# Intro. to Electro-physics

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# Syllabus (1/2)

Lecture:	Introduction to Electro-physics (430.435) * <i>Prerequisite</i> : Semiconductor devices (430.312), (
Staff:	<ul> <li>Instructor: Jaesang Lee (email: jsanlgee@snu.ac.</li> <li>TAs: Heechan Lim (email: degris@snu.ac.kr), Hye</li> </ul>
Textbook:	S. H. Simon, "The Oxford Solid State Basics", Oxf + Ashcroft & Mermin, "Solid State Physics", Ce + C. Kittel, "Introduction to Solid State Physics"
Homework:	<ul> <li>Total 6 sets</li> <li>A problem set (HW) will be assigned approxima</li> <li>Submission due date: <i>a week after the assign</i></li> <li>Scan and upload it on eTL until 11:59 PM</li> <li>There will be a penalty for late submission!</li> </ul>
Grading Policy:	Attendance (10 %) Homework (42 %) Midterm (24 %): "Presentation on achievements b Final (24 %): "Presentation on free topics related s

Quantum mechanics (430.326), Electromagnetics (430.203A)

c.kr)

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ford University Press, 2019 (Reprint)

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ately **every two weeks**.

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by Nobel laureates/key contributors in solid-state physics" solid-state physics



## Syllabus (2/2)

### **Class schedule 1. Intro to Electro-physics (w1)**

### 2. The study of metals

- The classical theory (Drude model) (w2)
- The quantum theory (Sommerfeld model) (w3~4)

### 3. Chemical bonding [w5]

Ionic / covalent / van der Waals / Metallic / Hydrogen bonding

### 4. Crystal vibrations [w6~7]

• Monatomic / diatomic chain

- 6. Crystal structure [w9]
- 7. Reciprocal lattice and X-ray diffraction (XRD) [w10]
- 8. Electrons in periodic potential [w11]
- 9. Insulator, Semiconductor and metal [w12]
- **10. Semiconductor Physics and devices [w13]**
- **11. Magnetic properties of solids [w14]**

[Final: "Presentation on free topics related solid-state physics"]

[Midterm: "Presentation on achievements by Nobel laureates/key contributors in solid-state physics"]



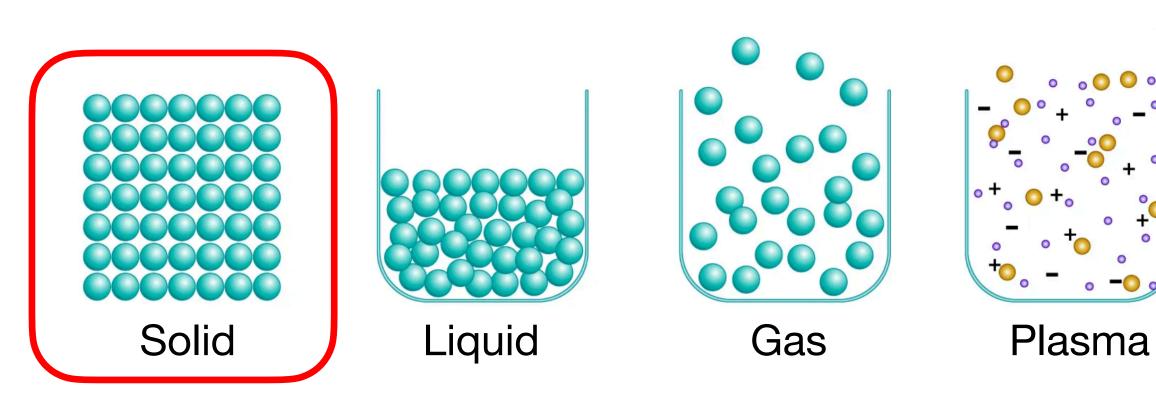
### **Course definition**

- What is electro-physics?

