denmark UK Korea								
Supply rate('0	7)	5.0%	3.4%	8.6%	18.1%	2.4%	2.4%		
Goal 10.9%('30) 20%('20) 18%('20) 30%('20) 15%('20) 11%('30) Data : Energy Balance of OECD Countries('09), IEA									

Set the goal of renewable energy supply and under continuous efforts



R&D Strategy of Major Advance Countries

" Establish New Growth Power Driving Strategy through Innovational Energy Technology"

America

American Recovery and Reinvestment Act('09.2)

26 billion dollars of budget for ARPA- E

*470% of Budget of DOE is increased compare to 2008

Japan

New National Energy Strategy ('06.5) Improve 30% of energy efficiency, and achieve 40% of oil development by 2030

Announcement of Cool Earth ('08.4)

Announced 21 Innovation technology

ΕU

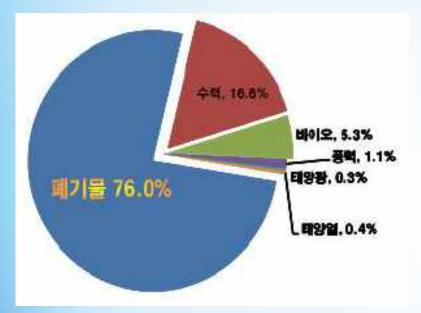
- An Energy Policy for Europe('07.1)
 Improve energy efficiency
- SET Plan('07. 11)
 Long term plan for clean energy society based on low carbon technology

Focus the capability Nation securing energy technology to prepare climate changes and dominating the global market



Policy of Renewable Energy in Domestic

Composition of Renewable Energy>



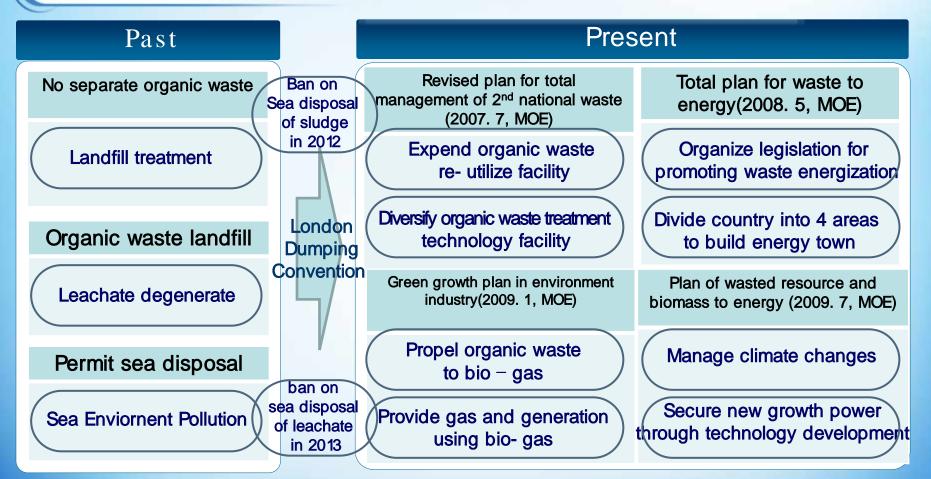
- Renewable energy using Bio is 5.3% of total energy
- Plan to increase rate of bio- energy up to 30% by 2030
- Production cost of bio energy among national renewable energy is similar with 10% of solar and 70% of wind

(단위 : Won/kwh)

Div.	Solar	Wind	Water	Waste	Bio			
Unit cost of production	716	107	70	71	75			
Establish goal plan in supply structure								
of Bio- Energy among renewable energy								



Policy Change of Organic Waste in Domestic



Change in paradigm for management policy of making resource recirculation society

2. Treatment Technology of Organic Waste



Treatment Technology of Organic Waste in Domestic

Feeds	Installation cost is cheap but expensive production cost, odor from dehydration process and difficult to look for source of fodder demand					
Compost	Many facilities are already installed because of easy access, but low additional value of by- product and need a wide area					
Combine to sewage treatment	Low installation cost because of using the existing sewage plant, but low operation result and need high- level treatment process					
	"Waste to Energy" National policy change					
Anaerobic Energization	Low odor and possible to be energization by manufacturing bio- gas Low operation cost , preparing for climate changes and suitable for low carbons policy					
Change into A	Anaerobic energization is coincide with government policy					

3. Characteristic of Organic Waste



Characteristic of Organic Waste in Domestic

Organic waste				Organi	c wast	e leach	nate			
Div.	Moisture (%)	TS (%)	VS (%)	VS/TS (%)	Div.	BOD (mg/L)	COD (mg/L)	SS (mg/L)	T- N (mg/L)	T- P (mg/L)
K city	82.35	17.65	14.33	81.19	K city	97,856	138,417	68,042	8,289	672
D city	83.49	16.51	14.29	86.55	l city	83,617	141,393	42,653	3,246	498
I city	76.26	23.77	17.45	73.41			,			
Literature	74 ~ 85	15 ~ 26	13 ~ 19	73 ~ 86	Literature	61,097 ~ 82,501	136,570 ~ 160,146	16,385 ~ 50,984	2,527 ~ 2,835	226 ~ 656

Data : K city, D city, I city, basic design report, literature : feasibility study of biogas development using organic waste, 2008, SLC

Moisture and VS rate is high, High concentration leachate is generated



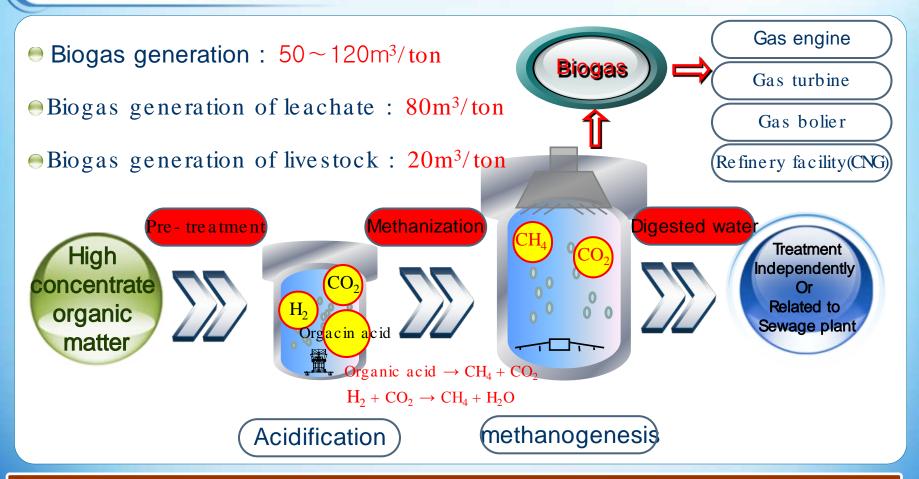
Characterization of Organic Waste in Domestic



Need high efficiency energization facility

4. Technology for Organic Waste Energization DOHWA

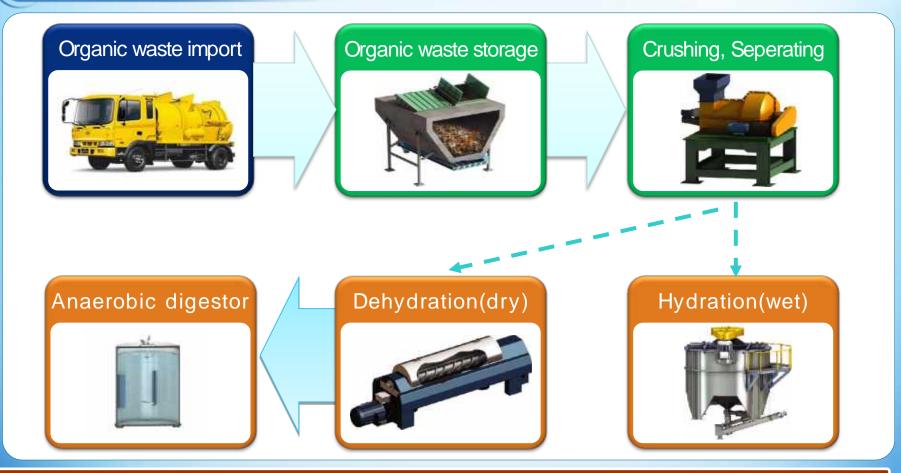
Concept of anaerobic digestion technology



Biogas is generated by anaerobic digestion process



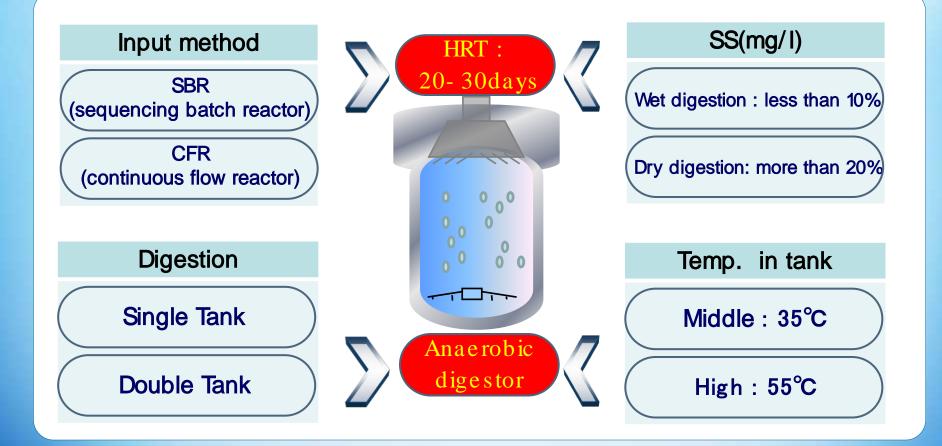
Input & Pre- Treatment Process



Process for micro- organism to be easy to use



Anaerobic Digestion Process Types



Select Anaerobic digestion process according to operation condition



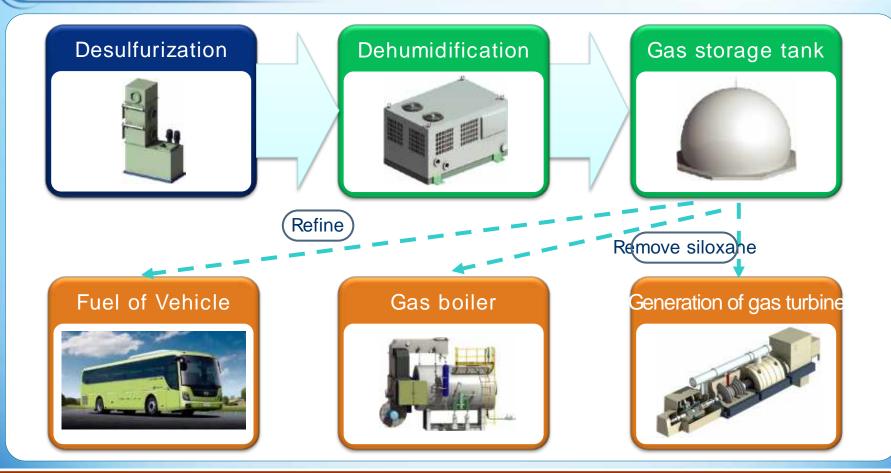
Operating Condition of Anaerobic Digestor

Div.	Condition	Remarks
Temp.	middle: 30 [~] 40°C high: 50 [~] 60°C	Additional heating cost of reactor when temperature increases
HRT	15~ 30 days	CSTR process
рН	Near 7.0	Optimal condition of methanogen
ORP	Less than - 300 mV	Organic carbon's reduction condition
Removal rate of VS	70~ 85%	Differences depend on characteristics of organic waste
Removal rate of COD	40~ 95%	Big differences depend on characteristics of organic waste

Basic conditions to maintain high activity of Anaerobic micro- organism



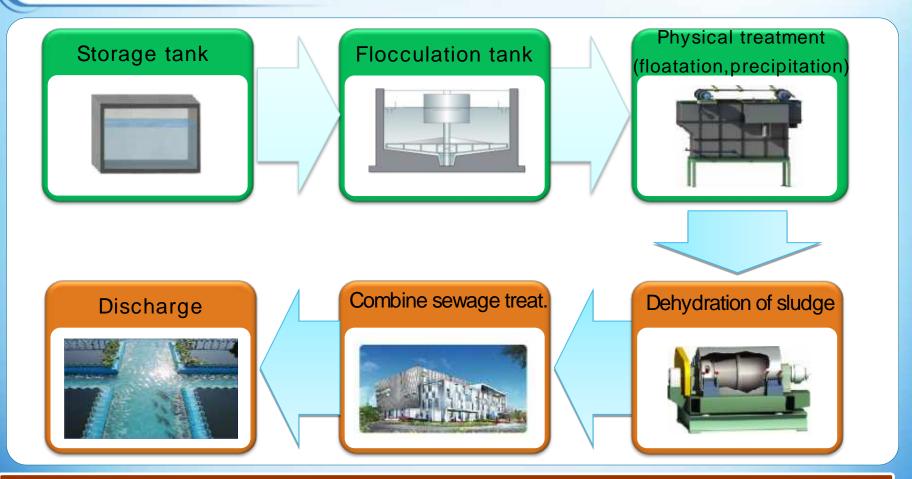
Biogas Utilization Process



Generated biogas can be utilized to various energy sources



Waste water treatment process



Digested waste water are treated independently or ties to sewage plant



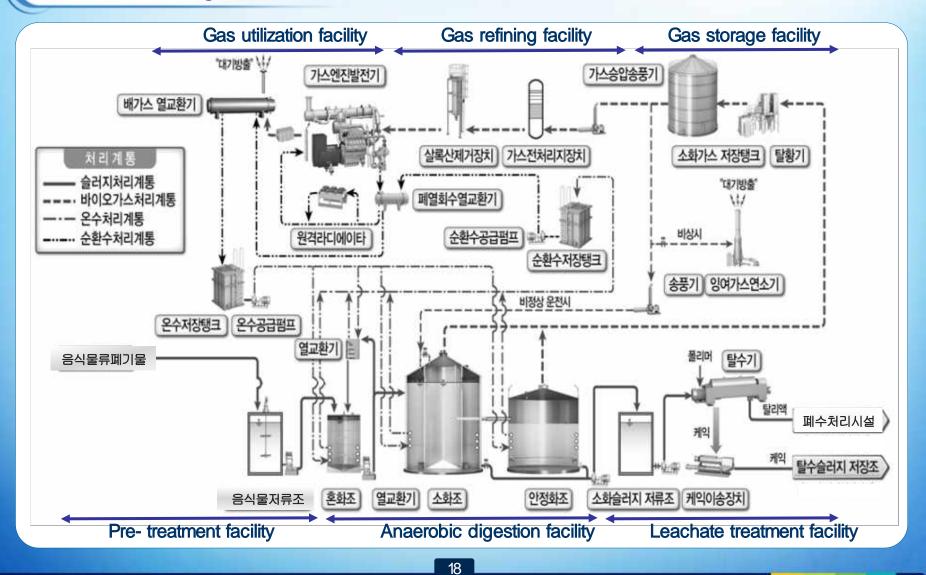
Odor Treatment Process



Removal odor by separating high and low concentration

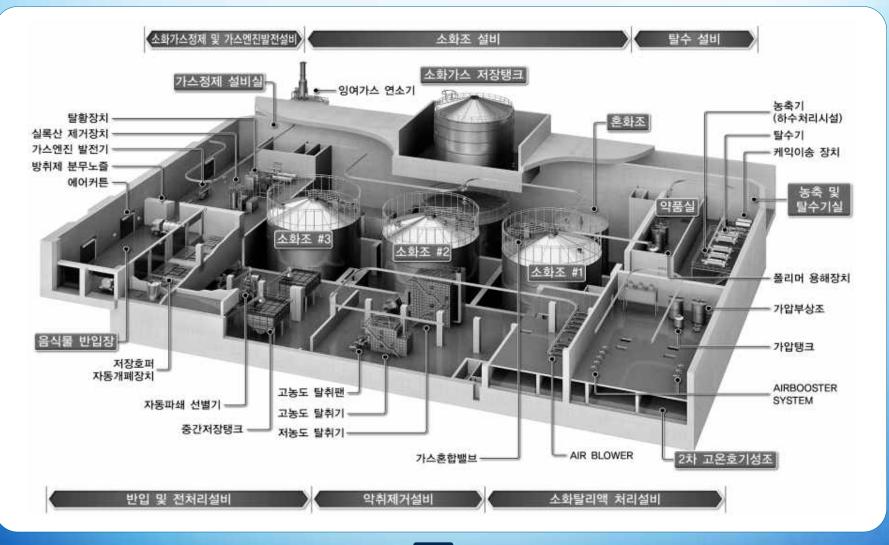
5. Example of Organic waste to energy in domestic DOHWA

Process diagram





Case of underground(P city- Private investment business)





Introduction Techniques of Domestic

	Div.	Dae- woo	Seo- hee	Hallasanup	Ecoday
Te	Technology Own technology OWS(E		OWS(Belgium)	OWS(Belgium)	Own technology
	Name	DASB	Dranco Process	Double wet middle temp. digestor	E. PFR- 2 SYSTM
	Waste	Leachate, livestock waste water	Organic waste	Organic waste, livestock waste water	Organic waste, Leachate
	Process	Double wet	Single dry	Double wet	Double wet
	Temp.	middle(35~40°C)	high(55± 2°C)/ middle(35± 2°C)	middle(within 35°C)	middle(within 35°C)
	Time	25 ~ 30 days	Within 30 days	(1 st 3 ~ 5days, 2 nd 15 ~ 20days)	Within 15 days
	CH ₄	Within 60%	60 ~ 70%	60 ~ 75%	Within 75%
	naracter of process	No- power stirring by gas pressure	No hydrolysis process	pH control by returned discharge water	High load, Fast treatment
	Result	•Nambu waste treatment 1,700 ㎡ (leachate) •Asan(100 ton/day)	 Busan(200 ton/day) Dongdaemun(98 ton/d) 	 Paju(30 ton/day) milyang(20 ton/day) Sokcho(20 ton/day) 	•Paju(30 ton/day)



Introduction Techniques of Overseas

	Div.	ARROWBIO	BTA	OWS	HESE
Technology Own technology		Own technology	Own technology Own technology		
Name		Double wet anaerobic	Single and double wet anaerobic	DRANCO Process	Double wet anaerobic digestion
	Waste	Organic Waste	Organic waste, livestock waste, sludge	Organic waste	Organic waste, livestock waste
	Method	Double wet	Single dry, double wet	Single dry, double wet	Double wet
	Temp.	middle(35 [~] 40°C)	middle(35 [~] 40°C)	high(50~65°C)	middle(within 35°C)
	Time	HRT : 1 ~ 3days SRT : 80 ~ 90days	Single : 14 ~ 16days Double : 5 ~ 7days	15 ~ 30days	19days
	CH ₄	81%	65 ~ 75%	50 ~ 60%	Within 60%
Cha	racteristic of process	UASB	No- power stirring by gas pressure	Directly supply steam to reactor	Maintain Aerobic at hydrolysis
Performance		•Tel Aviv, Israel(100 Ton/day)	•Kirchstockach, Germany(20,000 Ton/yr) etc.	•Rome, Italy (40,000Ton/yr) •Leonberg, Germany (30,000 Ton/yr) etc.	•Leicestershire, UK (40,000 Ton/yr) etc.



