Chapter 7
Cells Maintain Their Internal Environments

## Cell Membrane

## - Components of cell membrane

- Lipid: phospholipids, sterols
- Embedded proteins : receptor proteins, adhesion proteins, recognition proteins, transport proteins
- Transmembrane proteins
- Attachment to cytosolic or exterior face of membrane



## Cell Membrane

- Membrane-spanning domains of membrane proteins
- Hydrophobic surface and hydrophilic core
- Transport of sugars, amino acids and ions (hydrophilic) through the hydrophilic core of transport proteins



## Transport Across Membrane

- Hydrophobic substances and very small molecules can cross the membrane unassisted.
- Diffusion
- Free diffusion by concentration gradient
- Hydrophobic substance, nonpolar molecules $\left(\mathrm{O}_{2}\right.$, $\mathrm{CO}_{2}$ ), small polar molecules (water, ethanol)


## Transport Proteins

- Channel proteins
- Transport of ions ( $\left.\mathrm{Na}^{+}, \mathrm{K}^{+}, \mathrm{Ca}^{2+}, \mathrm{Cl}^{-}\right)$
- Along their concentration gradients
- Aquaporin: channel for water (much faster than the diffusion across the membrane)
- Gated channel



## Transport Proteins

- Carrier proteins
- Escort energy substrates and metabolic building blocks, such as glucose, amino acids, and nucleosides
- Along the concentration gradient
- Slower than simple diffusion



## Active Transport

- Pump
- One type of proteins that uses energy to move substances
- Transport against the concentration gradient
- Different concentrations between intracellular and extracellular fluids are maintained through pumps.

Table 7.1 Approximate concentrations of ions in intracellular and extracellular fluids

| Ion ${ }^{\boldsymbol{a}}$ | Intracellular concn (mM) | Interstitial concn (mM) |
| :--- | :---: | :---: |
| Sodium $\left(\mathrm{Na}^{+}\right)$ | 10 | 145 |
| Potassium $\left(\mathrm{K}^{+}\right)$ | 150 | 5 |
| Calcium $\left(\mathrm{Ca}^{2+}\right)$ | 0 | 3 |
| Chloride $\left(\mathrm{C}^{-}\right)$ | 5 | 110 |

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## Pump

- Two different energy sources
- ATP
-- e.g. Pumping Na+ and $\mathrm{K}^{+}$against their gradients using ATP
( $\mathrm{Na}^{+} / \mathrm{K}^{+}$ATPase)
- Energy inherent in gradient



## Transport Proteins in Animals

## - Nerve Impulses

(a) Multipolar interneuron

(b) Motor neuron


## Nerve Impulses

- Key players: $\mathrm{Na}^{+}, \mathrm{K}^{+}$
- Resting membrane potential of -70 mV
- (Some K+ channels are open.)
- Opening of $\mathrm{Na}^{+}$channel by stimulation
- Generation of action potential 50 mV
(a) Resting state (cytosolic face negative)

Exterior

(b) Depolarized state (cytosolic face positive)


- Opening of voltage-gated $\mathrm{K}^{+}$channel
- Repolarization of membrane potential
- Restoration of membrane potential by $\mathrm{Na}^{+} / \mathrm{K}^{+}$ATPase


## Muscle Contraction

- Key player: $\mathrm{Ca}^{2+}$
- Inside muscle cells, $\mathrm{Ca}^{2+}$ are packed into a membrane-bound compartment called the sarcoplasmic reticulum (SR).
- When the nerve impulse (powered by $\mathrm{Na}^{+} / \mathrm{K}^{+}$ gradients) reaches the muscle cells, it triggers $\mathrm{Ca}^{2+}$ channel in the SR to open.
- Opening of $\mathrm{Ca}^{2+}$ channel in SR
$\rightarrow$ Release of $\mathrm{Ca}^{2+}$
$\rightarrow$ Released $\mathrm{Ca}^{2+}$ binding to troponin (protein)
$\rightarrow$ Muscle contraction


## When Gradients Fail

Electrocardiogram (ECG)

- Long QT (LQT) syndrome
- Long recovery periods before new heart contraction
- Cell to cell variation of recovery periods
- Can cause arrhythmia (lack of rhythm)
- Defects in $\mathrm{K}^{+}$or $\mathrm{Na}^{+}$channels



## When Gradients Fail

## - Inherited heart failure

- Mutation in the regulatory protein of $\mathrm{Ca}^{2+}$ channel in SR (The pump can not transport $\mathrm{Ca}^{2+}$ back into SR)

Regulatory protein inhibits the calcium pump.