  - Useful materials studied in ECE
    - : Semiconductors, metals, insulators, magnets, superconductors, ...
- Electro-physics ~ Solid-state physics!
- What is solid-state physics?
  - The study of matter in its solid state
  - The largest sub-field of "Condensedmatter physics"

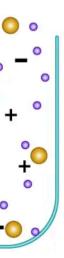
### - The study of physics of electrical phenomena observed in practically useful materials







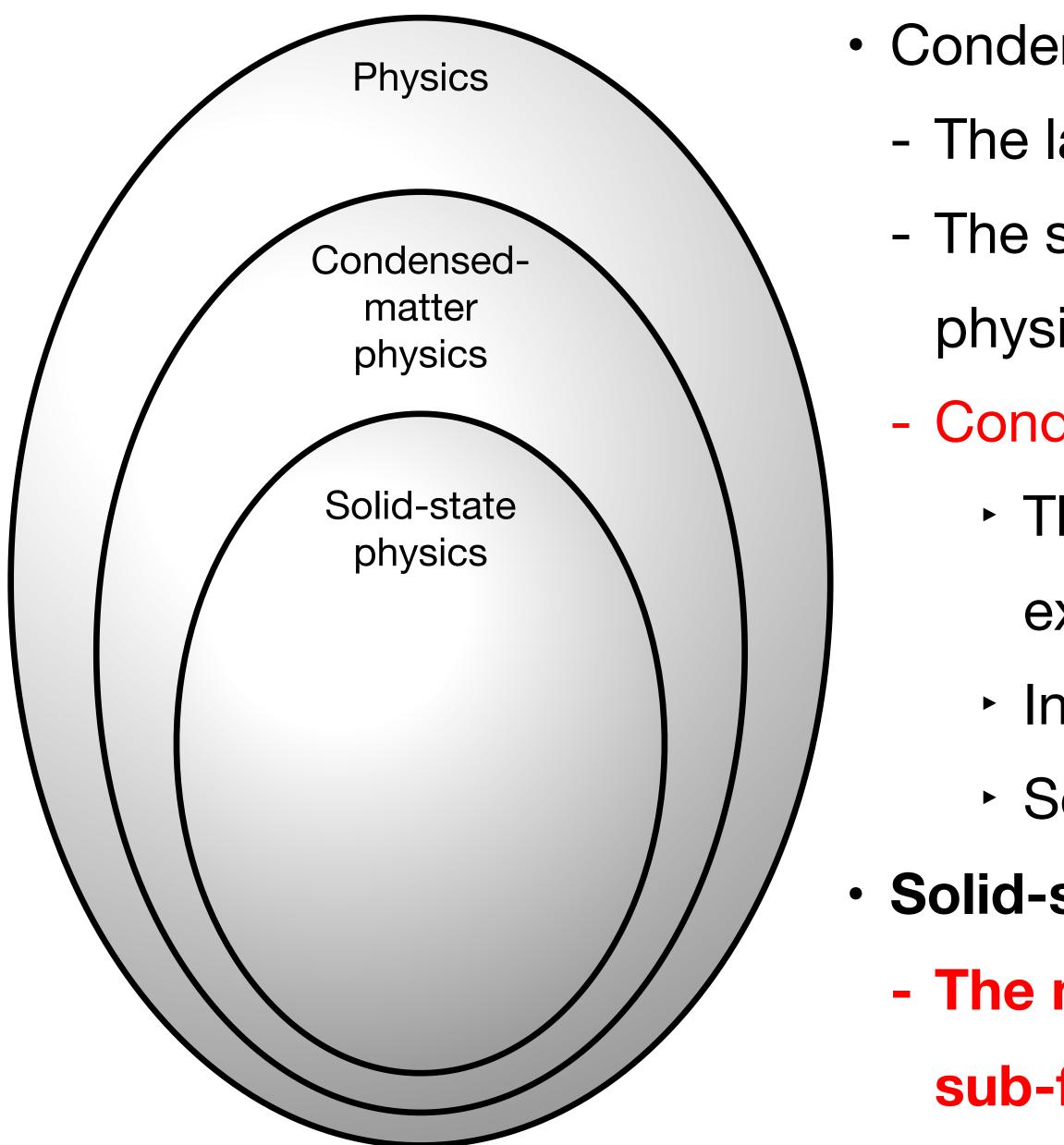








### **Condensed-matter, solid-state physics?**



- Condensed-matter physics
  - The largest sub-field of physics (>60%)
  - The study that deals with macro- and microscopic
    - physical properties of matter in condensed phase
  - Condensed phase
    - The number of constituents in a system is
      - extremely large
    - Interactions among them are very strong
    - Solids and liquids
- Solid-state physics
  - The most successful and technologically useful
    - sub-field of condensed matter physics!





