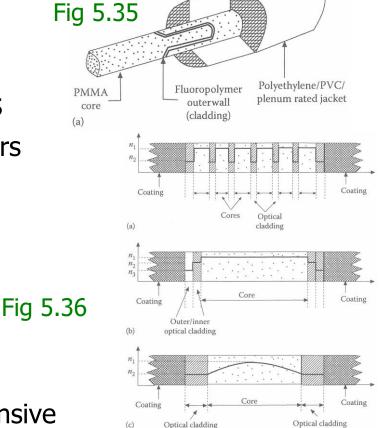
POF

- compared to GOF
 - flexible ~ workable
 - larger diameter ~ connectable, dispersion (→ low bandwidth, slow)
 - high loss ~ short-range
- core/cladding/jacket
 - core ~ PMMA popular
 - clearer than PC, more ductile than PS
 - cladding ~ lower RI ~ fluoropolymers
 - jacket ~ PE, nylon, --
- for high bandwidth POF
 - double-step index, graded-index
- for low-loss POF
 - low RI, heavy elements
 - PMMA-d8 or fluoropolymers ~ expensive



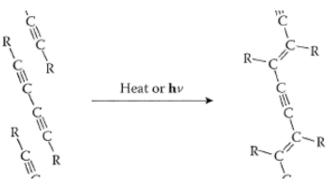


Polymers for NLO

polarization [P] of material by electric field [E]

$$\Delta P = \chi^{(1)}E + \chi^{(2)}EE + \chi^{(3)}EEE + \cdots$$

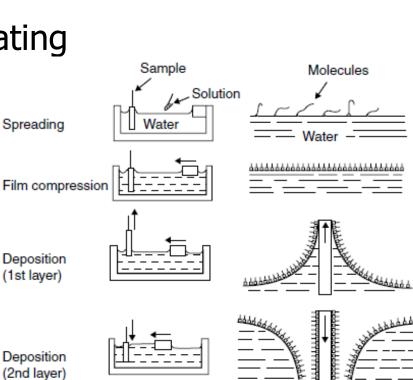
- linear ~ change in RI
 - transportation, waveguide
- (2nd) non-linear
 - 2nd harmonic generation = change in freq
 - freq doubler, amplification, optical mixing, hologram
 - need asymmetric structure
- NLO material
 - inorganic ~ LiNbO₃, ---
 - polymeric
 - organic-dispersed, main-chain, side-chain
 - processability, adhesion, cheap

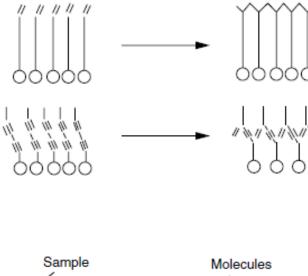


(b)

LB films

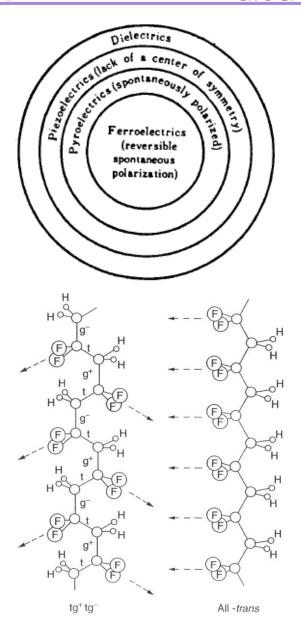
- for well-defined thin film
- monolayer at air/water interface
 - deposited and polymerized
 - polymerized and deposited
- deposition by dipping or rotating
- applications
 - NLO
 - e-beam resists
 - molecular electronics
 - sensors, ---





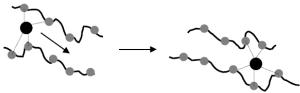
Piezo-, pyro-electric polymers

- piezo-, pyro-, ferro-electric
- PVDF and copolymers
- poling
 - poling [polarization] at high T
 - cooled with polar orientation
- compared to ceramics
 - workable, cheap, large size
 - low piezoelectric coefficient
- applications
 - heat detector
 - earphone, speaker, microphone
 - touch button



