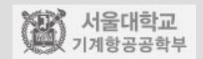
# Power and Refrigeration Systems – Gaseous Working Fluids (Lecture 12)

2021년 1학기 열역학 (M2794.001100.002)

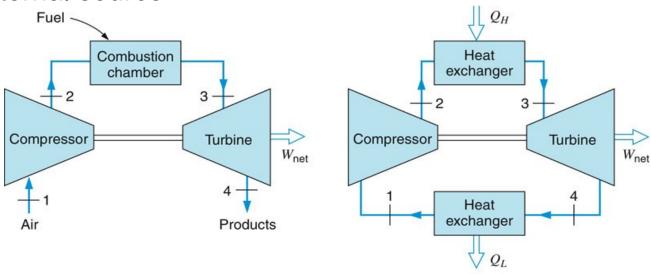
송한호

(\*) Some texts and figures are borrowed from Sonntag & Borgnakke unless noted otherwise.



## 10.1 Air-Standard Power Cycles

- → Air-standard cycle is devised to approximate an open (or actual) cycle.
- Assumptions:
  - → A fixed mass of air is used as working fluid throughout the cycle.
  - → Ideal gas, internally reversible processes.
  - → The combustion process is replaced by heat transfer from an external source.



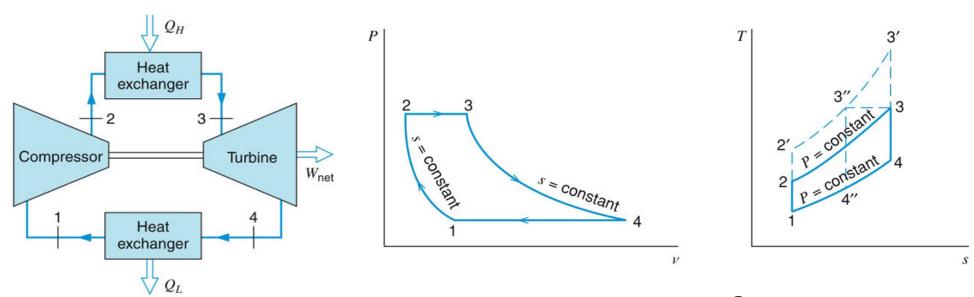
Open (or actual cycle)

Air-standard cycle



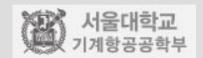
# 10.2 The Brayton Cycle

- Brayton cycle is an ideal cycle for the simple gas turbine.
- → The cycle consists of two constant-pressure heat transfer and two isentropic work transfer processes (same as in Rankine cycle).



**Brayton cycle** 

$$\eta_{\rm th} = 1 - \frac{Q_L}{Q_H}$$



For constant specific heats,

$$\eta_{\text{th}} = 1 - \frac{Q_L}{Q_H} = 1 - \frac{C_p \left( T_4 - T_1 \right)}{C_p \left( T_3 - T_2 \right)} = 1 - \frac{T_1 \left( T_4 / T_1 - 1 \right)}{T_2 \left( T_3 / T_2 - 1 \right)}$$

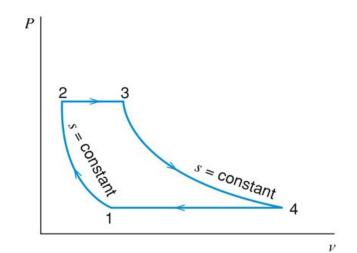
$$\frac{P_3}{P_4} = \frac{P_2}{P_1} \qquad \frac{P_2}{P_1} = \left(\frac{T_2}{T_1}\right)^{\frac{k}{k-1}} = \frac{P_3}{P_4} = \left(\frac{T_3}{T_4}\right)^{\frac{k}{k-1}}$$

