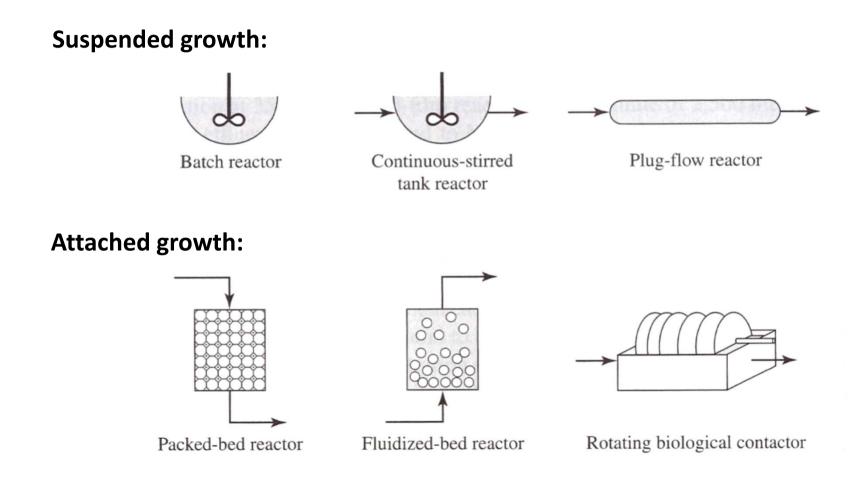
Reactors I

Today's class

- Types of bioreactors
- Generic approach for reactor analysis
- Reactor analysis example: batch reactor
 - Batch reactor analysis for 1st order reaction
 - Batch reactor analysis for Monod kinetics
 (with some knowledge buildup for numerical analysis)

Bioreactors



Anaerobic Fluidized Bed Membrane Bioreactor for Wastewater Treatment

JEONGHWAN KIM,' KIHYUN KIM.' HYOUNGYOUNG YE,' EUNYOUNG LEE,' CHUNGHEON SHIN,'

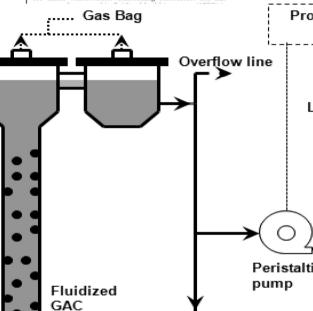
PERRY L. MCCARTY, * AND JAEHO BAE*.* Department of Environmental Engineering, Inha University, Department of Environmental Engineering, Inna Oniversity, Namgu, Yonghyun dong 253, Incheon, Republic of Korea, and Department of Civil and Environmental Engineering, Stanford University, Stanford, California 94305, United States

Received August 9, 2010. Revised manuscript received November 23, 2010. Accepted November 30, 2010.

Anaerobic membrane bioreactors have potential efficient treatment of domestic and other wastewate fouling being a major hurdle to application. It wa fouling can be controlled if membranes are place contact with the granular activated carbon (GAC) in fluidized bed bioreactor (AFMBR) used here for treatment of effluent from another anaerobic rea

tive newer approach is the use of anaerobic membrane bioreactors (AMBR), which are capable of achieving high effluent quality (4, 5). However, membrane fouling is a concern, as it increases operating and energy costs (5-9). Membranes are commonly placed external to the bioreactor, where a high cross-flow velocity is used to reduce fouling, or submerged within the reactor itself, where extensive gas scouring is generally used. In an extensive review, Liao et al. (6) summarized that energy usage was 3-7.3 kWh/m³ with external cross-flow membranes and 0.25-1.0 kWh/m3 with internal submerged membranes, quantities that are little different than with their aerobic counterparts. Such highenergy requirements reduce the potential advantage of anaerobic over aerobic systems.