Cases of Organic Waste to Biogas in Domestic

Div.	Gwangju	SLC	Dongdaemun gu
Picture			
Capacity (Ton/day)	150	1,300	98
Completion	2007. 02	2008. 04	2010.10
Treatment	Leachate	Leachate	Organic waste
Utilization of biogas	Heat reactor	Heat reactor and air conditioning and heating	Electronic and steam generation for onsite
Anaerobic Process	Wet high temp.	Wet double phase	Dry single phase
Remarks	Anaerobic digestion of organic waste leachate	Largest in national of anaerobic digestor	Undergrounding and Making park



Project Plan of Organic Waste to Biogas

Local Government	Treatment	Period (year)	Capacity (T/d)	Case of underground
Sokcho	Organic waste	09- 10	40	-
Daegu	Organic waste	09- 12	300	Treatment facility(underground), Park(ground)
Goyang	Organic waste	09- 12	260	Treatment facility(underground), Park(ground)
Kimhae	Leachate	09- 12	100	_
Jinju	Leachate	10- 11	150	-
Unpyoung	Organic waste	10- 12	100	Treatment facility(underground), Park(ground)
Gwangju	Leachate	10- 12	300	Treatment facility(underground), Park(ground)
Ulsan	Leachate, Livestock	10- 12	150	-
Chungju	Leachate	10- 12	203	



Case of Changing Facility into Park(K city- 300 Ton/day, Detail design)



Possible to make the park being designed with efficiency and environment friendly



Case of Changing Facility into Park (P city- 200 ton/day, on going private investment business)





Thank you

Waste to Energy Status of Biomass Fired Power plant

Indonesia

KOREA

2013. 11. 4





/ . POLICY and TREND of RENEWABLE ENERGY

//. BIOMASS as CLEAN ENERGY

///. BIOMASS ENERGY MARKET OUTLOOK

IV. BIOMASS ENERGY TECHNOLOGY TRENDs

Case: KTH Biomass Plant

I. POLICY and TREND of RENEWABLE ENERGY



I. POLICY and TREND of RENEWABLE ENERGY

Background of Renewable Energy is

Annual change in oil prices	Year	Cost (\$/bbl)
(Change Rate)	2003	26.80
350%	2004	33.77
300%	2005	49.37
250% 200%	2006	61.55
150% 130% 155%	2007	68.43
100% 84% 131%	2008	94.29
50% 26% 0% 0%	2009	61.92
2003 2004 2005 2006 2007 2008 2009 2010 2011 2012	2010	78.13
International oil price rapidly increase	2011	105.98
recently 3 years	2012	109.03

International oil price is 107.93 \$/bbl(2013.09)

> 10th largest energy consumer of the world \Rightarrow relies on imports for 97%

EU	The goal of renewable energy supply in 2020 is 20 % of total energy 34 % of generation , 10 % of transportation fuel						
Japan	Reopen to give	The goal of renewable energy supply in 2020 is 20 % of total energy (MOE, '10.1) Reopen to give solar energy subsidy('09.1) Mandatory for purchase remain solar energy ('09.11)					
USA	Provide renev (Announceme	•••	which is 25% c Government)	of eletric powe	r in 2025		
Chaina	(300GW of Wa	iter, 30GW of V		of Solar, 30GV	Primary Energ / of Biomass)	У	
Germany	-	newable energet ation amount)	gy supply in 2)	2020 is 18% of	Final Energy		
Div.	USA Japan Germany Denmark UK Korea						
Supply rate('07)	5.0%	5.0% 3.4% 8.6% 18.1% 2.4% 2.4%					
Goal	10.9%('30)	20%('20)	18%('20)	30%('20)	15%('20)	11%('30)	
Data : Energy	Data : Energy Balance of OECD Countries('09), IEA						

Set the goal of renewable energy supply and under continuous efforts

R&D Strategy of Major Advance Countries

"Establish New Growth Power Driving Strategy through Innovational Energy Technology"

America

American Recovery and Reinvestment Act('09.2)

26 billion dollars of budget for ARPA-E

*470% of Budget of DOE is increased compare to 2008

Japan

- New National Energy Strategy ('06.5) Improve 30% of energy efficiency, and achieve 40%
 - of oil development by 2030
- Announcement of Cool Earth ('08.4)

Announced 21 Innovation technology

ΕU

An Energy Policy for Europe('07.1)

Improve energy efficiency

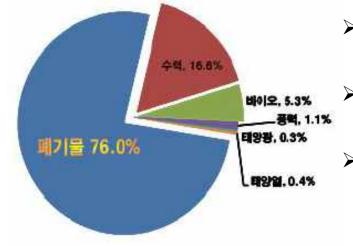
SET Plan('07. 11)

Long term plan for clean energy society based on low carbon technology

Focus the capability Nation securing energy technology to prepare climate changes and dominating the global market

Policy of Renewable Energy in Domestic

<Composition of Renewable Energy>



- Renewable energy using Bio is 5.3% of
 - total energy
- Plan to increase rate of bio-energy up to 30% by 2030
- Production cost of bio energy among national renewable energy is similar with 10% of solar and 70% of wind

Div.	Solar	Wind	Water	Waste	Bio
Unit cost of production	716	107	70	71	75

Establish goal plan in supply structure of Bio-Energy among renewable energy

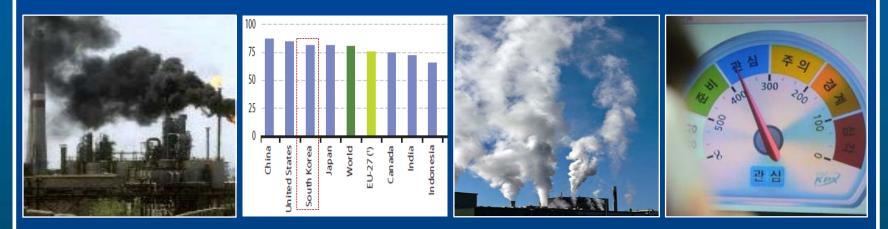
II. BIOMASS as CLEAN ENERGY



II. BIOMASS as CLEAN ENERGY

Alternative Clean Energy in Domestic

- To address global warming, the development of alternative clean energy source like biomass must accelerate to reduce our dependence on fossil fuel.
- Korean power demand ranked third place of electricity consumption rate among global top 8 major countries (International Energy Agency(IEA))
- A steep increase of power demand especially on specific time (summer and winter) can lead to electricity crisis like Blackouts



Domestic Policy for Alternative Clean Energy

- Government needed alternative clean energy and introduced the "Renewable Energy Portfolio Standard(RPS) " to satisfy power demand and reduce greenhouse gas emissions and opened the new RECs(Renewable Energy Certificates) market
 - -RPS duty supply: 2.0% by 2012, 2.5% by 2013, 3.0% by 2014, 10% after 2022 of total electricity generated
 - -2012year result: KEPCO subsidiaries(6 companies) are carrying out 64% (3,808 GWh out of total 5,911 GWh)
- REC Performance in 2012

Unit(GWh)

Constator	Total amount of duty	Performa	ormance result Per		Penalty
Generator	(photavoltaic)	Self-supply	Outside purchase	Implementation delay	(0.1 billion won)
Kosep(남동)	834(43)	62(18)	302(23)	470(2)	105
Komipo(중부)	738(43)	89(6)	303(36)	346(1)	59
Kowepo(서부)	761(43)	72(6)	366(33)	323(4)	45
Kospo(남부)	834(43)	115(13)	451(29)	268(1)	8
EWP(동서)	734(43)	72(13)	351(30)	311(0)	44
KHNP(한수원)	2,010(43)	1,291(2)	333(37)	386(4)	0
Total	5,911(258)	1,702(58)	2,106(188)	2,103(12)	261
\Rightarrow Increased	d using biomass	s fuel for the	effective imr	plementation of	the RPS

Increased using biomass fuel for the effective implementation of the RPS to avoid penalty

Biomass?

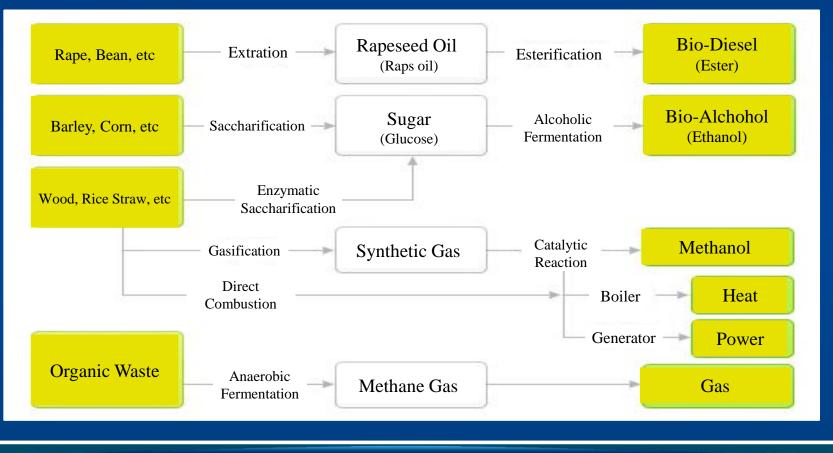
 Biomass is derived from sources of various types, such as agricultural, forestry, fishery, stockbreeding, and waste resources, and the technologies to use those various types also vary widely.





Biomass as a Renewable Energy Source

 Biomass can either be used directly via combustion to produce heat, or indirectly after converting it to various forms of biofuel, wood chip, pellet, biodiesel, bioethanol and biogas.



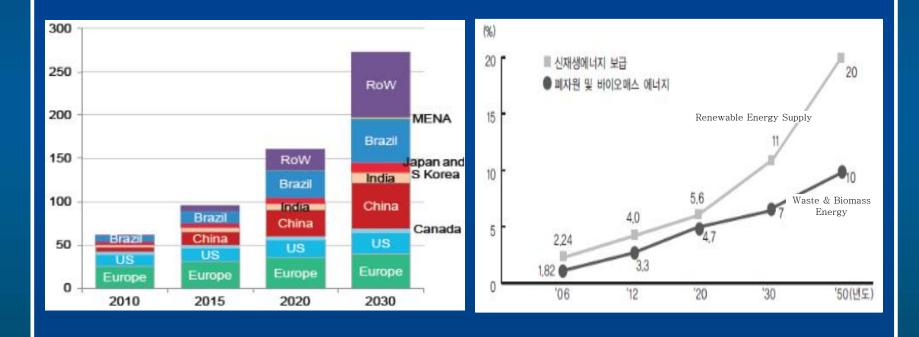




Ш. BIOMASS ENERGY MARKET OUTLOOK

Biomass Energy Demand Forecast (1)

 All over the world, biomass power plant is expected to continue high rate of increase by planning to supply 150GW around developing countries and BRICs such as Brazil, India, China and other countries by 2020.



Biomass Energy Demand Forecast (2)

 Korea completed the test of mixed fuel power plant of biomass energy and plans to mixed fuel in available plants with 3~10% of biomass content.



(unit : 1,000TOE)

Remark	2020	2030	2050
Primary Energy Demand Forecast	287,976	300,417	373,872
Long-term goal of renewable energy supply	5.6%(16,241)	11.0%(33,027)	20.0%(74,774)
Goal of waste resources and biomass energy supply	4.7%(13,383)	7.0%(21,000)	10.0%(37,387)

Secure Biomass Fuel

- Biomass production capacity is only 10,000 tons/year in domestic (requirement : 3.2 million tons)
- Indonesia has a lot of various forest resources
- Korean private capital are investing on Indonesian forest resources for longterm development with high technology.

Region	Company Name	Production	Area(10,000ha)
Sumatra	PT. HAN	Wood pellet	3.3
Zawa	National Forestry Cooperatives Federation	Timber	1.0
Control Kalimantan	Korindo	Wood pellet	6.0
Central Kalimantan	Taeyoung Global	Timber	6.0
	SK Networks co., Ltd.	Timber	3.0
Southern Kalimantan	National Forestry Cooperatives Federation	Wood pellet, Timber	8.0
Danua	Korindo	Palm oil, Timber	16.0
Papua	LG International Corp.	Timber	17.0

*** source : Korea-Indonesia Forest Cooperation Center internal data(2012)**

- Recently Biomass Fuel Global Market is expanding.
- Construction of Biomass Power Plant is increasing world-wide as well Korea.

IV. BIOMASS ENERGY TECHNOLOGY TRENDs



IV. BIOMASS ENERGY TECHNOLOGY TRENDS

Biomass Energy Production Technology

- Biomass energy production technology is divided into biogas production, solid derived fuel, bio-derived fuel technologies
 - \rightarrow solid derived fuel has been the most available in domestic.
- Solid derived fuel technology includes both Solid Refused Fuel(SRF) technology to manufacture wood pellets or charcoal and direct combustion technology to produce hot water and electricity by burning SRF.

ater heate

Heat Recovery Forms by Direct Combustion

In direct combustion Generator facility, 3 forms(electricity, Combustion Heat hot water, steam) are Steam exchanger aas energy (boiler) (water heater) applied for heat recovery. Combustion gas Combustion Hot water gas energy

Electricity

Hot water

Steam

Hot water

Types of Biomass Boiler

- Boiler has been most widely used in biomass energy conversion of direct combustion method.
- The forms of boiler are stoker grate, circulation fluidized-bed and bubble fluidized-bed.