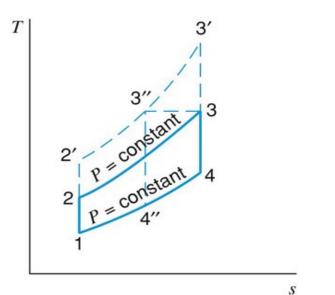
$$\frac{T_3}{T_4} = \frac{T_2}{T_1} \rightarrow \frac{T_3}{T_2} = \frac{T_4}{T_1} \rightarrow \frac{T_3}{T_2} - 1 = \frac{T_4}{T_1} - 1$$

$$\therefore \eta_{th} = 1 - \frac{T_1}{T_2} = 1 - \left(\frac{P_1}{P_2}\right)^{\frac{k-1}{k}} = 1 - \frac{1}{\left(\frac{P_2}{P_2}\right)^{\frac{k-1}{k}}}$$

Higher 
$$P_2$$
:  $1-2'-3'-4-1$ :  $T_{H,avg} \uparrow \rightarrow \eta \uparrow$ 

$$T_3$$
"= $T_3$ :  $1-2'-3''-4''-1$ :  $T_{H,avg} \uparrow \rightarrow \eta \uparrow$ , work decrease per kg

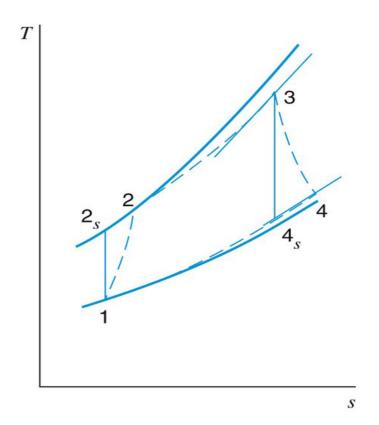




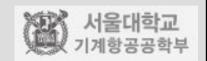


→ For real cycle, we need to consider isentropic efficiencies.

$$\eta_{comp} = \frac{h_{2s} - h_1}{h_2 - h_1} 
\eta_{turb} = \frac{h_3 - h_4}{h_3 - h_{4s}}$$

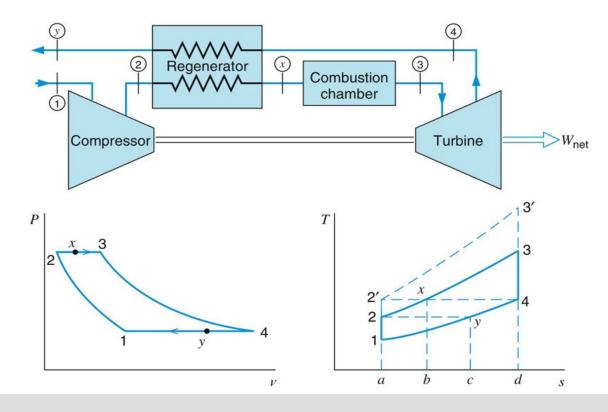


→ In Brayton cycle, the compressor work is 40~80 % of the turbine work.
(cf. In Rankine cycle, the pump work is 1~2 % of the turbine work.)



## 10.3 The Simple Gas-Turbine Cycle with a Regenerator

- → The efficiency of the gas-turbine cycle can be improved by introducing a regenerator.
- → Heat transfer from the turbine exit stream (4) to post-compressor stream (2).



T<sub>avg,H</sub> increases

**→** Efficiency increases.

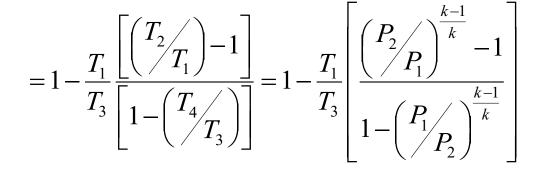
If  $P_H$  increases, there is some point where no more regeneration may occur.



