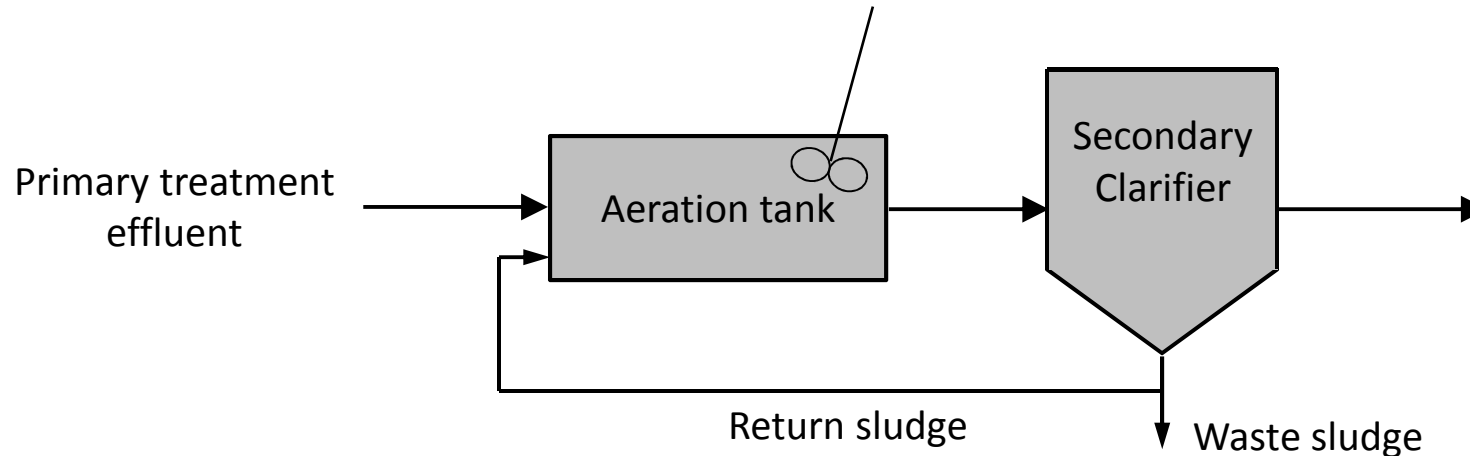
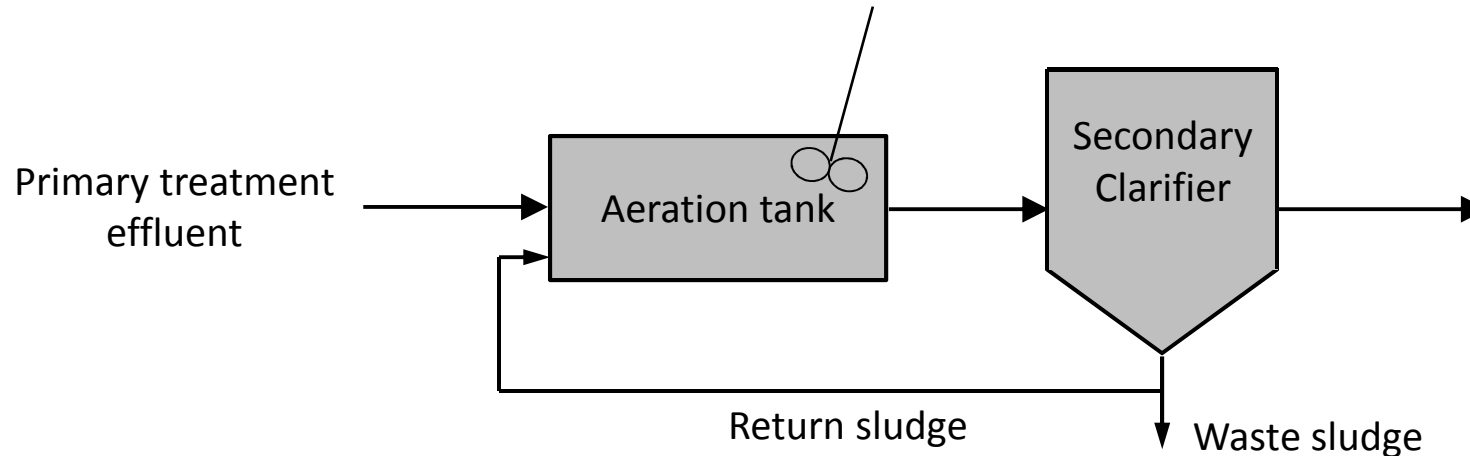


