

# Structure of polymers

Ch 10 sl 23

- chemical structure
  - atomic structure (of repeat unit)
  - isomers ~ configuration
    - cis-trans isomer
      - 1,2-, cis-1,4-, trans-1,4-PBD
    - tacticity Chapt 15
      - isotactic/syndiotactic/atactic
- skeletal structure
  - linear
  - branched
  - crosslinked

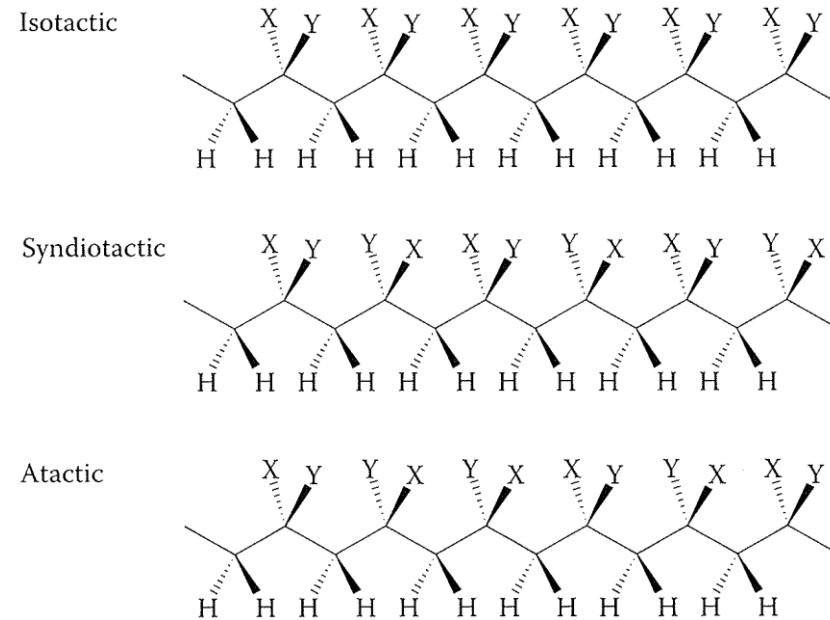
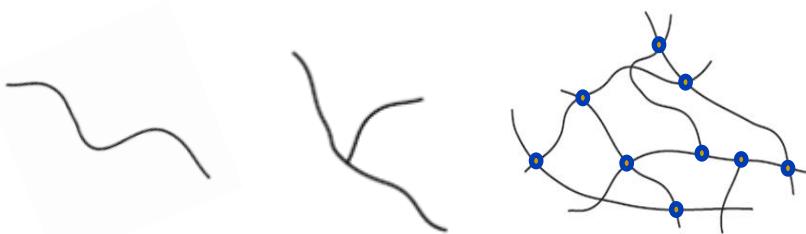


Fig 6.1 p148

□ physical structure

□ single-chain structure

■ conformation

→ size and shape of single chain



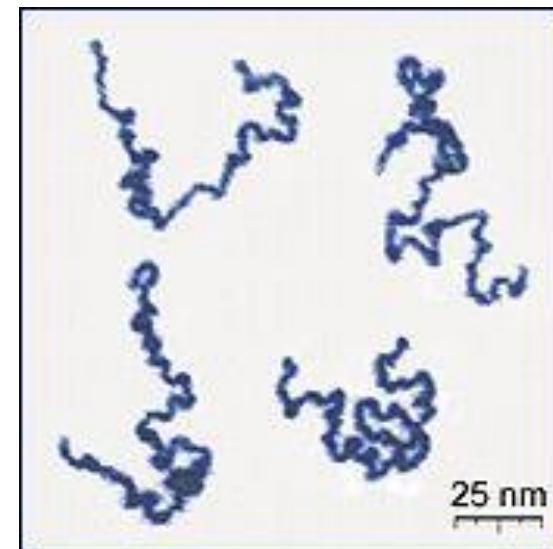
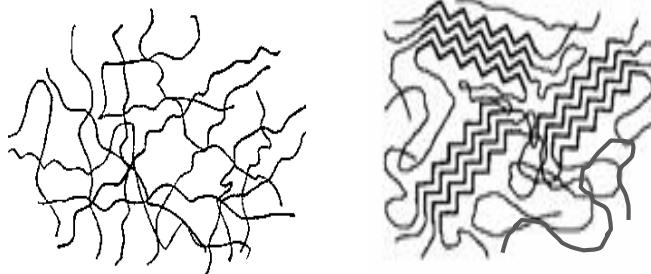
□ aggregation structure