Division	Туре	Fundamentals	Remarks
Stoker Grate		 Incineration take place by air coming from the bottom on stoker grate placing fuel. 	Europe
Circulation Fluidized—Bed		 Fluidized bed will circulate externally by repeating discharge and supply of sand. 	Domestic
Bubble Fluidized-Bed		 Fuel will incinerate quickly on bubled sand bed by mixing with air injected by perforated plate at the bottom of the furnace. 	Japan

KTH Biomass Fired Power Plant

Indonesia

KOREA

2013. 11. 11



CONTENTS

/. Summary

//. Project Process

///. Equipment List up & Drawings

IV. Shipping & Inspections

V. Construction

V/. Commissioning





I. Summary

Project	 KTH Biomass fired Power Plant
Location	• Kumai, Kalimantan, Indonesia
Site Area	• About 10,000 m ²
Facilities	 Boiler Capacity : 33 Ton/hr Power Capacity : 7.3MW(Net 6.5MW)
Client	• PT Korintiga Hutani(Korindo Group + OJI Paper)
Duty Scope	• EPC Work
Total EPC Amount	• USD 23,000,000
Period of Project	• 2011. 08. 29 ~ 2013. 03. 31 (19Months)

I. Summary

Location







II. Project Process

- 2010. 08. 28. : Submit Technical Proposal
- 2010. 11. 16. : Submit Commercial Proposal
- 2011. 02. 10. : Negotiation
- 2011. 05. 10. : Awarding
- 2011. 07. 21. : Contract EPC
- 2011. 09. 01. : Basic & Detail Design Start
- 2012. 04. 25. : Submit Final Detailed Design
- 2012. 05. 30. : Finished Pile Work
- 2012. 06. 20. : Shipping Material (Equipment, Raw Material etc.)
- 2012. 08. 18. : Start to Install Equipment
- 2013. 02. 10. : Mechanical Completion
- 2013. 03. 31. : Commissioning Complete

П. Project Process

Project Schedule

Actual	:
Plan	

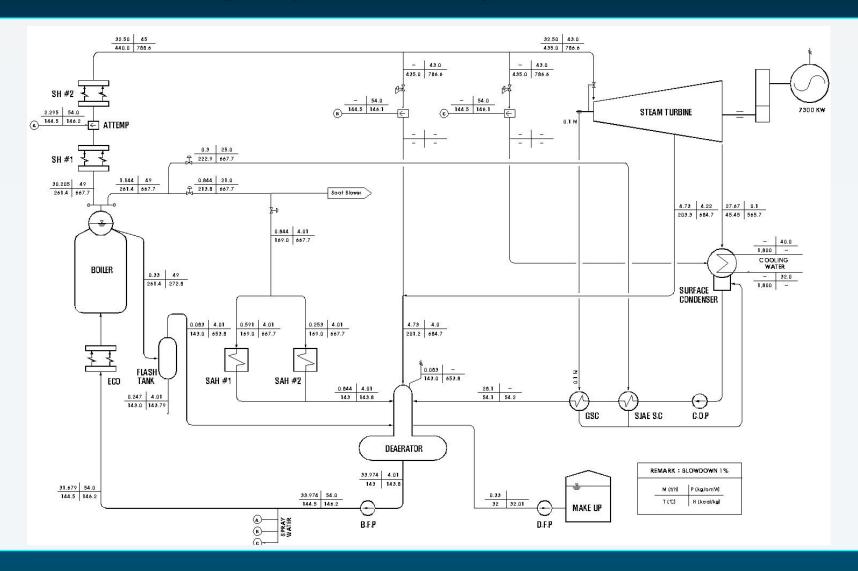
		2012							2013			
ITEM	DESCRIPTION	5	6	7	8	9	10	11	12	1	2	3
	MOBILIZATION	10	50	85	100	100	100	100	100	100	100 🥖	100
PREPARATION	SITE OFFICE & MESS	75	100	100	100	100	100	100	100	100	1.0	100
	MATERIAL HANDLING				35	100	100	100	100	100	1,0	100
	BOILER				5	10.15	33.2	75.55	92.13	96.86	6.86	100
	STG						9.75	21.6	64.76	94.79	100	100
	FUEL & ASH HANDLING									57	100	100
	FLUE GAS TREATMENT					15	59.67	80.9	81.3	92.18	100	100
	WATER TREATMENT							20.8	73.22	95.9+	100	100
	COOLING TOWER				5	10	10	40	46	.9	100	100
MECHANICAL	AIR COMPRESSOR									12	100	100
	TANK						2.5	53.9	92.45	100	100	100
	BFP						45	100	100	100	100	100
	PUMPS							22.65	38. 3	85	100	100
	CRANE								٥0	97.5	100	100
	CHEMICAL DOSING & SAMPLING							50	100	100	100	100
	STEEL STRUCTURE				7.5	40	75	87.5	97.5	100	100	100
PIPING	PIPING				2.5	15	31.41	44.2	65.17	81.81	100	100
	INSULATION									30	90	100
	TRANSFORMER									50	100	100
	SWITCHGEAR & MCC								7.5	77.5	100	100
	DC & UPS SYSTEM									50	100	100
ELECTRICAL	CABLE TRAY & CONDUIT							32.5	62.5	88.75	100	100
LLEOTRICAL	GROUNDING SYSTEM					40	7	92.5	98.75	100	100	100
	PAGING									2.5	100	100
	CABLE									10	100	100
	PLC									75	100	1000
INSTRUMENT	FIELD INSTRUMENT								15	75	100	100
	TUBING & WIRING									37.5	87.5	100
TEST	TEST & COMMISSIONING										37.5	100



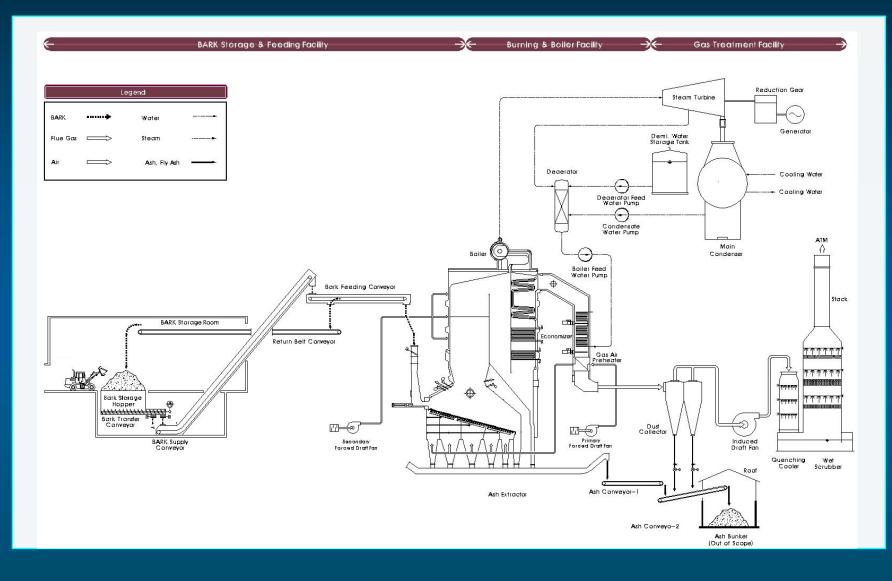
Equipment list

Description	Capacity	Remarks
Stoker Grate (Standardkessel GmbH in Germany)	 Fuel : Bark Chip Design Firing Capacity : 27.3 MW Design Fuel Flow : 10.8 t/h Fining Efficiency : 98% Annual Fuel Throughput : 165,000 ton/a Annual Operation Hours : 8,000 h/a 	
Boiler (Sookook Corporation In Korea)	 Boiler steam flow (MCR) : 32,760 kg/hr Super heater outlet temperature: 45 kg/cm².a Boiler steam temperature: 440 °C Boiler efficiency: 86 % Fuel(bark) consumption base on LHV : 2,184kcal/kg, 11,283 kg/hr 	
<mark>Steam Turbine</mark> (Shinnippon in Japan)	 Inlet steam Pressure : 43 kg/cm².a Temperature : 435 °C Flow : 32,460 kg/h Exhaust steam Pressure : 0.1 kg/cm²a Temperature : 45.45 °C Flow : 27,770 kg/h Output at generator terminal : 7.3MW 	

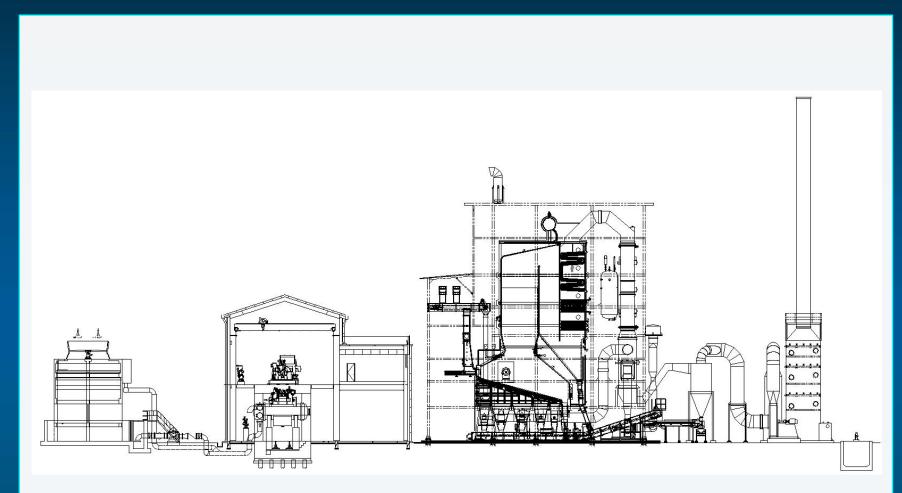
Heat Balance Diagram(Blowdown 1%)



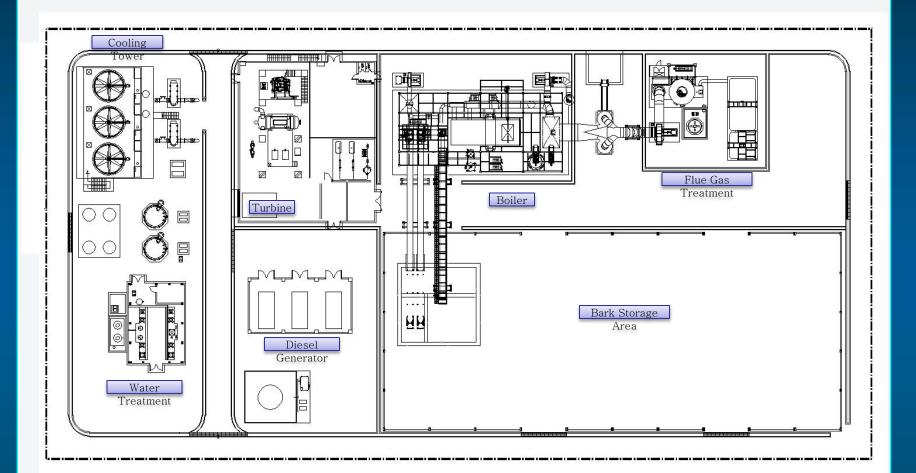
Process Flow Diagram



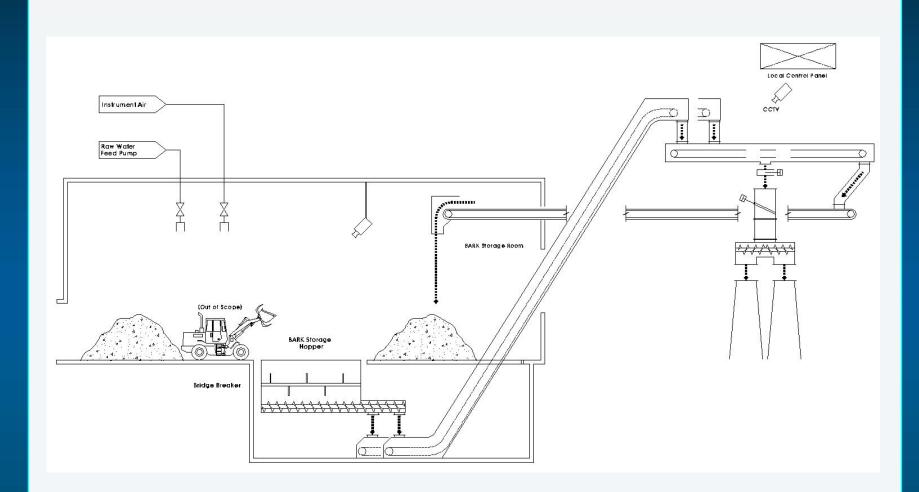
Section View



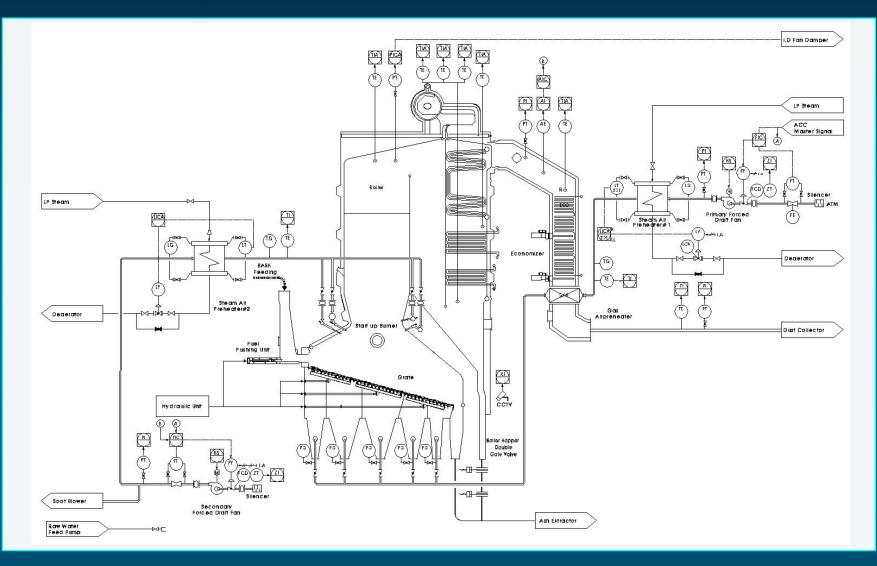
📙 Plot Plan



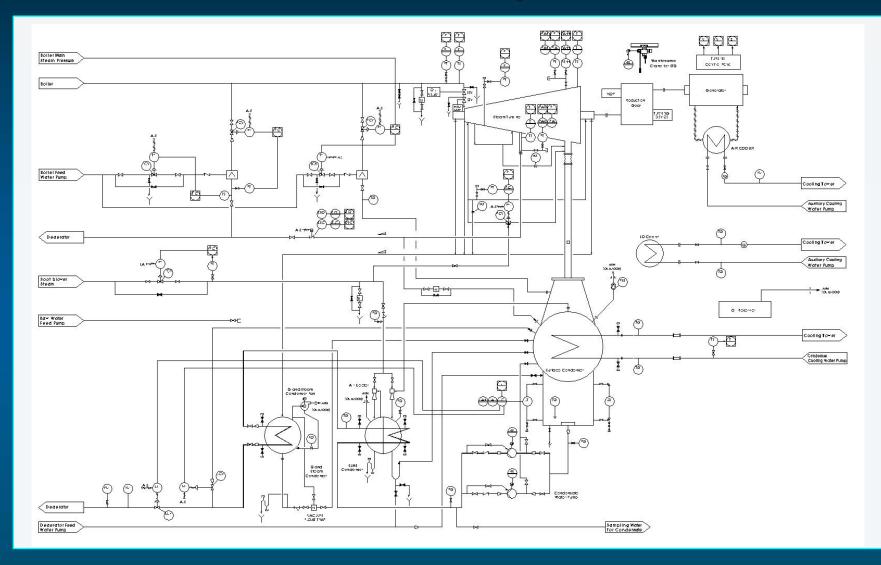
P&ID : Fuel Handling System



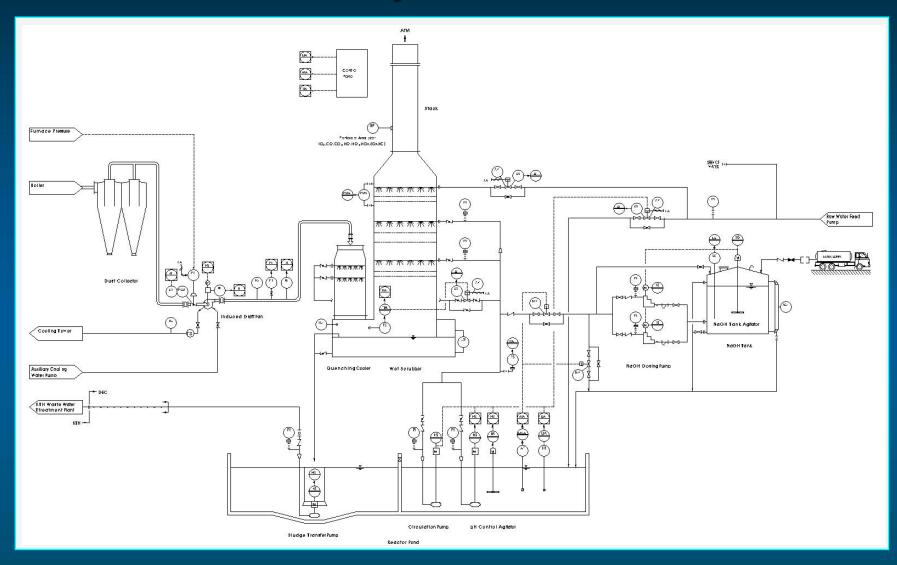
P&ID : Boiler System



P&ID : Steam Turbine & Condensate System



P&ID : Flue Gas Treatment System



IV. Shipping & Inspection



IV. Shipping & Inspection

Shipping













IV. Shipping & Inspection

Inspections













V. Construction

 Pile Punching & Head Treatment



Reinforcement Work



Column Reinforcement
 Arrangement Work



Boiler Steel Structure Erection







V. Construction

• STG 1st Slab - Install of Floor Post & Forms









V. Construction

STG Foundation - Removal of Forms







• Erection Boiler



Cleaning Access Floor For Control Room(STG Building)



Install Formwork for Cooling Tower







VI. Commissioning

After Completion of Commissioning



Thank you.

Organic waste-to-energy: Reduction and Recycling

Ki-Hoon Kang

Technology R&D Institute

DAELIM Industrial Co., Ltd.