### **CMP at SNU Physics**

- Faculty members in SNU physics
  - # of professors studying condensed matter phys.:

 $24/38 \simeq 63\%!$ 

- Research topics

New materials, emerging phenomena, correlation and topological effects, <u>compound semiconductors</u>, high-T superconductors, surfaces, interfaces, synthetic metals, transition and rare-earth oxides, quantum dots & wells, magnetism, spintronics and low-T properties …















Hong, Seunghui



Kim, Changyoung

Kim, Dohu



Lee, Takhee



Noh, Tae Won





Park, Je-Geun





Yi, Gyu-Chu

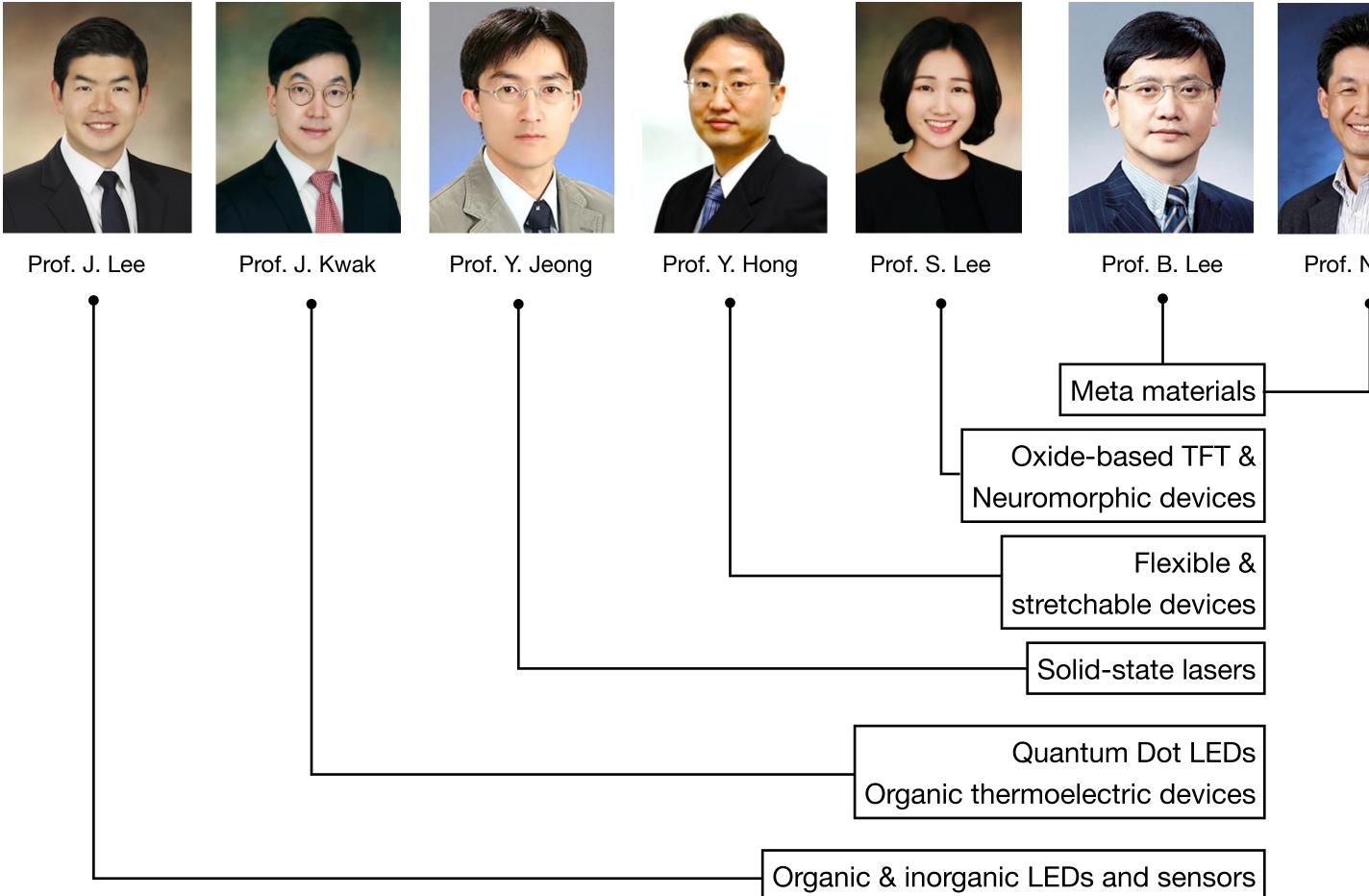






### **Solid-state devices at SNU ECE**

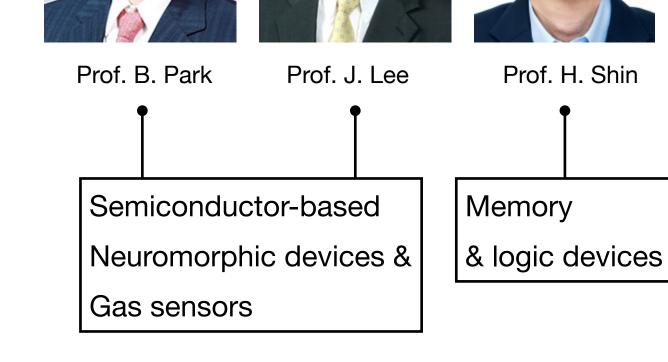
ECE faculty members working on solid-state devices & applications

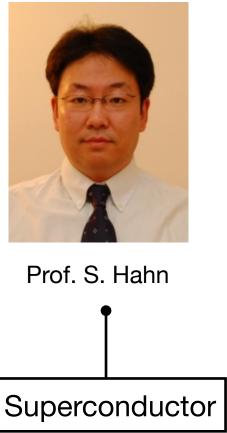


• What we do: based on understanding of electrical, optical, chemical properties of solid-state matters, we make technologically useful devices and applications!