Polymer electrolytes for battery

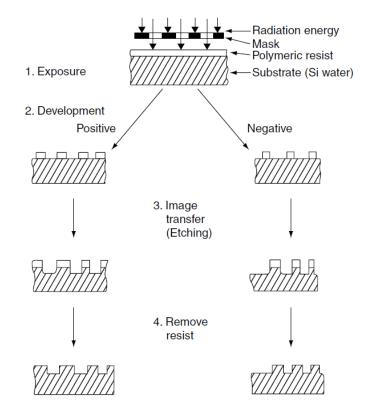
- to replace liq electrolyte in LIB
 - leakage, explosion
 - intrinsic solid polymer electrolytes
 - polymer w/ heteroatom + alkali metal salt [PEO/LiClO₄]
 - highly stable
 - low room temperature ion conductivity
 - Li moves through amorphous region
 - modifications
 - blend, composite, crosslinking, plasticizing
 - polymer gel electrolytes
 - crosslinked polymer + liquid electrolyte [PVDF/EC]
 - high conductivity, low stability



- □ for fuel cell membrane
 - PEMFC, DMFC
 - proton conductivity, separating fuel from O₂
 - fluoropolymers like Nafion
 - heat resistant polymers with polar groups
 - □ PI, PSF, PBI, PPO, ---
 - high thermomechanical stability

Polymers for lithography

- photoresist
 - coat expose w/ mask develop etch remove
 - □ positive ~ exposed area removed ← degrad'n, solubilize
 - □ negative ~ exposed area remains ← polym'n, Xlinking



- photoresist (cont'd)
 - better resolution with lower λ light

Critical Dimension(nm) =
$$k_1$$
 (λ /NA)

- near-, mid- (350> λ >280 nm), deep-UV
- excimer laser ~ 248 (KrF), 193 (ArF), 157 (F₂) nm
- resolution enhancement
 - CD $\sim 2 \lambda \rightarrow$ CD $\sim (1/3) \lambda \rightarrow ?$
- extreme UV (13.5 nm)
 - needs reflective system (← lens)

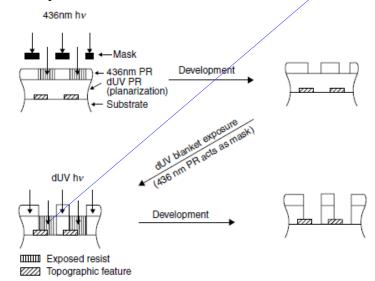
- negative PR
 - self-photocrosslinking
 - near-UV

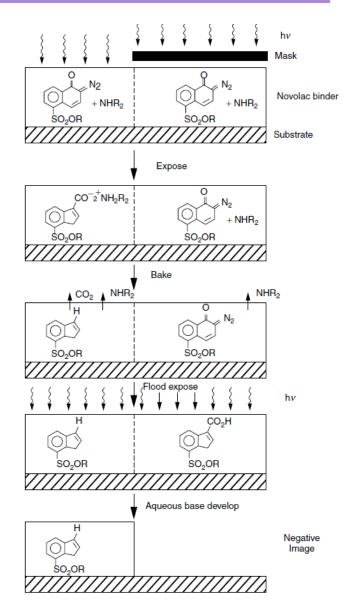
- crosslinking by photosensitizer
 - near-UV

positive PR

- enhancing solubility
 - cresol novolac/DNS
 - DNS ~ insoluble in alkali, binding polymer photoreactive to alkali-soluble
 - near- to deep-UV

- image reversal
 - convert posi image to nega image
 - amine-treated DNS system
 - deep-UV negative
- PCM (portable conformable mask)
 - multilayer mode
 - \blacksquare expose to different λ