Nov. 11, 2013

[Contents]

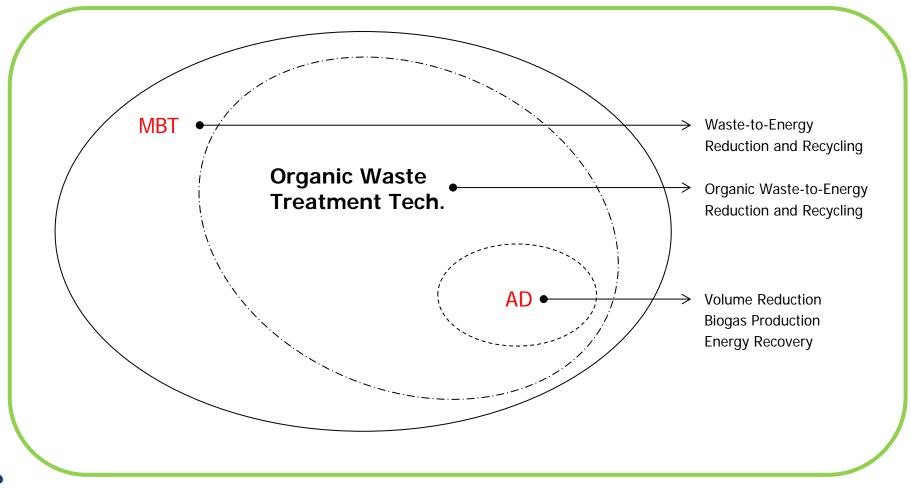
- Introduction
- Recycling as Resource (MBT)
- Waste-to-Biogas
- Pretreatment Methods for Enhanced Biogas Production
- Reduction of Sludge Production





RELATION OF THE TECHNOLOGIES in this material

Abbreviation
 MBT Mechanical-Biological Treatment
 AD Anaerobic Digestion





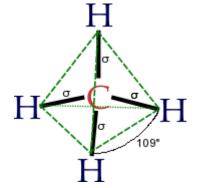
Introduction





DEFINITION

- Organic compounds and organic materials
 - Matters in its various forms that contain $\underline{C \ \& \ H} \ \underline{atoms}$
 - Structure of organic <u>methane</u> molecule: the simplest hydrocarbon compound



- Important constituents of many products including <u>plastics</u>, <u>drugs</u>, <u>petrochemicals</u>, <u>food</u>, <u>explosive material</u>, and <u>paints</u>.
- In this material, we limit meaning of term "organic waste" which is anything that comes from plants, animals or by-products from facilities that is biodegradable such as <u>food waste</u> or <u>sludge</u>.



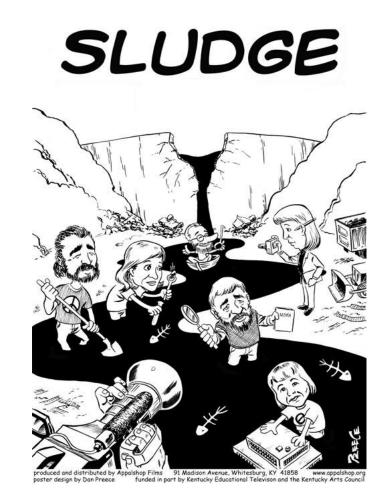






THE PROBLEM

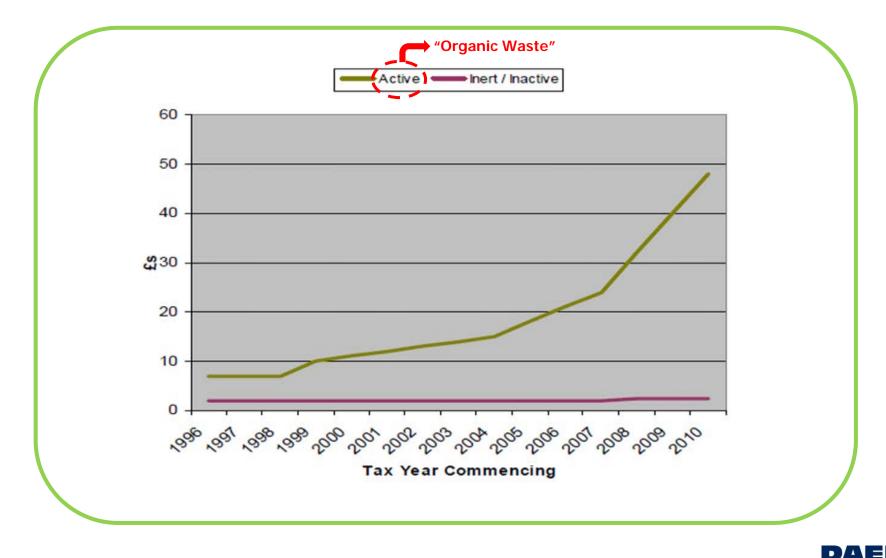
- The main problem is the sheer volume of waste being produced and how we deal with it.
- The disposal of waste activated sludge (WAS) has been one of major issues.
 - Generation rate in Korea : 8,292 ton/day (2009)
 - expected to increase to 10,936 ton/day by 2013
- WAS comprises 19% of the total amount of industrial waste production, which causes a big burden environmentally as well as economically.
- Although ocean dumping had been major disposal method of WAS in Korea, an appropriate alternative is required due to the prevention of ocean dumping.





Introduction

- In England, landfill tax is differentiated between un-treated active waste and treated residual inert
 - In 2011, £56/ton for active waste, £8/ton for inert residual





THE SOLUTION

- More appropriate and sustainable approaches to waste need to be adopted. To be sustainable we need to move the emphasis toward a system that makes use of low-tech, low-energy systems.
- It is focused on waste reduction and recycling.
 - ✓ Waste minimization is an approach that aims to reduce the production of waste through education and the adoption of improved processes.
 - Recycling, separating certain materials within the waste stream and reprocessing them. The recycling of many materials is currently not financially viable.
 - Waste processing is treatment and recovery (use) of materials or energy from waste through thermal, chemical, or biological means.







Recycling as Resource (MBT)



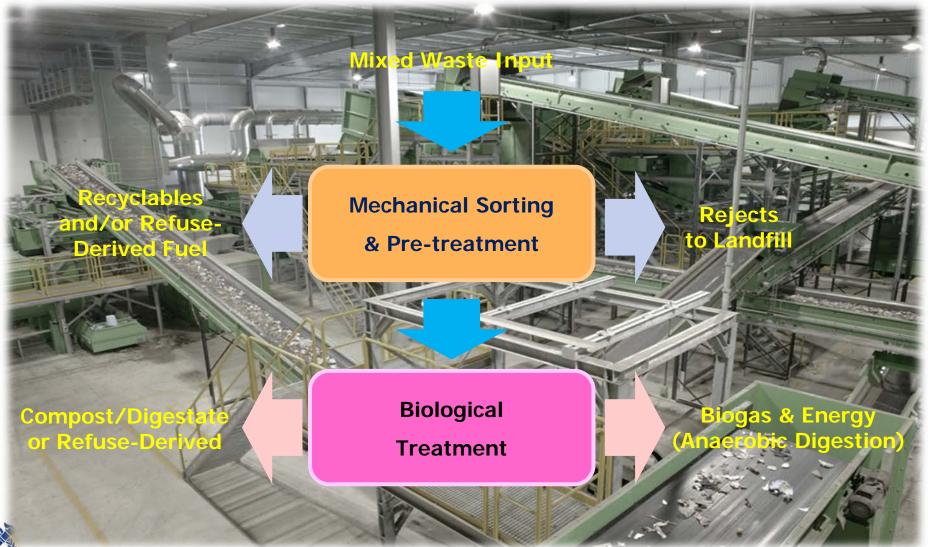
MBT OVERVIEW

- MBT was started to developed in Germany
 - as an alternative of MSW pretreatment prior to landfill
 - in compliance with EU Landfill Directive (1999/31/EC)
 - stepwise reduction of the landfilling of biodegradable waste
- MBT in Germany (2010)
 - 46 MBT plants with a capacity of 6 mil. Mg/y are operating.
 - Approx. 25% of MSW are treated by MBT
 - The realized process concepts are varying strongly and can not compared easily.
- MBT in Europe
 - 330 MBT plants were constructed during 2005 to 2011
 - Within next 5 yrs No. of MBT plants will be increased to 450 plants





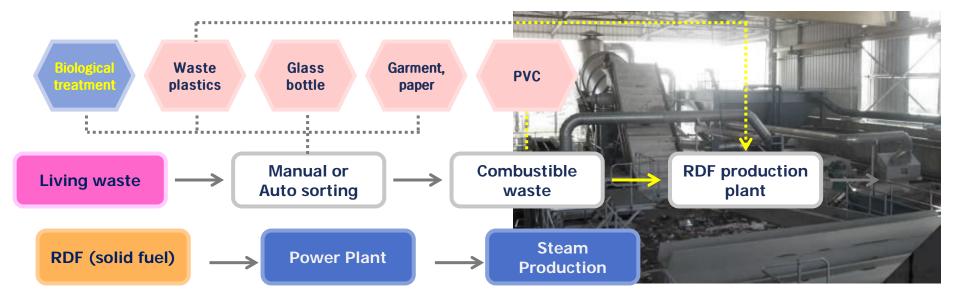
CONCEPTS OF MBT PROCESS



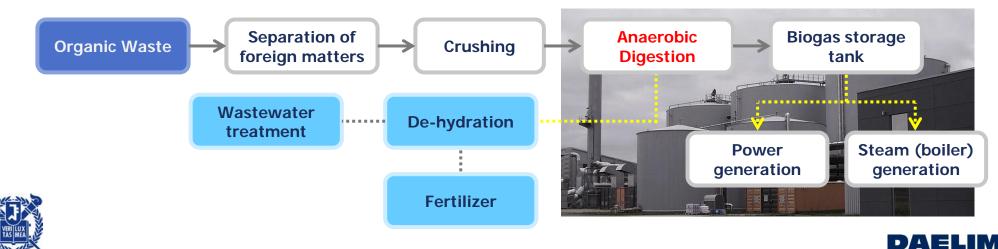




Mechanical Treatment (MT)



Biological Treatment (BT)



GOALS OF MBT

Abbreviation LFG Landfill Gas RDF Refuse Derived Fuel SRF Solid Refuse Fuel

- Volume reduction of landfilling wastes to minimize the necessary landfill capacity and to prolong the operating life of a landfill.
- Reduction of the biodegradable fraction of landfilling wastes so that the uncontrolled LFG generation is minimized as far as possible.
- Reduction of dangerous substances which will be leached in the landfill and can be led to a groundwater contamination
- Recovery of materials and energy Metals, RDF/SRF, Biogas

"The process concepts will be changed by the principal goal of localities"



Waste-to-Biogas



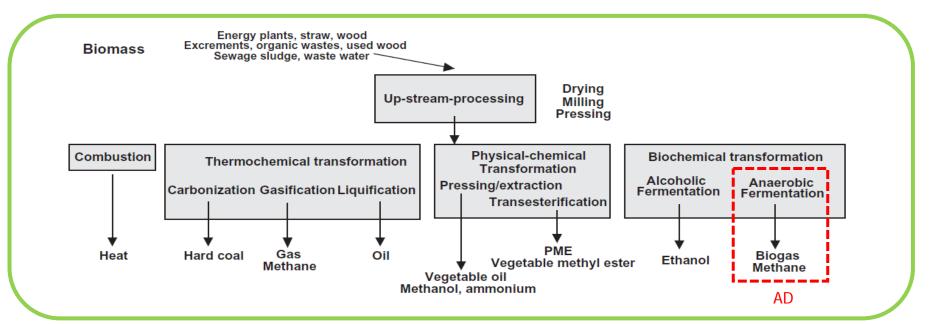


TECHNOLOGIES

Abbreviation

AD Anaerobic Digestion

- There are several processes to transform organic waste into solid, liquid, or gaseous secondary energy sources and reduce its mass.
- Processes : combustion, thermochemical transformation via carbonization, liquefaction or gasification, physico-chemical transformation by compression, extraction, transesterification, and biochemical transformation by fermentation with alcohol or anaerobic digestion





[Applied technologies to transform biomass(organic waste) into secondary energy sources]

THEORY

 The theoretical maximum yield of methane: assuming the elementary composition as a base

 $\begin{array}{l} C_{c}H_{h}O_{o}N_{n}S_{s}\,+\,yH_{2}O\,\,\rightarrow\,\,xCH_{4}\,+\,(c\,-\,x)\,\,CO_{2}\,+\,nNH_{3}\,+\,sH_{2}S\\ \\ \text{where,}\quad x\,=\,0.125\,\,(4c\,+\,h\,-\,2o\,-\,3n\,+\,2s)\\ y\,=\,0.250\,\,(4c\,-\,h\,-\,2o\,+\,3n\,+\,2s)\\ \\ \text{or, simplified:}\\ \\ C_{c}H_{h}O_{o}\,\rightarrow\,(2c\,+\,8h\,-\,4o)\,\,CH_{4} \end{array}$

Environmental requirements

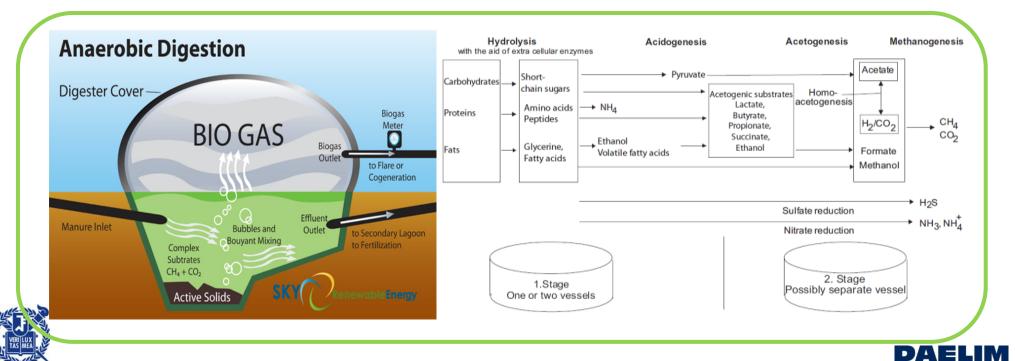
Parameter	Hydrolysis/acidogenesis	Methane formation		
Temperature (°C)	25–35	Mesophilic: 32–42		
		Thermophilic: 50–58		
pH value	5.2-6.3	6.7–7.5		
C:N ratio	10-45	20-30		
DM content (%)	<40	<30%		
Redox potential (mV)	+400 to -300	<-250		
Required C:N:P:S ratio	500:15:5:3	600:15:5:3		
Trace elements	No special requirements	Essential: Ni, Co, Mo, S		





ANAEROBIC DIGESTION (AD)

- Methane fermentation : hydrolysis, acidogenesis, acetogenesis, and methanation
- The individual phases : partly stand in syntrophic interrelation and place different requirements on the environment
- Two stages: the first 2 phases and the last 2 phases are linked closely with each other



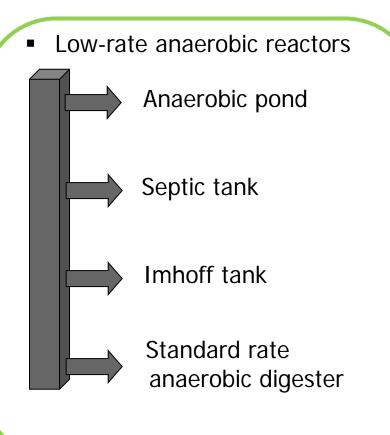
Abbreviation

AD Anaerobic Digestion

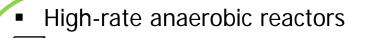
Abbreviation

AD Anaerobic Digestion

AD REACTORS



Slurry type bioreactor, temperature, mixing, SRT or other environmental conditions are not regulated. <u>Loading : 1-2 kg COD/m³-day</u>





Anaerobic contact process



- Anaerobic filter (AF)
- Upflow anaerobic sludge blanket (UASB)



Fluidized bed reactor



Hybrid reactor: UASB/AF

Anaerobic sequencing batch reactor (ASBR)

Able to retain very high concentration of active biomass in the reactor. Thus extremely high SRT could be maintained irrespective of HRT. Loading: 5-20 kg COD/m³-d





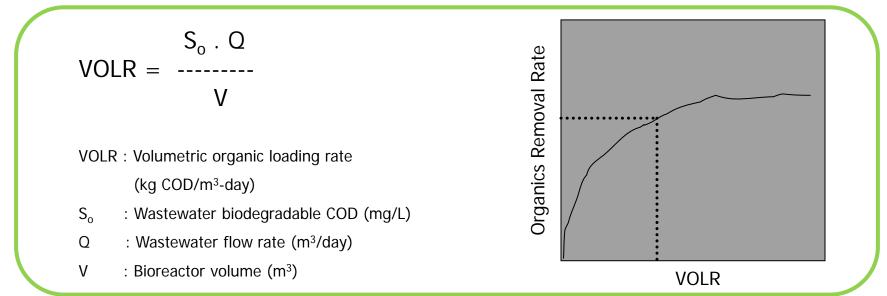
AD DESIGN

Abbreviation

AD Anaerobic Digestion

SRT Solids Retention Time

Design based on volumetric organic loading rate (VOLR)