Activated sludge process



- A biological wastewater treatment technique using suspended microorganisms (dispersed growth)
- Aeration tank: a mixture of wastewater and microorganisms is agitated and aerated
- Wastewater BOD is removed by active microorganisms

Activated sludge process



- Secondary clarifier: the microorganisms (also called biosolids or sludge) are separated from water by gravity
- Most of the settled sludge is returned to the aeration tank (Why? - We need a high population of microorganisms)
- A fraction of the settled sludge is wasted (Why? – microorganisms grow!)

Solids retention time (SRT)

- Recall hydraulic retention time
 - t_o = the time that fluid elements stay in the system
 - $= V/Q$
- Solids retention time (or mean cell residence time)
 - θ_c = the time that microorganisms stay in the system
- $t_o \neq \theta_c$ if sludge is returned to the aeration tank
(Why??)

Suspended solids in “mixed liquor”

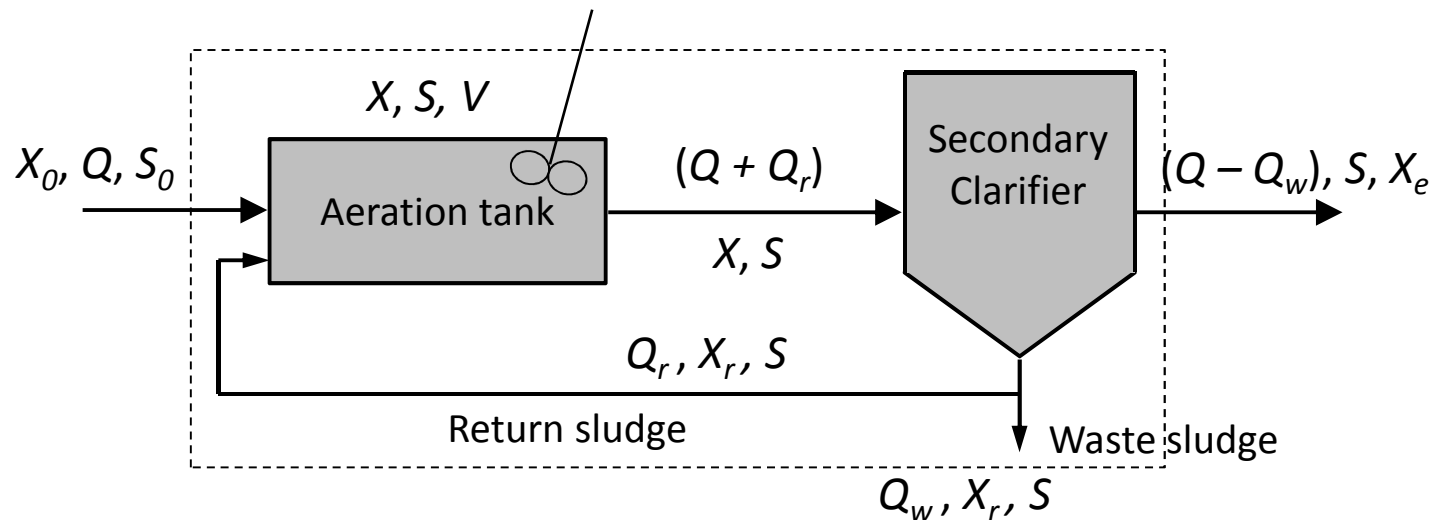
- The mixture of microorganisms and wastewater in aeration tank is called “mixed liquor”



Analyzing activated sludge process

- Let's analyze the activated sludge process using two basic knowledge:
 - Monod kinetics (the reaction)
 - The system configuration (mass balance)
- We have two substances to analyze:
 - BOD (=substrate=food): the performance of the activated sludge process to treat wastewater
 - Microorganisms (=MLVSS): those that consume BOD; also related to sludge production

Analyzing activated sludge process



X, X_e, X_r : MLVSS concentrations in aeration tank, effluent, and return sludge

Assumption:

- i) Steady-state
- ii) The aeration tank is a CMFR
- iii) All reactions occur in the aeration tank

Analyzing activated sludge process

- Mass balance for substrate:

$$QS_0 - V \frac{\mu_m SX}{Y(K_s + S)} = (Q - Q_w)S + Q_w S$$

- Mass balance for microorganisms:

$$QX_0 + V \left(\frac{\mu_m SX}{K_s + S} - k_d X \right) = (Q - Q_w)X_e + Q_w X_r$$

Analyzing activated sludge process

Additional assumption: The influent and effluent MLVSS is negligible

- Mass balance for microorganisms

$$QX_0 + V \left(\frac{\mu_m SX}{(K_s + S)} - k_d X \right) = (Q - Q_w)X_e + Q_w X_r$$

Analyzing activated sludge process

With some math, we get:

$$\frac{Q_w X_r}{V X} = \frac{Q Y}{V X} (S_0 - S) - k_d$$

When the effluent MLVSS is negligible, we find:

$$\theta_c = \frac{\text{MLVSS in the aeration tank}}{\text{MLVSS mass flow out of the system}} = \frac{V X}{Q_w X_r}$$

Analyzing activated sludge process

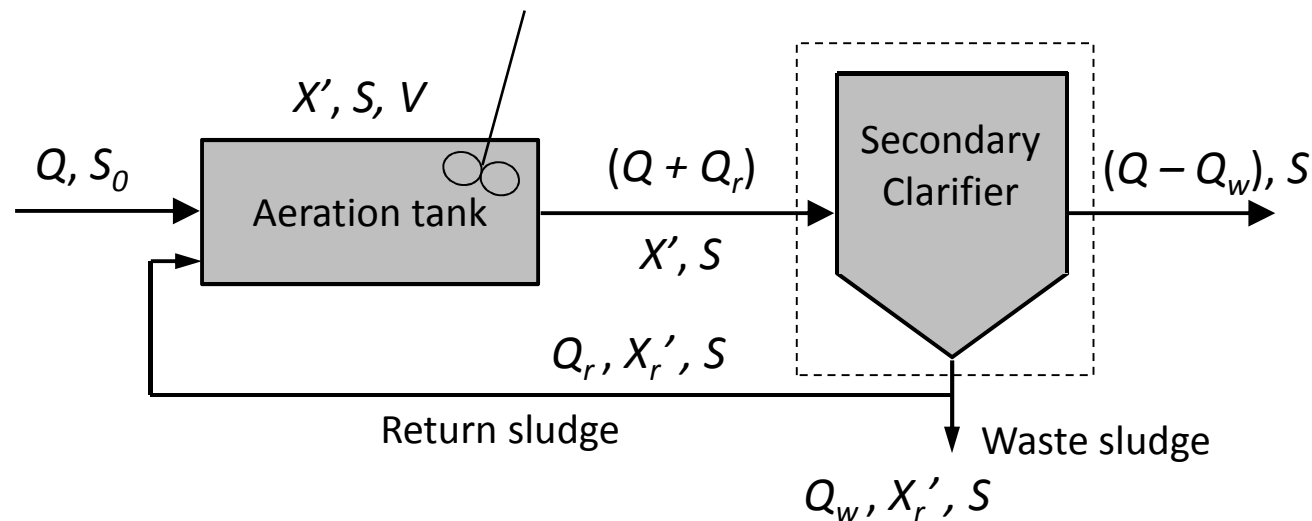
Solutions:

$$S = \frac{K_s(1 + k_d\theta_c)}{\theta_c(\mu_m - k_d) - 1} \quad X = \frac{\theta_c Y(S_0 - S)}{t_0(1 + k_d\theta_c)}$$

- SRT (θ_c) is a key design and operation parameter
- The effluent substrate concentration, S , is independent of the influent substrate concentration, S_0
- Higher $S_0 \rightarrow$ higher MLVSS in the aeration tank \rightarrow more substrate biodegradation \rightarrow same S

Sludge return

- Goal: to maintain a sufficient concentration of activated sludge (=microorganisms) in the aeration tank
- Mass balance for **MLSS** in the **secondary clarifier** (neglect effluent MLSS)



X', X_r' : **MLSS** concentrations in aeration tank and return sludge

Sludge return

- The return sludge flow rate, Q_r , to achieve the MLSS concentration in the aeration tank, X' :

$$Q_r = \frac{QX' - Q_w X_r'}{X_r' - X'}$$

The return sludge MLSS can be estimated prior to process operation by a settling test (called "SVI test")

This solution is under assumption that the effluent MLSS is negligible

(We will keep this assumption for this class, but it is not always true!)