■ amorphous state Chapt 16

← unperturbed-state conformation

■ semicrystalline state Chapt 17

← preferred conformation



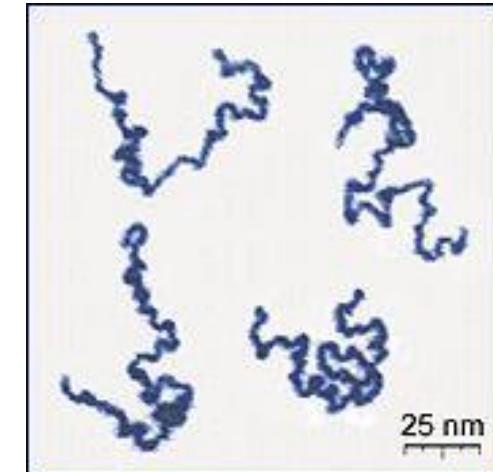
can be  
a chain in different times,  
a chain at different Temp's,  
different chains in the bulk,  
or  
different polymers  
(different chain stiffness)

# Modeling a polymer chain

Ch 10 sl 25

- chain dimension  $\leftarrow$  stiffness  $\leftarrow$  structure
  - comparison of modeling and experiments

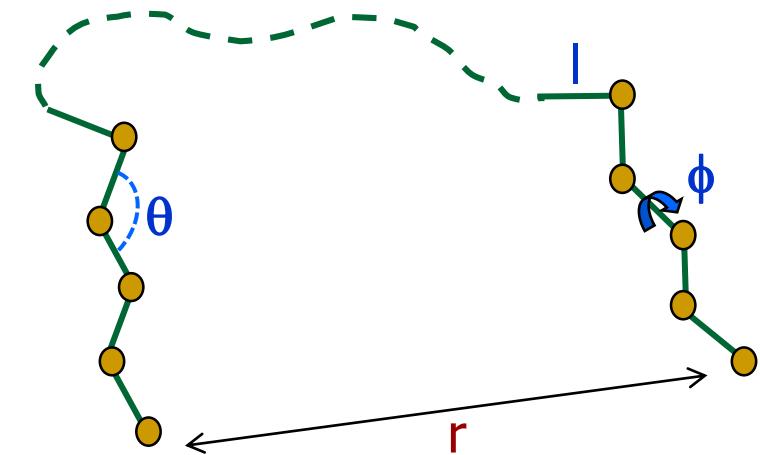
- a polymer chain with
  - $n$  bonds of length  $l$



- $n-1$  bond angles  $\theta$
- $n-2$  rotational angles  $\phi$
- end-to-end distance  $r$  = a measure of dimension
  - contour length  $nl$  does not reflect structure

- modeling

- freely jointed chain  $\sim l$  only fixed
- freely rotating chain  $\sim l$  and  $\theta$  fixed
- chain with hindered rotation
  - $\sim l, \theta$  fixed;  $\phi$  restricted



# Freely jointed chain

Ch 10 sl 26

- bond length only fixed
- random-flight analysis ( $\leftarrow$  random-walk)
  - probability finding the end at  $(x, y, z)$   $r$  away from O

$$W(x, y, z) = \left( \frac{\beta}{\pi^{1/2}} \right)^3 \exp[-\beta^2(x^2 + y^2 + z^2)] = \left( \frac{\beta}{\pi^{1/2}} \right)^3 \exp[-\beta^2 r^2]$$
$$\beta = [3/(2nl^2)]^{1/2}$$

- a **Gaussian** distribution function  $\rightarrow$  'Gaussian chain'  
'random coil'

- probability finding the end at distance  $r$

$$W(r)dr = W(x, y, z) \times 4\pi r^2 dr = 4\pi \left( \frac{\beta}{\pi^{1/2}} \right)^3 r^2 \exp(-\beta^2 r^2)$$