- S_o and Q can be measured easily and are known upfront VOLR can be selected!
- How do we select VOLR?
 - Conducting a pilot scale studies
 - Find out removal efficiency at different VOLRs
 - Select VOLR based on desired efficiency
 - Other important design factor : SRT

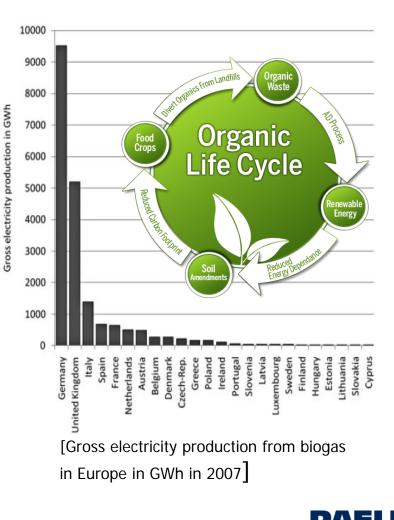




BIOGAS COMPOSITION PRODUCED BY AD

 The gas components: specified to the plant and substrate and should be checked regularly on a long-term basis

Component	Content	Effect
CO2	25–50% by vol.	Lowers the calorific value
		Increases the methane number and the anti-knock properties of engines
		Causes corrosion (low concentrated carbon acid). if the gas is wet
		Damages alkali fuel cells
H ₂ S	0–0.5% by vol.	Corrosive effect in equipment and piping systems (stress corrosion); many manufacturers of engines therefore set an upper limit of 0.05 by vol.%
		SO_2 emissions after burners or H_2S emissions with imperfect combustion-upper limit 0.1 by vol.%
		Spoils catalysts
NH ₃	0-0.05% by vol.	NO _x emissions after burners damage fuel cells
		Increases the anti-knock properties of engines
Water vapor	1-5% by vol.	Causes corrosion of equipment and piping systems
		Condensates damage instruments and plants
		Risk of freezing of piping systems and nozzles
Dust	>5µm	Blocks nozzles and fuel cells
N ₂	0-5% by vol.	Lowers the calorific value
		Increases the anti-knock properties of engines
Siloxanes	0-50 mg Nm-3	Act like an abrasive and damages engines





Pretreatment Methods

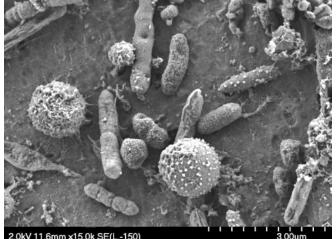
for Enhanced Biogas Production

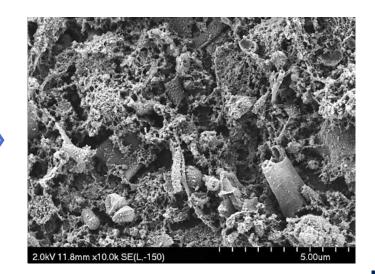




P. L. McCarty at Stanford University

- Looking for a way to improve anaerobic digestion (in the late 1970s)
 - Primary Sludge (PS) : easily digested
 - Waste Activated Sludge (WAS) : only 1/3 can be digested
- Waste Activated Sludge : mostly consisted of microbial cells
 - Protection by cell walls, most cellular material is unavailable to anaerobic microbes.
- Pretreatment Methods : focused on 'Cell Lysis"
 - Volatile solids reduction (VSR) can be increased 2~3 times.
 - Biogas production can be doubled.









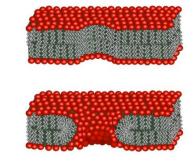
Methods of Cell Lysis

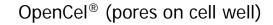
- High temperature (Hydrolysis)
 - Cambi[™] by Cambi AS, Norway : 165°C at 6 bar for 20~30 min
 - Exelys[™] by Veolia, France : 165°C at 9 bar for 30 min
- High Pressure
 - MicroSludge[®] by Paradigm Environment, Canada : homogenizer (60 bar)
- Physical Force
 - OpenCel® by OpenCel LLC, Atlanta : focused pulse
 - Ultrasound homogenizer
- Others : high or low pH, chemical oxidation, etc.





MicroSludge[®] Homogenizer







Cambi™



Carbon source production from sludge using Cell Lysis

- Methanol alternatives as C source for denitrification
 - Reduced sludge production & reduced
 - Utilization as C source has 10 times its value making biogas
 - \Rightarrow Reduced greenhouse gas emission
 - ⇒ Reduced O&M cost

Major Issues on Pretreatment of Sludge

- Cost can be higher than the benefit it can provide !!
 - Benefit (increased biogas production) should be greater than CAPEX + OPEX.
 - There are various unforeseeable and hidden costs.





Reduction

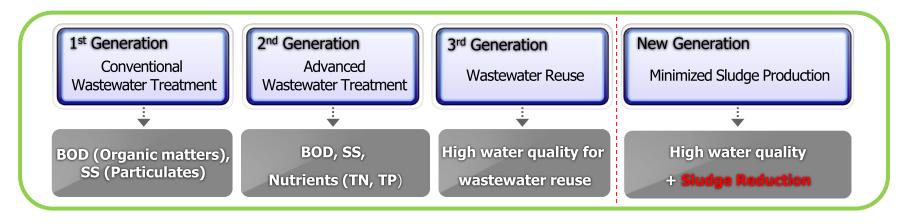
of Sludge Production



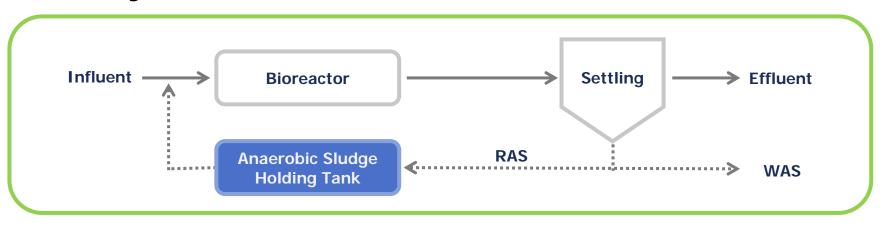


Paradigm Shift

 New technology for source reduction during wastewater treatment by manipulating microbial growth kinetics



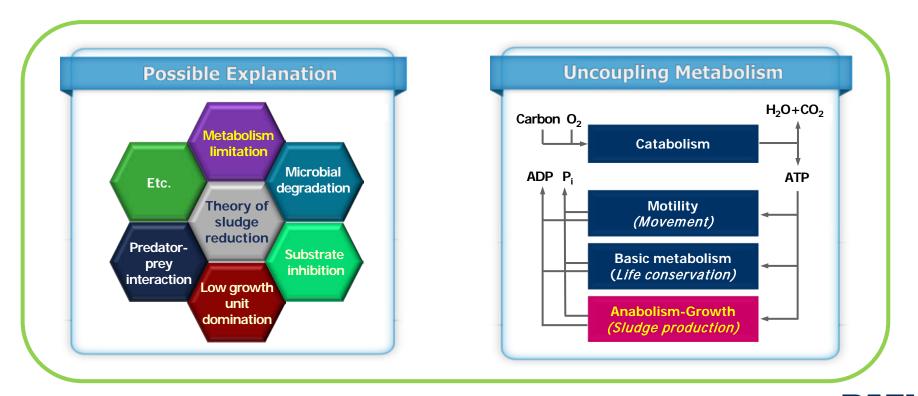
Process Diagram





PRINCIPLE OF SLUDGE REDUCTION

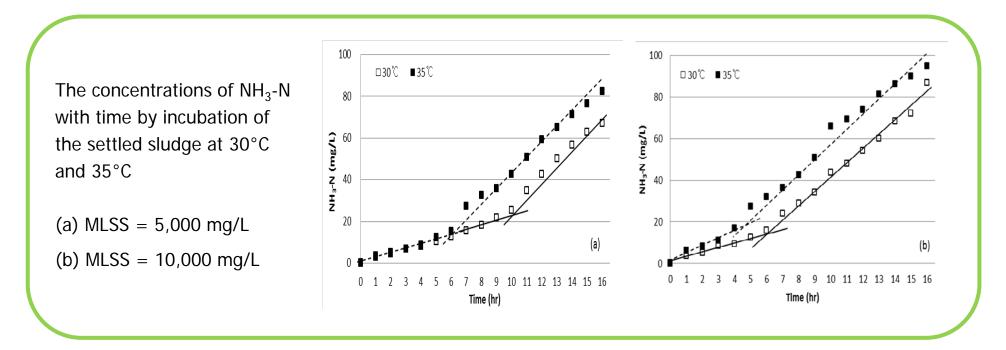
- The abrupt change in growth environments
 - Uncoupling metabolism occurs
 - Yield reduction while maintaining substrate uptake rate
 - The reduction of ATP production in microbial cells



Possible Explanation



EXPERIMENTAL RESULTS_ Optimal HRT



- Increase in the release rate = the increased degradation of microbial cells
- Optimal point : the start of full-fledged endogenous phase at each experimental

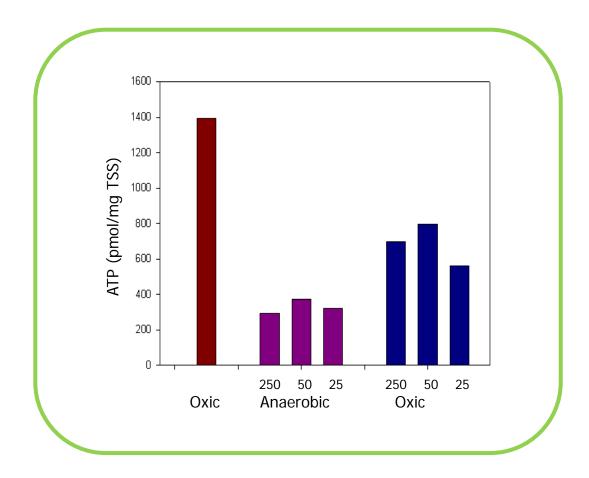
as the optimal hydraulic retention time for SHT.





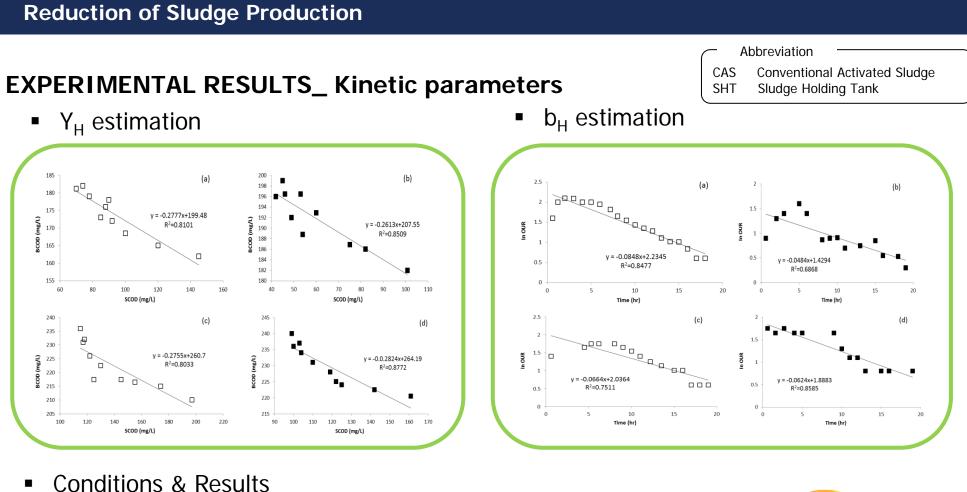
EXPERIMENTAL RESULTS_ Metabolism

Oxic-Anaerobic-Oxic (The change in growth environments)







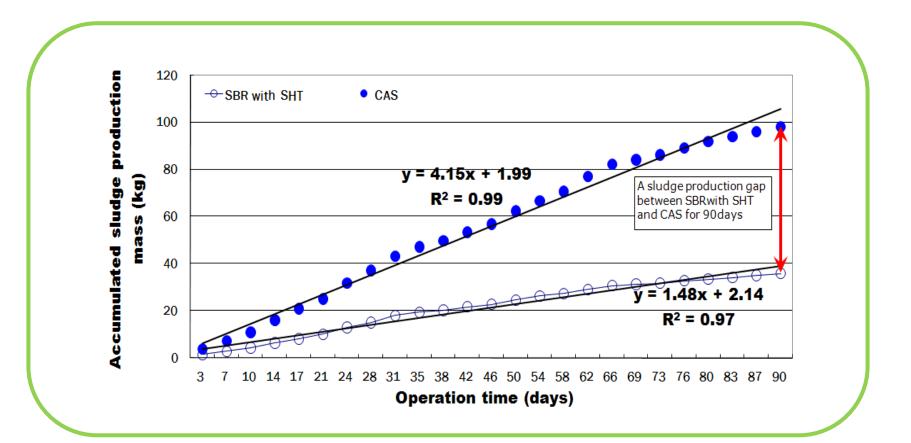


	Experii	(a)	(b)	(c)	(d)	
-	SHT operation	MLSS (mg/L)	5,000		10,000	
	condition	Temp. (°C)	30	35	30	35
_	Y _H		0.27	0.26	0.28	0.27
	b _H (day⁻¹)		0.11	0.06	0.09	0.08
TLUX	×					

The average values of kinetic parameters - Y_{H.avg.} = 0.27 (CAS ⇔ 0.67) $- b_{H.avg.} = 0.085 \text{ day}^{-1}$

EFFICIENCY

 WAS production rate can be reduced by 63.5% compared to conventional process with no deterioration of treatment efficiency.



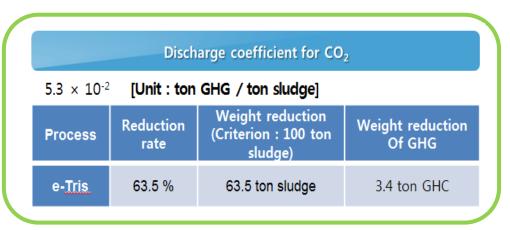




EFFECT OF TECHNOLOGY

- Excess Sludge Production : Reduction of 63.5% compared to CAS
- Operating costs of STP : Reduction of 30% operating costs
- GHG & Energy
 - : Eco-friendly to reduce GHG and energy

that occur in sludge treatment.





Abbreviation
 CAS Conventional Activated Sludge
 STP Sewage Treatment Plant
 GHG Green House Gas
 GT Green Technology

Q & **A**





Energy Management in

Wastewater Treatment Facility

Nov. 25. 2013

Dr. Hee-Jun Kim



R&D Center, Chief/Director

JIU Corporation









Current Status

Current Status of Sewage Treatment Facilities in Korea

- Number of sewage treatment facilities : 528 (in 2012) (Facilities with capacity lower than 500 m³/day are not included)
- Total amounts of sewage treated in facilities : 25 million m³/day
- 65 facilities have anaerobic digester, but only 57 facilities operate digesters actively
- Anaerobic digestion (AD) efficiency is quite lower than those in other countries

AD Efficiency and Sludge Reduction Data in Some Facilities in Korea

Facility	Anaerobic Digester Volume (m ³)	Digestion Efficiency (%)	Sludge Reduction (%)	
Α	82,776	35.3	27.3	
В	17,500	37.3	14.4	
С	25,120	25.1	35.9	
D	7,234	47.3	68.0	
Е	7,551	23.8	44.2	
F	12,565	42.3	29.1	
G	2,154	50.3	30.7	

Note) Anaerobic Digestion Efficiency =
$$\left(1 - \frac{FS_{in} \times VS_{out}}{FS_{out} \times VS_{in}}\right) \times 100$$

Current Status

Energy Consumption in Sewage Treatment Facilities

- Annual energy consumption in sewage treatment facilities : 395,121 TOE (in 2007)
- Among them, electric use occupies 98.6%
- Electric use per flow : 0.29 kWh/m³
- Electric use per BOD removal : 2.353 (kWh/kg BOD)
- Faction of electricity used in sewage treatment facilities reaches 0.5% of national electricity usage.
- Energy self-sufficiency of sewage treatment facilities is only 0.8%

Note) TOE : Tonnage of Oil Equivalent, the amount of energy released by burning one tonne of crude oil ≈ approximately 42 GJ (10⁷ Kcal)

Note) Energy self-sufficiency : (Renewable Energy production + Energy saving)/Annual electric use