Analyzing activated sludge process

Q1: A completely mixed activated sludge process receives a primary effluent at a flowrate of $0.150 \text{ m}^3/\text{s}$ with a BOD_5 of 84.0 mg/L . Calculate the solids retention time required to meet a secondary effluent standard of 10.0 mg/L BOD_5 .

Following microbial growth parameters apply:

$$K_s = 100 \text{ mg/L BOD}_5$$

$$\mu_m = 2.5 \text{ day}^{-1}$$

$$k_d = 0.050 \text{ day}^{-1}$$

$$Y = 0.50 \text{ mg VSS/mg BOD}_5$$

Analyzing activated sludge process

Q2: For the given activated sludge process, calculate the volume of the aeration tank to maintain MLVSS concentration of 2000 mg/L in the aeration tank.

Q3: Calculate the required return sludge pumping rate. Use following data.

$$MLSS = 1.43 \times MLVSS$$

$$\text{Return sludge concentration } (X'_r) = 10000 \text{ mg/L}$$

Effluent SS is negligible

Reading assignment

Textbook Ch 11 p. 538-546

Tertiary (advanced) treatment

- Goal: to improve the quality of the secondary treatment effluent
- Many of the Korean wastewater treatment plants now have advanced treatment process
- Further BOD and SS removal, nutrient removal, TDS removal, or the removal of refractory organic compounds
- Different processes can be used depending on the major target

Tertiary (advanced) treatment

- Available advanced treatment processes
 - **Granular filtration**
 - Additional removal of SS (including microorganisms)
 - Similar to water treatment
 - **Membrane filtration**
 - Additional removal of SS
 - Microfiltration is mostly commonly used for advanced treatment of wastewater
 - **Carbon adsorption**
 - Removal of non-biodegradable organic compounds (contributes to COD, but not to BOD)
 - Activated carbon is most commonly used as an adsorbent

Tertiary (advanced) treatment

- Available advanced treatment processes
 - **Chemical phosphorus removal**
 - Use chemical precipitants such as ferric chloride (FeCl_3), alum ($\text{Al}_2(\text{SO}_4)_3$), lime ($\text{Ca}(\text{OH})_2$)
 - **Chlorine disinfection**
 - Removal of pathogens
 - Required process in the U.S. → considered as a conventional process, but often categorized as tertiary treatment

