

Anaerobic processes for energy recovery from organic wastes

Today's lecture

- An anaerobic process that is under full-scale operation in the field -- *anaerobic digestion*
- An anaerobic process proposed as a candidate for mainstream domestic wastewater treatment -- *anaerobic fluidized bed membrane bioreactor*

Anaerobic digestion

- **Applications**

- Stabilization of waste sludge
- Treatment of high-strength organic wastes
- Pretreatment step for conventional biological treatment

- **Advantage**

- Low biomass yield
- Energy production in the form of methane (of recent interest!)
 - WWTP -- ~2% of total energy cost in USA
 - Target on energy positive treatment of wastewater

- **Disadvantage**

- Effluent quality usually not as good as aerobic treatment



Before construction (Jan 15)

Hongchun energy town, Korea
(animal manure + food waste)

After construction (Feb 18)





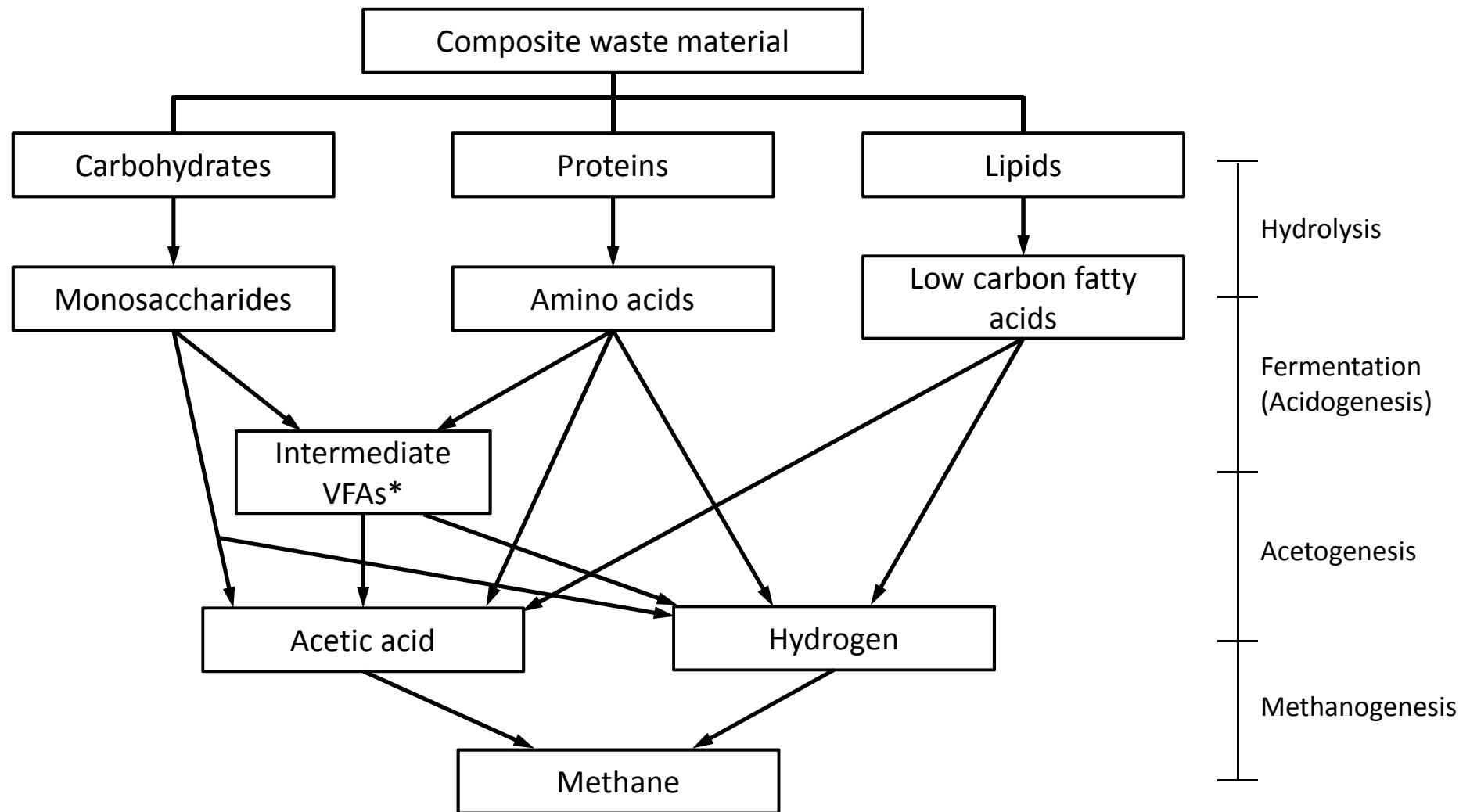
Anaerobic digestion tank

Suyeong wastewater treatment plant,
Busan, Korea
(sewage sludge)

Biogas storage tank



Pathway of anaerobic conversion of wastes



* Intermediate VFAs: Propionate ($C_2H_5COO^-$), Butyrate ($C_3H_7COO^-$), Valerate ($C_4H_9COO^-$)

Steps of anaerobic conversion (1)

- **Hydrolysis**

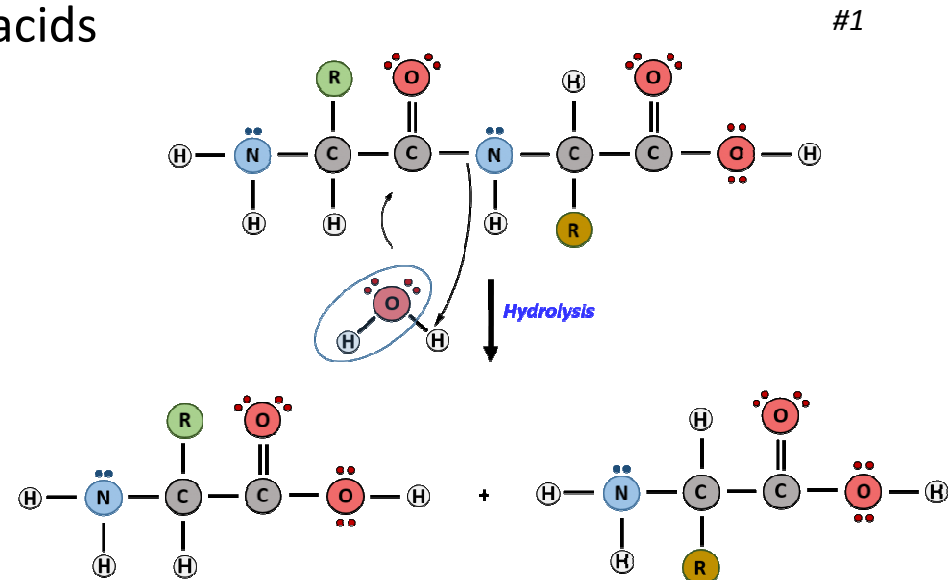
- Particulates - - - - → Soluble molecules - - - - → Monomers
- By extracellular enzymes