Prof. N. Park







# what is solid-state physics? (1/2)

- Solid-state physics: the study of rigid matters (i.e., solids)
  - It studies how the macroscopic properties of solids result from their atomic-scale properties
  - Macroscopic properties: electrical, mechanical, thermal, optical, magnetic, ...
  - Solid: "densely-packed" atoms that strongly interact altogether
  - Atomic-scale properties
    - Types of atoms that comprise a solid
    - Interaction forces (i.e., how they are bound to form a solid)
    - Arrangement pattern for the atoms (crystalline vs. amorphous)
    - ► etc.

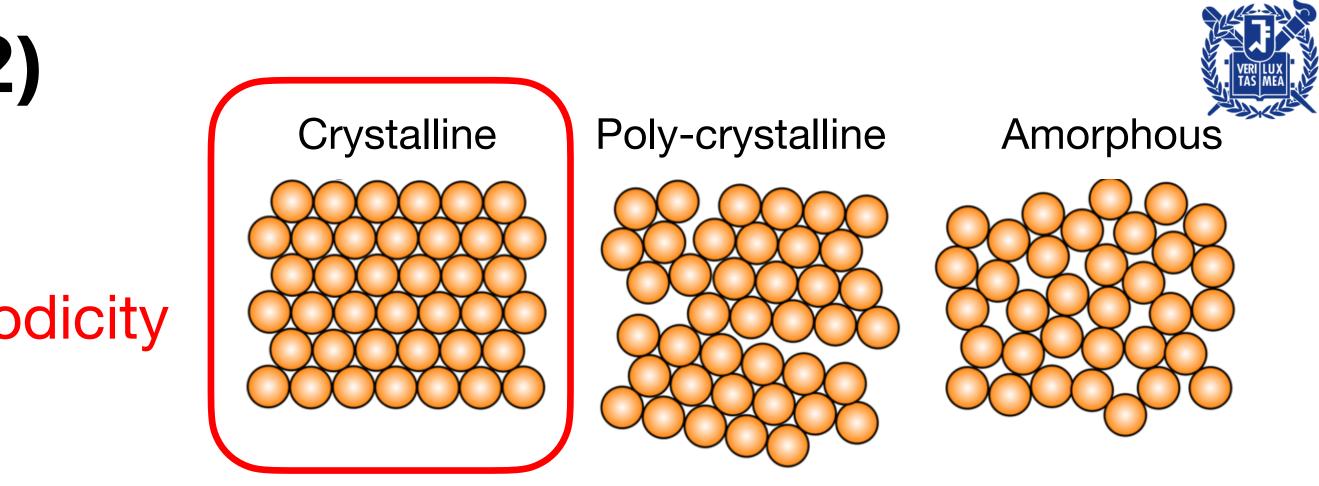


### what is solid-state physics? (2/2)

- Major focus: crystalline solids
  - Determining characteristics the periodicity of the comprising atoms
  - Easy to mathematically model their macroscopic properties
  - **Solids** (But, not always!)
- Interacting force between atoms