Sludge treatment

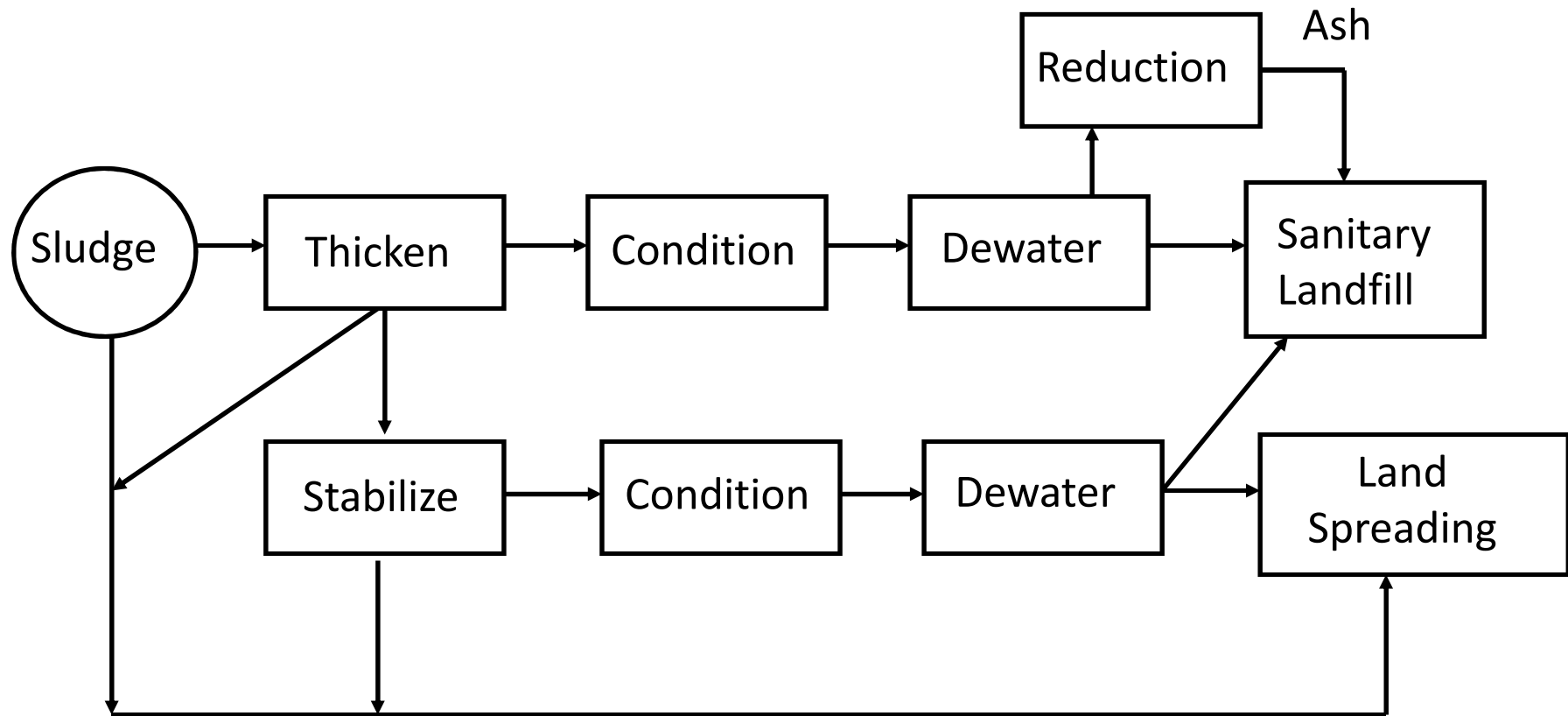
- Sources of solid waste from wastewater treatment
 - Bar racks & grit chamber
 - Inert, water can be easily removed
 - Generally not called as “sludge”
 - Truck directly to landfill after water removal
 - Primary and secondary treatment
 - Produces waste called “sludge”
 - High organic content → rapidly becomes anaerobic and putrefies
 - 3-8% solids for primary sludge & 0.5-2% solids for secondary sludge
 - Tertiary treatment: variable characteristics

Sludge treatment processes

- **Thickening:** separating as much as water possible from the raw sludge by gravity or flotation
- **Stabilization:** converting the organic solids to more inert forms
- **Conditioning:** treating the sludge with chemicals or heat so that the water can be readily separated
- **Dewatering:** separating water by vacuum, pressure, or drying
- **Reduction:** further reducing the solids and water when needed (ex: incineration)

Sludge treatment processes

- Organize the processes as needed



Sludge disposal

- **Land spreading:** can use nutrients and water in the sludge, but pathogen & heavy metal problem
- **Ocean disposal:** simple & easy, but not environmentally-friendly, now prohibited in Korea
- **Landfilling:** simple & easy, but takes a lot of landfill space
- **Composting:** use sludge as a valuable resource – but not well accepted by consumers

Wastewater as a resource

- A new paradigm: wastewater is not a WASTE, but a valuable RESOURCE
- Wastewater = water + nutrients + carbon (energy)
- Water reuse
 - Non-potable reuse: cooling water, irrigation, recreational use
 - Potable reuse: direct/indirect; can produce effluent with drinking water quality by reverse osmosis + UV disinfection
- Wastewater as a nutrient source
 - Irrigation of primary effluent
 - Use processed sludge as soil amendment
 - Precipitate nutrients to produce fertilizers

Wastewater as a resource

- Wastewater as an energy source
 - Wastewater treatment is a high energy process: accounts for 2-5% of the national energy consumption
 - Make the process “energy positive” → lots of energy savings!
 - Several ways to use energy in wastewater
 - CH₄ gas production from wastewater → heat / electricity generation
 - Electricity generation directly from wastewater (microbial fuel cells)
 - Using heat value of wastewater

Reading assignment

Textbook Ch 11 p. 562-571