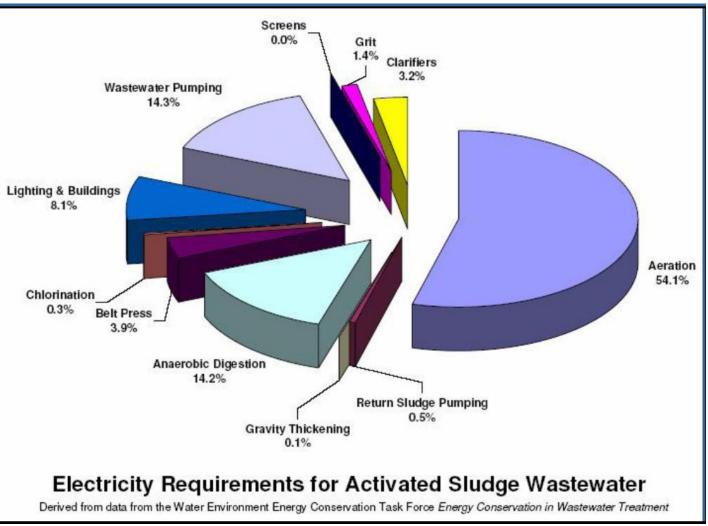
Energy Consumption in sewage treatment operations (2008)

Energy related operation	Aeration	Sewage pumping	Dewatering	Sludge pumping	Discharge pumping	Anaerobic Digestion (mixing)	Thickening, screen, etc.
Fraction (%)	40.1	21.3	6.4	3.6	2.3	1.4	23.9



Energy Used in Wastewater Treatment (US)

Energy Consumption in sewage treatment with AS Process





Electricity Use by Process (in kWh/MG)

	1 MGD (≈ 3,786 m³/d)	10 MGD (≈ 37,852 m³/d)	100 MGD (≈ 378,520 m³/d)
Wastewater Pumping	171	140	118
Screens	2	1	1
Aerated Grit Removal	49	13	12
Primary Clarifiers	15	16	16
Aeration	532	532	532
Biological Nitrification	346	345	340
Return Sludge Pumping	54	51	38
Secondary Clarifiers	15	16	15
Chemical Addition	80	55	42
Filter Feed Pumping	143	82	67
Filtration	137	39	34
Thickening	6	203	131
Digestion	1,200	170	155
Dewatering	0	46	25
Chlorination	1	3	3
Lighting and Buildings	200	80	30
Total Process	2,951	1,792	1,559



(from WEF M.O.P No. 32, "Energy Conservation in Water and Wastewater Facilities")

Electricity Used in WWTF

Annual Electric Use in Sewage Treatment Facilities

Year	2002	2003	2004	2005	2006	2007	2008	2009
Annual Expenses (million won)	365,084	424,366	468,660	511,082	584,635	649,582	696,934	782,225
Annual Electricity Cost (million won)	77,330	85,914	95,404	102,264	112,786	125,090	137,611	156,139
Electricity/Total (%)	21.2	20.2	20.4	20.0	19.3	19.3	19.7	20.0
Electricity Cost Growth (%)	-	11.1	11.0	7.2	10.3	10.9	10.0	13.5

Electric Use in Sewage Treatment Facility with Different Capacity

Capacity (m ³ /d)	Number of Facilities	Total Electric Cost (thousand won /year)	Average Electric Cost (thousand won /year)	Electricity Cost per Sewage Flow (won/m ³)	Electricity Consumption (kwh)	Electric Use per Flow (kwh/m ³)	
500 \sim 1,000	53	965,566	18,218	114.2	10,683,210	1.26	
1,000 ~ 5,000	102	4,391,042	44,354	72.7	71,014,710	1.18	
5,000 ~ 10,000	47	4,679,909	99,573	56.5	59,632,253	0.72	
10,000 ~ 50,000	87	20,331,289	239,191	37.8	331,990,571	0.62	
50,000 ~ 100,000	23	10,044,406	436,713	25.7	325,614,887	0.83	
100,000 ~ 500,000	33	43,710,493	1,324,560	23.2	863,180,857	0.46	
500,000 ~	14	57,850,387	4,132,171	17.5	847,361,345	0.26	

JU CORPORATION



Note) Average annual electric consumption of 1 household(4 persons) in Seoul city is about 4,800 kwh.

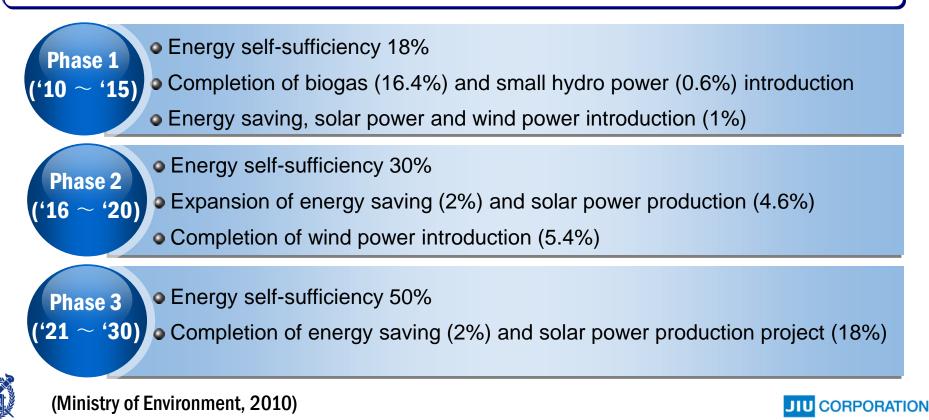
Basic Plan for Energy Self-Sufficiency

Basic Plan for Energy Self-Sufficiency in Sewage Treatment Facilities

Main Goal

Energy Self-Sufficiency in Sewage Treatment Facilities in year 2030

: 50% in 343 facilities



Basic Plan for Energy Self-Sufficiency

Some Strategies for Upgrade Energy Self-Sufficiency

Promoting Energy Saving

- Energy efficient operation
- Replacement to energy efficient equipments

Utilization of Unused Energy

- Improvement of biogas production and utilization
- Expansion of beneficial usage of small hydro power and heat energy in wastewater

Production of Natural Energy

Expansion of solar power and wind power

Basis Setting for Energy Self-Sufficiency

- Planning energy self-sufficiency scheme for every treatment facilities
- R&D, education, campaign for low-carbon green growth





Upgrading Energy Self-Sufficiency Example

Energy Self-Sufficiency Planning in Ansan WWTF, Korea



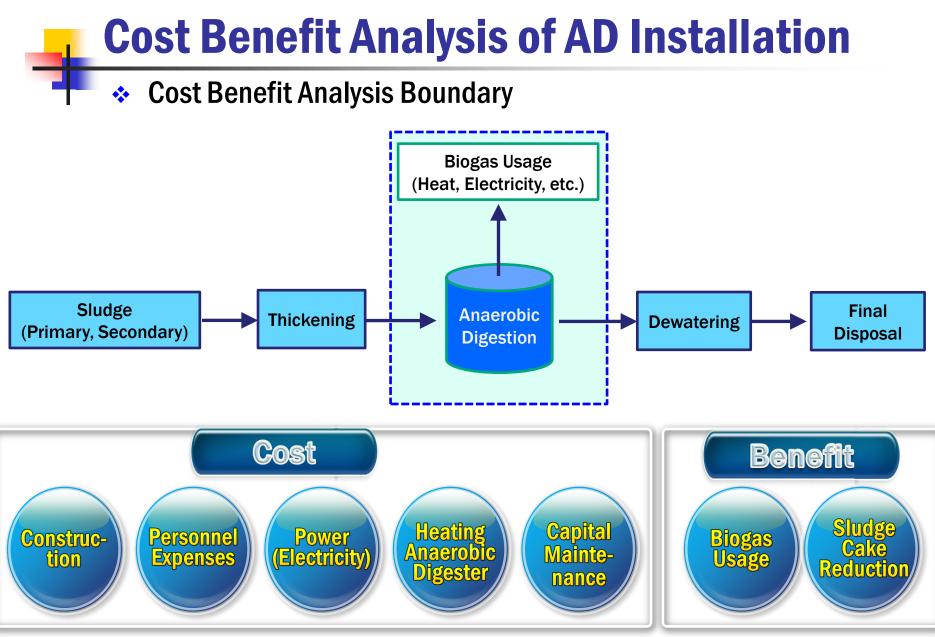




Is Anaerobic Digestion Always Economical in WWTF?









Cost Benefit Analysis Criteia

Benefit/Cost Ratio (B/C ratio)

$$B/C = \sum_{t=0}^{n} \frac{B_t}{(1+r)^t} / \sum_{t=0}^{n} \frac{C_t}{(1+r)^t}$$

- t:time
- n : period, 20 years for sludge treatment facility
- r : discount rate (5.5%)
- Present value of project benefits / present value of project costs
- If $B/C \ge 1.0$, the project is judged to be worthwhile in economic terms

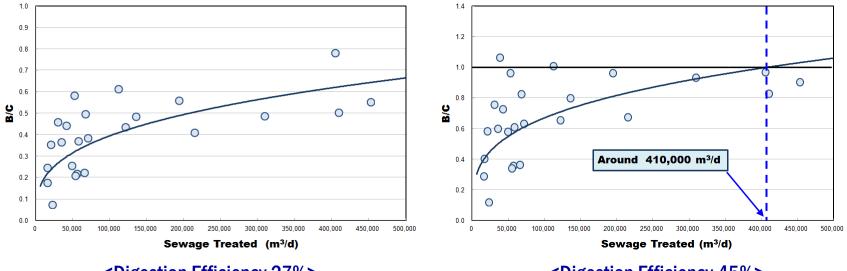
Calculation of B/C ratio

			C	ost				Benefit	
Period	Construction	Personnel Expenses	Electricity	Heating Energy	Maintenance	Sum	Biogas Usage	Sludge Cake Reduction	Sum
	(million won)	(thousand won/ year)	(thousand won/ year)	(thousand won/ year)	(thousand won/ year)	(million won)	(thousand won/ year)	(thousand won/ year)	(million won)
1	17,944	52,718	73,075	5 272,583	50,402		625,856	219,612	
2		49,970	69,265	5 258,373	47,774		593,228	208,163	
3		47,365	65,654	244,903	45,283		562,302	197,311	
•		•	•	•	•		•	•	
•		•	•	•	•			•	
•		•	•	•	•			•	
19		20,110	27,875	5 103,981	19,227		238,743	83,774	
20		19,062	26,422	98,560	18,224		226,296	79,407	
	17,944	664,654	921,300	3,436,638	635,447	23,602	7,890,575	2,768,791	10,6

• B/C ratio = 23,602/5,676 = 0.45

Cost Benefit Analysis Results

Effect of Digestion Efficiency



<Digestion Efficiency 27%>

<Digestion Efficiency 45%>

Note) 27% is average digestion efficiencies of 24 sewage treatment facilities

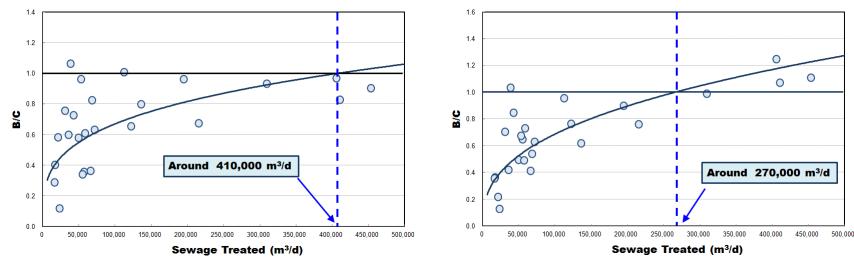
- At 27% of digestion efficiency, there was no facility with B/C ratio over 1.0.
- At 45% of digestion efficiency, B/C ratio exceeds 1.0 at wastewater treatment capacity over 410,000 m³/d.
- Increase in digestion efficiency raise B/C ratio due to the biogas production increase and reduction in sludge cake production.





Cost Benefit Analysis Results •••

Effect of Final Disposal Cost



<Using individual final disposal cost>

<Using average final disposal cost>

Note 1) Anaerobic digestion efficiency was assumed to be 45% at all treatment facilities

Note 2) Final disposal cost

Carbonization : 116,000 won/cake ton Drying Average

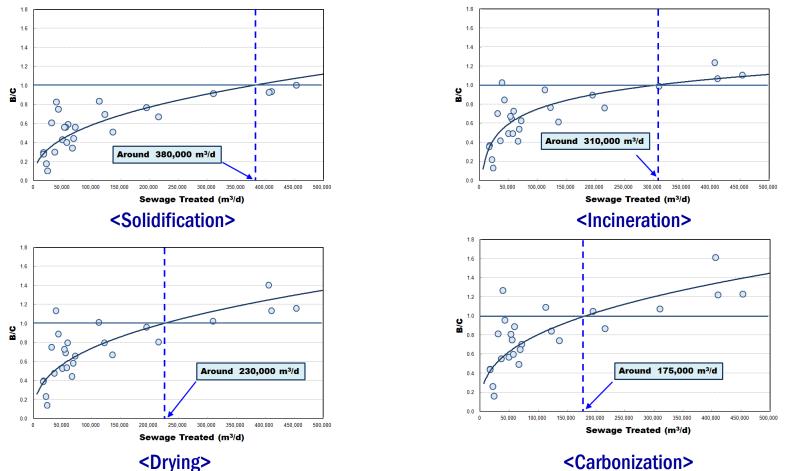
- : 100.000 won/cake ton
- 91,000 won/cake ton

- Incineration : 87,000 won/cake ton Solidification : 63,000 won/cake ton
- With average final disposal cost, the treatment capacity with B/C ratio 1.0 reduces to 270,000 m^3/d .
- Final disposal cost largely affects on B/C.



Cost Benefit Analysis Results

Effect of Final Disposal Methods



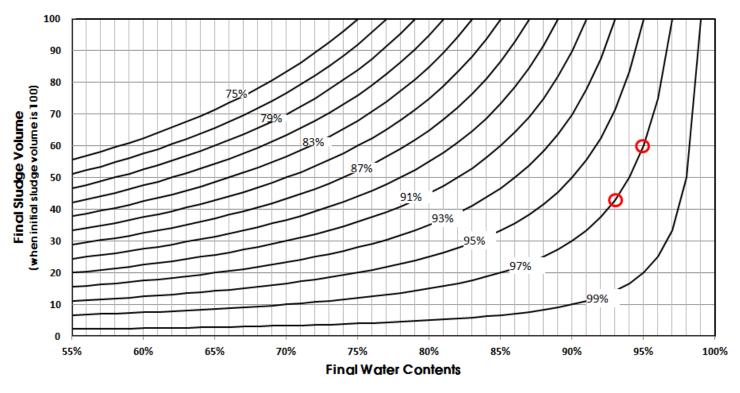


• AD installation is more economical at the facility using carbonization as the final sludge disposal option.

Cost Benefit Analysis of AD Installation

Cost Benefit Analysis Results

Effect of Sludge Thickening before AD



• Average water contents (W.C) of influent sludge is about 97%.

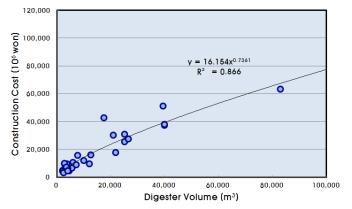
If W.C of sludge is reduced to 95% or 93%, the volume of sludge will be 60% or 42% of initial sludge volume, respectively. → We can build smaller anaerobic digester



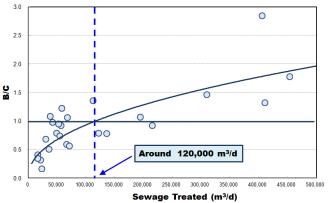
Cost Benefit Analysis of AD Installation

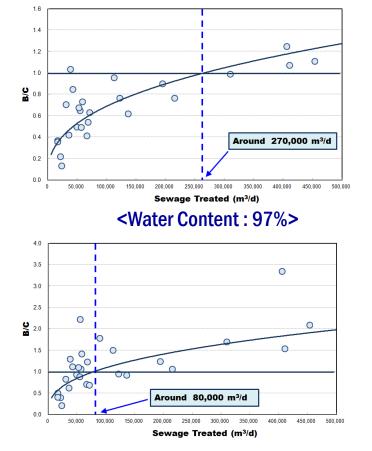
Cost Benefit Analysis Results

Effect of Sludge Thickening before AD



<AD construction cost vs AD volume>





<Water Content : 95%>

<Water Content : 93%>

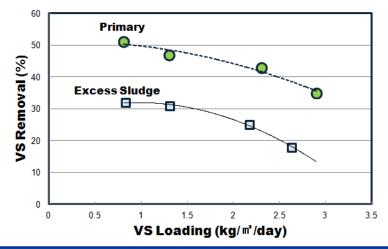


Higher solids contents leads to smaller capacity reaching B/C = 1.0 due to the lower construction cost.

Cost Benefit Analysis of AD Installation

Cost Benefit Analysis Results

Effect of Sludge Thickening on VS Removal



• Higher VS loading can deteriorate anaerobic digestion efficiency.

Some Findings from Cost Benefit Analysis Results

- There's specific anaerobic digestion capacity that can achieve economical benefit under given operational condition.
- If you want to gain economical benefit with smaller anaerobic digester (i.e. lower initial investment), mainly consider the measures to increase solids contents in sludge and anaerobic digestion efficiency.



+

Energy Management in

Wastewater Treatment Facility

Dec. 02. 2013

Dr. Hee-Jun Kim



R&D Center, Chief/Director

JIU Corporation





How to Improve Energy Self-Sufficiency in WWTF?