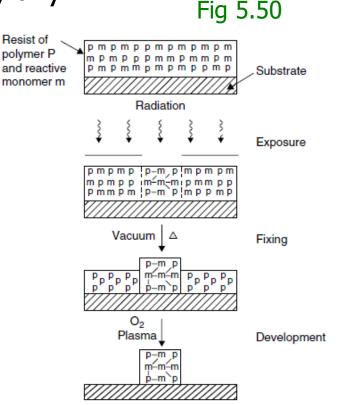
- CAR (chemically amplified resist)
 - photolabile acid generator/acid-labile polymer
 - I or S salt
 - t-butyl carbonate or esters
 - copolycarbonates
- self-developing
 - depolymerization
 - □ low T_c
- PR for higher resolution
 - □ Si, F, --- polymers

$$(CH_2-CH)_{\overline{n}} \xrightarrow{(i) h\nu} \xrightarrow{\phi_3S^+ SbF_6^-} \xrightarrow{(i) \Delta} -(CH_2-CH)_{\overline{n}} + CO_2 + CH_2 = CC_{CH_3}^{CH_3}$$

$$C-OC(CH_3)_3$$
(a) O

$$\begin{array}{c|c} & & & \\ & & \\ & & \\ & & \\ \end{array} \begin{array}{c} & \\ & \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}$$

- e-beam resist
 - shorter λ
 - positive ~ PMMA, polyolefin sulfone
 - negative ~ polymers with glycidyl, allyl
 - X-ray or ion-beam promising
- plasma developable PR
 - no-solvent, dry ~ no undercut
 - plasma-resistant polymers
 - aromatic or heteroatom
 - photolocking
- nanoimprinting, dip-pen [DPN]
- block copolymers

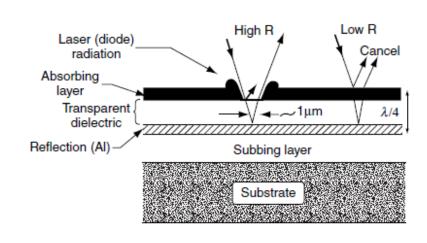


Photoresist for printing

- printing plates
 - replacing metal plates for newspaper etc
 - solid plates ~ photo-crosslinkable polymers
 - allyl-PU, acrylates, ---
- photoengraving
 - for illustrations, photographs
 - photosensitive coating on metal crosslinked and etched
- printed circuits
 - circuit/negative PR/Cu exposed and etched

Polymers for optical disc

- ROM ~ injection molding of PC on metal stamper
 - PC ~ transparent, low birefringence (low MW), low hygroscopic
- WORM
 - absorbing layer ~ metal/polymer or dye
- rewritable
 - inorganic ~ magnetic
 - organic
 - □ bump ~ rubber/thermoset
 - LCP
 - pyroelectric polymer
- □ DVD ~ high density, low λ laser



Adhesives

- liquid (to wet) to solid (for strength)
 - monomer or prepolymer polymerize
 - epoxy, cyanoacrylate, PF, ---
 - polymer at $T > T_m(T_q)$ or in solution
- solvent-based
 - PU ~ one- (high MW, shoe) or two-part (low MW, engineering)
 - substituted nylon, rubbers
- water-based
 - no VOC
 - water-soluble ~ PVA, --
 - emulsion or dispersion ~ PVAc, EVA, ----

- hot-melt
 - fast, no VOC
 - EVA, polyesters, polyamides, ---
 - low service T and strength
- radiation-curable
 - UV-curing of acrylate- or epoxy-terminated resin
 - epoxy, PU, polyester, ---
 - fast, no VOC
 - need transparency, shallow cure depth
 - e-beam curing ~ expensive
- pressure-sensitive adhesives ~ post-it

