

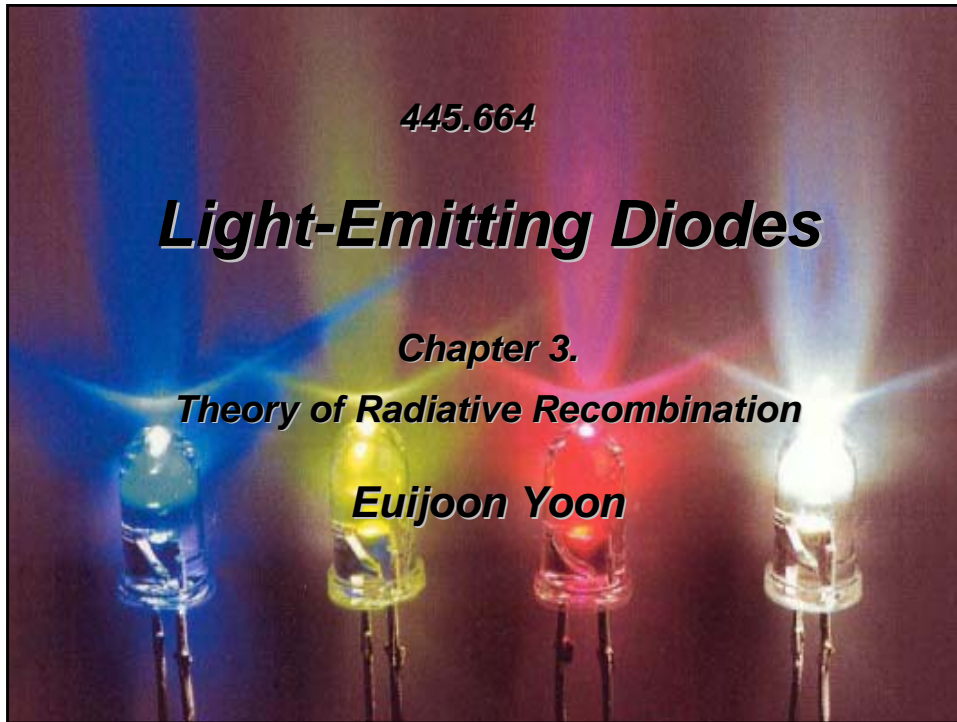
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Light-Emitting Diodes

Chapter 3.

Theory of Radiative Recombination

Euijoon Yoon



Theory of Radiative Recombination

**Semiclassical model of radiative recombination
based on equilibrium generation and recombination**

- **van Roosbroeck-Shockley model**

- Calculation of the spontaneous radiative recombination rate under equilibrium and non-equilibrium conditions

- **Einstein model**

- Calculation of the spontaneous and stimulated transitions in a two-level atom

The van Roosbroeck-Shockley Model

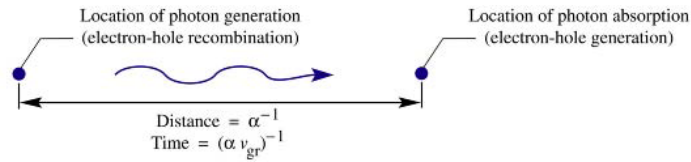


Fig. 3.1. Illustration of distance and elapsed time between a photon generation and absorption event.

$\alpha(\nu)$	absorption coefficient [cm⁻¹]
$\alpha(\nu)^{-1}$	mean distance that a photon travels before being absorbed
v_{gr}	group velocity of photons propagating in the semiconductor
$\tau(\nu)$	time that it takes for a photon to be absorbed $= (\alpha v_{gr})^{-1}$
ν	frequency of a generated photon

Photon Absorption Probability

group velocity of photons

$$v_{gr} = \frac{d\omega}{dk} = \frac{d\nu}{d(1/\lambda)} = c \frac{d\nu}{d(\bar{n}\nu)}$$

ω ($= 2\pi\nu$) **angular frequency**

k ($= \frac{2\pi}{\lambda}$) **wave vector**

\bar{n} ($= \frac{c}{\lambda\nu}$) **refractive index**

inverse photon lifetime

$$\frac{1}{\tau(\nu)} = \alpha(\nu) v_{gr} = \alpha(\nu) c \frac{d\nu}{d(\bar{n}\nu)}$$

→ photon absorption probability per unit time

Density of Photons

$$\bar{n} = \frac{c}{\lambda \nu} \Rightarrow \lambda = \frac{c}{\bar{n} \nu}$$

$$d\lambda = -\frac{c}{(\bar{n} \nu)^2} \frac{d(\bar{n} \nu)}{d\nu} d\nu$$

Density of photons per unit volume (equilibrium conditions)
by Planck's black body radiation formula

$$N(\lambda) d\lambda = \frac{8\pi}{\lambda^4} \frac{1}{e^{h\nu/kT} - 1} d\lambda$$

$$N(\nu) d\nu = \frac{8\pi \nu^2 \bar{n}^2}{c^3} \frac{d(\bar{n} \nu)}{d\nu} \frac{1}{e^{h\nu/kT} - 1} d\nu$$