- **Acidogenesis (fermentation)**

- Use: sugars, amino acids, fatty acids (both e⁻ donor & acceptor)
- Produce: VFAs, CO₂, H₂

- **Acetogenesis**

- Use: VFAs other than acetate
- Produce: acetate, H₂, CO₂



Steps of anaerobic conversion (2)

- **Methanogenesis**

- By methanogens (belongs to domain Archaea)
- Two groups of methanogens
 - *aceticlastic* methanogens: acetate \rightarrow CH₄ + CO₂
 - *hydrogenotrophic* methanogens: H₂ + CO₂ \rightarrow CH₄
- In anaerobic digestion process, ~72% methane from acetic acid & ~28% from H₂ (\rightarrow gas production of ~65% CH₄ & ~35% CO₂)

Syntrophic relationship

- Methanogens – acidogens & acetogens
 - Acidogens & acetogens: produce H_2 , acetate, etc.
 - Methanogens: cleans up the acido/acetogenesis end products
 - Acetogens require relatively low H_2 partial pressure
- Often called as “*Interspecies hydrogen transfer*”

COD balance for anaerobic process

$$(\text{COD utilized}) = (\text{Biomass COD}) + (\text{Methane COD})$$

- No e^- acceptor consumed!
- COD of methane = 64 g COD/mole CH_4
= 2.86 g COD/L CH_4 (@ 0°C, 1 atm)

COD balance for anaerobic process

Q: An anaerobic reactor, which is operated at 35°C and 1 atm, is used to process a wastewater stream with a flow of 3000 m³/d and a bCOD concentration of 5000 g/m³. At 95% bCOD removal and a net biomass yield of 0.04 g VSS/g COD, what is the amount of methane produced in m³/d?