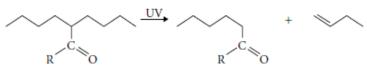
Degradable polymers

photodegradable

- environmental issue ~ time of degradation
 - UV-absorbing group like C=O
 - photosensitizer
- photoresist

biodegradable

- biodegradable polymers
 - aliphatic polyesters
 - PHB, PHV
 - PLA, PGA
 - PCL
 - aminoacid derivatives
 - polyorthoesters



$$\begin{pmatrix}
O & CH_3 & CH_5 \\
C & CH_2 & CH_2 \\
C & CH_2 & CH_3
\end{pmatrix}$$

$$\begin{pmatrix}
CH_2 & CH_2 \\
C & CH_2 \\
C & CH_2
\end{pmatrix}$$

$$CH_2 & CH_3$$

$$CH_3 & CH_4$$

$$CH_4 & CH_2$$

$$CH_5 & CH_5$$

$$CH_7 & CH_2$$

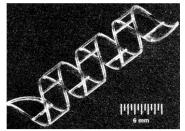
$$CH_7 & CH_9$$

$$CH_9 & CH_9$$

$$CH_9$$

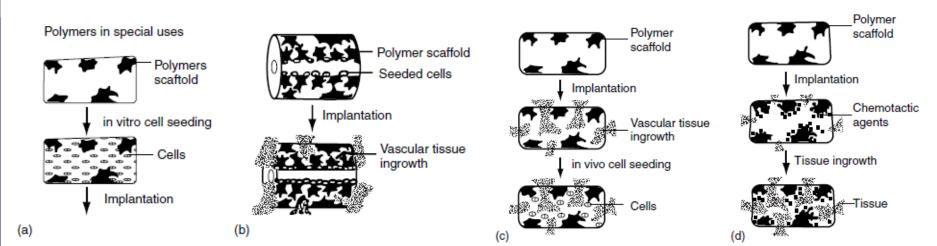
hydroxybutyrate unit hydroxyvalerate unit

- environmental issue ~ packaging
 - □ blending starch ~ 'biocollapsible'
 - □ PHB, PHV, P(HB-co-HV) ~ Biopol®
 - bio-produced and biodegradable ~ expensive
 - PLA
- for pesticides ~ controlled release
 - chelating or ester pendant group
 - encapsulation in starch, lignin, PHB, PLA
- DDS
 - encapsulation in PLGA (biodegradable), PEO, PHEMA (not)
 - oral, injection, or transdermal
- surgical instruments
 - suture, stent, screw, ---





- tissue engineering
 - organ regeneration or cell therapy
 - biodegradable polymer as scaffold
 - PLA (2 yrs), PGA (8 wks), PLGA
 - skin and cartilage commercialized
 - hydrogel
 - injectable
 - PEO, PVA, PHEMA copolymers



Ionic polymers

ion-containing polymers

- low ion conc'n ~ ionomer
- high ion conc'n
 - □ linear ~ water-soluble ~ polyelectrolyte ~ thickener, sizing
 - □ crosslinked ~ insoluble ~ ion-exchange resin

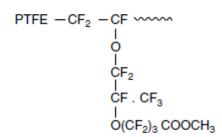
ionomers

- P(E-co-MAA) ~ random copolymer then ionized
 - □ ionic crosslinking ~ physical ~ processable
 - □ small crystallites ~ transparent, tough ~ packaging
- elastomeric ionomers
 - AA, MAA copolymerize with BD, EPDM, ---, then ionized
 - □ ionic vulcanization ~ higher strength, processable
 - □ not used as TPE ~ low service T, high stress relaxation
 - adhesives, coatings, ---

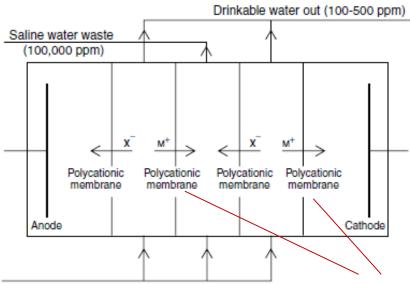
- Nafion ~ ion-exchange membrane
 - cation exchange
 - fuel cell, alkaline cell
 - electrodialysis
 - desalination

$$\begin{array}{c|c} \mathsf{PTFE} - \mathsf{CF}_2 - \mathsf{CF} & \\ & \mathsf{I} \\ & \mathsf{O} \\ & \mathsf{I} \\ & \mathsf{CF}_2 \\ & \mathsf{I} \\ & \mathsf{CF} \cdot \mathsf{CF}_3 \\ & \mathsf{I} \\ & \mathsf{O}(\mathsf{CF}_2)_2 \, \mathsf{SO}_2 \, \mathsf{F} \end{array}$$