For low-strength wastewater, Liao et al. (6) recommended more research on combining membranes with existing highrate reactor configurations already determined to be suitable for dilute wastewaters. Concerning bioreactors for this



Flow meter

-AFBR-

Prog Le Peristaltic С Recirculation pump Peristaltic pump -AFMBR-

Chemical Engineering Journal 426 (2021) 131912





Anaerobic membrane bioreactor model for design and prediction of domestic wastewater treatment process performance

Chungheon Shin^{a,b,c,*}, Sebastien H. Tilmans^{a,b,c}, Felipe Chen^c, Craig S. Criddle^{a,b,c}

^a Department of Civil and Environmental Engineering, Stanford University, 473 Via Ortega, Stanford, CA 94305, United States ^b National Science Foundation Engineering Research Center for Re-Inventing the Nation's Urban Water Infrastructure (ReNUWIt), 473 Via Ortega, Stanford, CA 94305, United States

^e Codiga Resource Recovery Center (CR2C), 692 Pampas Ln, Stanford, CA 94305, United States

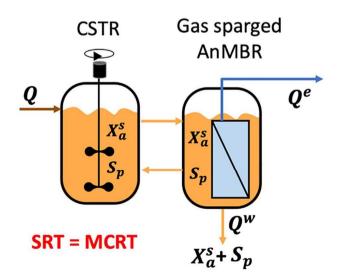
ARTICLEINFO

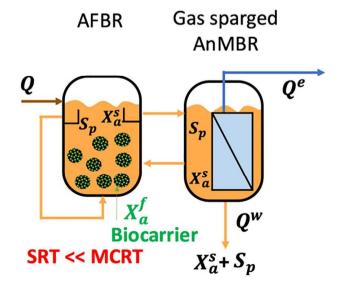
ABSTRACT

Keywords: Anaerobic membrane bioreactor Anaerobic treatment of municipal wastewater that produces energy and low levels of biosolids for disposal has the potential to replace conventional aerobic processes. Here we propose a generic model for anaerobic mem-

(a) CSTR-AnMBR







Suspended vs. attached growth





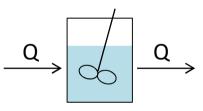
suspended growth

attached growth

Reactors for suspended growth

- Batch reactor
 - Bench-scale test systems
 - Some wastewater processes "sequencing batch reactors"
- Continuously-stirred tank reactor (CSTR)
 - Activated sludge
 - Flocculator
- Plug flow reactor (PFR)
 - Disinfection
 - Long river/canal
 - Pipeline/aqueduct







- 1. Draw schematics, define control volume, make assumptions if necessary
- 2. Set mass balance (for a <u>single</u> substance!!!) (mass rate of accumulation)

```
= (rate of mass in) – (rate of mass out)
```

```
+ (mass rate of gain/loss)
```

Any processes related to gain/loss, but here we are interested in reactions!

3. Rearrange/solve the equation to a useful form

Reactor analysis: Batch reactor, 1st order

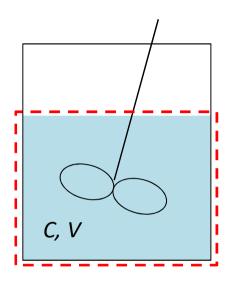
 Schematics, CV & assumption: entire reactor as CV, complete mixing, initial concentration = C₀

2) Set mass balance (for the contaminant)

$$\frac{dC}{dt} = -kC$$

3) Rearrange/solve

$$\frac{dC}{C} = -kdt \quad \Rightarrow \quad \int_{C_0}^C \frac{dC}{C} = -\int_0^t kdt \quad \Rightarrow \quad \ln C - \ln C_0 = -kt$$
$$C/C_0 = e^{-kt}$$



Reactor analysis: Batch reactor, Monod

1) Schematics, CV & assumption: initial substrate concentration = S^0 initial active biomass concentration = X_a^0

2) Set mass balance

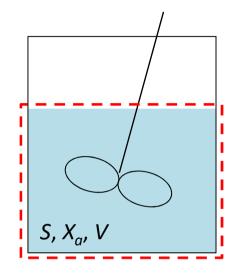
(one for substrate, one for active biomass)

[substrate mass balance]

$$\frac{dS}{dt} = r_{ut} = -\frac{\hat{q}S}{K+S}X_a$$

[active biomass mass balance]

$$\frac{dX_a}{dt} = \mu X_a = \left(Y\frac{\hat{q}S}{K+S} - b\right)X_a = r_{net}$$



 r_{net} = net rate of active biomass growth ($M_x L^{-3} T^{-1}$) **10**

Batch, Monod: Governing & initial eqs.