- max not at 0, but at  $r = (2/3)^{1/2} n^{1/2} l = 0.82 \langle r^2 \rangle^{1/2}$

Fig 10.3

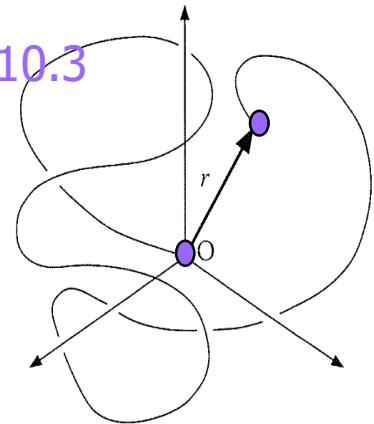
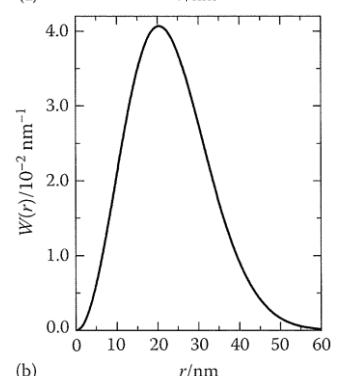
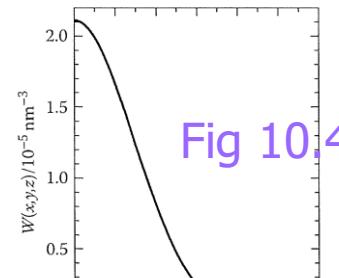


Fig 10.4



## □ random-flight analysis (cont'd)

□  $\langle r^2 \rangle = \int_0^\infty r^2 W(r) dr = nl^2$

< > average (over time or chains)

□  $\langle r^2 \rangle_f^{1/2} = n^{1/2}l$

■ root-mean-square (RMS) end-to-end distance

■  $n^{1/2}$  smaller than contour length [if  $n = 10^4$ , 1/100 smaller]

## □ vector analysis

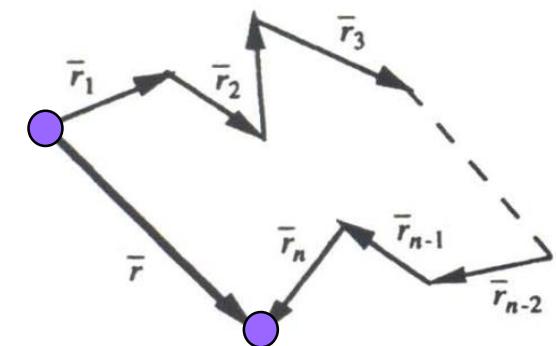
□ e-to-e vector  $\bar{r} = \sum_{i=1}^n \bar{r}_i$

□ scalar magnitude of  $\check{r}$ ,  $r = (\check{r} \cdot \check{r})^{1/2}$

□  $r^2 = \sum_{i=1}^n \bar{r}_i \cdot \sum_{j=1}^n \bar{r}_j = \sum_{i=1}^n \bar{r}_i^2 + 2 \sum_{i=1}^{n-1} \sum_{j=i+1}^n \bar{r}_i \bar{r}_j$

□  $\langle r^2 \rangle = nl^2 + 2 \sum_{i=1}^{n-1} \sum_{j=i+1}^n \langle \bar{r}_i \bar{r}_j \rangle = n l^2$