Nafion precursor



Flemion precursor

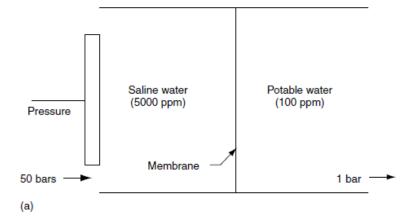


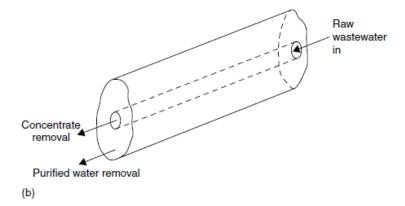
Saline water feed (5000 ppm)

polyanionic membrane

- PSF, PES ionomers
 - sulfonation of PSF, PES
 - reverse osmosis membrane
 - need not be ionic
 - should be hydrophilic
 - desalination, purification

$$-0 - \overline{\mathbb{R}} - \overline{\mathbb{C}} + \overline{\mathbb{C}} +$$



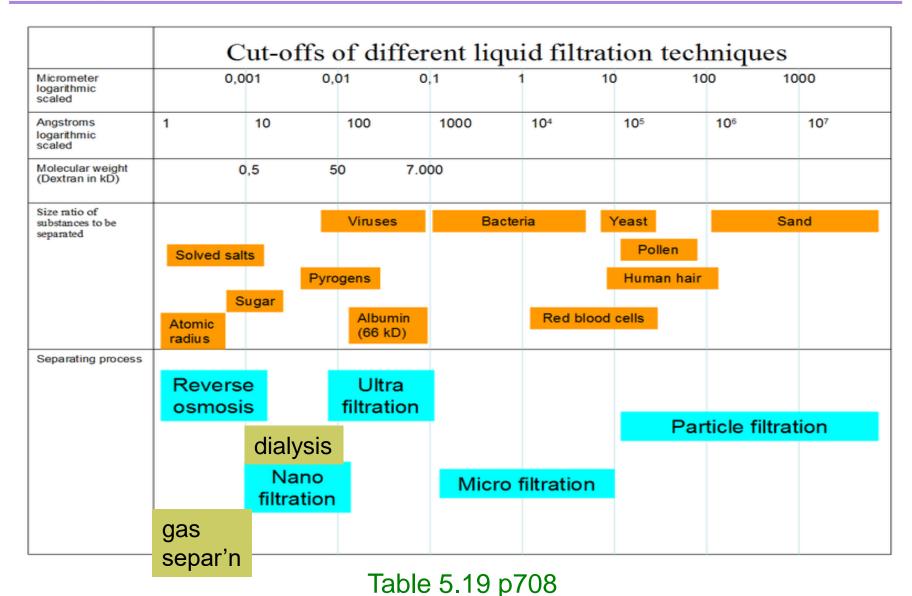


- □ ion-exchange resins Table 5.18
 - cation exchange $PSO_3^-H^+ + M^+X^- \rightleftharpoons PSO_3^-M^+ + H^+X^-$
 - □ P(S-co-DVB) sulfonated ~ regenerated by acid
 - anion exchange $\mathbb{P}_{NR_3^+OH^-} + H^+X^- \rightleftharpoons \mathbb{P}_{NR_3^-X^+} + H_2O$
 - P(S-co-DVB) chloromethylated/aminated ~ regenerate by alkali
 - ampholytic
 - cation and anion exchanger in one bead
 - $\ \ \, \text{regenerated by hot water} \quad \text{$\stackrel{COOH}{\mathbb{N}R_2}$} \quad \text{$\stackrel{COO^+Na^+}{\mathbb{N}}$}$
 - specific ~ modified to be selective to specific ion