Phosphorylated regulatory protein allows pump to operate.


Mutant regulatory protein cannot be phosphorylated; calcium pumping is blocked.


## Pumps, Carriers, and Nutrient Distribution

- Nutrient components must move from the intestine to the blood stream through intestinal epithelium.
- Epithelium
- The body's version of a cell's membrane
- Epithelial cells
- Cells cover body surfaces and line internal organs
- Intestinal epithelium
- Cells lining the digestive tract


## Intestinal Epithelium

- Microvilli
- Facing the intestinal track
- Enzymes and transport proteins are located.
- The enzymes break down complex sugars into simple sugars.

$$
\text { lactose, sucrose } \rightarrow \text { glucose }
$$

- Tight Junction between Cells
- Preventing transport of large molecules
- Barrier between the intestinal contents and the interstitial fluid
- Extracellular Matrix
- Supporting epithelial cells
- Tough network of extracellular proteins and carbohydrates


Matrix and interstitial fluid

## Transport of Nutrients across Epithelial Cells



## Transport of Nutrients across Epithelial Cells

- Intestinal side
- active transport of glucose powered by $\mathrm{Na}^{+}$gradient
- co-transport of two $\mathrm{Na}^{+}$ and one glucose molecule
- Interstitial fluid side
- glucose
-- by carrier proteins
- $\mathrm{Na}^{+}$
-- by Na+/K+ ATPase
- Capillary wall
- Glucose diffusion
- Designed to let all but the large molecules (e.g. blood proteins) cross over



## Cells, Salts, and Water Balance

- Movement of water across the cell
- Water movement to equalize the total concentration of solutes
- Osmosis: movement of water across membranes
- Osmotic balance: no net water movement


Cell is in osmotic balance with extracellular fluid.


Cell is higher in solutes than is extracellular fluid Water enters the cell.


Extracellular fluid is higher in solutes than higher in solutes
is the cell. Water is the cell. Water
leaves the cell.


- Cells in osmotic balance
- Cells contain many proteins, amino acids, and other small molecules.
- Concentration of total ions is higher outside than inside cells to keep the osmotic balance.


## Water follows salt

- Water in human body (75 kg man)
- 45 L of water
- 30 L : intracellular
- 3.75 L: blood plasma | Almost same solute (salt), since blood
- 11.25 L : extracellular fluid capillaries are permeable to small molecules
- Water balance
- Lactose intolerance
- Lack of lactase breaking lactose into glucose and galactose
- No digestion of lactose $\rightarrow$ movement of water into the intestine
- Metabolize of lactose by intestinal bacteria $\rightarrow$ gas production
- High-magnesium laxative : relieving constipation
- Cystic fibrosis (by impaired salt transport)
- Mutation in $\mathrm{Cl}^{-}$channel : reduced water secretion $\rightarrow$ thick mucus in epithelia of respiratory and gastrointestinal tracts


## Biotechnology

- Rehydration therapy
- Diarrhea: kill 2 million children/year by dehydration
- Solution of sugar and salt is effective to treat dehydration: e.g. sports drinks
- Enzyme treatments
- Lactose intolerance
- Add lactase ( $\beta$-galactosidase) in milk or dairy products
- When you eat a bean-rich meal in a Mexican restaurant
- Beans contain galactose-containing sugars (galactosides)
- Humans lack enzymes for breaking down galactosides.
- Microbial munching on galactosides $\rightarrow$ gas production
- Buy $\alpha$-galactosidase (Beano) in a drugstore


[^0]:    ${ }^{a}$ The most abundant ions in interstitial fluid are sodium and chloride ions, which are the components of table salt.