$\rightarrow 0 \leftarrow$  random disposition of bond vectors



# Freely rotating chain

Ch 10 sl 28

- bond length and bond angle fixed

- 'valence angle model'
  - reduced artificiality from FJC

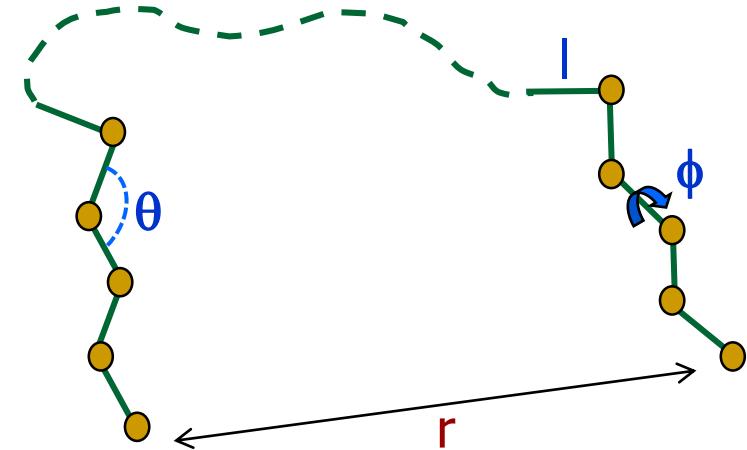
- chain dimension

$$\langle r^2 \rangle_{fa} = nl^2 \left( \frac{1 - \cos \theta}{1 + \cos \theta} \right)$$

- FRC with  $sp^3$  atoms [ $\theta=110^\circ$ ]

$$\langle r^2 \rangle_{fa} = 2nl^2$$

- FRC is  $2^{1/2}$  times larger than FJC.



For PE with 10000 bonds,  
 $n = 10^4$ ,  $l = .15 \text{ nm}$   
contour length  $nl = 1500 \text{ nm}$   
or  $nl \cos 35^\circ = 1230 \text{ nm}$   
 $r_{FJC} = 15 \text{ nm}$   
 $r_{FRC} = 21 \text{ nm}$

# Chain with restricted rotation

Ch 10 sl 29

- I and  $\theta$  fixed, and  
not all rotational angles allowed
  - only some **discrete states** allowed.
  - **rotational isomeric state [RIS]** model
- for butane

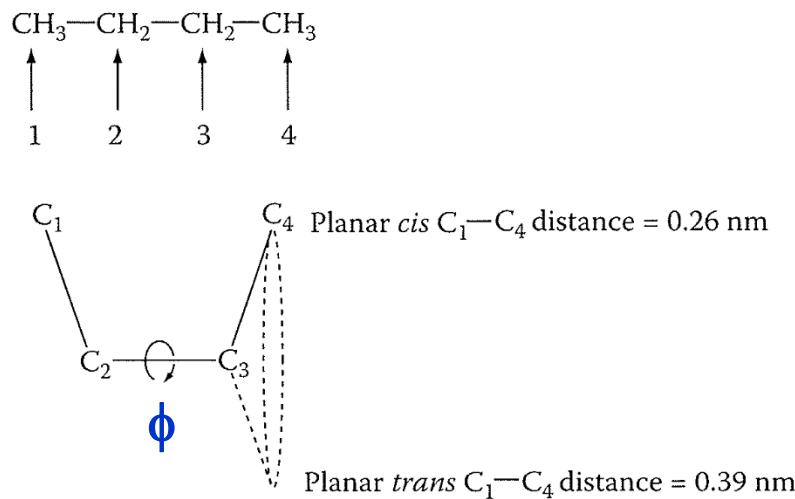
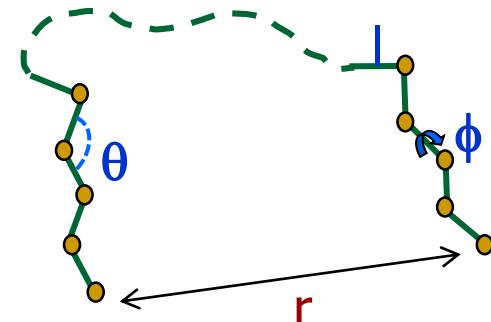
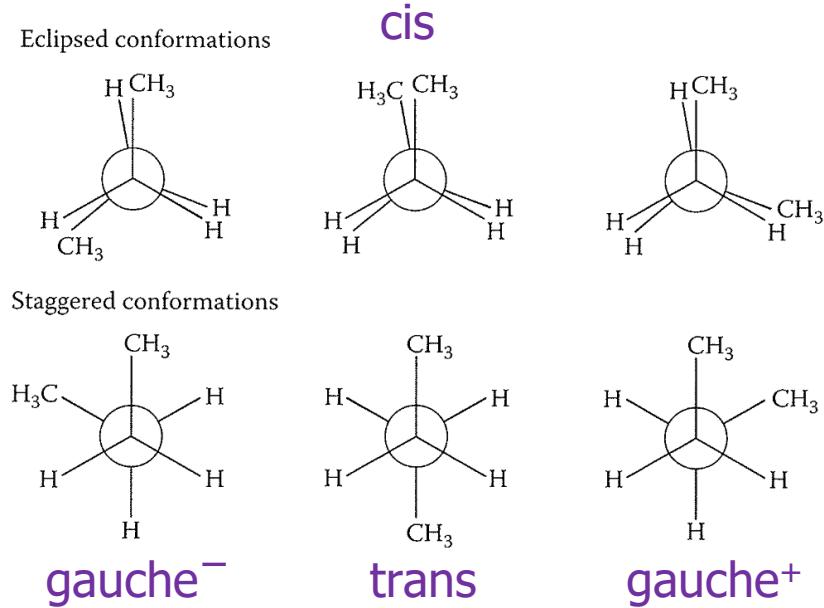