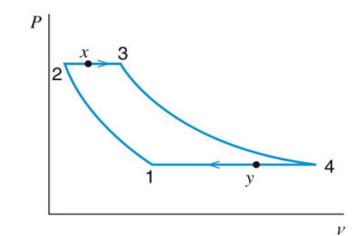
$$\eta_{th} = \frac{w_{net}}{q_H} = \frac{w_t - w_c}{q_H} \qquad q_H = C_p \left(T_3 - T_x\right) \\ w_t = C_p \left(T_3 - T_4\right)$$

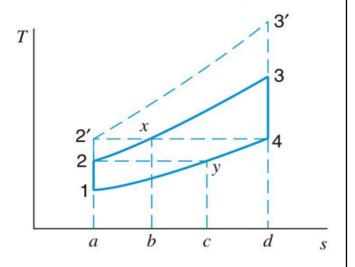
Ideal case:  $T_x = T_4 \rightarrow q_H = w_t$ 

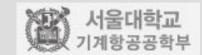
$$\eta_{th} = \frac{w_t - w_c}{w_t} = 1 - \frac{w_c}{w_t} = 1 - \frac{C_p (T_2 - T_1)}{C_p (T_3 - T_4)}$$



$$=1-\frac{T_1}{T_3}\left(\frac{P_2}{P_1}\right)^{\frac{k-1}{k}}\left|\frac{1-\binom{P_1}{P_2}^{\frac{k-1}{k}}}{1-\binom{P_1}{P_2}^{\frac{k-1}{k}}}\right| =1-\frac{T_1}{T_3}\left(\frac{P_2}{P_1}\right)^{\frac{k-1}{k}}=1-\frac{T_2}{T_3}$$



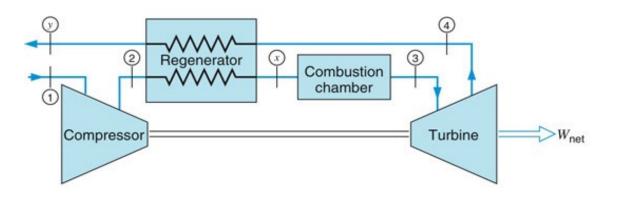


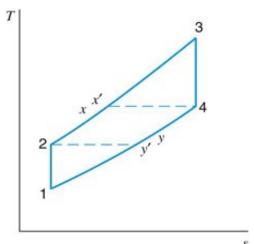


Regenerator efficiency is defined by,

$$\eta_{reg} = \frac{h_x - h_2}{h_{x'} - h_2}$$

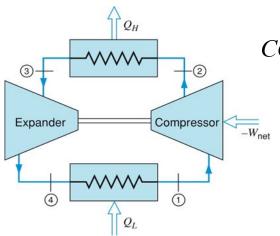
$$\eta_{reg} = \frac{T_x - T_2}{T_{x'} - T_2} \quad \text{if } C_p = const.$$





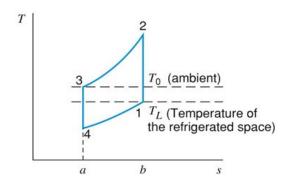
# 10.6 The Air-Standard Refrigeration Cycle

- → The air-standard refrigeration cycle is the reverse Brayton cycle.
  - → Expansion work is NOT negligible → Expander.
  - Used in the liquefaction of air, aircraft cooling systems, etc.



$$COP(\beta) = \frac{q_L}{w_{net}} = \frac{q_L}{w_C - w_E} = \frac{h_1 - h_4}{h_2 - h_1 - (h_3 - h_4)} \approx \frac{C_P(T_1 - T_4)}{C_P(T_2 - T_1) - C_P(T_3 - T_4)}$$

$$r_p = \frac{P_2}{P_1} = \left(\frac{T_2}{T_1}\right)^{k/(k-1)} = \frac{P_3}{P_4} = \left(\frac{T_3}{T_4}\right)^{k/(k-1)}$$



$$\beta = \frac{T_1 - T_4}{T_2 - T_1 - T_3 + T_4} = \frac{1}{\frac{T_2}{T_1} \frac{1 - T_3 / T_2}{1 - T_4 / T_1} - 1} = \frac{1}{\frac{T_2}{T_1} - 1} = \frac{1}{r_p^{(k-1)/k} - 1}$$