1) Calculate the amount of methane produced per day in terms of COD

$$(\text{COD utilized}) = (\text{Biomass COD}) + (\text{Methane COD})$$

$$(\text{Methane COD}) = (\text{COD utilized}) - (\text{Biomass COD})$$

$$= (\text{COD utilized}) \times (1 - 1.42 \times Y_n)$$

$$= 5000 \text{ g/m}^3 \times 3000 \text{ m}^3/\text{d} \times 0.95$$

$$\times \{1 - 1.42 \text{ g COD/g VSS} \times 0.04 \text{ g VSS/g COD}\}$$

$$= 1.34 \times 10^7 \text{ g COD/d}$$

**2) Calculate the conversion factor b/w methane COD & methane volume
@ 35 °C and 1 atm**

$$V/n = RT/P$$

$$= \frac{1 \text{ mole} \times 0.0821 \text{ atm-L/mole-K} \times 308 \text{ K}}{1 \text{ atm}}$$

$$= 25.3 \text{ L/mole}$$

$$\frac{25.3 \text{ L } CH_4/\text{mole } CH_4}{64 \text{ g COD}/\text{mole } CH_4} = 0.3953 \text{ L } CH_4/\text{g COD}$$

3) Calculate the amount of methane produced per day in terms of volume

$$\begin{aligned} 1.34 \times 10^7 \text{ g COD}/d &\times 0.3953 \text{ L } CH_4/\text{g COD} \\ &= 5.30 \times 10^6 \text{ L } CH_4/d \\ &= \mathbf{5300 \text{ m}^3/d} \end{aligned}$$