- applications
 - □ deionizing water ~ cation/anion
 - softening water ~ cation
 - metal recovery
- □ ionene ~ ion in backbone

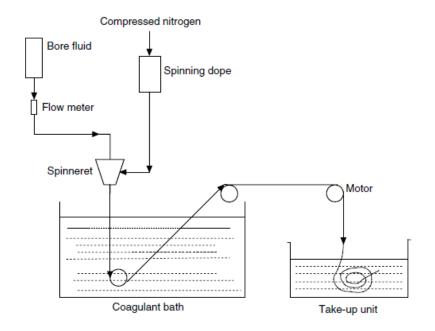
- □ scavenger resin ~ polymer with reagent
 - for purification in organic synthesis

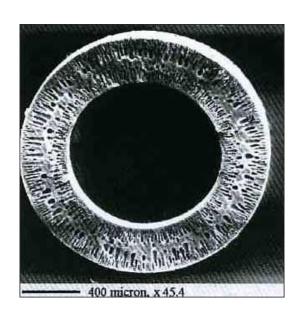
Polymer membranes



- morphology
 - isotropic (homogeneous; dense or porous)
 - anisotropic (asymmetric; dense to porous)
 - composite (dense/porous)
- preparation
 - melt extrusion
 - □ polymer only ~ dense, isotropic
 - □ polymer/diluent ~ temp-induced phase separation ~ porous
 - wet extrusion
 - polymer solution extruded and coagulated
 - to air, vapor, or liquid ~ isotropic or anisotropic

- polymerization/crosslinking/formation
 - prepolymer crosslinked between plates
 - □ for rubbery gas separation membranes ~ PEO, PDMS
- spinning
 - with bore fluid to coagulation bath
 - for hollow-fiber membranes





feed

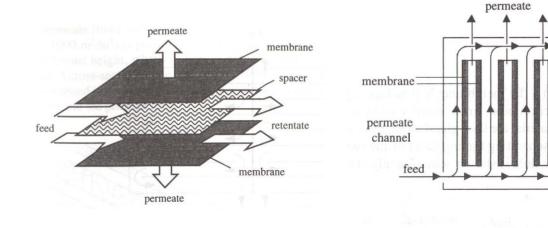
channel

concentrate

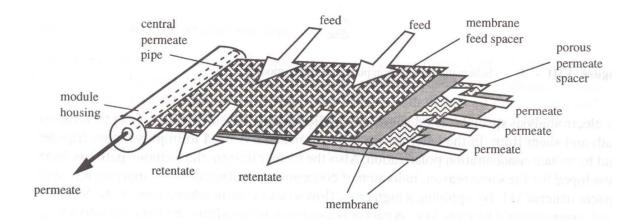
permeate

membrane modules

plate-and-frame



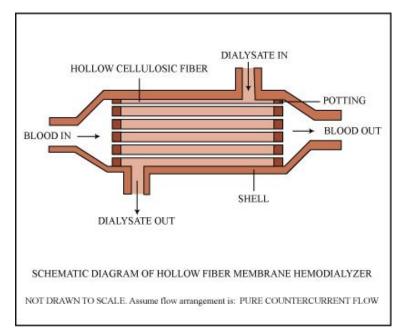
spiral wound



stop disc

- tube-in-shell
 - high flux, high solid content
- hollow fiber
 - large surface to volume
 - need clean feed (plugging)





applications

- hemodialysis, hemofiltration
 - cellulosics, PSF, PC-PE block
- plasmapheresis
 - cellulose acetate, PE, PP
- oil/water emulsion separation
 - □ hydrophilic preferred (PEO) ~ less adsorption
- gas separation
 - \bigcirc O₂ ~ medical, environmental, industrial (combustion)
 - \square N₂ ~ explosion, food
 - □ H₂ ~ energy
 - □ CO₂ ~ energy, environment
 - PP (porous)/PDMS (dense), PSF, PC