(how the atoms are held together to form solids)

- lonic bond (NaCl, LiF, …)
- Covalent bond (C, Si, GaAs, …)
- Metallic bond
- van der Waals bond (molecular solids)



- Knowledge derived from the crystalline solids "loosely" applicable to amorphous

## <Topic 1> Classical theory for metals

- Major interest of early solid-state physicists: Explaining electrical & heat conduction of a metal
- Earliest model by Paul Drude in 1900
  - Application of a kinetic theory of gases to electrons ( $\rightarrow$  "electron gas")
    - No electron-electron interaction ("Independent electron approx.")
    - No electron-ion interaction ("Free-electron approx.")
    - Electrons collide elastically with immobile ions and get scattered
    - Electrons follow Maxwell-Boltzmann distribution
  - (Roughly) explains electrical ( $\sigma$ ) and thermal ( $\kappa$ ) conductivities, Hall effects
  - Fails to explain electronic heat capacity\*

(\* The amount of heat to be supplied to a matter to produce a unit change in its temperature [J/K])

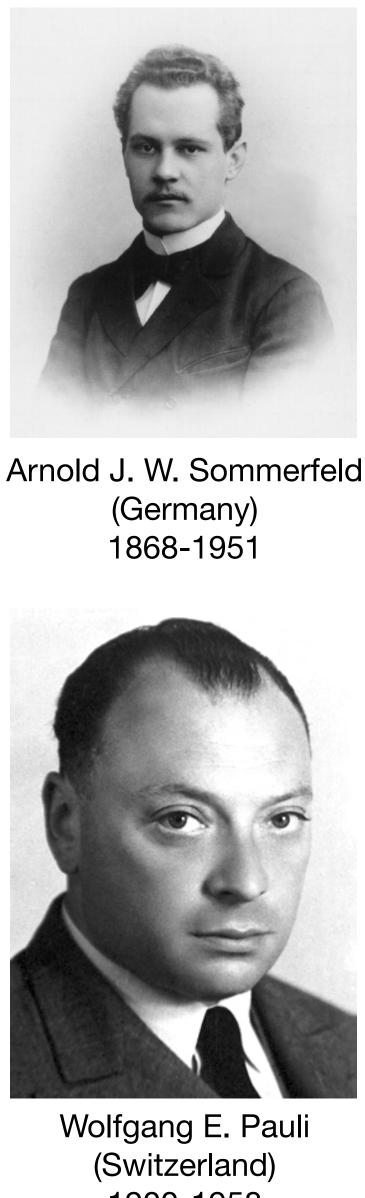


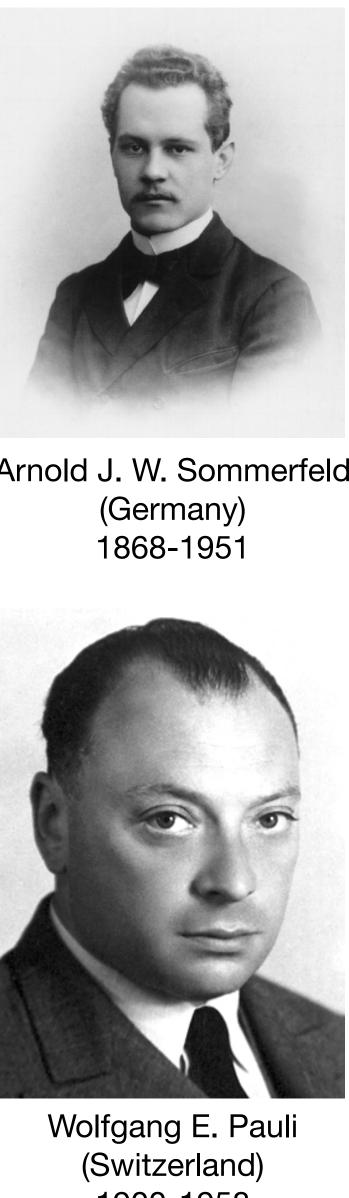
Paul K. L. Drude (Germany) 1863-1906

### <Topic 2> Semi-classical theory for metals

- Improved model by Arnold Sommerfeld
  - Drude model + a bit of quantum mechanics
    - Free-electron approximation (same)
    - Immobile positive ions act as scattering centers (same)
    - A gas of non-interacting electrons follows Fermi-Dirac distribution, **NOT** Maxwell-Boltzmann distribution (:: Electrons = Fermions that obey the Pauli Exclusion principle!)