Process kinetics

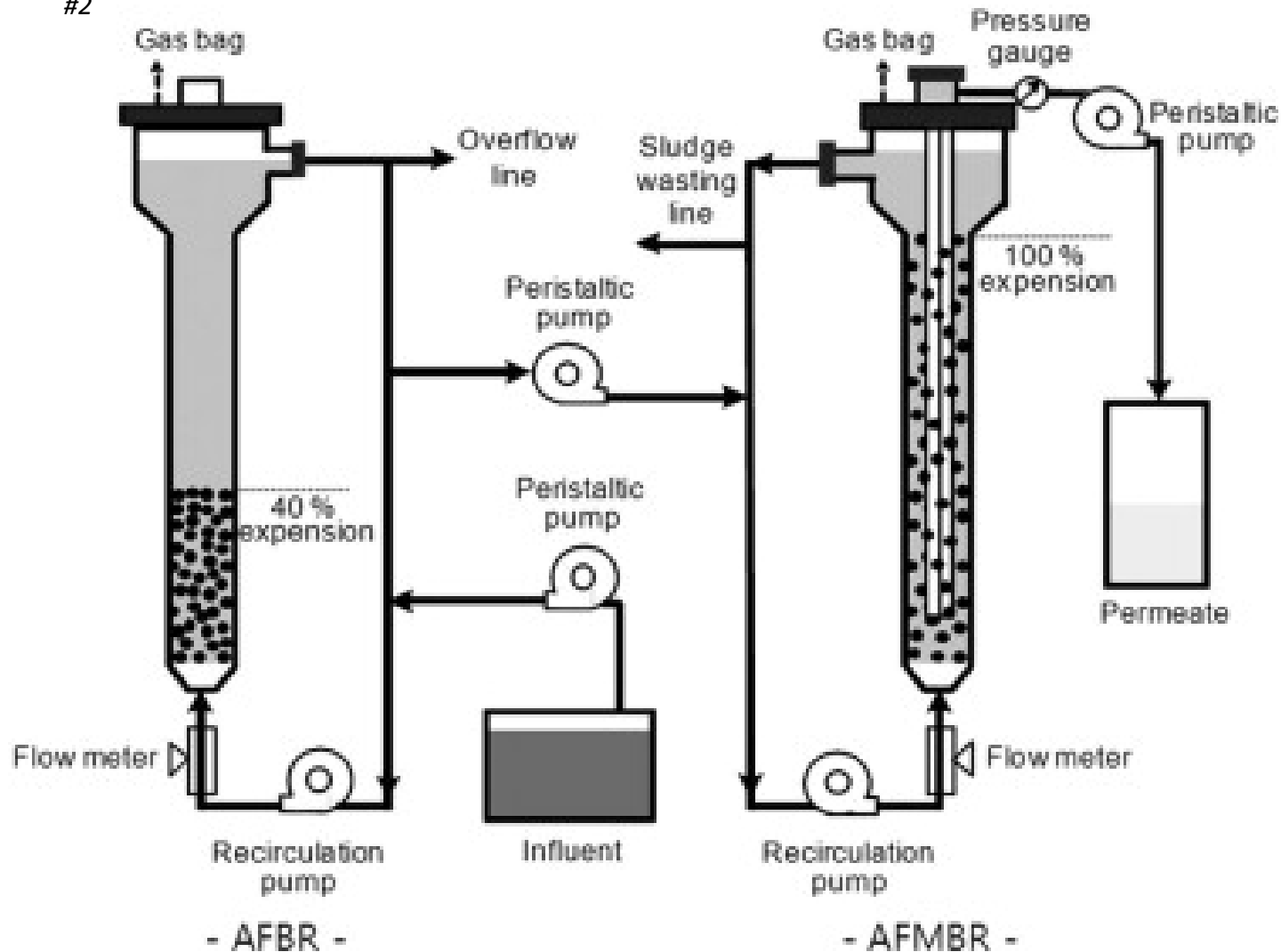
- Low yield coefficients
 - Low energy gain by chemical transformation
 - Fermentation: $Y \sim 0.06 \text{ g VSS/g COD}$; $b \sim 0.02 \text{ d}^{-1}$
 - Methanogenesis: $Y \sim 0.03 \text{ g VSS/g COD}$; $b \sim 0.008 \text{ d}^{-1}$
- Consider two steps:
 - Hydrolysis
 - Soluble substrate utilization for fermentation and methanogenesis
 - Methanogenesis the rate-limiting step
- High SRT is needed (around 40 d) due to slow degradation rate

Process stability

- Kinetics of VFA production is faster than utilization (methanogenesis)
- At steady state, sufficient methanogen population is established to maintain low VFA concentration ($<200 \text{ g/m}^3$) & $\text{pH} \geq 7.0$
- Unstable digester operation may develop under transient loading conditions (VFA production $>$ utilization): VFA accumulation & pH drop
- Low pH leads to decline in methanogenic activity: process failure
- Methanogenic inhibition can also occur by acetate accumulation (acetate conc. $> 3000 \text{ g/m}^3$)

Anaerobic Fluidized Bed Membrane Bioreactor

#2



AFMBR: Concept & advantages

- Concept
 - Two-step anaerobic process for low-strength wastewater
 - Granular activated carbon (GAC) as media for attached growth
 - Membrane filtration for solid/liquid separation
- Pilot scale demonstration successful, with acceptable effluent quality
- Advantages
 - Those for MBR apply
 - + Energy recovery (CH_4)
 - + No aeration requirements
 - + Removal of refractory organics via GAC adsorption

References

- #1) <https://wou.edu/chemistry/courses/online-chemistry-textbooks/ch450-and-ch451-biochemistry-defining-life-at-the-molecular-level/chapter-2-protein-structure/>
- #2) Kim, J., Kim, K., Ye, H., Lee, E., Shin, C., McCarty, P. L., Bae, J. (2011) *Environmental Science & Technology*, 45(2): 576-581.