Fig 10.5



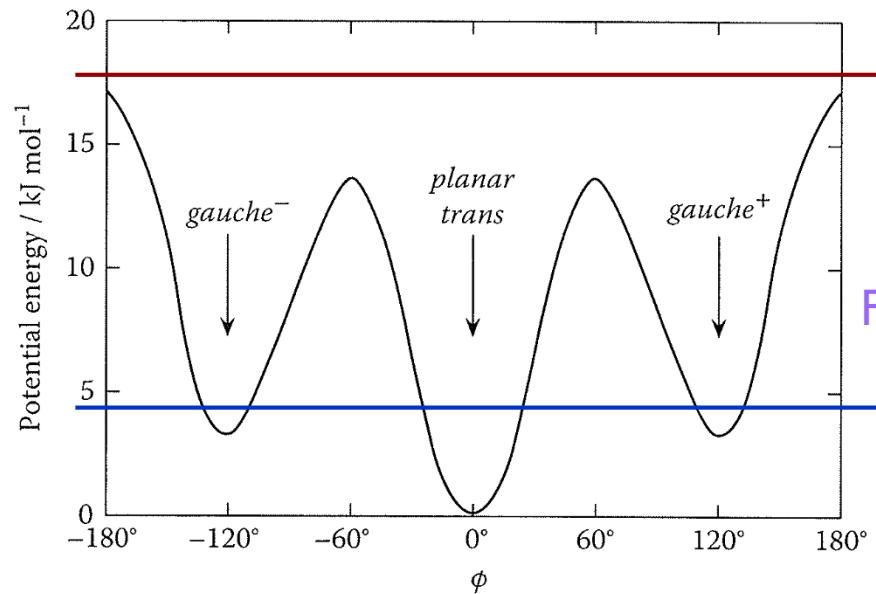
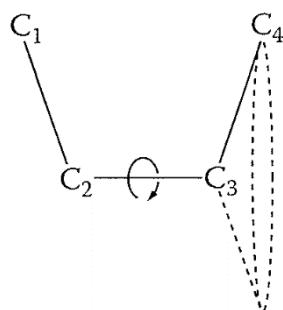
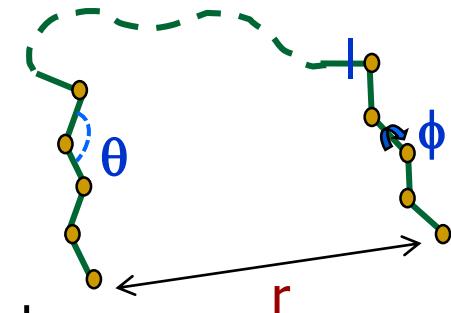


Fig 10.6

- at **high temp**, all  $\phi$  allowed  $\rightarrow$  FRC
- at **lower temp**, only  $-120^\circ$ ,  $0^\circ$ , and  $120^\circ$  allowed
  - due to **first-order [3-bond] interaction**
  - $g^-$ ,  $t$ , and  $g^+$  are the three discrete RIS's
- for polymer [PE] chain, 3 RIS's for each of  $n-2$  bonds