- Able to explain the electronic heat capacity
- Fails to explain the existence of insulators and semiconductors





1900-1958

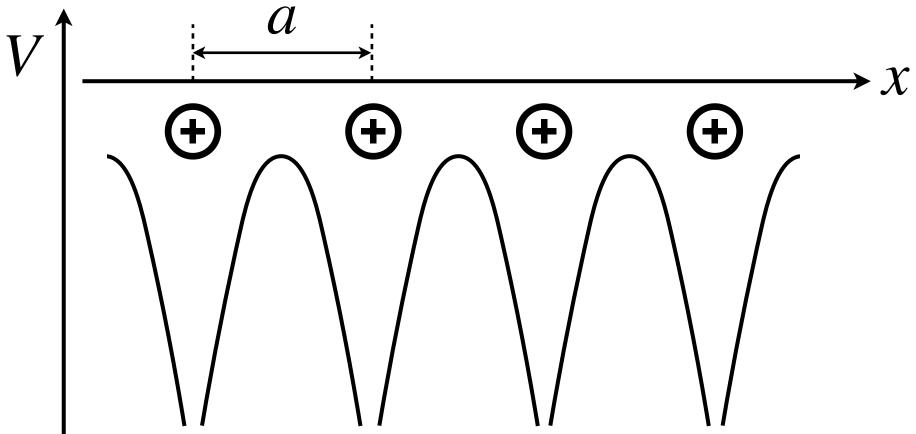
# <Topic 3> Quantum theory for solids (1/2)

- "Nearly" free electron model
  - In a crystalline solid, electrons move almost freely but subject to a
    - weak, periodic potential due to the periodically arranged ions
  - Quantum mechanics plays a role!
    - Schrödinger's equation for the periodic potential:  $-\frac{\hbar^2}{2m}\frac{d^2\psi(x)}{dx} + V(x)\psi(x) = E\psi(x), \text{ where } V(x)$
    - Solution  $\psi(x)$ : "Bloch's state"