The van Roosbroeck-Shockley Equation

Phonon absorption rate per unit volume
= Absorption probability × Photon density

Absorption rate per unit volume in the frequency interval ν and $\nu + d\nu$

$$R_0(\nu) d\nu = \frac{N(\nu) d\nu}{\tau(\nu)}$$

$$= \left(\frac{8\pi \nu^2 \bar{n}^2}{c^3} \frac{d(\bar{n} \nu)}{d\nu} \frac{1}{e^{h\nu/kT} - 1} d\nu \right) \cdot \left(\alpha(\nu) c \frac{d\nu}{d(\bar{n} \nu)} \right)$$

$$= \frac{8\pi \nu^2 \bar{n}^2}{c^2} \frac{\alpha(\nu)}{e^{h\nu/kT} - 1} d\nu$$

Absorption rate per unit volume

$$R_0 = \int_0^\infty R_0(\nu) d\nu = \int_0^\infty \frac{8\pi \nu^2 \bar{n}^2}{c^2} \frac{\alpha(\nu)}{e^{h\nu/kT} - 1} d\nu$$

van Roosbroeck-Shockley Equation

The Simplified van Roosbroeck-Shockley Model

Absorption coefficient

$$\alpha = \alpha_0 \sqrt{\frac{E - E_g}{E_g}} \quad \begin{array}{l} E_g \text{ band gap energy of the semiconductor} \\ \alpha_0 \text{ absorption coefficient at } h\nu = 2E_g \end{array}$$

Neglecting the frequency dependence of the refractive index

Using the absorption coefficient at the band edge ($\alpha = \alpha_0$)

$$x = h\nu / kT = E / kT$$

$$x_g = E_g / kT$$

$$dx = (h / kT) d\nu$$

$$R_0 = 8\pi c \bar{n}^2 \alpha \sqrt{\frac{kT}{E_g}} \left(\frac{kT}{ch}\right)^3 \int_{x_g}^{\infty} \frac{x^2 \sqrt{x - x_g}}{e^x - 1} dx$$

The simplified van Roosbroeck-Shockley Equation

Bimolecular Recombination Rate

Under equilibrium conditions,

$$\begin{array}{ccc} \text{photon absorption rate } (R_0) & = & \text{photon emission rate} \\ \parallel & & \parallel \\ \text{carrier generation rate} & = & \text{carrier recombination rate } (R) \end{array}$$

Equilibrium carrier recombination rate

$$R = B n p = B n_i^2 = R_0 \quad B \text{ bimolecular coefficient}$$

Material	E_g (eV)	α_0 (cm^{-1})	\bar{n} (-)	R_0 ($\text{cm}^{-3} \text{s}^{-1}$)	n_i (cm^{-3})	B ($\text{cm}^3 \text{s}^{-1}$)	τ_{spont} (s)
GaAs	1.42	2×10^4	3.3	7.9×10^2	2×10^6	2.0×10^{-10}	5.1×10^{-9}
InP	1.35	2×10^4	3.4	1.2×10^4	1×10^7	1.2×10^{-10}	8.5×10^{-9}
GaN	3.4	2×10^5	2.5	8.9×10^{-30}	2×10^{-10}	2.2×10^{-10}	4.5×10^{-9}
GaP	2.26	2×10^3	3.0	1.0×10^{-12}	1.6×10^0	3.9×10^{-13}	2.6×10^{-6}
Si	1.12	1×10^3	3.4	3.3×10^6	1×10^{10}	3.2×10^{-14}	3.0×10^{-5}
Ge	0.66	1×10^3	4.0	1.1×10^{14}	2×10^{13}	2.8×10^{-13}	3.5×10^{-6}

Direct band gap

$$B = 10^{-9} \sim 10^{-11} \text{ cm}^3/\text{s}$$

Indirect band gap

$$B = 10^{-13} \sim 10^{-15} \text{ cm}^3/\text{s}$$

Table 3.1. Bimolecular recombination coefficient at 300 K for different semiconductors as calculated from the energy gap, absorption coefficient, and refractive index at the bandgap energy. The spontaneous lifetime is given by $B^{-1} N_{D,A}^{-1}$ and it is calculated for a majority carrier concentration of 10^{18} cm^{-3} .

Einstein Model

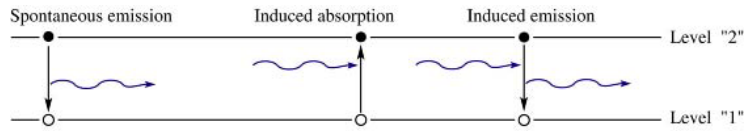


Fig. 3.2. Spontaneous emission, induced absorption, and induced emission events in the two-level atom model.

A Spontaneous transition rate

B Induced (stimulated) transition rate \propto photon density, $\rho(\nu)$

Probability per unit time

downward transition ($2 \rightarrow 1$)

upward transition ($1 \rightarrow 2$)

$$W_{2 \rightarrow 1} = B_{2 \rightarrow 1} \rho(\nu) + A$$

$$W_{1 \rightarrow 2} = B_{1 \rightarrow 2} \rho(\nu)$$

Induced emission

spontaneous emission

Induced absorption

Einstein showed that

1. $B = B_{2 \rightarrow 1} = B_{1 \rightarrow 2} \rightarrow$ Stimulated absorption and stimulated emission are complementary processes.
2. $A/B = 8\pi \bar{n}^3 h \nu^3 / c^3 =$ constant in an isotropic medium