## □ Dimension of chain with 1st-order interaction

$$\langle r^2 \rangle_0 = nl^2 \left( \frac{1 - \cos \theta}{1 + \cos \theta} \right) \left( \frac{1 + \overline{\cos \phi}}{1 - \cos \phi} \right)$$

$$\overline{\cos \phi} = \frac{u(t) \cos 0^\circ + u(g^+) \cos (120^\circ) + u(g^-) \cos (-120^\circ)}{u(t) + u(g^+) + u(g^-)}$$

$$u(\phi) = \exp [-E(\phi)/kT]$$

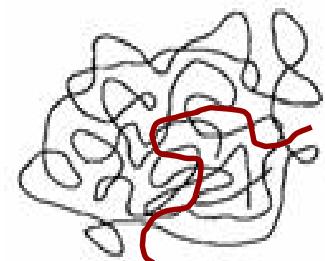
### □ for PE at 400 K

- $u(t) = 1, u(g) = .5 \rightarrow \overline{\cos \phi} = .25 \rightarrow \left( \frac{1 - \cos \theta}{1 + \cos \theta} \right) \left( \frac{1 + \overline{\cos \phi}}{1 - \overline{\cos \phi}} \right) = 3.4$

- experiments (light scattering, neutron scattering)  $\rightarrow 6.8$

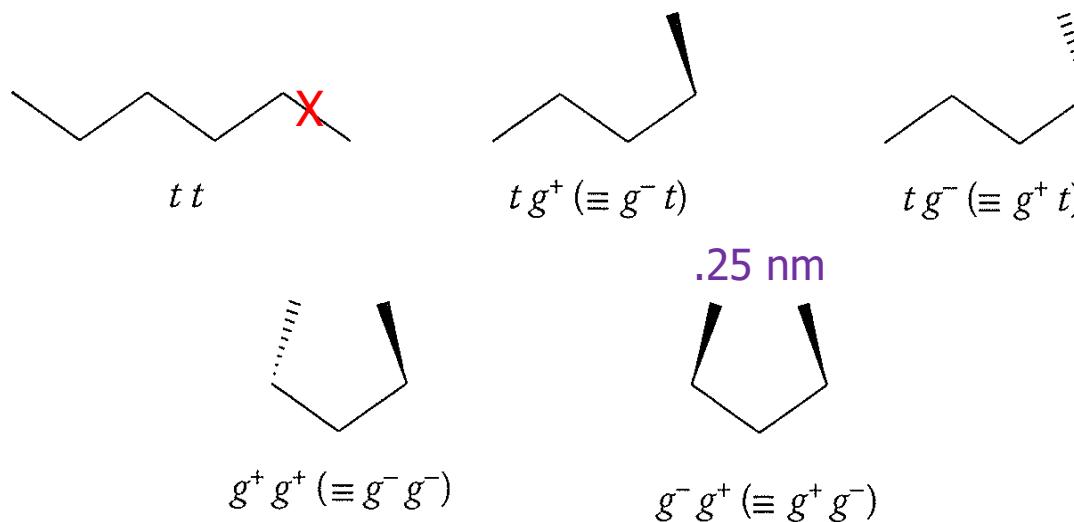
- for  $n = 10^4$ ,  $r_{\text{measured}} = 55 \text{ nm}$  not  $27 \text{ nm}$

- why the difference?



## □ 2nd-order [4-bond] interaction

- for pentane, 2-bond rotation
- 9 RIS's
- $g^-g^+$  and  $g^+g^-$  are not allowed (due to very high energy)
- pentane interaction, 'short-range interaction'
- Bond rotation should be considered pairwise.



## □ RIS calculation Nobel prize 1974

- taking pairwise potential
  - $u(\phi_i, \phi_{i+1})$  not  $u(\phi)$
- tensor transformation and calculation

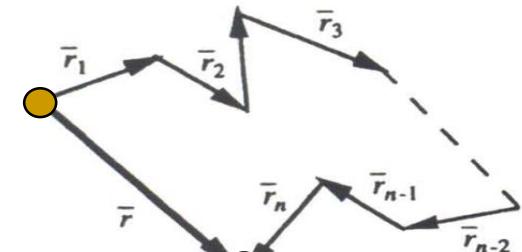
$$\square \langle r^2 \rangle_0 = \sigma^2 nl^2 \left( \frac{1 - \cos \theta}{1 + \cos \theta} \right) = \sigma^2 \langle r^2 \rangle_{\text{FRC}}$$