Erwin Schrödiı (Austria) 1887-1961

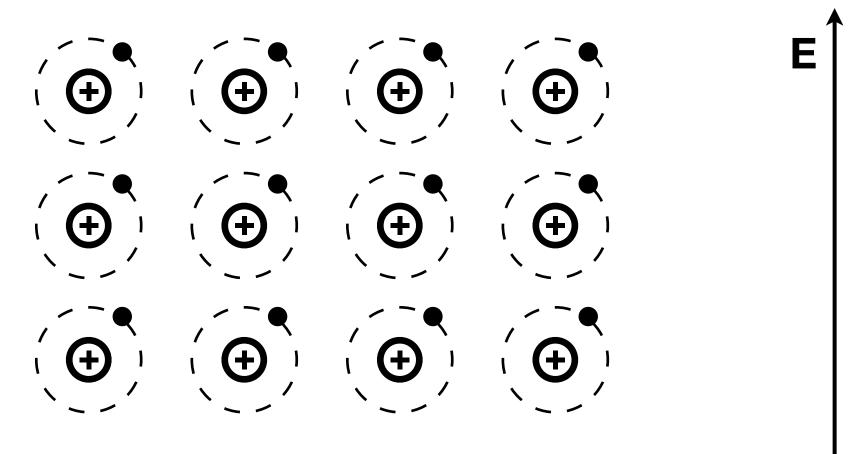
(x), where 
$$V(x + na) = V(x)$$



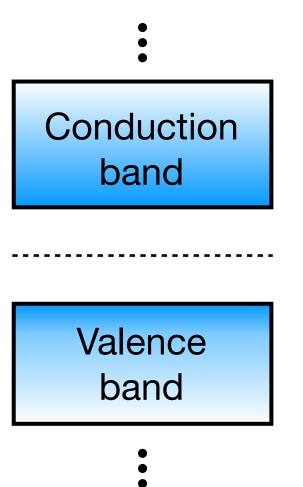
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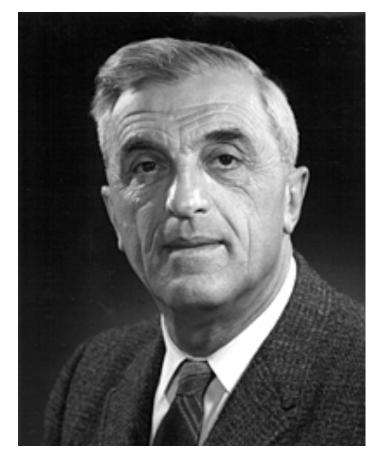
# <Topic 3> Quantum theory for solids (2/2)

- Bloch's state
  - Electrons are largely delocalized across the crystalline solid
  - Electrons occupy the closely-spaced electronic states ("Energy band")



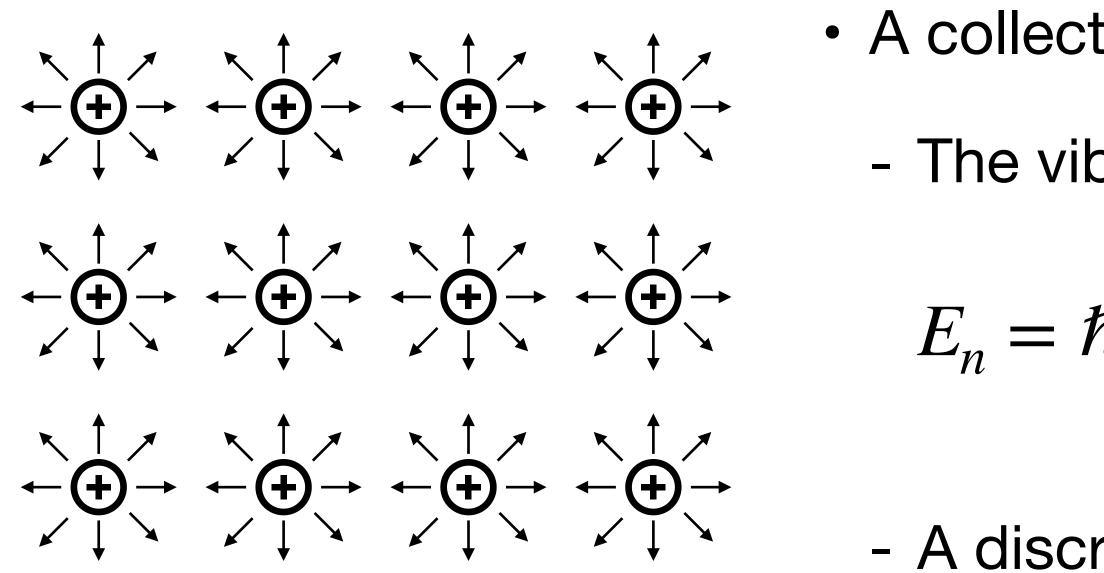
- Band theory
  - Enables to distinguish conductors, insulators and semiconductors
  - Forms a foundation of the understanding of solid-state devices (transistors, solar cells, LEDs, lasers, and etc)!





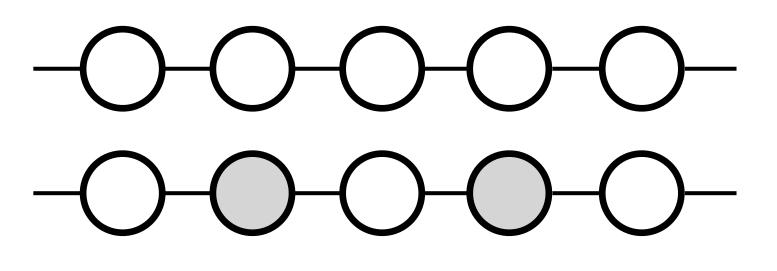
Felix Bloch (Switzerland) 1905-1983

### <Topic 4> Crystal vibrations in solids



c.f.) A discrete quantum of electromagnetic vibration: "photon"

- Phonon affects the electrical, thermal conductivities and many other properties!
- What we will study: A simple 1D model
  - Monatomic chain
  - Diatomic chain