3) Rearrange/solve

We need to get a solution for...

Governing equations

$$\frac{dS}{dt} = -\frac{\hat{q}S}{K+S}X_a$$
$$\frac{dX_a}{dt} = \left(Y\frac{\hat{q}S}{K+S} - b\right)X_a$$

$$S(t = 0) = S^0$$
 $X_a(t = 0) = X_a^0$

Our interest would be S vs t, X_a vs t

The math here is much more difficult than it was for 1st order reaction because:

- There are two variables which are inter-correlated
- The differential equations are nonlinear with respect to *S*

Two ways of solving a mathematical model:

- 1) Analytical solution an exact solution
 - such as S = f(t), $X_a = g(t)$; not always available
- 2) Numerical solution an approximate solution

Batch, Monod: Analytical solution

We need an assumption which is only occasionally acceptable that <u>decay is negligible</u>.

Then,

 $X_a = X_a^0 + Y(S^0 - S)$ (biomass growth) = (true yield) x (substrate utilized)

The two mass balance equations are reduced to one nonlinear differential eq.:

$$\frac{dS}{dt} = -\frac{\hat{q}S}{K+S} \left[X_a^{\ 0} + Y(S^0 - S) \right]$$

Using the best knowledge of math, we get:

$$t = \frac{1}{\hat{q}} \left\{ \left(\frac{K}{X_a^0 + YS^0} + \frac{1}{Y} \right) ln \left(X_a^0 + YS^0 - YS \right) - \left(\frac{K}{X_a^0 + YS^0} \right) ln \frac{SX_a^0}{S^0} - \frac{1}{Y} ln X_a^0 \right\}$$
[5.11]

We fail to get an explicit solution of **S** as a function of **t**

Batch, Monod: Numerical solution

$$\frac{dS}{dt} = -\frac{\hat{q}S}{K+S}X_a \qquad \frac{dX_a}{dt} = \left(Y\frac{\hat{q}S}{K+S} - b\right)X_a$$

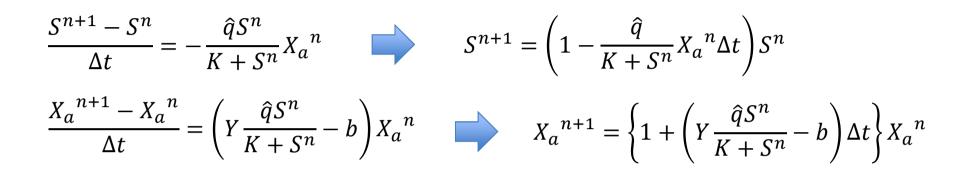
Divide the time range into finite time steps with a length of Δt . Then, between nth and n+1th time step, the 1st derivatives can be approximated as:

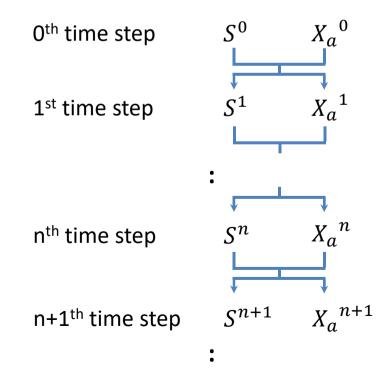
$$\frac{dS}{dt} \approx \frac{S^{n+1} - S^n}{\Delta t} \qquad \frac{dX_a}{dt} \approx \frac{X_a^{n+1} - X_a^n}{\Delta t} \qquad \frac{S^n \& X_a^n: S \& X_a \text{ values at}}{n^{th} \text{ time step, respectively}}$$

If nth time step data are used for the right hand sides of the equations it is called as an "explicit" method.

cf) "implicit" method uses n+1th time step

Let's try explicit method:





15