Energy self-sufficiency =

(Renewable Energy production + Energy saving)

Annual electric use

Basic Strategies to Enhance Energy Self-Sufficiency

- Improve renewable energy production
 - : mainly, enhancing biogas production in AD

(co-digestion of food waste or night soil can be considered)

- : introducing solar power, small hydropower, wind power, etc.
- Focus on biggest energy consumers at WWTF (aeration, pumping, etc)
- Tailor operations to meet seasonal and diurnal changes
- Consider equipment life and energy usage to guide repair and replacement





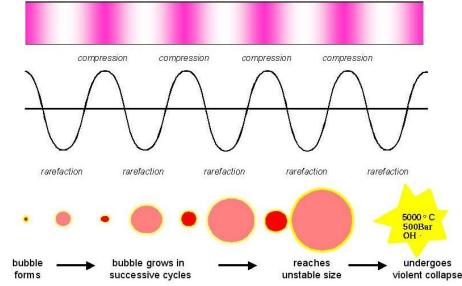
Pretreatment Before Anaerobic Digestion

- What Can We Expect from Pretreatment Before Anaerobic Digestion
- Faster hydrolysis of particulate
- Decrease of retention time in anaerobic digestion
- Enhancement of biogas production
- Improvement of dewatering characteristic of sludge
- Types of Pretreatment Methods
 - Mechanical : homogenizer, stirred ball mills, <u>cavitation</u>, etc.
 - Chemical : alkaline/acid hydrolysis, ozonation
 - Biological : enzyme addition, thermophilic bacteria injection, etc.
 - Thermal : thermal hydrolysis & Freeze-Thawing
 - Combined : thermal-chemical, ultrasonic-chemical, etc.
 - Others : electron beam, microwave, **focused pulsed electricity** etc.



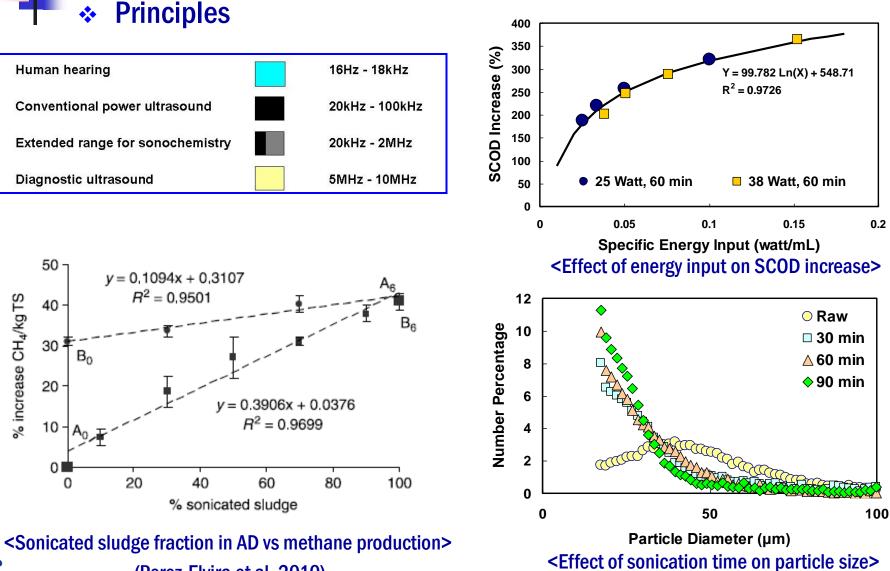
Principles

- At the lower end of ultrasound (20kHz to 10MHz), the compaction and rarefraction waves generated by ultrasound lead to the formation of cavitation bubbles in the fluid, which implode creating high mechanical shear forces.
- The implosions create localized hot spots (temperatures up to 5,000°C) and pressures up to 500 bar (7,250 psig) and jet-stream (400 km/hr)
 - \rightarrow Shear forces can be used for disintegrating solids in the fluid





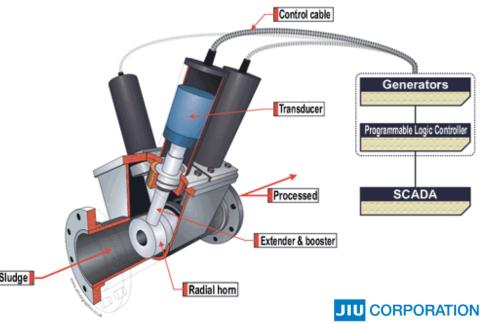
(Perez-Elvira et al, 2010)



- Commercial Process Sonix
- Description
 - English company, Sonico
 - Usually 3 to 5 sonotrodes are installed in 1 unit (6kW/unit)
 - VS reduction and gas production increase by up to $30 \sim 50\%$









- Commercial Process Ultrawaves
- Description
 - German company, Ultrawaves
 - Usually 5 sonotrodes are installed in 1 unit (2kW/unit)
 - WAS treatment fraction is 30 to 50%
 - Ultrasound energy : 0.04 ~ 0.05 kWh/kg DS
 - VS reduction and gas production increase
 by up to 20 ~ 50%
 - Applied in one Korean WWTF (Ulsan Yongyeon)







Commercial Process – Reverse Flow Disintegration Unit

Description

- Austrian company, VTA Technologie GmbH
- Sewage sludge continuously flows top down through the disintegration reactor
- Mixing with agitator installed in reactor ($30 \sim 120$ rpm)
- Disintegration degree can be controlled with the residence time of the sludge in the reactor, the flow rate, the rotation speed of the agitator and the energy input

of the integrated ultrasonic elements

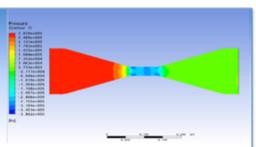
• VS reduction and gas production increase by up to $20 \sim 50\%$

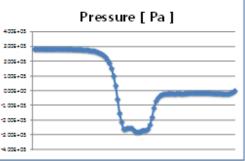




Hydrodynamic Cavitation Pretreatment

- Commercial Process Crown Disintegration
- Description
 - German company, Biogest
 - Cavitation bubbles produced in the constriction region (venturi throat) due to the pressure drop below vapour pressure and rapidly collapses (implosion) in the expansion region → producing shock waves
 - Homogenizer, progressive cavity pump, disintegrator and control panel
 - Gas production increase by up to 30%

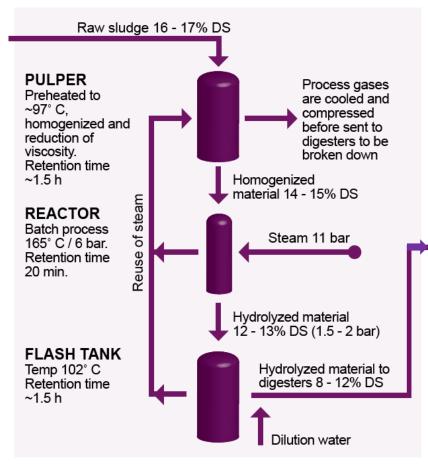








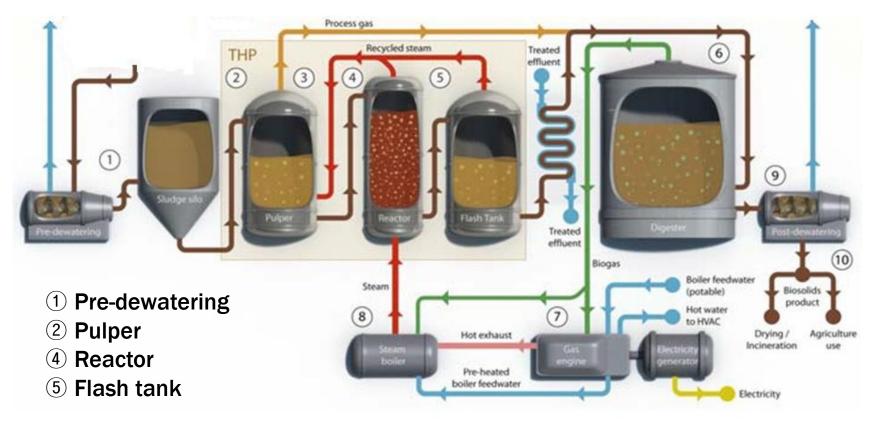
- Commercial Process Cambi Process
- Description
 - Norwegian company, Cambi
 - First full scale demonstration plant
 - : HIAS WWTP in Hamar, Norway
 - Using high-pressure steam : 6 bar, 165°C
 - Process configuration
 - : Pulper Reactor Flash Tank
 - Batch process
 - Need pre-dewatering process
 - : TS contents $16 \sim 17\%$
 - Increase gas production up to 30 ~ 100%





Commercial Process - Cambi Process

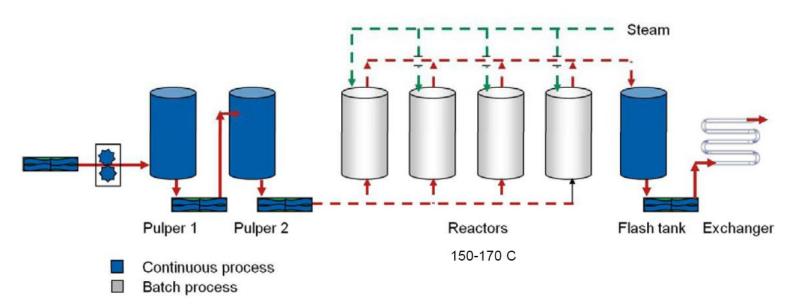
Flow Diagram





Commercial Process - Cambi Process

Operation



	15 min	15 min	30 min		15 min	15 min
Reactor 1	Fill	Steam In	React		Steam out	Empty
Reactor 2	Empty	Fill	Steam In	React		Steam out
Reactor 3	Steam out	Empty	Fill	Steam In	React	





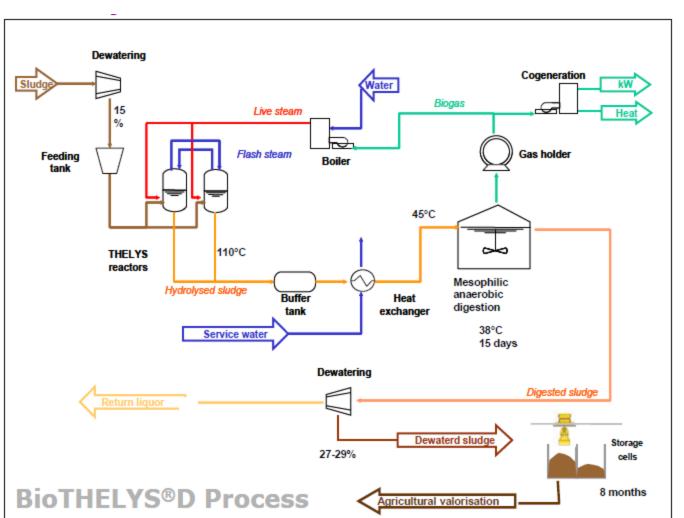
- Commercial Process Biothelys Process
- Description
 - French company, Veolia
 - Thermal hydrolysis (Thelys process) + anaerobic digestion
 - Using high-pressure steam (7 \sim 9 bar, 150 \sim 170 $^{\circ}$ C)
 - Batch process
 - Retention time 30 ~ 60 min
 - Need pre-dewatering process
 - : TS contents around 15%
 - Reduce quantity of sludge production by up to 80%
 - Tergnier (1,600ton DS/y) WWTF, France





Commercial Process - Biothelys Process

🥥 Flow Diagram





Commercial Process - Biothelys Process

Operation 20 min 25 min 30 min 20 min 25 min 5 min 25 min REACTOR 1 Flash dea Live Flash steam Sludge Sludge feeding inlet steam Reaction outlet Emptying feeding Flash steam Fresh steam Pause REACTOR 2 Sludge Flash steam Flash steam Flash steam outlet feeding Inlet Fresh steam out/et Emptying React/on Emptying Reaction Pause Pause Raw Raw Flash Raw Live Live Flash Live sludge sludge sludge steam steam steam steam steam -В В В В В А А A Α В А А А В А В 110°C 160°C 75°C 15°C 160°C 15°C 160°C 15°C 160°C 75°C 110°C ٠ 160°C 160°C 110°C 75°C 75°C 110°C 160°C Hydrolyse sludge Hydrolysed sľudge

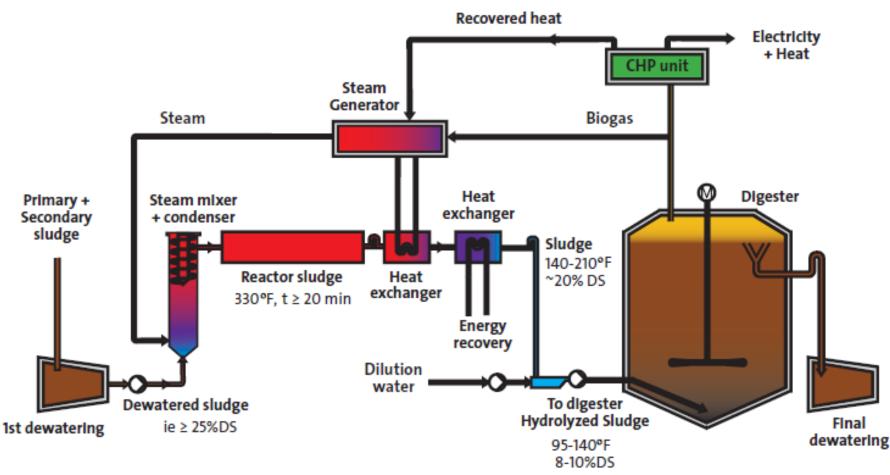
- Commercial Process Exelys Process
- Description
 - French company, Veolia
 - Using high-pressure steam (9 bar, 165°C)
 - Tube type reactor : plug flow
 - Continuous process
 - Need pre-dewatering process
 - : TS contents over 22%
 - Increase gas production up to 20 ~ 40%





Commercial Process - Exelys Process

🥥 Flow Diagram





Electric Pulse Pretreatment

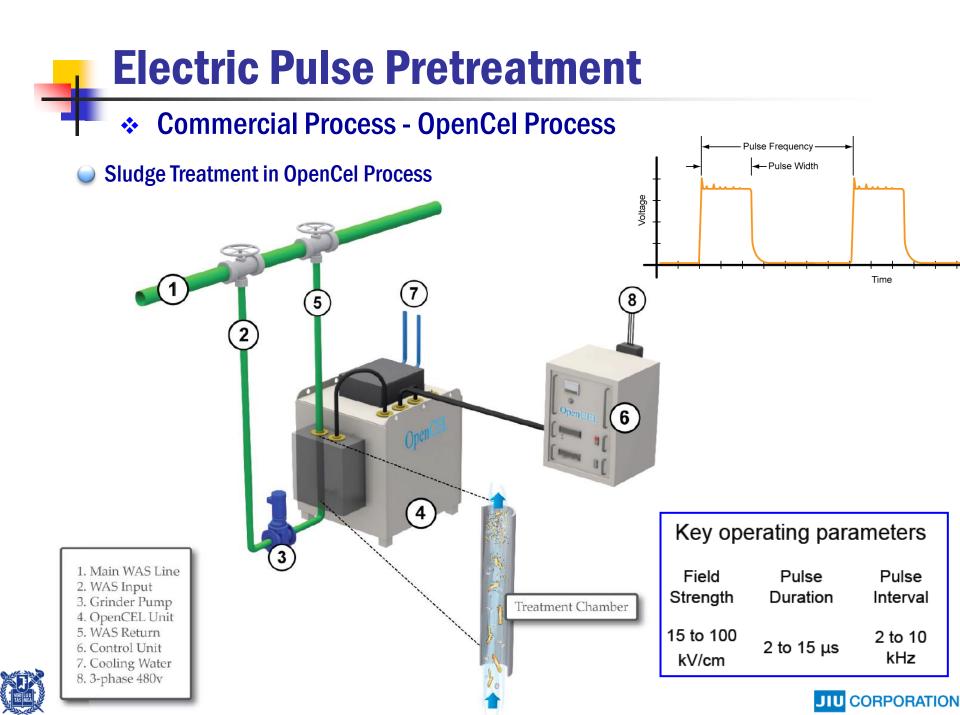
- Commercial Process OpenCel Process
- Description
 - American company, OpenCel
 - Using pulsed electric field (PEF) technology to the sludge to achieve a process known as irreversible "electroporation"



- Electroporation : the pores in the microorganisms cell wall open in the pulsing electric field → when higher electrical power is applied the pores, they do not close in a reversible fashion. Instead, the cell membranes become permeable to the influx of small molecules from carrier medium, leading to swelling and rupture of the cell
- Pre-thickening is required : TS 6%
- Increase gas production up to 30 ~ 50%







Combined Pretreatment

- Commercial Process Microsludge Process
- Description
 - Canadian company, Paradigm
 - Chemical + mechanical process
 - Alkaline treatment with NaOH : pH 8~9
 - Screening(800µm) before mechanical treatment
 - Pressure drop : 12,000 psi (pprox 816 atm) to 50 psi
 - VSR in AD after pretreatment : 78% (with HRT 13day)
 - Canada Chilliwack WWTF