- A collective vibration of "periodically arranged" atoms
  - The vibration at a given frequency  $(\omega)$  is quantized:

$$\hbar\omega\left(n+\frac{1}{2}\right), \ (n=0,1,2,\ \cdots)$$

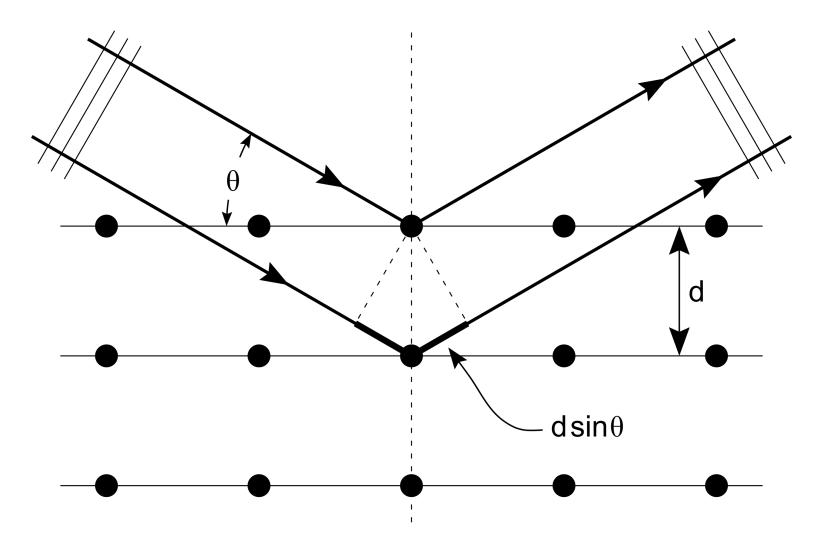
- A discrete quantum of crystal vibration: "phonon"

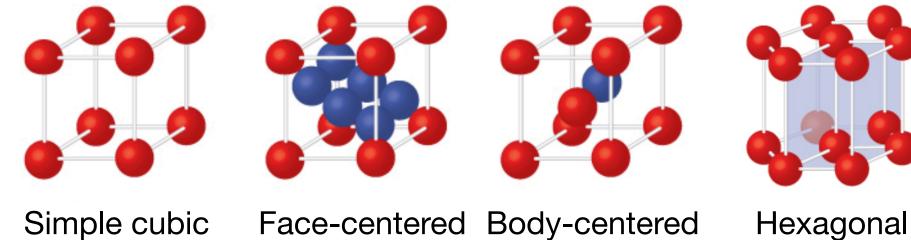




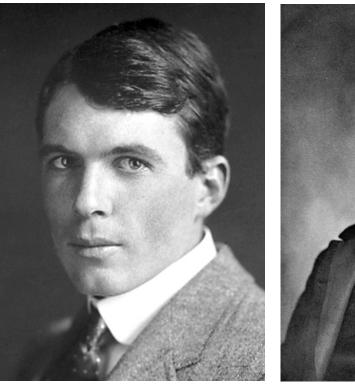
### <Topic 5> Crystal structure

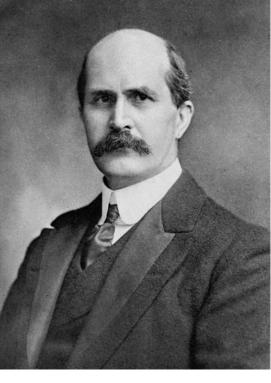
- Important terminology in crystals
  - Lattice: A periodic arrangement of points in space
  - Basis: A component that fills the lattice (e.g. atom, ion, molecule,  $\cdots$ )
- A method of identifying the crystal structure : X-Ray Diffraction (XRD)
- Bragg's law :  $n\lambda = 2d\sin\theta$ , where  $n = 1, 2, 3, \cdots$





**14 Different lattice structures** (Not all shown here; opentextbc.ca)





Lawrence Bragg (Australia) 1890-1915 Son

William Bragg (United Kingdom) 1862-1942 Father







### ...and further topics

<Topic 6> Chemical bonding

<Topic 7> Basics of semiconductor physics and devices

<Topic 8> Magnetic properties of solids

< Topic 9> Superconductor and its applications (if time allows...)