■  $\sigma$  = steric parameter  $\sim$  rarely used

$$\square \langle r^2 \rangle_0 = C_\infty nl^2 \quad C_\infty = \frac{\langle r^2 \rangle_0}{nl^2} = \langle r^2 \rangle / \langle r^2 \rangle_{\text{FJC}}$$

■  $C_\infty$  = characteristic ratio  $\sim$  a measure of chain stiffness  $\leftarrow$  structure

$$\langle r^2 \rangle = nl^2 + 2 \sum_{i=1}^{n-1} \sum_{j=i+1}^n \langle \bar{r}_i \bar{r}_j \rangle$$



Polymer	Temperature / °C	$\sigma$	$C_\infty$	
Polyethylene	140	1.8	6.8	$C_\infty = 5 - 10$
Isotactic polypropylene	140	1.6	5.2	$C_\infty^{1/2} = 2 - 3$
Poly(vinyl chloride)	25	1.8	6.7	
Polystyrene	25	2.3	10.8	
Polystyrene	70	2.1	9.2	Table 10.3

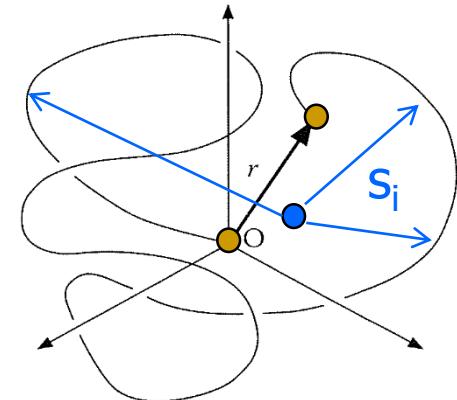
# Quantities characterizing chain

Ch 10 sl 34

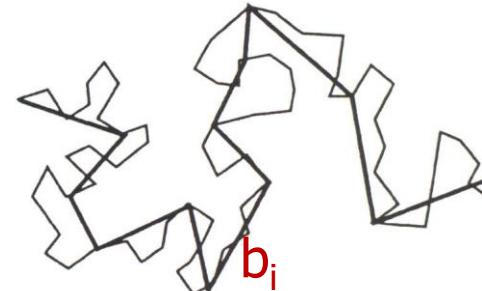
## □ dimension

- RMS end-to-end distance,  $\langle r^2 \rangle^{1/2}$
- RMS radius of gyration,  $\langle s^2 \rangle^{1/2}$ 
  - RMS distance from center of mass

$$\langle r^2 \rangle^{1/2} = \frac{\langle s^2 \rangle^{1/2}}{6^{1/2}}$$

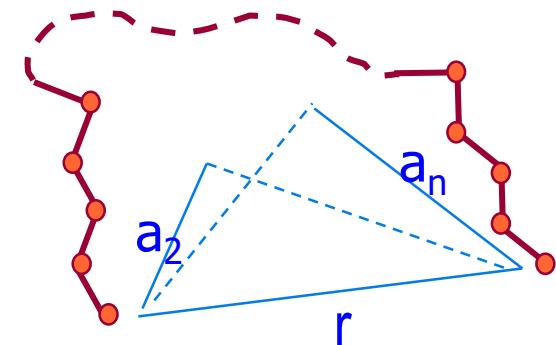


- contour length
  - $nl$  or  $nl \sin(\theta/2)$



## □ stiffness

- characteristic ratio,  $C_\infty$
- Kuhn length,  $b = C_\infty l$  or  $C_\infty l / \sin(\theta/2)$ 
  - Kuhn chain = statistically equivalent FJC
- persistence length,  $a = [(C_\infty + 1)/2] l$



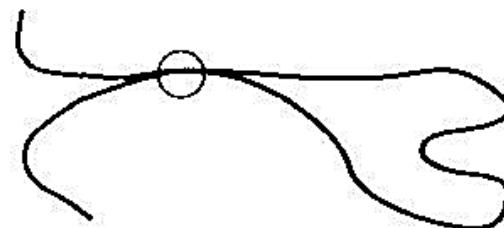
# Chain with excluded volume

Ch 10 sl 35

- intrachain volume exclusion

- = 'long-range interaction'