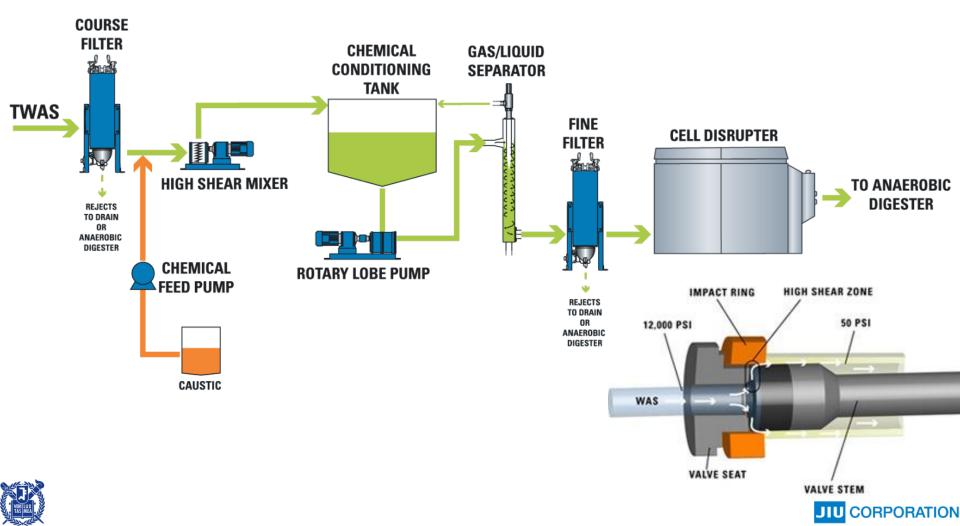




Combined Pretreatment

Commercial Process – Microsludge Process

Flow Diagram



Combined Pretreatment

Commercial Process – Microsludge Process

Example O&M Costs of Microsludge for WAS Pretreatment before AD

Item	Unit Cost	Unit	Cost/dry ton TWAS (4% TS) ¹	$\frac{\text{Cost/dry ton}}{\text{TWAS} (6\% \text{ TS})^2}$
Electricity	\$0.06	/kWh	(\$46)	(\$30)
Caustic	\$0.60	/L	(\$13)	(\$9)
Maintenance			(\$15)	(\$10)
Total O&M			(\$73)	(\$50)
Natural gas offset	\$8.50	/GJ	\$78	\$78
Heat from homogenization	\$8.50	/GJ	\$18	\$12
Biosolids Management Savings	\$120	dt	\$49	\$49
Total Savings			\$145	\$139
Net O&M Value			\$72	\$90

¹ Based on data from Chilliwack WWTP trial. Assumes CH₄ at 64%, biosolids management at \$30/wt, 25% TS cake. Does not include cost of handling screenings. Performance assumptions to be adjusted based on final data from Los Angeles trial. ² Estimated





Pretreatment Comparison

Comparison between Some Pretreatment Methods

Methods	Heat treatment	Ultrasonic	Ozonization	Cavitation
Outlines	Use of enhancing dehydra- tion through releasing water in a cell by heating at high temperature and pressor condition	Use of ultrasonic cavita- tion caused by ultrasonic generator	Adjustment of direct reaction by ozone and indirect reaction with OH-radical produced through a degradation by ozone	Usage of cavitation pro- duced by a turbulent flow of liquid
Sludge reduction	60~70%	30~50%	30~40%	20~80%
Advantages	 Excellent dehydration Extinction of bacteria and filamentous fungus High rate of sludge removal 	 Simple facility Small facility area Low cost for construction of reactor 	 Extinction of bacteria and filamentous fungus Improvement of sludge precipitation and dehydration No chemical reagents 	 Possibility of sludge homogenization
Dis- advantages	 High cost of operation and maintenance 	 No intermittent operation but only continuous operation Need of a double cover for a shock noise 	 Low economical efficiency for excess ozone treatment A fluctuation of treatment effi- ciency by pH 	 Low persistence Large power consumption



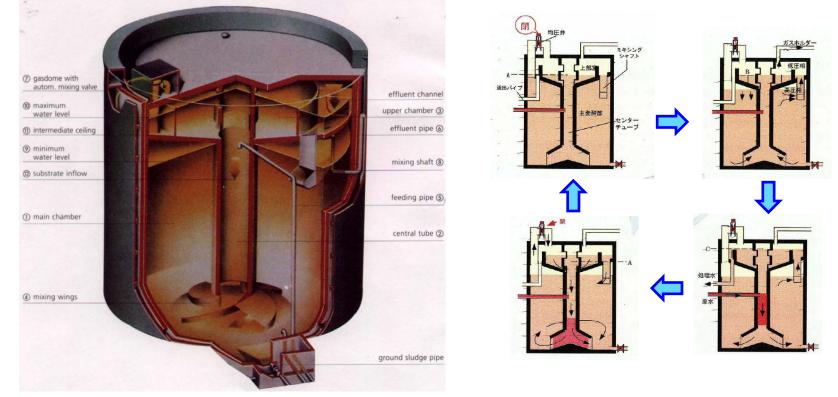
(Namgung and Chon, 2010)



Energy Saving in Process Configuration

Saving Mixing Energy in Anaerobic Digestion

Biogas Induced Mixing Arrangement (BIMA)



- Self-mixing hydraulic digester systems → No need for mechanical equipment such as agitator, circulation pumps or gas injection for mixing the digester contents.
- The 2-chamber system uses the produced biogas to create a level difference in the chambers
- The turbulent mixing occurs against the biogas production in intervals of 4-10 times a day.

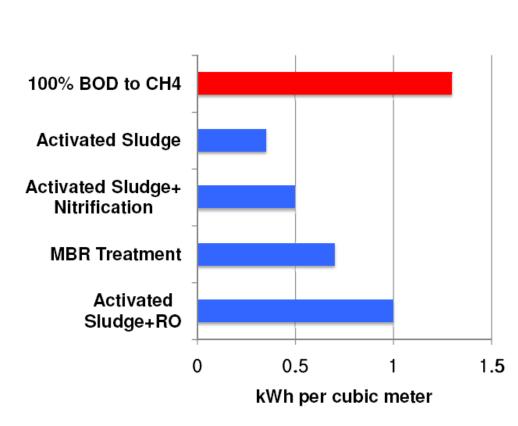


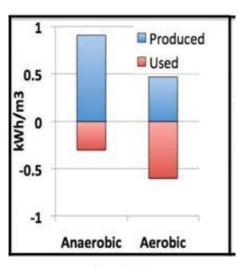
Energy Saving in Process Configuration

Anaerobic Treatment of Wastewater

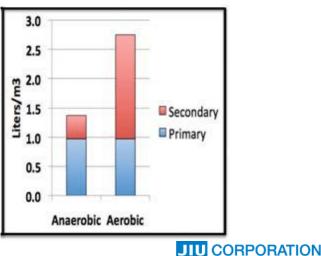
Energy Consumption or Production

<Energy>







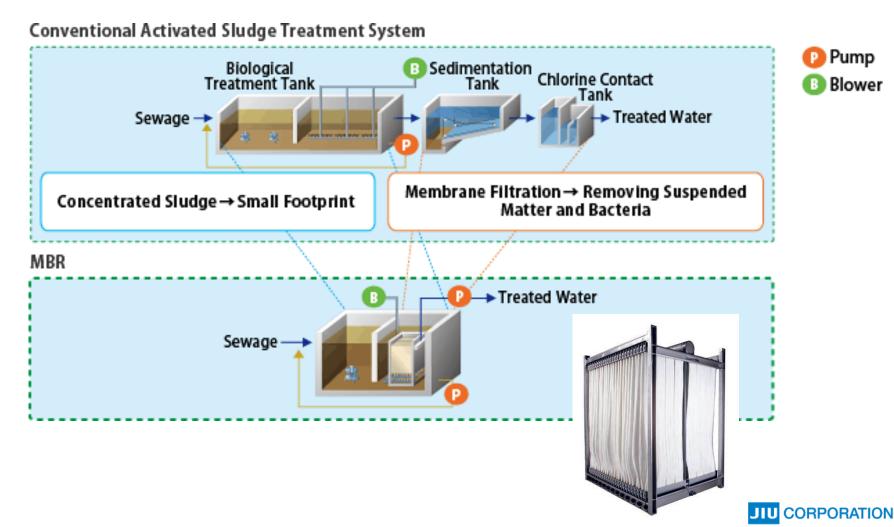




Energy Saving through Tailored Operation

Saving Aeration Energy in MBR Process

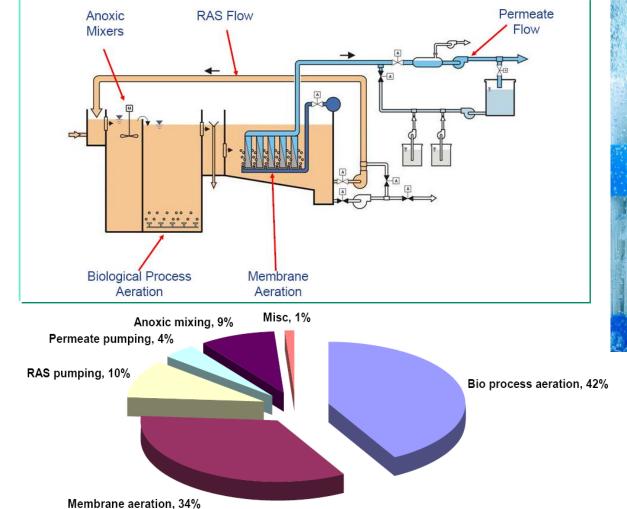
A Basic MBR Process

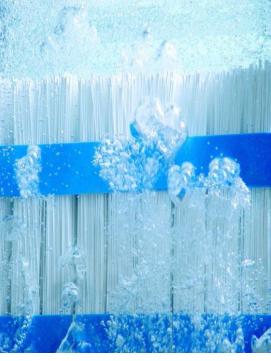


Energy Saving through Tailored Operation

Saving Aeration Energy in MBR Process

Energy User in MBR System





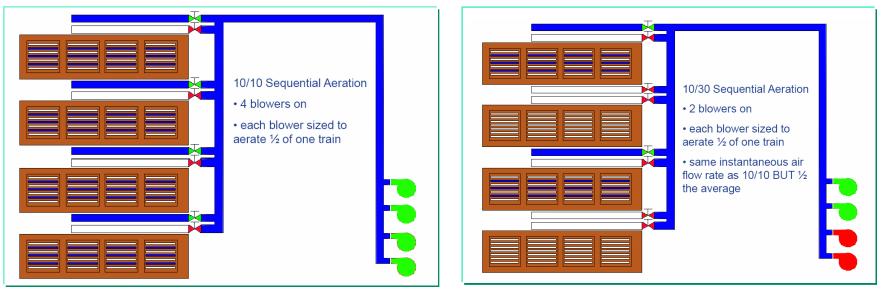
<Air scouring>



Energy Saving through Tailored Operation

Strategy for Air Scouring

10/10 Air Scour and 10/30 Air Scour (GE, Zenon)



- 10/10 air scour : cycled air on and off in 10 second intervals
- 10/30 air scour

- for 10 seconds, 24 of the 48 modules in a given cassette receive air scour. For the next 10 seconds this cassette does not receive air scour, but air scour is being used in other cassettes. For the next 10 seconds, the other 24 modules in the cassette receive air scour. For the last 10 seconds of the cycle, the cassettes do not receive air scour. A given cassette receives air ½ the time, and a given module receives air ¼ of the time. 50% savings compared to 10/10.

- Maintain 10/10 aeration at or above average daily flow
- Run at 10/30 aeration below average daily flow



Replacement to Energy Saving Equipments

- Oxygen Requirements in Biological processes
- BOD oxidation

: For solids retention times of 5-10 days, the kg of oxygen per kg of BOD usually varies from 0.92-1.07. A conservative value of 1.1 kg O_2/kg BOD is used on occasion. Higher values are valid for long detention times with low organic loadings and additional sludge oxidation.

- Ammonia oxidation
 - : Usually one kg of ammonia requires 4.3-4.6 kg of oxygen.
- Endogenous respiration
 - : 0.05-0.15 kg O_2 /kg MLVSS/d
- DO concentration maintenance for side-stream (internal and external recycle, etc.) loading
- The AOR(actual oxygen requirement) demand is the sum of the above sources
- SOR : standard oxygen requirement





Replacement to Energy Saving Equipments

Oxygen Requirements in Biological processes

$$SOR = \frac{AOR \cdot Csat_{20}}{\left[\beta \cdot Csat_{T}\left(\frac{P_{field}}{P_{msi}}\right) - DO_{field}\right]\alpha \cdot \theta^{T-20}}$$
(1)

Where AOR = actual oxygen requirements (field conditions) SOR = standard oxygen requirements (standard conditions)

$$=\frac{KL_a \text{ waste water}}{KL_a \text{ waste water}}$$

 β = saturation factor

α

P_{field} = barometric pressure at the treatment site

P_{msi} = barometric pressure at mean sea level

T = operating temperature of wastewater (°C)

 C_{sat20} = surface DO saturation concentration at 20°C and standard conditions for the particular aeration equipment at the design submergence

 C_{satT} = Surface DO saturation concentration at design temperature T and

14.7 PSIA for the particular equipment at the submergence

DO_{field} = dissolved oxygen in wastewater

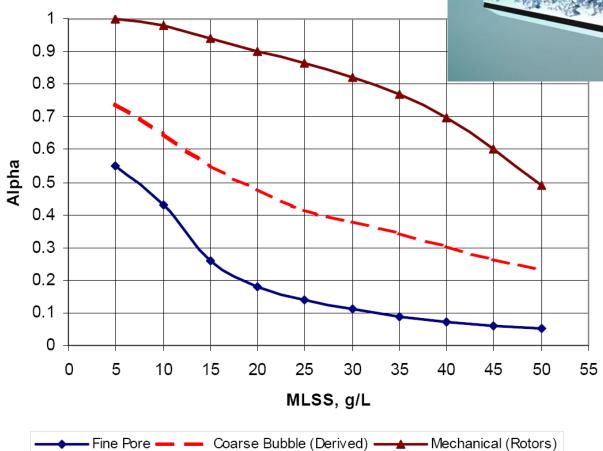
 θ = temperature correction factor

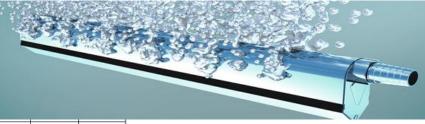




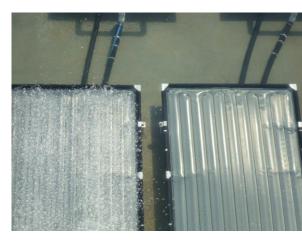
Replacement to Energy Saving Equipments

Effect of Aeration Methods





<Coarse bubble aerator>



<Ultra-fine pore membrane diffuser>





Replacement to Energy Saving Equipments

Pulsed Air Mixing of Anoxic and Anaerobic Zones - BioMIx

- Efficient mixing in anaerobic and anoxic zones with no significant oxygen transfer.
- Intermittent release of bursts of compressed air at the bottom of the water column zones.
- Testing at F. Wayne Hill Water Resource Center in Buford, GA to compare effectiveness, compatibility with anaerobic and anoxic environments, and power requirements vs. a conventional submersible propeller mixer.
 - Dye tracer tests showed similar mixing for the BioMIx and submersible mixer systems.
 - Continuous oxidation reduction potential (ORP) measurements over periods of 12 to 28 hours showed 95th percentile ORP values of less than -150 millivolts (mv), which is indicative of anaerobic environments.
 - Power analyzer readings taken simultaneously showed that energy (in kW) required to mix one anaerobic cell using the BioMIx system was 45 percent less than the energy required by a submersible mixer.



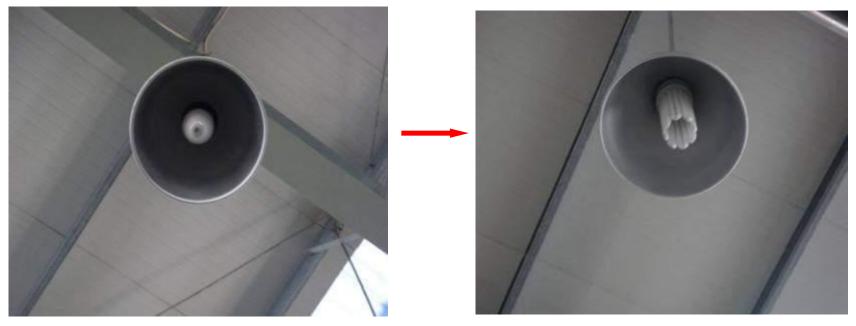


<Typical BioMIx installation>



Replacement to Energy Saving Equipments

Light Tubes



• Utilize energy saving light tubes, save about 1/3-1/4 electricity use





Thank You for Your Attention!



Questions or Comments ? hjkim@jiuene.com