- makes chain larger



- $\langle r^2 \rangle^{1/2} = \alpha_r \langle r^2 \rangle_0^{1/2}$

- $\alpha \sim$  expansion parameter

- in a good solvent

- repulsion(polymer-polymer) > repulsion(polymer-solvent)

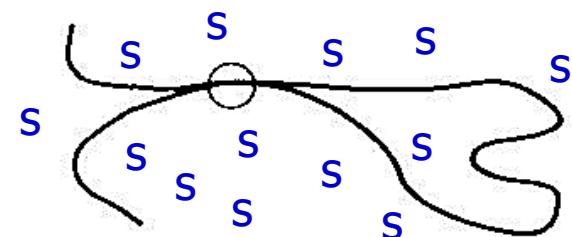
- chain expands,  $\alpha > 1$

- in a poor solvent

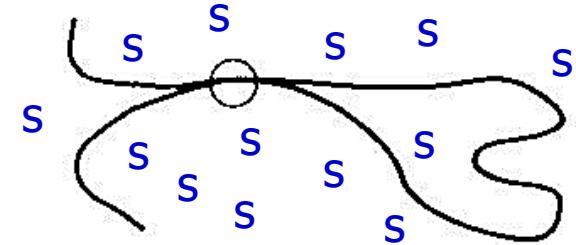
- repulsion(polymer-polymer) < repulsion(polymer-solvent)

- chain shrinks,  $\alpha < 1$

- Actually,  $\alpha \approx 1$  and chain precipitates.



- $\langle r^2 \rangle^{1/2} = \alpha_r \langle r^2 \rangle_0^{1/2}$  (cont'd)
- in a certain poor solvent with  $\alpha = 1$ 
  - repulsion(polymer-polymer) = repulsion(polymer-solvent)
  - polymer = solvent; solution behaves ideally
  - Chain neither expands nor shrinks.
  - 'phantom' or 'ghost' chain
  - theta [ $\Theta$ ] condition
    - in a theta solvent/temperature
    - $\beta_e = C_e \left[ \left( \frac{\theta}{T} \right) - 1 \right] = 0$
    - $\beta_e$  = excluded volume integral = volume excluded by segment
    - 'unperturbed state' or unperturbed dimension [ $\langle r^2 \rangle_0$ ]
      - unperturbed by environment [solvent] or EV effect
      - $\langle r^2 \rangle = \langle r^2 \rangle_{EV} = \langle r^2 \rangle_{RIS} = \langle r^2 \rangle_0$  at theta [ $\Theta$ ] condition

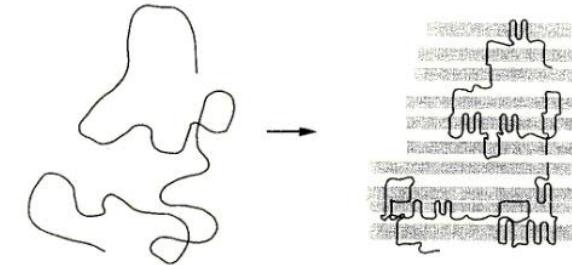
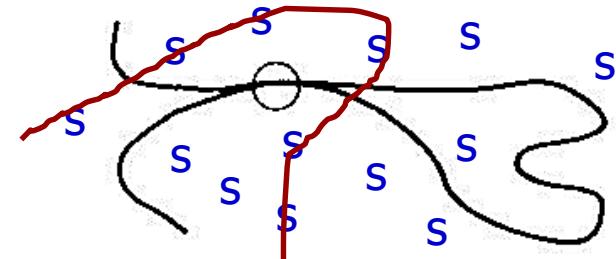


# Real chain in bulk

Ch 10 sl 37

## □ in bulk amorphous state

- The neighbor is the polymer chain with the same type.
  - instead of solvent
- repulsion(polymer-polymer, intra)  
= repulsion(polymer-polymer, inter)
- Chains are in **unperturbed** states.
- proposed by Flory
- proved by SANS exp't later
- valid for melt and semicrystalline states



## □ RIS model describes the state of single chain in bulk (melt, amorphous, semicrystalline).

- chain in crystal ~ conformation with the lowest energy